

HOKKAIDO UNIVERSITY

Title	Synthesis and Functions of Neoglycolipids Based on the Glycoblotting Method
Author(s)	石田, 純也
Citation	北海道大学. 博士(生命科学) 甲第11399号
Issue Date	2014-03-25
DOI	10.14943/doctoral.k11399
Doc URL	http://hdl.handle.net/2115/55634
Туре	theses (doctoral)
File Information	Junya_Ishida.pdf



Hokkaido University Collection of Scholarly and Academic Papers : HUSCAP

Synthesis and Functions of Neoglycolipids Based on the Glycoblotting Method

(糖鎖捕捉反応を利用した新奇糖脂質の合成と機能)

Doctoral Thesis

2014

Junya Ishida

Graduate School of Life Science, Hokkaido University

CONT	ENTS	2
ABBR	EVIATIONS	5
СНАР	TER 1	
Genera	al Introduction	
1-1	Glycolipids	9
1-2	Glycoblotting Method	11
1-3	Objective of Thesis	12
1-4	References	14

CHAPTER 2

Design & Synthesis of Functionalized Ceramide Derivatives

2-1 Introduction	
2-2 Results & Discussions	
2-2-1 Design and Synthesis of Methoxyamino-Functionalized Ceramide Derivation	ti ves 1 19
2-2-2 Design of Other Functional Ceramide Derivatives	23
2-2-3 Synthesis of Aminooxy-Derivative 15 & Methylaminooxy-Derivative 16	24
2-2-4 Synthesis of Hydrazide-Derivative 17	26

2-2-5 Synthesis of Aminooxy-Serine-Derivative 18 & Methylaminooxy-Serine-Derivative 19.27

2-3 Conclusion	
2-4 Experimental Section	

CHAPTER 3

Glycoblotting Reaction for Construction of Neoglycolipid Library	
3-1 Introduction63	
3-2 Results and Discussions64	
3-2-1 Glycoblotting Methoxyamino-Derivative 164	
3-2-2 Glycoblotting Reaction Using Other 5 Derivatives72	
3-3 Conclusion	
3-4 Experimental Section79	
3-5 References	

CHAPTER 4

Inhibitory Activity of N-LacCer Derivatives for EGCase II

4-1 Introduction	
4-2 Results & Discussions	105

4-2-1 Synthesis of O-Lactosylceramide 53 for a Positive Control	105
4-2-2 Primitive Hydrolysis Assay	106
4-2-3 Optimization of Park & Johnson Method to Evaluate Inhibitory Activity	107
4-2-4 Inhibition Assay of Recombinant EGCase II	109
4-3 Conclusion	113
4-4 Experimental Section	114
4-5 References	121

CHAPTER 5

Further Functionalization of Neoglycolipids

5-1 Introduction124
5-2 Results & Discussions126
5-3 Conclusion
5-4 Experimental Section135
5-5 References
CHAPTER 6
Concluding Remarks151

ACKNOWLEDGEMENTS 154

Abbreviations

Ac	Acetyl
AO	Aminooxy
Boc	tert-butylbutoxycarbonyl
Cel	Cellobiose
CG	Ceramide glycanase
CSA	10-camphorsulfonic acid
DCC	N,N-Dicyclohexylcarbodiimide
DCM	Dichloromethane
DHB	Dihydroxybenzoicacid
DIAD	Diisopropyl Azodicarboxylate
DIEA	N,N-diisopropylethylamine
DMF	N,N-dimethylformamide
DMSO	Dimethylsulfoxide
DPPA	Diphenylphosphoryl azide
EEDQ	N-Ethoxycarbonyl-2-ethoxy-1,2-dihydroquinoline
ESI	Electro spray ionization
Fmoc	9-fluorenylmethylcarbonyl
Fuc	Fucose
Gal	Galactose
Glc	Glucose
GlcNAc	N-acetyl-D-glucosamine

Glu	Glutamic acid
GSLs	Glycosphingolipids
HOBt	1-hydroxybenzotriazole
HSQC	¹ H-detected single quantum coherence spectrum
HRMS	High resolution mass spectrometry
Lac	Lactose
LacCer	Lactosylceramide
Mal	Maltose
MALDI	Matrix Assisted Laser Desorption/Ionization
Me	Methyl
Mel	Melibiose
NMR	Nuclear magnetic resonance
PC-3 cell	Prostate cancer cell
Ph	Phenyl
Phth	Phthaloyl
PG	Protecting Group
PS	Phytosphingosine
Pyr	Pyridine
rEGCase	Recombinant Endo-glycoceramidase
RT, r.t.	Room Temperature
Su	Succinimidyl
TBAF	Tetrabutylammonium Fluoride
TBS	tert-butyldimethylsilyl
TBTA	Tris[(1-benzyl-1H-1, 2, 3-triazol-4-yl)methyl]amine

TEMPO	2,2,6,6-tetramethylpiperidine 1-oxyl
TFA	Trifluoroacetic acid
THF	Tetrahydrofuran
TLC	Thin layer chromatography
TOF	Time of flight
Tr, Trt	Trytyl
Trp	Tryptophan
UV	Ultra violet
Xyl	Xylose
WSC	1-ethyl-3-(3-dimethyl aminopropyl) carbodiimide hydrochloride
Z	Benzyloxycarbonyl

Chapter 1

General Introduction

1-1 Glycolipids

Glycosylation is one of the most important posttranslational modifications of proteins in eukaryotes and this step is essential to modulate a wide range of protein and lipid functions both on the cellular surfaces and within the cells. In mammals, glycans are made up of monosaccharide, such as glucose, galactose, mannose, xylose, *N*-acetylglucosamine, *N*-acetylgalactosamine, fucose, and the negatively charged *N*-acetyl neuraminic acids and glucuronic acids. These nine monosaccharides can give rise to a large number of oligosaccharide structures, which can cause structural diversity mediating biological processes involving an infection, cell adhesion, immunity, differentiation, and quality control of proteins^[1,2] (Fig 1-1).

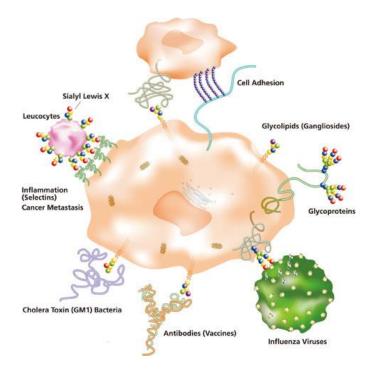


Fig. 1-1. Glycans play various important roles in biologicalphenomena.

Sphingolipids, a complex and ubiquitous group of membrane lipids in eukaryote^[3], play important roles in diverse biological phenomena, such as cell differentiation, cell– cell interactions, apoptosis, infections, and immune responses^[4]. Glycosylation is the most complex modification process of the sphingolipids to form glycosphingolipids (GSLs) and exert various functions on the cell membrane^[4,5]. Generally, the GSLs are classified to four groups, cerebrosides, sulfatides, neutral GSLs, and acidic GSLs. Although the complexity of the GSLs is the source of multiple functions, the variety of the glycan structure make complex to elucidate the function of each structures (Fig. 1-2).

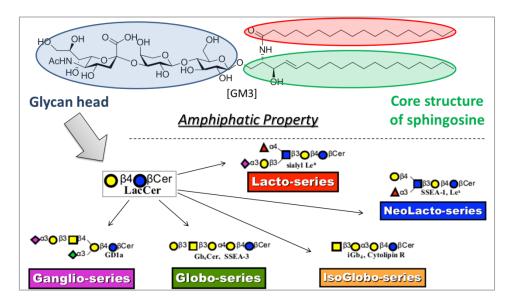


Fig. 1-2. Structural feature and glycan diversity of GSLs

Last three decades, many efforts have been paid to synthesize the natural and mimetic glycosphingolipids to elucidate the functions and their application^[5-7]. However, the preparation of GSLs is still skillful and time spending task (Fig. 1-3). One-step chemical ligation is an attractive alternative for such conventional step-wise synthetic strategy to conjugate biomolecules^[8-10].

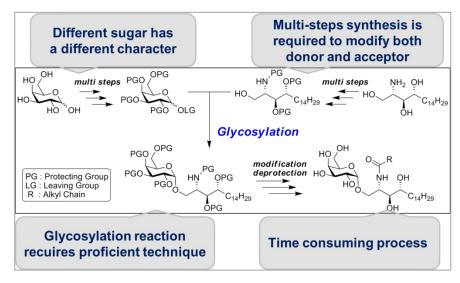


Fig. 1-3. Problems of conventional chemical synthesis of GSLs

1-2 Glycoblotting Method

Oxime formation between aminooxy-functionalized compound and reducing sugars, named glycoblotting^[11], has a potential advantage of rapid preparation of glycan-conjugated compounds without any modification of sugar residue before the conjugation (Fig. 1-4). Our laboratory has been applied this glycoblotting strategy to various glycomic studies such as glycan finger printing^[12], glycan array^[13], and synthesis of glycopeptide library^[14].

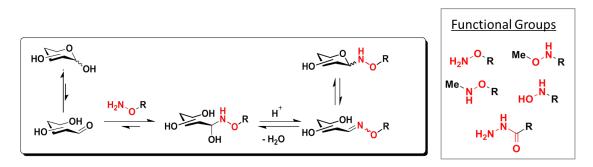


Fig. 1-4. Reaction mechanism and available functional group of glycoblotting

From the point of view to synthesize mimetic compounds of glycoconjugate, the method has great an advantage due to its simplicity. Actually, it has contributed to librarization of various glycoconjugates to date^[15-21] (Fig. 1-5).

