Efficient \((R)\)-3-hydroxybutyrate production using acetyl-CoA regenerating pathway catalyzed by coenzyme A transferase

Ken'ichiro Matsumoto*, Takehiro Okei, Inori Honma, Toshihiko Ooi, Hirobumi Aoki

Division of Biotechnology and Macromolecular Chemistry, Graduate School of Engineering, Hokkaido University, N13W8, Kita-ku, Sapporo, 060-8628, Japan, and

Corporate R&D Center, Showa Denko K.K. 5-1, Okawa-cho, Kawasaki-ku, Kawasaki, Kanagawa, 210-0858, Japan

Abstract

\((R)\)-3-hydroxybutyrate \([(R)\text{-}3\text{HB}]\) is a useful precursor in the synthesis of value-added chiral compounds such as antibiotics and vitamins. Typically, \((R)\text{-}3\text{HB}\) has been microbially produced from sugars via modified \((R)\text{-}3\text{HB}\)-polymer-synthesizing pathways in which acetyl-CoA is converted into \((R)\text{-}3\text{hydroxybutyryl-coenzyme A} \ [(R)\text{-}3\text{HB-CoA}\] by \(\beta\)-ketothiolase (PhaA) and acetoacetyl-CoA reductase (PhaB). \((R)\text{-}3\text{HB-CoA}\) is hydrolyzed into \((R)\text{-}3\text{HB}\) by modifying enzymes or undergoes
degradation of the polymerized product. In the present study, we constructed a new (R)-3HB-generating pathway from glucose by using propionyl-CoA transferase (PCT). This pathway was designed to excrete (R)-3HB by means of a PCT-catalyzed reaction coupled with regeneration of acetyl-CoA, the starting substance for synthesizing (R)-3HB-CoA. Considering the equilibrium reaction of PCT, the PCT-catalyzed (R)-3HB production would be expected to be facilitated by the addition of acetate since it acts as an acceptor of CoA. As expected, the engineered Escherichia coli harboring the phaAB and pct genes produced 1.0 g L\(^{-1}\) (R)-3HB from glucose, and with the addition of acetate into the medium, the concentration was increased up to 5.2 g L\(^{-1}\), with a productivity of 0.22 g L\(^{-1}\) h\(^{-1}\). The effectiveness of the extracellularly added acetate was evaluated by monitoring the conversion of \(^{13}\)C carbonyl carbon labeled acetate into (R)-3HB using gas chromatography/mass spectrometry. The enantiopurity of (R)-3HB was determined to be 99.2% using chiral liquid chromatography. These results demonstrate that the PCT pathway achieved a rapid co-conversion of glucose and acetate into (R)-3HB.

Keywords: chiral synthesis, enantioselective, production rate, polyhydroxybutyrate, polyhydroxyalkanoate

**Introduction**

Microbial cell factories afford a powerful tool for converting renewable biomass into value-added chemicals under mild physicochemical conditions. In particular, the fine synthesis of stereoisomers is an advantage over chemical processes. The microbial production of (R)-3-hydroxybutyrate [(R)-3HB] is a typical example. This chiral compound is known as a monomeric constituent of bacterial polyester P(3HB), and is
used as a precursor for synthetic antibiotics (Iimori and Shibasaki 1986), vitamins, pheromones, and also as a building block of biobased polyesters such as P(3HB) (Ren et al. 2010, Tokiwa and Ugwu 2007).

(R)-3HB has been typically obtained by two main approaches, the degradation of P(3HB) and CoA-removal from the intermediate compound (R)-3HB-CoA that is generated in the P(3HB) biosynthetic pathway. P(3HB) is synthesized from acetyl-CoA via three successive reactions (Fig. 1). β-ketothiolase (PhaA) catalyzes the condensation of two acetyl-CoA molecules into acetoacetyl-CoA, which is subsequently converted into (R)-3HB-CoA by NADPH-dependent acetoacetyl-CoA reductase (PhaB). Finally, polyhydroxyalkanoate synthase (PhaC) polymerizes (R)-3HB-CoA into a polyester.

Coexpression of intracellular P(3HB) depolymerase (PhaZ) together with P(3HB) biosynthetic enzymes (PhaABC) enables the excretion of monomeric (R)-3HB into the medium. The PhaZ pathway has reportedly achieved up to 7.3 g L$^{-1}$ (R)-3HB production in a shake flask culture (Shiraki et al. 2006). This method has an advantage over (R)-3HB synthesis by the acidic hydrolysis of P(3HB) that generates acid derivatives (de Roo et al. 2002, Shiraki et al. 2006). However, the PhaZ pathway requires a two step fermentation, accumulation of the P(3HB) polymer and also its degradation. Therefore, the culture time required is relatively long (100 h) and this has significantly added to the cost of production.

