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REGIOSELECTIVE SYNTHESIS OF TRIFLUOROMETHYL GROUP SUBSTITUTED PYRAZOLE DERIVATIVES FROM 1-ARYL-3,4,4,4-TETRAFLUORO-2-BUTEN-1-ONES

Keisuke Sano and Shoji Hara*

Graduate School of Engineering, Hokkaido University, Sapporo 060-8628, Japan
E-mail: shara@eng.hokudai.ac.jp

Abstract – Trifluoromethyl group substituted pyrazole derivatives were prepared from hydrazines and 1-aryl-3,4,4,4-tetrafluoro-2-buten-1-ones obtained by the deoxyfluorination of β-diketones. The reaction proceeded regioselectively and 5-aryl-3-trifluoromethyl-1-phenyl-1H-pyrazole was obtained from phenylhydrazine. On the other hand, when methylhydrazine was used, 3-aryl-5-trifluoromethyl-1-methyl-1H-pyrazole was selectively formed.

INTRODUCTION
Pyrazoles with a trifluoromethyl substituent are of considerable interest because they are present in pharmacologically and agrochemically important compounds.1 A pyrazole ring was generally prepared from β-diketones with hydrazines.2 However, when unsymmetrical β-diketones such as 1-aryl-4,4,4-trifluoro-1,3-butanones were used, a mixture of regioisomers were formed and it was difficult to obtain the desired regioisomer selectively.3 Recently, we reported the regioselective synthesis of 1-aryl-3,4,4,4-tetrafluoro-2-buten-1-ones (1) by the deoxyfluorination reaction of 1-aryl-4,4,4-trifluoro-1,3-butanones with N,N-diethyl-α,α-difluoro-(m-methylbenzylamine) (DFMBA).4 In this paper, we report the regioselective synthesis of a trifluoromethyl group substituted pyrazole derivative (2 or 3) by the reaction of 1-aryl-3,4,4,4-tetrafluoro-2-buten-1-ones (1) with mono-substituted hydrazines (Scheme 1).5

Dedicated to Professor Akira Suzuki on the occasion of his 80th birthday.
RESULTS AND DISCUSSION

The reaction of 3,4,4,4-tetrafluoro-1-phenyl-2-buten-1-one (1a) with phenylhydrazine was performed in ether under reflux for 24 h and 3-trifluoromethyl-1,5-diphenyl-1H-pyrazole (2a) was obtained in 88% yield (Entry 1 in Table 1). Interestingly, its regioisomer, 5-trifluoromethyl-1,3-diphenyl-1H-pyrazole, was not formed at all. On the other hand, when methylhydrazine was used, 5-trifluoromethyl-1-methyl-3-phenyl-1H-pyrazole (3a) was obtained in 95% yield and its regioisomer, 3-trifluoromethyl-1-methyl-5-phenyl-1H-pyrazole, was not formed (Entry 2). The selectivity is not dependent on the type of aryl group in 1, and 5-aryl-3-trifluoromethyl-1-phenyl-1H-pyrazole (2a-d) was formed selectively from phenylhydrazine (Entries 1, 3, 4, and 6). On the other hand, from methylhydrazine, 3-aryl-5-trifluoromethyl-1-methyl-1H-pyrazole (3a,c) was selectively obtained (Entries 2 and 5). Recently, the regioselective synthesis of 1-aryl-3-trifluoromethyl-1H-pyrazoles 2 by the reaction of 1-aryl-4,4,4-trifluoro-1,3-butanones with phenylhydrazines was reported. However, the selective synthesis of 3-aryl-5-trifluoromethyl-1-methyl-1H-pyrazole 3 from 1-aryl-4,4,4-trifluoro-1,3-butanones and methylhydrazine has not yet been performed. Therefore, our method is useful for the selective synthesis of various pyrazole derivatives with trifluoromethyl substituent.
This high selectivity can be explained by the difference between the reactivity of phenylhydrazine and that of methylhydrazine. In phenylhydrazine, \(-\text{NH}_2\) is more nucleophilic than \(-\text{NHPH}\) due to the electron-withdrawing effect of the Ph group\(^6\) and \(-\text{NH}_2\) attacks the C3 carbon of 1 to yield the intermediate 4.\(^7\) The subsequent addition of \(-\text{NHPH}\) to the carbonyl group yields a cyclic imminium salt 5 that changes to the pyrazole 2. On the other hand, in the reaction with methylhydrazine, \(-\text{NHMe}\) is more reactive than \(-\text{NH}_2\) due to the electron-donating effect of the methyl group and \(-\text{NHMe}\) attacks the C3 carbon of 1 to yield an intermediate 6.\(^6\) The subsequent cyclization proceeds by the attack of \(-\text{NH}_2\) on the carbonyl group to give the pyrazole 3, selectively. As the formation of 4 or 6 by the 1,4-addition reaction of the hydrazines to 1 is irreversible, the regioselectivity of the reaction is determined at the