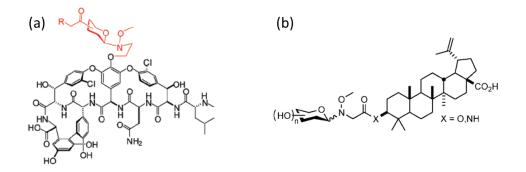


Fig. 1-5. Reported glycoconjugates synthesized by glycoblotting reaction

However, there is no example of glycosphingolipid library constructed by the glycoblotting strategy, although some glycolipid analogues prepared by oxyme formation were reported^[22, 23].

1-3 Objective of Thesis

To elucidate the functions of glycolipids has been demanded for a long time. However, the method for rapid and easy preparation has not been developed although many researchers made efforts.

In this thesis, the author challenged to synthesize functional ceramides which can capture free sugars and construct neoglycolipid library via glycoblotting reaction. In chapter 2, I described design and synthesis of 6 ceramide derivatives having different functional groups. Then, glycoblotting reaction was conducted using them in chapter 3. Additionally, we tried to evaluate whether synthesized neoglycoside work as a mimetic compound of glycosphingolipids in chapter 4. In chapter 5, further functionalization of neoglycolipids was demonstrated.

1-4 References

- A. Varki, "Biological roles of oligosaccharides all of the theories are correct", *Glycobiology*, **1993**, 3, 97-130
- H. Lis and N. Sharon, "Protein glycosylation -Structual and functional aspects-", *Eur. J. Biochem.*, 1993, 218, 1-27
- 3. W. Curatolo, "Glycolipid function", Biochim. Biophys. Acta., 1987, 906, 137-160.
- 4. A. H. Merrill, Jr., "Sphingolipid and Glycosphingolipid Metabolic Pathways in the Era of Sphingolipidomics", *Chem. Rev.*, **2011**, 111, 6387-6422
- T. Wennekes, R. J. B. H. N. van den Berg, R. G. Boot, G. A. van der Marel, H. S. Overkleeft, and J. M. F. G. Aerts "Glycosphingolipids—Nature, Function, and. Pharmacological Modulation", *Angew. Chem. Int. Ed.*, 2009, 48, 8848-8869
- Y. D. Vankara and R. R. Schmidt "Chemistry of glycosphingolipids—carbohydrate molecules of biological significance", *Chem. Soc. Rev.*, 2000, 29, 201-216
- S.-I. Nishimura, "Combinatorial syntheses of sugar derivatives", *Curr. Opn. Chem. Biol.*, 2001, 5, 325-335
- L. L. Kiessling and R. A. Splain, "Chemical Approaches to Glycobiology", *Annu. Rev. Biochem.*, 2010, 79, 619-653
- J. A. M.-Serna, O. Boutureira, Y. Diaz, M. I. Matheu, and S. Castillon, "Recent advances in the glycosylation of sphingosines and ceramides", *Carbohydr. Res.*, 2007, 342, 1595-1612
- S. E. Cervigni, P. Dumy, and M. Mutter, "Synthesis of Glycopeptides and Lipopeptides by Chemoselective Ligation", *Angew. Chem. Int. Ed.*, **1996**, 35, 1230-1232

- S.-I. Nishimura, K. Niikura, M. Kurogochi, T. Matsushita, M. Fumoto, H. Hinou, R. Kamitani, H. Nakagawa, K. Deguchi, N. Miura, K. Monde, and H. Kondo, "High-Throughput Protein Glycomics: Combined Use of Chemoselective Glycoblotting and MALDI-TOF/TOF Mass Spectrometry", *Angew. Chem. Int. Ed.*, 2005, 44, 91-96
- H. Shimaoka, H. Kuramoto, J-i. Furukawa, Y. Miura, M. Kurogochi, Y. Kita, H. Hinou, Y. Shinohara, and S.-I. Nishimura, "One-Pot Solid-Phase Glycoblotting and Probing by Transoximization for High-Throughput Glycomics and Glycoproteomics", *Chem. Eur. J.*, 2007, 13, 1664-73
- T. Matsushita, W. Takada, K. Igarashi, K. Naruchi, R. Miyoshi, F. Garcia-Martin, M. Amano, H. Hinou, and S.-I. Nishimura, "A straightforward protocol for the preparation of high performance microarray displaying synthetic MUC1 glycopeptides", *Biochim. Biophys. Acta.*, 2014, 1840, 1105-1116
- M. Fumoto, H. Hinou, T. Ohta, T. Ito, K. Yamada, A. Takimoto, H. Kondo, H. Shimizu, T. Inazu, Y. Nakahara, and S.-I. Nishimura, "Combinatorial Synthesis of MUC1 Glycopeptides: Polymer Blotting Facilitates Chemical and Enzymatic Synthesis of Highly Complicated Mucin Glycopeptides" *J. Am. Chem. Soc.*, 2005, 127, 11804-11818
- J. M. Langenhan, B. R. Griffith, and J. S. Thorson, "Neoglycorandomization and chemoenzymatic glycorandomization: two complementary tools for natural product diversification", *J. Nat. Prod.*, 2005, 68, 1696–1711
- R. D. Goff and J. S. Thorson, "Enhancing the divergent activities of betulinic acid via neoglycosylation", *Org. Lett.*, 2009, 11, 461–464

- J. M., Langenhan, N. R. Peters, I. Guzei, F. M. Hoffmann, and J. S. Thorson, "Enhancing the anticancer properties of cardiac glycosides by neoglycorandomization", *Proc. Nat. Acad. Sci. USA*, 2005, 102, 12305–12310
- F. Peri, P. Dumy and M. Mutter, "Chemo- and Stereoselective Glycosylation of Hydroxylamino Derivatives : A Versatile Approach to Glycoconjugates", *Tetrahedron*, **1998**, 54, 12269-12278
- Y. Liu, AS Palma, and T. Feizi, "Carbohydrate microarrays: key developments in glycobiology", *Biol. Chem.*, 2009, 390, 647-656
- B. R. Griffith, C. Krepel, X. Fu, S. Blanchard, A. Ahmed, C. E. Edmiston, and J. S. Thorson, "Model for antibiotic optimization via neoglycosylation: synthesis of liponeoglycopeptides active against VRE" *J. Am Chem Soc.*, 2007, 129, 8150-8155
- A. K. V Iyer, M. Zhou, N. Azad, H. Elbaz, L. Wang, D. K. Rogalsky, and J. M. Langenhan, "A Direct Comparison of the Anticancer Activities of Digitoxin MeON-Neoglycosides and O-Glycosides: Oligosaccharide Chain Length-Dependent Induction of Caspase-9-Mediated Apoptosis", ACS Med. Chem. Lett., 2010, 1, 326-330
- 22. R. D. Goff, S. Singh, and J. S. Thorson, "Glycosyloxyamine neoglycosylation: a model study using calicheamicin", *ChemMedChem*, **2011**, 6, 774-776
- 23. Y. Liu, T. Feizi, M. A. Campanero-Rhodes, R. A. Childs, Y. Zhang, B. Mulloy, P. G. Evans, H. M. I. Osborn, D. Otto, P. R. Crocker, and W. Chai, "Neoglycolipid probes prepared via oxime ligation for microarray analysis of oligosaccharide-protein interactions", *Chemistry & Biology*, 2007, 14, 847-859

Chapter 2

Design & Synthesis of Functionalized Ceramide Derivatives

2-1 Introduction

Glycolipids, especially glycosphigolipids (GSLs), play an important role not only on cell surface but as a secretion^[1-5]. There is no doubt for the importance to elucidate the functions of glycosphingolipids.

As mentioned in chapter 1, there has been no example of glycosphingolipid library constructed by the glycoblotting strategy although some glycolipid analogues prepared by oxyme formation were reported^[6, 7]. This made me focus on the synthetic study of construction of GSL library using glycoblotting reaction.

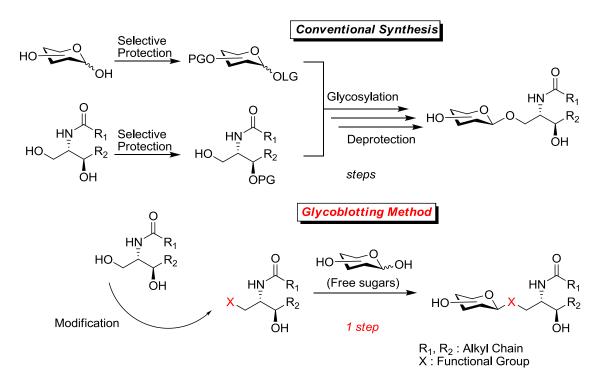


Fig. 2-1-1. Comparison of glycoblotting method with conventional method

for synthesis of glycoipid derivatives.

Compared with conventional chemical and chemo-enzymatic synthesis^[8-11] (Fig. 2-1-1), glycoblotting method has greater advantages to synthesize glycoconjugates, where glycosyl linkage can be formed by only mixing free sugar and aglycon with no protection or modification. However, it was inevitable to say that the neoglycosylated bond possibly form several type of linkage (Fig. 2-1-2). To classify them into 2 groups briefly, one of them is ring-open structure and another is ring-closed structure similar to natural glycoconjugates^[12-14]. While this diversity is occasionally troublesome, it can be regulator of bioactivities. That is why some researches tried to conjugate sugars with one aglycon having different functional groups and indicated that activity profiles differed depending on the functional groups if they had the same glycan on their head^[14].