To accelerate the production rate, alternative (R)-3HB-producing pathways from (R)-3HB-CoA have been investigated. Gao et al. reported that the coexpression of phosphotransbutyrylase (Ptb) and butyrate kinase (Buk) together with PhaA and PhaB in E. coli resulted in 2.0 g L$^{-1}$ (R)-3HB production in a shake flask (Gao et al. 2002). Similarly, (R)-3HB-CoA can be directly hydrolyzed into (R)-3HB by thioesterase
(TesB) \textit{(TesB pathway} in Fig. 1), an approach that yielded up to 4.0 g L$^{-1}$ \((R)\)-3HB production in a shake flask (Liu et al. 2007, Tseng et al. 2009). These processes produced \((R)\)-3HB within a shorter culture time period (24-48 h). However, the \((R)\)-3HB concentrations currently obtained are still lower than those of the PhaZ pathway.

The aim of this study is therefore to design a new \((R)\)-3HB-synthesizing system with higher productivity. To meet this goal, we attempted to generate \((R)\)-3HB from \((R)\)-3HB-CoA using a CoA transferase (referred to as propionyl-CoA transferase, PCT). PCT can transfer a CoA moiety to other short-chain organic acids. It should be noted that this reaction allows the release of \((R)\)-3HB from \((R)\)-3HB-CoA and simultaneous coupling with the generation of acetyl-CoA from acetate \textit{(PCT pathway} in Fig. 1). The regeneration of acetyl-CoA should facilitate the \((R)\)-3HB synthesis, because acetyl-CoA is a precursor of \((R)\)-3HB-CoA, as described above. In this paper, we performed efficient \((R)\)-3HB synthesis from glucose using the PCT pathway, and obtained evidence for the regeneration acetyl-CoA from acetate. PCT from \textit{Megasphaera elsdenii} was used for the production of \((R)\)-3HB because the enzyme is reported to functionally catalyze the CoA transfer reaction between \((R)\)-3HB and acetate (Jossek et al. 1998). The PCT pathway achieved a higher \((R)\)-3HB concentration and hourly productivity by co-utilizing glucose and acetate than either of these obtained by the TesB pathway.

\textbf{Materials and Methods}

Plasmid, strain and culture condition
pTV118NpctC1AB(STQK) (Taguchi et al. 2008) was digested with XbaI/PstI, blunted with T4 DNA polymerase and self-ligated to eliminate the phaC1 gene, and an EcoRI/SmaI fragment of the pct gene from Megasphaera elsdenii (accession No. YP_004765791), which was synthesized by Operon Biotechnologies (Japan), was inserted into the EcoRI/SmaI sites of the plasmid. The resulting plasmid containing the pct gene from M. elsdenii together with the phaA and phaB genes from Ralstonia eutropha was designated pTVpctAB (Fig. 2). pTVpctAB was digested with EcoRI/SmaI, blunted with T4 DNA polymerase and self-ligated to eliminate the pct gene. The resultant plasmid was designated pTVAB.

E. coli BW25113 was purchased from National BioResource Project, Japan (NBRP). The cells harboring pTVpctAB and pTVAB were aerobically cultured in a test tube for 24 h at 30 °C using 1.5 mL LB medium containing 10 g L⁻¹ glucose, various concentrations of acetate, and 100 µg L⁻¹ ampicillin. Acetate was neutralized to pH 7.0 by sodium hydroxide and passed through a 0.2 µm cellulose acetate filter for sterilization.

Analytical methods

The cell-free supernatant of the culture medium was prepared by centrifugation and passage through 0.2 µm cellulose acetate filter. The 3HB, acetate and glucose in the medium were analyzed using HPLC (Jasco, Japan) equipped with an Aminex HPX-87H column (Bio-Rad) and an RI detector. Samples were eluted in the isocratic mode using 0.014N H₂SO₄ as the mobile phase. The enantiopurity of 3HB was determined using HPLC equipped with a Sumichiral OA-6100 column (Sumika Chemical Analysis
Samples were eluted in isocratic mode using 1 mM CuSO₄ as the mobile phase with detection at \( \lambda = 254 \) nm. \((R)\)- and \((S)\)-3HB were purchased from Kanto Chemical (Japan) and used as the standards.