---

**Table 1. Synthesis of trifluoromethylpyrazoles\(^a\)**

<table>
<thead>
<tr>
<th>Entry</th>
<th>Substrate(^b)</th>
<th>Hydrazine, R</th>
<th>Time (h)</th>
<th>Product</th>
<th>Yield(%)(^c)</th>
</tr>
</thead>
</table>
| 1     | \[
\begin{array}{c}
\text{Ph} \\
\text{CF}_3 \\
\end{array}
\] \[\begin{array}{c}
\text{O} \\
\text{F} \\
\hline
\text{Ph} \\
\text{CF}_3 \\
\end{array}\] 1a | Ph | 24 | \[
\begin{array}{c}
\text{N} \\
\text{N} \\
\end{array}
\] \[\begin{array}{c}
\text{Ph} \\
\text{CF}_3 \\
\hline
\text{N} \\
\text{N} \\
\end{array}\] \[\begin{array}{c}
\text{Ph} \\
\text{CF}_3 \\
\hline
\text{Ph} \\
\text{CF}_3 \\
\end{array}\] 2a | 88 |
| 2     | 1a | Me | 12 | \[
\begin{array}{c}
\text{N} \\
\text{N} \\
\end{array}
\] \[\begin{array}{c}
\text{Ph} \\
\text{CF}_3 \\
\hline
\text{Me} \\
\text{CF}_3 \\
\end{array}\] 3a | 95 |
| 3     | \[
\begin{array}{c}
\text{N} \\
\text{N} \\
\end{array}
\] \[\begin{array}{c}
\text{Ph} \\
\text{CF}_3 \\
\hline
\text{Ph} \\
\text{CF}_3 \\
\end{array}\] 1b | Ph | 24 | \[
\begin{array}{c}
\text{N} \\
\text{N} \\
\end{array}
\] \[\begin{array}{c}
\text{Ph} \\
\text{CF}_3 \\
\hline
\text{Ph} \\
\text{CF}_3 \\
\end{array}\] 2b | 94 |
| 4     | \[
\begin{array}{c}
\text{N} \\
\text{N} \\
\end{array}
\] \[\begin{array}{c}
\text{F} \\
\text{CF}_3 \\
\hline
\text{Ph} \\
\text{CF}_3 \\
\end{array}\] 1c | Ph | 24 | \[
\begin{array}{c}
\text{N} \\
\text{N} \\
\end{array}
\] \[\begin{array}{c}
\text{Ph} \\
\text{CF}_3 \\
\hline
\text{O} \\
\text{CF}_3 \\
\end{array}\] 2c | 89 |
| 5     | 1c | Me | 12 | \[
\begin{array}{c}
\text{N} \\
\text{N} \\
\end{array}
\] \[\begin{array}{c}
\text{Me} \\
\text{CF}_3 \\
\hline
\text{Me} \\
\text{CF}_3 \\
\end{array}\] 3c | 86 |
| 6     | \[
\begin{array}{c}
\text{N} \\
\text{N} \\
\end{array}
\] \[\begin{array}{c}
\text{F} \\
\text{CF}_3 \\
\hline
\text{Ph} \\
\text{CF}_3 \\
\end{array}\] 1d | Ph | 24 | \[
\begin{array}{c}
\text{N} \\
\text{N} \\
\end{array}
\] \[\begin{array}{c}
\text{Ph} \\
\text{CF}_3 \\
\hline
\text{S} \\
\text{CF}_3 \\
\end{array}\] 2d | 90 |