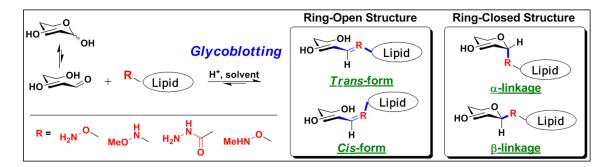


Fig. 2-1-2 Representative structures after glycoblotting reaction

2-2 Results & Discussions

2-2-1 Design and Synthesis of Methoxyamino-Functionalized Ceramide Derivatives 1

To construct neoglycolipid library, selection of three parts for synthesis of functional ceramide derivative was needed. One of them was the type of functional group and the

others were the structure of based lipid and acyl chain. As the lipid moiety, we chose phytosphingosine (PS) skeleton which is abundant in yeast and plants^[15,16] although there are many kinds of sphingosine structure, D/L erythro- or threosphingosine, lysosphingosine, and so on. That was because various bioactivities of phytosphingolipid derivatives have been reported to date such as induction of apoptosis^[17,18] and activation of immunological responses^[19]. For functionalization of the sphingolipid toward glycoblotting reaction, we firstly chose O-methylhydroxylamino group to maintain ring-closed glycan structure^[20-23] and same bond number of natural GSLs after the glycoblotting. As the acyl part, stearic acid which has 18 carbon atoms was selected to be condensed with amine group of sphingosine. However, because of many papers which indicated that the structure of acyl chain affected various biological activities, we considered the acylation step had to be performed later in the total synthesis. Considering that, we conducted retrosynthetic analysis of methoxyaminoderivative 1 starting from abundant phytosphingosine 4 (Fig. 2-2-1). Advantages of this synthetic strategy were clear because of the poor solubility towards common organic solvents of the ceramide derivatives bearing a long N-acyl chain and N-acyl structure was changeable easily. Besides, the most important advantage is a much similar structure to natural glycoceramide can be expected after glycoblotting reaction (Fig. 2-2-2).

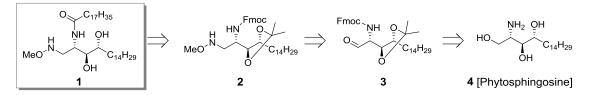


Figure 2-2-1. Design and retrosynthetic analysis of ceramide derivative **1** containing methoxyamino functional group

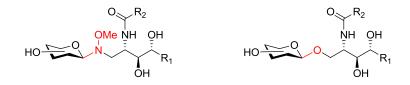
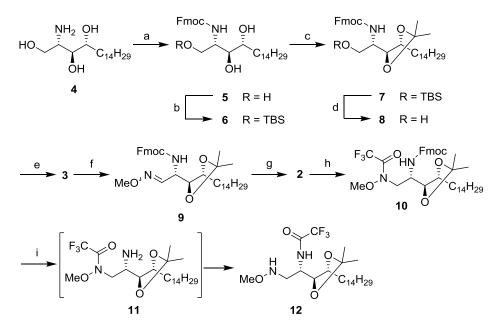


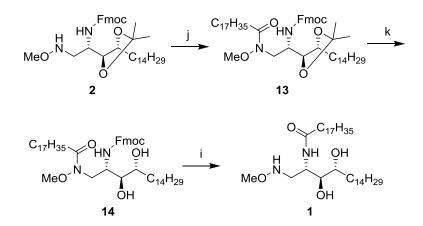
Figure 2-2-2. Comparison of structural similarity between an expected structure of compound **1** after glycoblotting reaction and natural glycosylceramide

Synthesis of compound **1** was started from commercially available phytosphingosine **4**. As shown in Scheme 2-2-1, amino group of phytosphingosine **4** was protected by Fmoc group at first. Interestingly, this is the first example that Fmoc group was introduced directly to phytosphingosine **4** to give **5** which showed good solubility to organic solvents compared to *N*-acylated one. TBS protection of primary alcohol to give **6** and following isopropyliden protection of secondary alcohol provided the all protected compound **7**. Then, selective deprotection of TBS group by HF-pyr complex gave compound **8**. TEMPO oxidation of the primary hydroxyl group^[24] of **8**, followed by oximization by *O*-methylhydroxylamine gave compound **9**, and reduction by sodium cyanoborohydride in acidic condition^[25] afforded methoxyamino derivative **2**. Then the methoxyamino group of **2** was protected by trifluoroacetyl group to give **10** for selective *N*-acylation with fatty acid. Surprisingly, a complete *N*-*N*² acyl migration was occurred during Fmoc deprotection process to give compound **12** as sole product.



Scheme 2-2-1. Synthesis of ceramide derivative 1. Reagents and conditions: a) FmocOSu, THF, RT, 2.5 h, 97%; b) TBSCI, imidazole, THF, RT, 3 h, 98%; c) 2-methoxypropane, CSA, THF, RT, 5 h, 98%; d) HF-pyr, THF, RT, overnight, 95%; e) TEMPO, KBr, *t*BuOCI, NaHCO₃, Na₂CO₃, CH₂Cl₂, 0 °C, 40 min; f) MeONH₃Cl, pyr, THF/MeOH (2:3), 2.5 h, 97% (*E*:*Z* = 8:1, 2 steps); g) NaBH₃CN, THF/AcOH (1:2), 0 °C to RT, 1 h, 51%; h) (CF₃CO)₂O, NaHCO₃, THF, RT, 1 h, 68%; i) piperidine, THF, RT, 3 h, 70%

This unanticipated migration made me deduce that compound **1** could be obtained from fatty acyl modified methoxyamino derivative via this N(-OMe) to N' acyl migration concurrently with Fmoc deprotection. Although any example of such N-N'acyl migration for synthetic strategy has never been reported, the occurrence itself was conjectured by Carrasco et al. in the synthesis of a neoglycopeptide by using methoxyamine modified serine residue^[26]. They speculated that the migration was equilibrium reaction and the ratio was dependent on the pH and under basic condition, acyl chain attached to methoxyamine migrated to its α -amino group completely in their HPLC analysis. This report supported that the basic condition for deprotection of Fmoc group was suitable for my *N*-*N*' migration strategy. That was why I revised the synthetic scheme (Scheme 2-2-2). By using activated ester reagent^[27], stearoyl group was attached to compound **2** to give compound **13** and TFA treatment gave compound **14**. At the last step, the *N*-*N*' acyl migration proceeded successfully and immediately after deprotection of Fmoc group by piperidine treatment, to afford desired ceramide derivative **1** in 14 % overall yield.



Scheme 2-2-2. Synthesis of ceramide derivative 1 via acyl migration reaction; j) $H_{35}C_{17}COOSu$, THF/pyr (2:1), 50 °C, 2 d, 82%; k) TFA, CH_2CI_2/H_2O (10:1), RT, 30 min, 57%.; i) piperidine, THF, RT, 3 h, 66%

2-2-2 Design of Other Functional Ceramide Derivatives

After completion of the synthesis of compound **1**, I designed other ceramide derivatives based on methoxyamino derivative **1** to compare effects of functional group for neoglycosylation (Fig. 2-2-3).

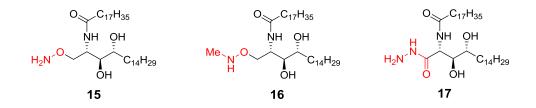


Fig. 2-2-3 Design of other functional ceramide derivatives based on compound 1

Additionally, we designed two of serine-based ceramide mimics (Fig. 2-2-4). That was because our laboratory had clarified that serine-based glucosylceramide could be a substrate of ceramideglycanase (CG) and improve skin function^[28]. The result corresponded with my strategy to synthesize mimetics of glycosphingolipid.

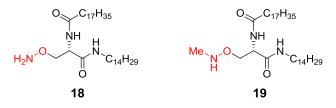
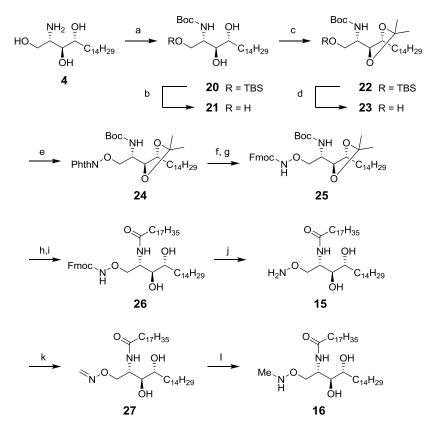


Fig. 2-2-4 Design of serine-based functional ceramide derivatives

2-2-3 Synthesis of Aminooxy-Derivative 15 & Methylaminooxy-Derivative 16

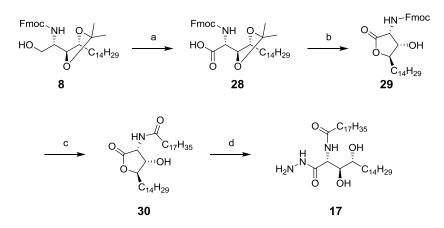
Synthesis of aminooxy derivative **15** and methylaminooxy derivative **16** was shown in Scheme 2-2-3^[29]. Boc protection to phytosphingosine afforded compound **20** and selective TBS protection was performed at primary hydroxyl group. Remained two hydroxyl groups were capped by isopropyliden in the similar way of Fmoc derivative. After removal of TBS group, Mitsunobu reaction gave N-hydroxyphthalimide-substituted compound **24**. Treatment of MeNH₂ followed by Fmoc protection afforded compound **25**. Boc and isopropyliden groups were removed by TFA solution and stearoyl group was attached to amino group by using activated reagent. Aminooxy-functionalized compound **15**, which was one of desired ceramide derivatives, was obtained by adding piperidine in 28% total yield. Through the next 2 steps, oximization and reduction, *N*-methylated compound **16** was obtained in 18% overall yield.