The conversion of acetate to 3HB was monitored using \(^{13}\)C-labeled acetic acid having \(^{13}\)C-enriched (99%) carbonyl carbon (purchased from Taiyo Nissan, Japan). The cell-free culture medium was dried in vacuo, and 3HB was derivatized into ethyl-3HB by heating with ethanol and hydrochloric acid in chloroform as described previously (Arai et al. 2002). The ester which was obtained was analyzed using a gas chromatograph/mass spectrometer (GCMS-QP2010Plus, Shimazu, Japan) to determine the \(^{13}\)C-enrichment in 3HB.

Results

PCT-catalyzed 3HB production

The *E. coli* cells harboring pTVAB and pTVpctAB were grown on glucose (Table 1). The cells harboring pTVpctAB produced 1.0 g L\(^{-1}\) 3HB, while no 3HB was detected in the absence of PCT, indicating that PCT essentially contributed to the 3HB production. The 3HB synthesis increased glucose consumption, indicating the link between the flux in the glycolysis pathway and 3HB production. Additionally, the production of 3HB was associated with a decrease in acetate concentration in the medium (2.5 to 1.8 g L\(^{-1}\)). This result indicates that the designed PCT pathway functioned as expected (Fig. 1, PCT pathway).

If acetate acts as an acceptor for the CoA transferring reaction, the presence of a
high concentration of acetate would increase the production of 3HB. To examine this possibility, acetate was added to the medium used for the production of 3HB. In the absence of PCT, 3HB was not produced (Table 1). In contrast, in the presence of PCT, the 3HB production was increased depending on the concentration of acetate, reaching 5.2 g L\(^{-1}\) when 6.6 g L\(^{-1}\) acetate was supplemented (Table 1). Since the cultivation time was 24 h, the productivity of this condition was calculated to be 0.22 g L\(^{-1}\) h\(^{-1}\). With a higher concentration of acetate, 3HB production was somewhat decreased. The glucose consumption was potently increased under the 3HB-producing conditions. The pH of the medium was shifted to alkaline, probably because of the conversion of two acetic acid molecules into one 3HB molecule catalyzed by PCT (Fig. 1).

Flux analysis from acetate toward 3HB

The results in Table 1 suggest the functionality of the PCT-assisted metabolic route converting acetate into 3HB (Fig. 1). In order to clearly demonstrate this, the flux from the extracellular acetate toward 3HB was estimated by adding \(^{13}\)C carbonyl carbon labeled acetic acid into the medium, and then the \(^{13}\)C enrichment in the synthesized 3HB was monitored. With the addition of 3.3 g L\(^{-1}\) \(^{13}\)C carbonyl carbon labeled acetate, 4.7±0.4 g L\(^{-1}\) 3HB and 1.6±0.1 g L\(^{-1}\) acetate were detected in the medium at a level that was similar to the result obtained with non-labeled acetate (Table 1). The glucose consumption at this condition was 9.3±0.1 g L\(^{-1}\). Then 3HB was derivatized into an ethyl ester and analyzed using GC/MS. Ethyl 3HB generates an ion fragment in which m/z = 117. The ratio of \(^{13}\)C in 3HB was determined by measuring the relative intensity of the fragments m/z = 117, 118 and 119. As shown in Fig. 3, 3HB molecules
synthesized under the presence of $^{13}$C carbonyl carbon labeled acetate contained larger amounts of the heavier fragments (m/z = 118 and 119), clearly indicating that the 3HB molecule was partly derived from exogenous acetic acid. This result indicates that the extracellular acetate was converted into 3HB by the PCT-assisted acetyl-CoA regenerating pathway. According to the relative peak areas of the fragments with m/z = 118 and 119, the ratio of the acetate-derived carbon in 3HB was estimated to be 30±4%.

Enantiomer analysis of 3HB

Because of the enantioselectivity of PhaB, the 3HB produced by *E. coli* was expected to be composed of the *R*-form. To confirm this, the supernatant of the culture medium of *E. coli* harboring pTVpetAB under the presence of 3.3 g L$^{-1}$ acetate was subjected to chiral HPLC. As a result, 3HB was eluted as a major peak (Fig. 4C) at the same retention time of (*R*)-3HB standard (Fig. 4A), and the combination with the standard was eluted as a single peak (Fig. 4D), indicating that the 3HB produced by engineered *E. coli* was composed of almost entirely the *R*-form. In addition, there was a minor peak at 9 min, which was close to the retention time of the (*S*)-3HB standard (Fig. 4B). This small peak was eluted together with the externally added (*S*)-3HB standard (Fig. 4E), indicating that the sample contained a small amount of (*S*)-3HB. According to the peak areas, the enantiopurity of (*R*)-3HB was estimated to be as high as 99.2 mol%.