\(^a\) The reaction was carried out in ether under reflux. \(^b\) A mixture of stereoisomers was used. \(^c\) Isolation yield based on 1.
initial step, attack of nitrogen to C3, and largely influenced by the nucleophilicity of the nitrogen on the hydrazines. On the other hand, in the reaction with β-diketones, the initial addition of the nitrogen to the carbonyl group is reversible and the regioselectivity is determined at the dehydration step.²,³,⁶c Therefore, the regioselectivity in the reaction with β-diketones is not so influenced by the nucleophilicity of the nitrogen on the hydrazines as in the reaction with 1 and a similar selectivity was not observed (Scheme 2).

For the synthesis of 3-trifluoromethyl-5-methyl-1-phenyl-1H-pyrazole (2e), a volatile 4,5,5,5-tetrafluoro-3-penten-2-one (1e) is required. Therefore, we synthesized 2e from β-diketone without isolating 1e. The reaction of 1,1,1-trifluoropentane-2,4-dione with DFMBA was carried out at room temperature for 24 h. From the ¹⁹F NMR analysis of the crude mixture, deoxyfluorination was found to occur at the C2 position selectively and 4,5,5,5-tetrafluoro-3-penten-2-one (1e) was formed in 70% yield. Crude 1e was used for the reaction with phenylhydrazine, and 2e was selectively formed in 54% yield from the diketone (Scheme 3).

![Scheme 2 Diagram](image-url)

**Scheme 2**

![Scheme 3 Diagram](image-url)

**Scheme 3**
Celecoxib, 4-[5-(4-methylphenyl)-3-trifluoromethyl-1H-pyrazol-1-yl]benzenesulphonamide, is widely used as an anti-inflammatory drug for the treatment of osteoarthritis, rheumatoid arthritis, and acute pain.\(^8\) Using our method, we prepared Celecoxib from 4,4,4-trifluorobutane-1-(p-tolyl)-1,3-dione.\(^9\) The deoxyfluorination of the diketone with DFMB was performed in CH\(_2\)Cl\(_2\) at 20 °C for 24 h to produce 3,4,4,4-tetrafluoro-1-(p-tolyl)-2-buten-1-one (1f) in 91% yield as a mixture of stereoisomers. The reaction of 1f with \(p\)-aminosulfonylphenylhydrazine hydrochloride was performed in a mixture of DMF and H\(_2\)O (4:1) at 40 °C for 12 h to give Celecoxib in 80% yield and the formation of its regioisomer was not observed (Scheme 4).

![Scheme 4](image)

EXPERIMENTAL

4.1. General

The melting points were measured with a Yanagimoto micro melting-point apparatus. The IR spectra were recorded using a JASCO FT/IR-410. The \(^1H\) NMR (400 MHz) spectra, \(^19\)F NMR (376 MHz) spectra, and \(^13\)C NMR (100 MHz) were recorded in CDCl\(_3\) on a JEOL JNM-A400II FT NMR and the chemical shift, \(\delta\), is referred to TMS (\(^1H, \(^13\)C) and CFCl\(_3\) (\(^19\)F), respectively. The EI-high-resolution mass spectra were measured on a JEOL JMS-700TZ. Methylhydrazine, phenylhydrazine, 4-hydrazinobenzenesulfonamide hydrochloride, and diketones were purchased from Tokyo Kasei Kogyo Co., Ltd. 4,4,4-Trifluoro-1-(p-tolyl)-butane-1,3-dione was prepared according to a literature.\(^9\) 1-Aryl-3,4,4,4-tetrafluoro-2-buten-1-ones 1a, 1b, 1c, and 1d were prepared from the corresponding diketones by the reported method.\(^4\) DFMB was donated from Mitsubishi Gas Chemical Company Inc.