Scheme 2-2-3. Synthesis of ceramide derivative **15** and **16**. Reagents and conditions: a) Boc_2O , NaHCO₃, THF, RT, 3.5 h, quantitatively; b) TBSCI, imidazole, DCM, RT, 1 h, 92%; c) 2-methoxypropane, CSA, THF, RT, 2 h, 88%; d) TBAF, THF, RT, 2 h, 90%; e) HONPhth, Ph₃P, DIAD, THF, 50 °C, 9 h, 95%; f) MeNH₂, THF/MeOH, 7 h; g) FmocOSu, THF, RT, 9 h, 82%, 2 steps; h) TFA, DCM, H₂O, RT, 40 min; i) $H_{35}C_{17}COOSu$, THF, RT, 2 d, 75%, 2 steps; j) piperidine, THF, RT, 6 h, 66%; k) HCHO, DCM, MeOH, AcOH, RT, overnight, 81%; I) NaBH₃CN, AcOH/THF (4:1), RT, 1 h, 80%.

2-2-4 Synthesis of hydrazide-derivative 17

Hydrazide derivative **17** was synthesized by using Fmoc derivative **8**, which was intermediate for methoxyamino derivative **1**. Long time oxidation of the primary hydroxyl group gave carboxylic acid **28**. TFA treatment removed isopropyliden and caused cyclization which afforded lactone **29**. After removal of Fmoc group by piperidine, stearoyl attached compound **30** was obtained. Desired hydrazide **17** was gained by cleavage of lactone bond in 25% overall yield from phytosphingosine **4**.



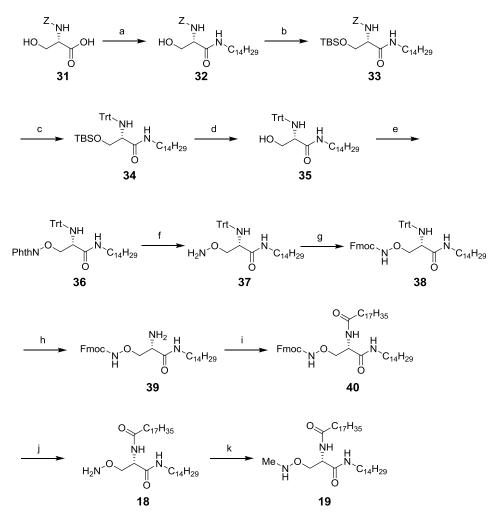
Scheme 2-2-4 Synthesis of hydrazide derivative 17

Reagents and conditions: a) TEMPO, KBr, *t*BuOCI, NaHCO₃, Na₂CO₃, CH₂CI₂, 0 $^{\circ}$ C to RT, overnight, 53%; b) TFA, DCM, H₂O, RT, 3 h, 96%; c) i. piperidine, THF, RT, 30 min; ii. C₁₇H₃₅COOSu, THF, RT, overnight, 60%, 2 steps; d) H₂NNH₂· H₂O, CHCI₃, MeOH, RT, 1.5 h, 91%

2-2-5 Synthesis of Aminooxy-Serine-derivative **18** & Methylaminooxy-Serine-derivative

19

Serine derivatives were synthesized from Z-serine 31. Condensation of the starting material with tetradecylamine gave fatty amide 32 and then hydroxyl group was protected by TBS group. Succeeding two steps reaction afforded trityl protected 34. reaction^[30] After of TBS Mitsunobu removal group, gave N-hydroxyphthalimide-substituted compound **36**. Treatment of MeNH₂ followed by Fmoc protection afforded compound 38. Trityl group was removed by TFA solution and stearoyl group was attached to amino group by using activated reagent. Aminooxy-functionalized compound 18, which was one of desired ceramide derivatives, was obtained by adding piperidine in 31% overall yield. Through the next 2 steps, oximization and reduction, N-methylated compound 19 was obtained in 7% total yield.



Scheme 2-2-5 Synthesis of serine-based functional ceramide derivatives 18, 19 Reagents and conditions: a) WSC, HOBt, $H_2NC_{14}H_{29}$, RT, 2 d, 89%; b) TBSCI, imidazole, DCM, RT, overnight, 91%; c) i. H_2 , Pd/C, THF, RT, 5 h; TrCl, DIEA, THF, RT, 11 h, 86%, 2 steps; d) TBAF, THF, RT, 45 min, 89%; e) HONPhth, Ph₃P, DIAD, THF, 0 °C to 50 °C, 6 h, 84%; f) MeNH₂, THF/MeOH, 2.5 h, 95%; g) FmocOSu, THF, RT, 2 h, 86%; h) TFA, DCM, MeOH, RT, 1 h, 93%; i) $H_{35}C_{17}COOSu$, THF, RT, 3 d, 86%; j) piperidine, THF, CHCl₃, MeOH RT, 2 h, 91%; k) i. HCHO, DCM, MeOH, RT,3 h; ii. NaBH₃CN, TFA, THF, RT, 30 min, 23%, 2 steps.

2-3 Conclusion

In summary, we synthesized 6 ceramide derivatives which were expected to imitate natural glycosphingolipids after glycoblotting reaction. In the synthesis of methoxyamino derivative 1, we came across an unpredicted acyl migration. This finding motivated us to establish new synthetic pathway toward the targeted methoxyamino-functionalized ceramide 1 using N-N' acyl migration as a key reaction. As a result, I have developed an efficient protocol for the synthesis of methoxyamino-functionalyzed ceramide 1 using a specific N-N' acyl migration to allow for the selective modification at $N\alpha$ position of the methoxyamino-functionalized phytosphingosine derivatives, notably the precursors of methoxyamino-functionalized ceramide derivatives. This discovery will contribute to construct synthetic strategy of methoxyamino derivative.

2-4 Experimental Section

General Information

All reactions were carried out under a nitrogen atmosphere with dry solvents under anhydrous conditions, unless otherwise noted. Proton and carbon NMR was recorded at 298K with Varian UnityInova 500 MHz (Agilent Inc., USA; ¹H: 500 MHz, ¹³C: 125 MHz) or Bruker AVANCE DRX 600, equipped with a cryoprobe (Bruker BioSpin Co., Germany; ¹H: 600 MHz, ¹³C: 150 MHz). Chemical shifts are given in ppm and referenced to internal TMS ($\delta_{\rm H}$ 0.00 in CDCl₃), CHCl₃ ($\delta_{\rm H}$ 7.26 in CDCl₃), pyridine-*m*-H $(\delta_{\rm H}7.22 \text{ in } d_5$ -Pyridine), d_5 -Pyridine-*m*-C ($\delta_{\rm C}123.87$) or CDCl₃ ($\delta_{\rm C}77.00$). Assignments in ¹H NMR were made by first-order analysis of the spectra by using ACD/NMR processor software (Advanced Chemistry Development, inc.) and were verified by H-H COSY and HSQC experiments. High/low resolution electrospray ionization mass spectra (ESI-MS) were recorded by JMS-700TZ (JEOL, Japan). TLC was performed on Merck pre-coated plates (20 cm \times 20 cm; layer thickness, 0.25 mm; Silica Gel 60F₂₅₄); of spots were visualized by spraying a solution 90:5:5 (v/v/v)MeOH-p-anisaldehyde-concentrated sulfuric acid and heating at 250 °C for ca. 1/2 min, a solution of 95: 5 (v/v) MeOH-concentrated sulfuric acid and heating at 180°C for ca. 1/2 min, and by UV light (256 or 365 nm) when applicable. Column chromatography was performed on Silica Gel N60 (spherical type, particle size 40-50 µm; Kanto Chemical Industry) with the solvent systems specified, and the ratio of solvent systems was given in v/v. The reaction progress of enzymatic hydrolysis was measured by Park and Johnson method using a microplate reader (SpectraMaxM5, Molecular Devices Co.,

Sunnyvale, CA). The following abbreviations were used to explain the multiplicities: s = singlet, d = doublet, dd = double doublet, t = triplet, m = multiplet, br = broad. In addition to those specified above, the following abbreviations, designations and formulas are used throughout the Supporting Information: MeOH = methanol, H_2O = water, EtOAc = ethyl acetate, DCM = dichloromethane, DMF = dimethylformamide, CHCl₃ = Chloroform, Et₃N = triethylamine, NaHCO₃ = sodium bicarbonate, MgSO₄ = magnesium sulfate, aq. = aqueous, sat. = saturated, 1N HCl = 1 normal hydrogen chloride solution

Materials

Phytosphingosine was purchased from Tokyo Chemical Industry Co., Ltd. (Tokyo, Japan). Z-serine was purchased from Wako Pure Chemical Industries, Ltd. (Osaka, Japan). Solvents and other reagents for the chemical syntheses were purchased from Sigma-Aldrich Co., Tokyo Chemical Industry Co., Ltd., and Wako Pure Chemical Industries, Ltd. (Osaka, Japan) and used without further purification.

Experimental procedure for the chemical synthesis of ceramide mimic 1

(2S, 3S, 4R)-2-[N-(9-fluorenylmethyloxycarbonyl)amino]octadecane-1,3,4-triol (5).