**Discussion**

In this study, a new pathway for (*R*)-3HB production by co-conversion of glucose and acetate was constructed using CoA transferase (Fig. 1). The productivity of
(R)-3HB was enhanced by addition of acetate in the medium, achieving the maximum concentration of 5.2 g L\(^{-1}\) (R)-3HB with a productivity of 0.22 g L\(^{-1}\) h\(^{-1}\). These values were higher than those obtained by the TesB pathway (4.0 g L\(^{-1}\) and 0.17 g L\(^{-1}\) h\(^{-1}\)) (Liu et al. 2007). Although this concentration was lower than that obtained by the PhaZ pathway (7.2 g L\(^{-1}\) and 0.07 g L\(^{-1}\) h\(^{-1}\)) (Shiraki et al. 2006), the PCT pathway had an advantage over the PhaZ pathway in terms of productivity. In addition, The (R)-3HB productions using UV mutagenized native P(3HB)-producing strains have been reported. For examples, the mutagenized *R. eutropha* produced 3.4 g L\(^{-1}\) (R)-3HB with a productivity of 0.07 g L\(^{-1}\) h\(^{-1}\) from gluconate (Vollbrecht and Schlegel 1979), and the mutagenized *Azohydromonas lata* produced 6.5 g L\(^{-1}\) (R)-3HB with a productivity of 0.07 g L\(^{-1}\) h\(^{-1}\) from glucose (Ugwu et al. 2011). These results also indicate that the PCT pathway in *E. coli* is particularly effective to achieve the higher production rate of (R)-3HB. In terms of (R)-3HB production via P(3HB) hydrolysis using recombinant *E. coli*, Lee et al. reported the production of 9.6 g L\(^{-1}\) 3HB (Lee and Lee 2003). This is of great interest, but also requires careful confirmation, because there is a counter-claim that the high (R)-3HB production was not reproducible, based on both theoretical grounds and experimental evidence (Uchino et al. 2008).

The key mechanism in enhancing (R)-3HB production is the regeneration of acetyl-CoA by the CoA transfer reaction (Fig. 1). The \(^{13}\)C carbonyl carbon labeling analysis demonstrated that extracellular acetate has been converted into (R)-3HB, strongly supporting the functionality of the proposed pathway. Under this condition, 52 mM glucose was consumed and 45 mM (R)-3HB was produced. Because the ratio of the glucose-derived (R)-3HB was approximately 70%, the carbon yield from glucose to (R)-3HB was estimated to be 61%. To further increase the concentration of (R)-3HB in
the medium, a strategy of feeding glucose and acetate using a jar-fermentor would be effective, based on the previous studies (Liu et al. 2007, Shiraki et al. 2006).

Highly stereoselective synthesis of stereoisomers such as 3HB is a great advantage biological processes have over chemoprocesses. It was reported that nearly 100% of (R)-3HB was obtained using PhaZ pathway. A high degree of enantiopurity was achieved by the very strict stereoselectivity toward (R)-3HB-CoA exhibited by PhaC (Rehm 2003). Using the PCT pathway, however, a small amount of S-isomer was detected (Fig. 4). The (S)-3HB may be synthesized by β-oxidation enzymes such as enoyl-CoA hydratase/3-hydroxyacyl-CoA oxidase (FadB) and its homologue FadJ (formerly YfcX). To further increase the enantiopurity, the use of a dual knockout strain of these genes would be effective (Tappel et al. 2012). On the other hand, because of the non-stereoselective substrate specificity of PCT (Tung and Wood 1975), the PCT pathway is applicable to (S)-3HB production using (S)-3HB-CoA dehydrogenase reported by Tseng et al. (Tseng et al. 2009).

Acknowledgements We thank J.M. Nduko for technical assistance of HPLC analysis. E. coli strain was provided by National BioResource Project (NBRP), Japan. This work was financially supported by Showa Denko K. K. (Japan). Pacific Edit reviewed the manuscript prior to submission.