**Preparation of 1-Aryl-3,4,4,4-tetrafluoro-2-buten-1-one\(^4\) (1)**

A mixture of 1-aryl-4,4,4-trifluorobutane-1,3-dione (1 mmol), DFMB (426 mg, 2 mmol), and CH\(_2\)Cl\(_2\) (1
in a reaction vessel made of Teflon™ FEP with a tight screw cap was stirred at 20 °C for 24 h. The mixture was poured into water, neutralized with sat. aq. NaHCO₃, and extracted with Et₂O (20 mL x 3). The combined organic layer was dried over MgSO₄ and concentrated under reduced pressure. Purification by column chromatography (silica gel/hexane-Et₂O) gave 1.

**Preparation of 5-Aryl-3-trifluoromethyl-1-phenyl-1H-pyrazole (2)**

To 1 (0.48 mmol) in Et₂O (8 mL) was added at rt phenylhydrazine (64 mg, 0.59 mmol) and the mixture was stirred under reflux for 24 h. Solid material was removed by filtration and the filtrate was concentrated under reduced pressure. Purification by column chromatography (silica gel/hexane-Et₂O) gave 2.

**Preparation of 3-aryl-5-trifluoromethyl-1-methyl-1H-pyrazole (3)**

To 1 (0.48 mmol) in Et₂O (8 mL) was added at rt methylhydrazine (50 mg, 1.08 mmol) and the mixture was stirred under reflux for 12 h. Solid material was removed by filtration and the filtrate was concentrated under reduced pressure. Purification by column chromatography (silica gel/hexane-Et₂O) gave 3.

**3-Trifluoromethyl-1,5-diphenyl-1H-pyrazole (2a)**

A pale yellow solid; mp 87–88 °C (lit., 10 87–88 °C): IR (KBr): 1496, 1236, 1122 cm⁻¹. ¹H NMR δ 7.37–7.21 (m, 10H), 6.76 (s, 1H). ¹³C NMR δ 144.6, 143.1 (q, J = 38.1 Hz), 139.1, 129.1, 129.0 (2C), 129.0, 128.8 (2C), 128.7 (2C), 128.4, 125.4 (2C), 121.3 (q, J = 268.9 Hz), 105.5 (q, J = 1.9 Hz). ¹⁹F NMR δ –62.82 (s, 3F) {lit., 11 –62.6 (s)}.

**5-Trifluoromethyl-1-methyl-3-phenyl-1H-pyrazole (3a)**

clear oil: IR (neat): 1441, 1275, 1204, 1126 cm⁻¹. ¹H NMR δ 7.78–7.76 (m, 2H), 7.43–7.32 (m, 3H), 6.89 (s, 1H), 4.04 (s, 3H). ¹³C NMR δ 150.3, 133.1 (q, J = 39.1 Hz), 132.0, 128.8 (2C), 128.3, 125.5 (2C), 120.0 (q, J = 268.3 Hz), 104.5 (q, J = 2.7 Hz), 38.1 (q, J = 1.9 Hz). ¹⁹F NMR δ –61.10 (s, 3F) {lit., 6c –60.9 (s)}.

**3-Trifluoromethyl-5-(2-naphtyl)-1-phenyl-1H-pyrazole (2b)**

A highly viscous liquid: IR (neat): 1599, 1486, 1124 cm⁻¹. ¹H NMR δ 7.84–7.74 (m, 2H), 7.55–7.49 (m, 2H), 7.43–7.35 (m, 5H), 7.22–7.20 (m, 1H), 6.87 (s, 1H). ¹³C NMR δ 144.6, 143.2 (q, J = 38.6 Hz), 139.2, 132.9, 132.9, 129.1 (2C), 128.4, 128.3 (2C), 128.1, 127.7, 127.0, 126.7, 126.4, 125.7, 125.3 (2C), 121.3 (q, J = 268.9 Hz), 105.8 (q, J = 1.9 Hz). ¹⁹F NMR δ –62.79 (s, 3F) {lit., 10 –62.3 (s)}.