FmocOSu (405 mg, 1.20 mmol) was added to a suspension of phytosphingosine **4** (318 mg, 1.00 mmol) in THF (10 mL), and the mixture was stirred at room temperature for 2.5 h. MeOH was added to the reaction mixture, and the mixture was concentrated to a half volume. After silica gel was added to the mixture, solvent was removed *in vacuo*

completely. The residue was purified by flash column chromatography on silica gel (hexane-EtOAc, 3:2~1:2) to yield the desired compound **5** as a white solid (525 mg, 97%): ¹H NMR (500 MHz, d_5 -pyridine): δ 8.44 (d, J = 8.91 Hz, 1H; NH), 7.86 (d, J = 7.54 Hz, 2H; Ar-H), 7.71 (d, J = 7.54 Hz, 2H; Ar-H), 7.40 (t, J = 7.20 Hz, 2H; Ar-H), 7.26 (m, 2H; Ar-H), 4.89 (m, 1H; H-2), 4.63 (m, 1H; Fmoc-CH₂a), 4.55-4.59 (m, 2H; H-1a, Fmoc-CH₂b), 4.51 (m, 1H; H-1b), 4.46 (br.s, 1H; H-3), 4.32-4.39 (m, 2H; H-4, Fmoc-CH), 2.27 (m, 1H; H-5a), 1.87-2.03 (m, 2H; H-5b, H-6a), 1.71 (m, 1H; H-6b), 1.35-1.50 (m, 2H; CH₂), 1.17-1.35 (m, 20H; CH₂), 0.87 (t, J = 7.20 Hz, 3H; CH₃); ¹³C NMR (125 MHz (HSQC), d_5 -pyridine): δ 128.40, 127.84, 126.06, 120.77 (Ar), 77.02 (C-3), 73.25 (C-4), 67.04 (Fmoc-CH₂), 62.44 (C-1), 55.90 (C-2), 48.24 (Fmoc-CH), 34.23 (C-5), 32.49-23.30 (CH₂), 14.65 (CH₃); HRMS (ESI) Calcd. for C₃₃H₄₉NO₅Na [M+Na]⁺ 562.35084, found 562.35158

(2*S*,3*S*,4*R*)-1-(*tert*-butyldimethylsilyloxy)-2-[*N*-(9-fluorenylmethyloxy-carbonyl)ami no]octadecane-3,4-diol (6).

To a solution of compound **5** (7.00 g, 13.0 mmol) and imidazole (2.66 g, 39.0 mmol) in THF (130 mL) was added *tert*-butylchlorodimethylsilane (2.94 g, 19.5 mmol). The solution was stirred at room temperature for 3 h. The reaction mixture was diluted with EtOAc, washed with sat. NaHCO₃ aq. and brine, dried over MgSO₄, and concentrated under reduced pressure. Purification of the crude product by flash column chromatography on silica gel (hexane-EtOAc, 6:1~3:1) yielded compound **6** (8.30 g, 98%) as an amorphous white solid; ¹H NMR (500 MHz, CDCl₃): δ 7.75 (m, 2H; Ar-*H*), 7.56-7.60 (m, 2H; Ar-*H*), 7.38 (t, *J* = 7.46 Hz, 2H; Ar-*H*), 7.30 (m, 2H; Ar-*H*), 5.53 (d, *J*

= 8.70 Hz, 1H; N*H*), 4.38 (m, 2H; Fmoc-C*H*₂a,b), 4.21 (t, *J* = 7.15 Hz, 1H; Fmoc-C*H*), 3.93(m, 2H; *H*-1a, *H*-2), 3.79 (m, 1H; *H*-1b), 3.62 (m, 2H; *H*-3, *H*-4), 3.30 (d, *J* = 7.77 Hz, 1H; 4-O*H*), 2.90 (d, *J* = 6.84 Hz, 1H; 3-O*H*), 1.69 (m, 1H; *H*-5a), 1.42-1.58 (m, 2H; *H*-5b, *H*-6a), 1.19-1.41 (m, 23H; *H*-6a, C*H*₂), 0.91 (br.s, 9H; *t*-*Bu*), 0.88 (t, *J* = 6.84 Hz, 3H; C*H*₃), 0.11 (d, *J* = 1.55 Hz, 6H; Si-C*H*₃); ¹³C NMR (125 MHz (HSQC), CDCl₃): δ 127.64, 126.97, 124.98, 119.93 (Ar), 75.96 (C-3), 73.09 (C-4), 66.84 (Fmoc-CH₂), 62.54 (C-1), 52.01 (Fmoc-CH), 47.17 (C-2), 33.37 (C-5), 31.88-29.32 (CH₂), 25.80 (C-6), 25.76 (*t*-*Bu*), 22.65 (CH₂), 14.08 (CH₃), -5.55, -5.70 (Si-CH₃); HRMS (ESI) Calcd. for C₃₉H₆₃NO₅SiNa [M+Na]⁺ 676.43732, found 676.43409

(2*S*,3*S*,4*R*)-1-(*tert*-butyldimethylsilyloxy-2-[*N*-(9-fluorenylmethyloxy-carbonyl)ami no]-3,4-*O*-isopropylidene-octadecane-3,4-diol (7).

Compound **6** (2.48 g, 3.79 mmol) was mixed with 2-methoxypropene (721 µL, 7.58 mmol) in THF (30 mL). CSA (44 mg, 190 µmol) was added and the reaction mixture was stirred at room temperature. After 5 h the reaction was quenched with Et₃N and concentrated under reduced pressure. Purification of the crude product by flash column chromatography on silica gel (hexane-EtOAc, 20:1) yielded compound **7** (2.59 g, 98%) as a colorless oil; ¹H NMR (500 MHz, CDCl₃): δ 7.75 (d, *J* = 7.57 Hz, 2H; Ar-*H*), 7.57 (d, *J* = 7.33 Hz, 2H; Ar-*H*), 7.39 (t, *J* = 7.57 Hz, 2H; Ar-*H*), 7.29 (t, *J* = 7.57 Hz, 2H; Ar-*H*), 5.02 (d, *J* = 9.77 Hz, 1H; N*H*), 4.38 (d, *J* = 7.08 Hz, 2H; Fmoc-CH₂), 4.23 (t, *J* = 7.08 Hz, 1H; Fmoc-C*H*), 4.10 (m, 2H; *H*-3, *H*-4), 3.89 (d, *J* = 9.77 Hz, 1H; *H*-1a), 3.83 (br.s, 1H; *H*-2), 3.67 (d, *J* = 8.31 Hz, 1H; *H*-1b), 1.53 (br.s, 2H; *H*-5a,b), 1.42 (s, 3H; (RO)₂CCH₃a), 1.32 (s, 3H; (RO)₂CCH₃b), 1.32-1.19 (m, 24H; CH₂), 0.91 (s, 9H; *t*-Bu),

0.88 (t, J = 6.97 Hz, 3H; CH₃), 0.06 (s, 6H; Si-CH₃); ¹³C NMR (125 MHz (HSQC), CDCl₃): δ 127.65, 126.96, 124.96, 119.94 (Ar), 77.77 (C-3), 75.34 (C-4), 66.79 (Fmoc-CH₂), 62.76 (C-1), 51.33 (C-2), 47.02 (Fmoc-CH), 31.90-22.66 (CH₂), 28.28 ((RO)₂CCH₃a), 25.91 ((RO)₂CCH₃b), 25.85 (*t*-Bu), 14.10 (CH₃), -5.49, -5.52 (Si-CH₃); HRMS (ESI) Calcd. for C₄₂H₆₇NO₅SiNa [M+Na]⁺ 716.46862, found 716.46882

(2*S*,3*S*,4*R*)-2-[*N*-(9-fluorenylmethyloxy-carbonyl)amino]-3,4-*O*-isopropylidene-octa decan-1,3,4-triol (8).

Compound **7** (9.54 g, 13.7 mmol) was dissolved in THF (80 mL) and was added HF-pyridine (4 mL, 154 mmol). The reaction mixture was stirred vigorously at room temperature overnight then diluted with EtOAc. The solution was washed with 1 N aqueous HCl, sat. NaHCO₃ aq. and brine. The organic phase was dried over MgSO₄ and concentrated under reduced pressure. Purification by precipitation (hexane-EtOAc, 4:1~3:2) gave compound **8** as a white solid (7.57 g, 95%); ¹H NMR (500 MHz, CDCl₃): δ 7.76 (d, *J* = 7.64 Hz, 2H; Ar-*H*), 7.58 (m, 2H, Ar-*H*), 7.40 (m, 2H; Ar-*H*), 7.32 (m, 2H; Ar-*H*), 5.18 (m, 1H; N*H*), 4.43 (d, *J* = 6.72 Hz, 2H; Fmoc-CH₂), 4.21 (t, *J* = 7.03 Hz, 1H; Fmoc-CH), 4.17 (br.s, 1H; *H*-4), 4.12 (t, *J* = 6.11 Hz, 1H; *H*-3), 3.89 (m, 1H; *H*-1a), 3.82 (br.s, 1H; *H*-2), 3.71 (br.s, 1H; *H*-1b), 2.21 (br.s, 1H; OH), 1.56 (m, 3H; *H*-5a,b, *H*-6a), 1.46 (s, 3H; (RO)₂CCH₃a), 1.43 (br.s, 1H; *H*-6b), 1.34 (s, 3H; (RO)₂CCH₃b), 1.33-1.21 (m, 22H; CH₂), 0.88 (t, *J* = 7.03 Hz, 3H; CH₃); ¹³C NMR (125 MHz (HSQC), CDCl₃): δ 127.70, 127.03, 125.02, 119.97 (Ar), 78.15 (*C*-3), 77.68 (*C*-4), 66.80 (Fmoc-CH₂), 63.47 (*C*-1), 51.58 (*C*-2), 47.27 (Fmoc-CH), 31.91-22.68 (CH₂), 27.55

$((RO)_2CCH_3a)$, 25.21 ((RO)₂CCH₃b), 14.11 (CH₃); HRMS (ESI) Calcd. for $C_{36}H_{53}NO_5Na [M+Na]^+ 602.38214$, found 602.38025

(2*R*,3*S*,4*R*)-3,4-dihydroxy-2-[*N*-(9-fluorenylmethyloxy-carbonyl)amino]-3,4-*O*-isop ropylidene-octadecane-1-al (3).