References

intermediates of β-oxidation. Plant Cell Physiol 43:555-562


Tokiwa Y, CU Ugwu (2007) Biotechnological production of (R)-3-hydroxybutyric acid


Figure captions

Figure 1. Metabolic pathways for production of 3-hydroxybutyrate (3HB) from glucose. PhaA, β-ketothiolase; PhaB, acetoacetyl-CoA reductase; PhaC, polyhydroxyalkanoate synthase; PhaZ, P(3HB) depolymerase; TesB, thioesterase; Ptb, phosphotransbutyrylase; Buk, butyrate kinase; PCT, CoA transferase.

Figure 2. Plasmids used for 3HB production. Plac, lac promoter. pct, Propionyl-CoA transferase gene of *M. elsdenii*. PRe, phb operon promoter of *R. eutropha*. phaA and phaB, β-ketothiolase and acetoacetyl-CoA reductase genes of *R. eutropha*.

Figure 3. GC/MS analysis of 3HB produced by recombinant *E. coli* harboring pTVpctAB under the presence of 3.3 g L⁻¹ acetic acid (a) and ¹³C carbonyl carbon labeled acetic acid (b). Three ion fragments from ethyl 3HB (m/z = 117, 118 or 119)
were monitored. Asterisk indicates carbon atoms, which could be $^{13}$C-enriched, when $^{13}$C-labeled acetic acid was converted into 3HB via the proposed PCT pathway (Fig. 1).

Figure 4. Chiral HPLC analysis of 3HB produced by engineered *E. coli*. A: 10 µg (injected amount) ($R$)-3HB standard. B: 10 µg ($S$)-3HB standard. C: 10 µg 3HB produced by recombinant *E. coli*. D: Sample C + 10 µg ($R$)-3HB standard. E: Sample C + 0.1 µg ($S$)-3HB standard.
Table 1. 3HB production by recombinant *E. coli* expressing propionyl-CoA transferase \(^a\)

<table>
<thead>
<tr>
<th>Plasmid</th>
<th>Acetate concentration (g/L)</th>
<th>OD(_{600})</th>
<th>pH</th>
<th>Organic acids (g L(^{-1}))</th>
<th>Glucose consumed (g L(^{-1}))</th>
<th>3HB productivity (g L(^{-1}) h(^{-1}))</th>
</tr>
</thead>
<tbody>
<tr>
<td>None</td>
<td>0</td>
<td>5.2±0.3</td>
<td>4.5</td>
<td>nd</td>
<td>2.6±0.1</td>
<td>2.6±0.5</td>
</tr>
<tr>
<td>pTVAB</td>
<td>0</td>
<td>5.1±0.3</td>
<td>4.6</td>
<td>nd</td>
<td>2.5±0.1</td>
<td>2.7±0.3</td>
</tr>
<tr>
<td></td>
<td>3.3</td>
<td>6.3±0.3</td>
<td>5.0</td>
<td>nd</td>
<td>5.3±0.1</td>
<td>2.9±0.1</td>
</tr>
<tr>
<td></td>
<td>6.6</td>
<td>7.8±0.5</td>
<td>5.3</td>
<td>nd</td>
<td>8.2±0.2</td>
<td>3.4±0.5</td>
</tr>
<tr>
<td>pTV(_{pc}AB)</td>
<td>0</td>
<td>6.5±0.1</td>
<td>4.6</td>
<td>1.0±0.1</td>
<td>1.8±0.1</td>
<td>3.6±0.1</td>
</tr>
<tr>
<td></td>
<td>3.3</td>
<td>13.7±0.5</td>
<td>7.8</td>
<td>4.6±0.2</td>
<td>2.1±0.1</td>
<td>7.6±0.3</td>
</tr>
<tr>
<td></td>
<td>6.6</td>
<td>13.4±1.6</td>
<td>8.4</td>
<td>5.2±0.1</td>
<td>4.2±0.0</td>
<td>9.0±0.5</td>
</tr>
<tr>
<td></td>
<td>9.9</td>
<td>10.2±0.2</td>
<td>7.9</td>
<td>4.1±0.2</td>
<td>6.9±0.1</td>
<td>6.6±0.3</td>
</tr>
</tbody>
</table>

\(^a\) Cells were grown on LB medium containing 10 g L\(^{-1}\) glucose and indicated concentrations of acetate at 30 °C for 24 h; OD\(_{600}\) and pH were determined at 24 h; nd, not detected; Data are averages of three trials.
Figure 1
Figure 2
Figure 3
Figure 4