**3-Trifluoromethyl-5-(fur-2-yl)-1-phenyl-1H-pyrazole (2c)**

A clear liquid; IR (neat): 1496, 1247, 1134 cm⁻¹. ¹H NMR δ 7.51–7.42 (m, 6H), 6.91 (s, 1H), 6.35–6.34 (m, 1H), 5.96 (d, J = 2.5 Hz, 1H). ¹³C NMR δ 143.2, 143.2 (q, J = 38.2 Hz), 143.0, 139.3, 136.1, 129.5, 129.3 (2C), 126.0 (2C), 121.1 (q, J = 268.9 Hz), 111.4, 109.8, 103.7 (q, J = 2.4 Hz). ¹⁹F NMR δ –62.95
5-Trifluoromethyl-3-(fur-2-yl)-1-methyl-1H-pyrazole (3c)
clear liquid; IR (neat): 1274, 1126 cm⁻¹. ¹H NMR δ 7.46 (s, 1H), 6.82 (s, 1H), 6.68 (d, J = 3.6 Hz, 1H), 6.47 (dd, J = 3.2, 1.8 Hz, 1H), 4.02 (s, 3H). ¹³C NMR δ 147.3, 142.8, 142.2, 132.8 (q, J = 39.1 Hz), 119.7 (q, J = 269.0 Hz), 111.3, 106.4, 104.2 (q, J = 1.9 Hz), 38.0 (q, J = 1.9 Hz). ¹⁹F NMR δ −61.26 (s, 3F) {lit., 6c−61.1 (s, 3F)}.

3-Trifluoromethyl-1-phenyl-5-(thien-2-yl)-1H-pyrazole (2d)
yellow solid; mp 65–66 °C (lit., 82–83 °C); IR (KBr): 1469, 1251, 1129 cm⁻¹. ¹H NMR δ 7.46–7.39 (m, 5H), 7.33–7.32 (m, 1H), 6.96 (dd, J = 5.0, 3.7 Hz, 1H), 6.87–6.86 (m, 1H), 6.81 (s, 1H). ¹³C NMR δ 143.0 (q, J = 38.1 Hz), 138.8, 138.7, 129.5, 129.2, 129.2 (2C), 128.0, 127.5, 127.4, 126.3 (2C), 121.1 (q, J = 269.0 Hz), 105.2 (q, J = 1.9 Hz). ¹⁹F NMR δ −62.94 (s, 3F).

3-Trifluoromethyl-5-methyl-1-phenyl-1H-pyrazole (2e)
A mixture of 1,1,1-trifluoropentane-2,4-dione (74 mg, 0.48 mmol), DFMBA (212 mg, 1 mmol) and CH₂Cl₂ (2 mL) in a reaction vessel made of Teflon® FEP with a tight screw cap was stirred at rt for 24 h. Then, the mixture was poured into sat. aq. NaHCO₃ (20 mL) and extracted with Et₂O (10 mL). The ethereal solution of 4,5,5,5-tetrafluoro-3-penten-2-one was transferred into a reaction vessel and phenylhydrazine (45.5 mg, 0.42 mmol) was added. The mixture was stirred under reflux for 24 h. The solid material was removed by filtration and filtrate was concentrated under reduced pressure. Purification by column chromatography (silica gel/hexane–Et₂O) gave 2e (59 mg) in 54% overall yield from 1,1,1-trifluoropentane-2,4-dione. clear oil: IR (neat): 1489, 1384, 1240, 1140 cm⁻¹. ¹H NMR δ 7.53–7.43 (m, 5H), 6.46 (s, 1H), 2.35 (s, 3H). ¹³C NMR δ 142.6 (q, J = 37.6 Hz), 140.7, 138.8, 129.2 (2C), 128.7, 125.2 (2C), 121.4 (q, J = 268.9 Hz), 104.8 (q, J = 1.9 Hz), 12.2. ¹⁹F NMR δ −62.85 (s, 3F) {lit., 10−62.5 (s)}.