To a solution of compound 8 (1.75 g, 3.02 mmol) and NaHCO₃ (505 mg, 6.04 mmol) in CH₂Cl₂ (20 mL) and H₂O (15 mL), TEMPO (25 mg, 151 µmol), KBr (40 mg, 302 μmol), and Na₂CO₃ (15 mg, 151 μmol) were added at 0°C. The solution was stirred vigorously at 0°C for 5 min then added dropwise t-BuOCl (500 μ L, 4.53 mmol). After 40 min, the reaction was quenched by MeOH, diluted with EtOAc, and washed with brine. The organic phase was dried over MgSO₄ and concentrated under reduced pressure. Crude product was used in the next reaction without further purification. Analytical sample was purified by flash column chromatography on silica gel (hexane-EtOAc, $20:1 \sim 3:1$) and compound **3** was yielded as a white solid; ¹H NMR (500MHz, CDCl₃): δ 9.75 (s, 1H; CHO), 7.76 (d, *J* = 7.49 Hz, 2H; Ar-*H*), 7.60 (m, 2H; Ar-H), 7.39 (t, J = 7.86 Hz, 2H; Ar-H), 7.33 (m, 2H; Ar-H), 5.65 (br.s, 1H; NH), 4.51 (m, 1H; H-2), 4.40 (m, 3H; H-3, Fmoc-CH₂a,b), 4.31 (m, 1H; H-4), 4.22 (t, J = 7.11 Hz, t)1H; Fmoc-CH), 1.77 (m, 2H; H-5a,b), 1.60 (m, 1H; H-6a), 1.48 (m, 1H; H-6b), 1.40 (s, 3H; (RO)₂CCH₃a), 1.38-1.16 (m, 25H; CH₂, (RO)₂CCH₃b), 0.88 (t, *J* = 6.92 Hz, 3H; *CH*₃); ¹³C NMR (150 MHz (HSQC), CDCl₃): δ 198.13 (*C*HO), 127.76, 127.10, 125.07, 120.01 (Ar), 78.85 (C-3), 77.53 (C-4), 67.32 (Fmoc-CH₂), 61.04 (C-2), 47.13 (Fmoc-CH), 31.94-22.71 (CH2, (RO)₂CCH₃a,b), 14.14 (CH₃); HRMS (ESI) Calcd. for $C_{36}H_{51}NO_5Na [M+Na]^+ 600.36649$, found 600.36715

(2*S*, 3*S*, 4*R*)- 2-[*N*-(9-fluorenylmethyloxycarbonyl)amino]-3,4-*O*-isopropylidene -1-[*N*-(methoxy)imino]octadecane-3,4-diol (9).

The crude product containing compound 3 was dissolved in THF (10 mL) and MeOH (10 mL). The solution was stirred at room temperature and added another solution of MeONH₃Cl (500 mg, 5.98 mmol) and pyridine (5.0 mL, 6.20 mmol) in MeOH (5 mL). With the progress of reaction, white solid was precipitated and then, the mixture was re-dissolved by diluting with EtOAc after 2.5 h. The solution was washed with sat. NaHCO₃ aq. and brine. The organic phase was dried over MgSO₄ and concentrated under reduced pressure. Purification by precipitation (EtOAc-Hexane) gave a white solid 9 (1.38 g) with a mixture of E/Z isomer (8:1, estimated by ¹H NMR). After filtration, the solution was concentrated, purification of the product by flash column chromatography on silica gel (hexane-EtOAc, 20:1~4:1) yielded compound 9 (400 mg) as a white solid. Totally, product 9 was obtained in 97% yield in 2 steps; (*E* conformer): ¹H NMR (500 MHz, CDCl₃): δ 7.76 (d, J = 7.64 Hz, 2H; Ar-H), 7.60 (m, 2H; Ar-H), 7.47 (d, *J* = 3.67 Hz, 1H; N=C*H*), 7.40 (t, *J* = 7.33 Hz, 2H; Ar-*H*), 7.32 (t, *J* = 7.64 Hz, 2H; Ar-H), 5.51 (d, J = 8.55 Hz, 1H; NH), 4.49 (m, 1H; H-2), 4.42 (m, 1H; Fmoc-CH₂a), 4.35 (m, 1H; Fmoc-CH₂b), 4.24 (m, 3H; H-3, H-4, Fmoc-CH), 3.88 (s, 3H; OMe), 1.65 (m, 2H; H-5), 1.54 (m, 1H; H-6a), 1.42 (s, 3H; (RO)₂CCH₃a), 1.40-1.22 (m, 26H; CH₂, $(RO)_2CCH_3b$, 0.88 (t, J = 7.03 Hz, 3H; CH_3); ¹³C NMR (150 MHz (HSQC), CDCl₃): δ 147.37 (N=CH), 127.70, 127.04, 125.14, 119.96 (Ar), 79.14 (C-3), 77.35 (C-4), 67.11 (Fmoc-CH₂), 61.92 (OMe), 51.44 (C-2), 47.18 (Fmoc-CH), 31.93-22.70 (CH₂, (RO)₂CCH₃a,b), 14.12 (CH₃); (Z conformer): ¹H NMR (500 MHz, CDCl₃): δ 7.76 (d, J = 7.56 Hz, 2H; Ar-*H*), 7.58 (m, 2H; Ar-*H*), 7.40 (t, *J* = 7.37 Hz, 2H; Ar-*H*), 7.32 (m, 2H;

Ar-*H*), 6.82 (d, J = 5.67 Hz, 1H; N=C*H*), 5.33 (d, J = 7.94 Hz, 1H; N*H*), 4.84 (m, 1H; *H*-2), 4.41 (m, 3H; *H*-3, Fmoc-C*H*₂a,b), 4.22 (t, J = 7.18 Hz, 1H; Fmoc-C*H*), 4.16 (br.s, 1H; *H*-4), 3.90 (s, 3H; O*Me*), 1.58 (m, 2H; *H*-5a,b), 1.47 (s, 3H; (RO)₂CC*H*₃a),1.35 (s, 3H; (RO)₂CC*H*₃b) 1.50-1.20 (m, 25H; C*H*₂, *H*-6a,b), 0.88 (t, J = 7.18 Hz, 3H; C*H*₃) ¹³C NMR (150 MHz (HSQC), CDCl₃, 298 K): δ (ppm) = 147.63 (N=CH), 127.71, 127.04, 125.01, 119.98 (Ar), 77.43 (*C*-3, *C*-4), 66.96 (Fmoc-CH₂), 62.16 (O*Me*), 48.06 (*C*-2), 47.22 (Fmoc-CH), 31.93-22.70 (*C*H2, (RO)₂CCH₃a,b), 14.13 (*C*H₃); HRMS (ESI) Calcd. for C₃₇H₅₄N₂O₅Na [M+Na]⁺ 629.39304, found 629.39096

(2*S*, 3*S*, 4*R*)- 2-[*N*-(9-fluorenylmethyloxycarbonyl)amino]-3,4-*O*-isopropylidene -1-[*N*-(methoxy)amino]octadecane-3,4-diol (2).

NaBH₃CN (157 mg, 2.50 mmol) was added to a suspension of compound **9** (303 mg, 500 µmol) in THF (5 mL) and AcOH (10 mL) at 0°C and temperature was gradually increased to ambient in 1 h. The reaction mixture was diluted with EtOAc and washed with sat. NaHCO₃ aq. (3 times) and brine. The organic phase was dried over MgSO₄, and concentrated under reduced pressure. Purification of the crude product by flash column chromatography on silica gel (CH₂Cl₂-Et₂O, 40:1~30:1, Hexane-EtOAc, 3:1) yielded **2** (155 mg, 51%) as a white solid; ¹H NMR (500 MHz, CDCl₃): δ 7.76 (d, *J* = 7.56 Hz, 2H; Ar-*H*), 7.59 (m, 2H; Ar-*H*), 7.39 (t, *J* = 7.56 Hz, 2H; Ar-*H*), 7.31 (t, *J* = 7.25 Hz, 2H; Ar-*H*), 5.13 (d, *J* = 8.77 Hz, 1H; N*H*), 4.42 (m, 2H; Fmoc-CH₂a,b), 4.22 (t, *J* = 6.65 Hz, 1H; Fmoc-C*H*), 4.15 (br.s, 1H; *H*-4), 4.09 (m, 1H; *H*-3), 3.97 (br.s, 1H; *H*-2), 3.51 (s, 3H; OMe), 3.23 (d, *J* = 13.3 Hz, 1H; *H*-1a), 2.97 (dd, *J* = 13.91, 6.65 Hz, 1H; *H*-1b), 1.61 (m, 2H; *H*-5a,b), 1.52 (m, 1H; *H*-6a), 1.43 (s, 3H; (RO)₂CCH₃a), 1.43

(m, 1H; *H*-6b), 1.33 (s, 3H; (RO)₂CCH₃b), 1.37-1.15 (m, H; CH₂), 0.88 (t, J = 6.80 Hz, 3H; CH₃); ¹³C NMR (125 MHz (HSQC), CDCl₃): δ 127.68, 127.01, 124.98, 119.95 (Ar), 78.37 (C-3), 77.71 (C-4), 66.64 (Fmoc-CH₂), 61.45 (OMe), 52.92 (C-1), 49.52 (C-2), 47.30 (Fmoc-CH), 31.92-22.69 (CH₂), 28.90 (C-5), 27.33 ((RO)₂CCH₃a), 25.36 ((RO)₂CCH₃b), 14.12 (CH₃); HRMS (ESI) Calcd. for C₃₇H₅₆N₂O₅Na [M+Na]⁺ 631.40869, found 631.40900

(2*S*, 3*S*, 4*R*)- 2-[*N*-(9-fluorenylmethyloxycarbonyl)amino]-3,4-*O*-isopropylidene -1-[*N*-(methoxy)-*N*-(trifluoroacetyl)amino]octadecane-3,4-diol (10).

To a suspension of compound **2** (305 mg, 500 µmol) and NaHCO₃ (210 mg, 2.50 mmol) in THF (5 mL) was added trifluoroacetic anhydride (209 µL, 1.50 mmol) and the mixture was stirred at room temperature for 1 hour. After added MeOH, the solution was diluted with EtOAc and washed with sat. NaHCO₃ aq. and brine. The organic phase was dried over MgSO₄, and concentrated under reduced pressure. Purification of the crude product by flash column chromatography on silica gel (hexane-EtOAc, 10:1~5:1) yielded **10** (240 mg, 68%) as colorless oil; ¹H NMR (600 MHz, CDCl₃): δ 7.75 (d, *J* = 7.52 Hz, 2H; Ar-*H*), 7.56 (m, 2H; Ar-*H*), 7.39 (t, *J* = 7.43 Hz, 2H; Ar-*H*), 7.30 (t, *J* = 7.34 Hz, 2H; Ar-*H*), 4.91 (d, *J* = 9.54 Hz, 1H; N*H*), 4.41 (dd, *J* = 10.64, 6.97 Hz, 1H; Fmoc-CH₂a), 4.30 (m, 1H; Fmoc-CH₂b), 4.15-4.23 (m, 3H; *H*-2, *H*-4, Fmoc-C*H*), 4.14 (t, *J* = 6.24 Hz, 1H; *H*-3), 4.07 (dd, *J* = 14.67, 9.35 Hz, 1H; *H*-1a), 3.84 (dd, *J* = 15.04, 2.93 Hz, 1H; *H*-1b), 3.75 (s, 3H; OMe), 1.63-1.51 (m, 4H; *H*-5a,b, *H*-6a,b), 1.47 (s, 3H; (RO)₂CCH₃a), 1.34 (s, 3H; (RO)₂CCH₃b), 1.32-1.18 (m, 22H; CH₂), 0.88 (t, *J* = 7.06 Hz, 3H; CH₃); ¹³C NMR (150 MHz (HSQC), CDCl₃): δ 127.45, 126.87, 124.84, 119.70 (Ar), 77.82 (*C*-3), 77.43 (*C*-4), 66.98 (Fmoc-*C*H₂), 62.42 (O*Me*), 48.86 (*C*-2), 46.90 (Fmoc-*C*H), 46.24 (*C*-1), 31.89-22.33 (*C*H₂), 29.26 (*C*-5), 27.38 ((RO)₂C*C*H₃a), 24.98 ((RO)₂C*C*H₃b), 14.08 (*C*H₃); HRMS (ESI) Calcd. for C₃₉H₅₅F₃N₂O₆Na [M+Na]⁺ 727.39099, found 727.39308

(2*S*,3*S*,4*R*)-1-[*N*-(methoxy)amino]-2-[*N*-(trifluoroacetyl)-amino]-3,4-*O*-isopropylide ne-octadecane-3,4-diol (12).

Compound **10** (222 mg, 315 µmol) was dissolved in THF (4 mL) and added piperidine (1 mL) at room temperature. The solution was stirred for 3 h and silica gel was added before evaporated completely. The crude product was purified by flash column chromatography on silica gel (hexane-EtOAc, 10:1~3:1) yielded compound **12** (106 mg, 70%) as a white solid; ¹H NMR (600 MHz, CDCl₃): δ 6.95 (d, *J* = 8.62 Hz, 1H; N*H*), 4.23 (br.s, 1H; *H*-2), 4.16 (m, 2H; *H*-3, *H*-4), 3.49 (s, 3H; O*Me*), 3.24 (m, 1H; *H*-1a), 3.08 (dd, *J* = 13.94, 6.24 Hz, 1H; *H*-1b), 1.59 (m, 1H; *H*-5a), 1.52 (m, 2H; *H*-5b, *H*-6a), 1.42 (s, 3H; (RO)₂CCH₃a), 1.33 (s, 3H; (RO)₂CCH₃b), 1.36-1.18 (m, 23H), 0.86 (t, *J* = 6.60 Hz, 6H; CH₃); ¹³C NMR (150 MHz (HSQC), CDCl₃): δ 77.37 (*C*-3, *C*-4), 61.53 (O*Me*), 52.04 (*C*-1), 49.12 (*C*-2), 31.99-22.35 (CH₂), 28.85 (*C*-5), 27.08 ((RO)₂CCH₃a), 24.84 ((RO)₂CCH₃b), 14.25 (CH₃); HRMS (ESI) Calcd. for C₂₄H₄₆F₃N₂O₄ [M+H]⁺ 483.34097, found 483.34184

(2*S*, 3*S*, 4*R*)- 2-[*N*-(9-fluorenylmethyloxycarbonyl)amino]-3,4-*O*-isopropylidene -1-[*N*-(methoxy)-*N*-(octadecanoyl)amino]octadecane-3,4-diol (13).

To a solution of compound **2** (100 mg, 164 µmol) in THF/Pyr (2:1, 15 mL) was added *N*-hydroxysuccinyl octadecanoate (188 mg, 493 µmol) and the mixture was stirred at 50°C for 2 days. After silica gel was added to the mixture, solvent was removed *in vacuo* completely. Purification of the crude product by flash column chromatography on silica gel (hexane-EtOAc, 20:1~10:1) yielded **13** (117 mg, 82%) as a white solid; ¹H NMR (500 MHz, CDCl₃): δ 7.75 (d, *J* = 7.52 Hz, 2H; Ar-*H*), 7.58 (m, 2H; Ar-*H*), 7.39 (t, *J* = 7.22 Hz, 2H; Ar-*H*), 7.31 (t, *J* = 7.52, 2H; Ar-*H*), 5.20 (m, 1H; N*H*), 4.28 (m, 2H; Fmoc-C*H*₂), 4.16 (m, 5H; *H*-1a, *H*-2, *H*-3, *H*-4, Fmoc-C*H*), 3.69 (s, 3H; OMe), 3.58 (m, 1H; *H*-1b), 2.39 (m, 2H; NHCOC*H*₂a,b), 1.65 (m, 2H; *H*-5a,b), 1.57 (m, 3H; *H*-6a, NHCOCH₂C*H*₂a,b), 1.48 (s, 3H; (RO)₂CC*H*₃a), 1.35 (s, 3H; (RO)₂CC*H*₃b), 1.41-1.14 (s, 51H; C*H*₂, *H*-6b), 0.88 (t, *J* = 6.92 Hz, 6H; C*H*₃); ¹³C NMR (150 MHz (HSQC), CDCl₃): δ 127.46, 126.91, 125.06, 119.74 (Ar), 78.52 (C-3), 77.49 (C-4), 66.87 (Fmoc-CH₂), 61.70 (OMe), 49.72 (C-2), 46.96 (Fmoc-CH), 45.68 (C-1), 32.19-22.11 (CH2), 27.01 ((RO)₂CCH₃a), 24.89 ((RO)₂CCH₃b), 14.38 (CH₃); HRMS (ESI) Calcd. for C₅₅H₉₀N₂O₆Na [M+Na]⁺ 897.66966, found 897.67088

(2*S*,3*S*,4*R*)-2-[*N*-(9-fluorenylmethyloxycarbonyl)amino]-1-[*N*-(methoxy)-*N*-(octadec anoyl)amino]octadecane-3,4-diol (14).

To a suspension of compound **13** (525 mg, 600 μ mol) in CH₂Cl₂/H₂O (10:1, 2.2 mL) was added TFA (8 mL) and the mixture was stirred at room temperature for 30 minutes. The reaction mixture was diluted with EtOAc and washed with H₂O (2 times), sat. NaHCO₃ aq. (3 times) and brine. The organic phase was dried over MgSO₄, and concentrated under reduced pressure. Purification of the crude product by flash column

chromatography on silica gel (hexane-EtOAc, 10:1~3:1) yielded **14** (285 mg, 57%) as a white solid; ¹H NMR (500 MHz, CDCl₃): δ 7.76 (d, *J* = 7.59 Hz, 2H; Ar-*H*), 7.58 (d, *J* = 7.59 Hz, 2H; Ar-*H*), 7.40 (t, *J* = 7.27 Hz, 2H; Ar-*H*), 7.31 (m, 2H; Ar-*H*), 5.39 (d, *J* = 8.54 Hz, 1H; N*H*), 4.39 (m, 2H; Fmoc-C*H*₂), 4.20 (t, *J* = 6.96 Hz, 1H; Fmoc-C*H*), 4.14 (m, 1H; *H*-1a), 4.07 (m, 1H; *H*-2), 3.76 (m, 1H; *H*-1b), 3.67 (s, 3H; O*Me*), 3.61 (d, *J* = 5.38 Hz, 1H; *H*-4), 3.46 (m, 1H; *H*-3), 2.50 (m, 1H; NHCOC*H*₂a), 2.40 (m, 1H; NHCOC*H*₂b), 1.65 (m, 3H; *H*-5a, NHCOCH₂C*H*₂a,b),1.55 (m, 1H; *H*-6a), 1.44 (m, 1H; *H*-5b), 1.38-1.16 (s, 53H; C*H*₂, *H*-6b), 0.88 (t, *J* = 6.64 Hz, 6H; C*H*₃); ¹³C NMR (150 MHz (HSQC), CDCl₃): δ 127.36, 126.78, 124.79, 119.65 (Ar), 74.63 (*C*-3), 72.65 (*C*-4), 66.80 (Fmoc-CH₂), 61.32 (O*Me*), 52.45 (*C*-2), 47.06 (Fmoc-CH₂), 45.27 (*C*-1), 33.28-24.21 (*C*H₂), 14.08 (*C*H₃); HRMS (ESI) Calcd. for C₅₂H₈₆N₂O₆Na [M+Na]⁺ 857.63836, found 857.63707

(2*S*, 3*S*, 4*R*)-1-[*N*-(methoxy)amino]-2-[*N*-(octadecanoyl)amino]octadecane-3,4-diol (1).