Synthesis of Celecoxib
3,4,4,4-Tetrafluoro-1-(p-tolyl)-2-buten-1-one (1f)
A mixture of 1,1,1-trifluoro-4-(p-tolyl)pentane-2,4-dione (232 mg, 1.01 mmol), DFMBA (425 mg, 1.99 mmol) and CH₂Cl₂ (2 mL) in a reaction vessel made of Teflon™ FEP with a tight screw cap was stirred at 30 °C for 24 h. Then, the mixture was poured into sat. aq. NaHCO₃ (20 mL) and extracted with Et₂O (20 mL x 3). The combined organic layer was dried over MgSO₄ and the volatile part was removed under reduced pressure. Purification by column chromatography (silica gel/hexane-Et₂O) gave 1f (196 mg) in 84% yield as a mixture of stereoisomers (E:Z = 13:77, only Z isomer is isolable as a pure form). clear oil: (Z)-isomer: IR (neat): 2929, 1707, 1608, 1208 cm⁻¹. ¹H NMR δ 7.82 (d, J = 8.2 Hz, 2H), 7.32 (d, J = 7.9 Hz, 2H), 6.90 (d, J = 31.6 Hz, 1H, –CH), 2.45 (s, 3H). ¹³C NMR δ 185.9, 150.7 (dq, J = 282.0, 39.4 Hz),
145.6, 133.7, 129.6 (2C), 128.8 (2C), 117.8 (dq, $J = 40.3, 273.1$ Hz), 107.9-107.8 (m), 21.6. $^{19}\text{F NMR}$ δ –73.75 (d, $J = 9.8$, 3F), –118.28 (dq, $J = 31.4, 9.8$ Hz, 1F). HRMS(EI): calcd for C$_{11}$H$_8$OF$_4$; 232.0511. Found; 232.0514.

**Celecoxib**

To 1f (a mixture of the stereoisomers was used, 114 mg, 0.49 mmol) in a 4:1 mixture of DMF and H$_2$O (5 mL) was added 4-hydrazinobenzenesulfonamide hydrochloride (133 mg, 0.60 mmol) and the mixture was stirred at 40 °C for 12 h. After cooling to rt, H$_2$O (40 mL) was added to the mixture. The mixture was extracted with Et$_2$O (40 mL x 3), and the organic layer was washed with H$_2$O (40 mL x 3), and brine (40 mL). The organic layer was dried over MgSO$_4$, concentrated under reduced pressure. Purification by column chromatography (silica gel/hexane-Et$_2$O) gave Celecoxib (150 mg) in 80% yield; a white crystal; mp 154–156 °C (lit.,$^{8a}$ 157–159 °C): IR (KBr): 3267, 1164 cm$^{-1}$. $^1\text{H NMR}$ δ 7.91 (d, $J = 8.4$ Hz, 2H), 7.48 (d, $J = 8.4$ Hz, 2H), 7.18 (d, $J = 8.4$ Hz, 2H), 7.11 (d, $J = 8.0$ Hz, 2H), 6.75 (s, 1H), 4.85 (s, 2H), 2.39 (s, 3H). $^{13}\text{C NMR}$ δ 145.2, 144.1 (q, $J = 38.7$ Hz), 142.5, 141.2, 139.8, 129.7 (2C), 128.7 (2C), 127.5 (2C), 125.6, 125.5 (2C), 121.0 (q, J = 269.0 Hz), 106.3 (q, $J = 1.9$ Hz), 21.3. $^{19}\text{F NMR}$ δ –63.03 (s, 3F)

**ACKNOWLEDGEMENTS**

We are grateful to Mitsubishi Gas Chemical Company Inc. for their donation of DFMBA.

**REFERENCES AND NOTES**


5. As for the synthesis of the pyrazole rings by the reaction of β-fluoro-α,β-unsaturated ketone with the


7. In the reaction of the hydrazines with \(\alpha,\beta\)-unsaturated ketones, 1,4-addition reaction precedes 1,2-addition under non-acidic conditions, see: Y. A. Al-Farkh, F. H. Al-Hajjar, F. S. Al-Shamali, and H. S. Hamoud, *Chem. Pharm. Bull.*, 1979, **27**, 257.