To a solution of compound **14** (290 mg, 347 µmol) in THF (1.5 mL), piperidine (200 µL) was added and the mixture was stirred at room temperature for 3 hours. After evaporation, purification of the crude product by flash column chromatography on silica gel (toluene-EtOAc, 2:1-1:1-0:1) yielded **1** (140 mg, 66%) as a white solid; ¹H NMR (500 MHz, CDCl₃): δ 6.24 (d, *J* = 7.82 Hz, 1H; N*H*), 4.22 (m, 1H; *H*-2), 3.60 (m, 2H; *H*-3, *H*-4), 3.56 (s, 3H; OMe), 3.23 (m, 2H; *H*-1a,b), 2.20 (t, *J* = 7.57 Hz, 2H; NHCOCH₂a,b), 1.66 (m, 3H; *H*-5a, NHCOCH₂CH₂a,b), 1.52 (m, 1H; *H*-6a), 1.44 (m, 1H; *H*-5b), 1.38-1.19 (m, 51H; CH₂, *H*-6b), 0.88 (t, *J* = 6.84 Hz, 6H; CH₃); ¹³C NMR

(150 MHz (HSQC), CDCl₃): δ 76.13 (*C*-3), 73.20 (*C*-4), 61.26 (OMe), 51.03 (*C*-1),
49.78 (*C*-2), 36.71 (NHCOCH₂), 33.18 (*C*-5), 31.80-23.24 (*C*H₂), 13.99 (*C*H₃); HRMS
(ESI) Calcd. for C₃₇H₇₇N₂O₄ [M+H]⁺ 613.58833, found 613.58608

Experimental Procedure of Chemical Synthesis of H₂NO-/MeHNO-PS-derivatives 15, 16

(2*S*,3*S*,4*R*)-1-phthalimidyloxy-2-[(*tert*-butyloxycarbonyl)amino]-3,4-*O*-isopropylide ne-octadecane-3,4-diol (24).

Into a THF solution (150 mL) containing compound **23** (6.58 g, 14.4 mmol), Ph₃P (4.54 g, 17.3 mmol) and *N*-hydroxyphthalimide (2.82 g, 17.3 mmol), DIAD (3.42 mL, 17.3 mmol) was dropped at 0 degree. After 20 min, the reaction mixture was stirred at 50 degree for 9 hours. Then, the solution was evaporated under reduced pressure. Purification by flash column chromatography on silica gel (Hexane-EtOAc, 6:1-4:1) yielded **24** (8.27 g, 95%) as a white solid.; ¹H NMR (500 MHz, CDCl₃): δ 7.84 (m, 2H; Ar-*H*), 7.76 (m, 2H; Ar-*H*), 5.37 (d, *J* = 9.30, 1H; N*H*), 4.62 (m, 1H; *H*-1a), 4.26 (m, 2H; *H*-1b, *H*-3), 4.16 (dd, *J* = 12.40, 6.51, 1H; *H*-4), 3.92 (m, 1H; *H*-2), 1.60-1.20 (m, 29H; (RO)₂CCH₃b, CH₂), 1.41 (s, 9H; *t*-*Bu*), 1.38 (s, 3H; (RO)₂CCH₃a), 0.87 (t, *J* = 7.13, 3H; CH₃); ¹³C NMR (150 MHz (HSQC), CDCl₃): δ 134.54 123.53 (Ar) 78.52 (C-1) 77.92 (C-4) 76.28 (C-3) 49.83 (C-2) 31.92-22.69 (CH₂) 28.30 (*t*-*Bu*) 27.91 ((RO)₂CCH₃a) 25.47 ((RO)₂CCH₃b) 14.11 (CH₃)

(2*S*,3*S*,4*R*)-1-[(9-fluorenylmethylcarbonyl)aminooxy]-2-[(*tert*-butyloxycarbonyl)am ino]-3,4-*O*-isopropylidene-octadecane-3,4-diol (**25**).

Into a THF solution (150 mL) containing compound **24** (1.37 g, 3.0 mmol), MeNH₂ (40% in MeOH, 1.53 mL, 15.0 mmol) was added at room temperature and the reaction mixture was stirred for 7 hours. Then, the solution was diluted with toluene and evaporated under reduced pressure. This procedure was repeated two times to remove the remaining MeNH₂ completely. After evaporaton, the crude product was disolved in THF (30 mL) and FmocOSu (1.21 g, 3.6 mmol) was added to the solution. Reaction mixture was stirred at room temperature for 9 hours and evaporated under reduced pressure. Purification by flash column chromatography on silica gel (Hexane-EtOAc, 10:1-5:1) yielded **25** (1.71 g, 82%, 2 steps) as a white amorphous solid.

(intermediate)

¹H NMR (600 MHz, CHLOROFORM-*d*) δ ppm 5.54 (br. s, 2 H; ONH₂) 4.62 - 4.71 (m, 1 H; NH) 4.10 - 4.17 (m, 1 H; H-4) 3.95 - 4.05 (m, 2 H; H-2, H-3) 3.80 - 3.88 (m, 1 H; H-1a) 3.64 - 3.72 (m, 1 H; H-1b) 1.68 – 1.20 (m, 26 H; CH₂) 1.43 (s, 9 H; *tBu*) 1.42 (s, 3 H; (RO)₂CCH₃a) 1.31 (s, 3 H; (RO)₂CCH₃b) 0.87 (t, *J*=6.60 Hz, 3 H; CH₃); F1 (ppm) 77.92 (*C*-4) 77.48 (*C*-3)76.12 (*C*-1) 49.20 (*C*-2) 31.89-22.68 (*C*H₂) 28.35 (*t*-*Bu*) 27.67 ((RO)₂CCH₃a) 25.59 ((RO)₂CCH₃b) 14.10 (*C*H₃)

(compound **25**) ¹H NMR (500 MHz, CDCl₃): δ 8.26 (br.s, 1H; ON*H*Fmoc), 7.76 (m, 2H; Ar-*H*), 7.60 (m, 2H; Ar-*H*), 7.40 (m, 2H; Ar-*H*), 7.31 (m, 2H; Ar-*H*), 4.82 (m, 1H; N*H*), 4.43 (m, 2H; Fmoc-C*H*₂), 4.27 (t, *J* = 7.37, 1H; Fmoc-C*H*), 4.14 (m, 2H; *H*-1a, *H*-4), 4.04 (m, 2H; *H*-2, *H*-3), 3.83 (m, 1H; *H*-1b), 1.60-1.20 (m, 26H; C*H*₂), 1.44 (s, 9H; *tBu*), 1.42 (s, 3H; (RO)₂CC*H*₃a), 1.32 (s, 3H; (RO)₂CC*H*₃b), 0.87 (t, *J* = 7.13, 3H; C*H*₃); (ppm) 127.79 127.11 125.15 120.00 (Ar) 80.07 (C-4) 77.80 (C-1) 77.21 (C-3)

67.65 (FmocCH₂) 48.86 (C-2) 46.96 (FmocCH) 31.93-22.69 (CH₂) 28.33 (*tBu*) 27.51 ((RO)₂CCH₃a) 25.42 ((RO)₂CCH₃b) 14.12 (CH₃)

(2*S*,3*S*,4*R*)-1-[(9-fluorenylmethylcarbonyl)aminooxy]-2-[(octadecanoyl)amino] -octadecane-3,4-diol (26).

To a suspension of compound **25** (2.37 g, 3.41 mmol) in CH₂Cl₂/H₂O (10:1, 4.5 mL) was added TFA (15 mL) and the mixture was stirred at room temperature for 40 minutes. The reaction solution was neutralized with sat. NaHCO₃ aq. and extracted with CHCl₃4 times. The organic phase was dried over MgSO₄ and concentrated under reduced pressure. The crude product was directly used for succeeding reaction without further purification. The product and StearoylOSu were suspended in THF (40 mL) and the mixture was stirred at room temperature for 2 days. The reaction solution was evaporated under reduced pressure completely. After evaporation, obtained white solid was suspended in MeOH and filtrated. The filtercake was dissolved in mixed solvent (CHCl₃/MeOH, 2:1) and silica gel was added to the flask till the solution almost lost fluidity. Remained solvent was removed by evaporation and purification of the crude product by flash column chromatography on silica gel (hexane-EtOAc, 2:1, CHCl₃-*i*PrOH, 20:1) yielded **26** (2.12 g, 75%) as a white solid;

¹H NMR (500 MHz, CHLOROFORM-*d*) δ ppm 7.80 - 7.88 (m, 1 H; Fmoc-NHO) 7.77 (d, *J*=7.41 Hz, 2 H; Fmoc-Ar) 7.56 (d, *J*=7.41 Hz, 2 H; Fmoc-Ar) 7.41 (t, *J*=7.41 Hz, 2 H; Fmoc-Ar) 7.32 (t, *J*=7.41 Hz, 2 H; Fmoc-Ar) 6.57 (br. s, 1 H; NH) 4.49 (d, *J*=7.10 Hz, 2 H; Fmoc-CH₂) 4.17 - 4.26 (m, 3 H; Fmoc-CH, *H*-1a, *H*-2) 4.05 - 4.10 (m, 1 H; *H*-1b) 3.69 - 3.58 (m, 2 H; *H*-3, *H*-4) 2.21 (t, *J*=7.72 Hz, 2 H; NHCOCH₂) 1.97 - 1.18

(m, 56 H; CH₂) 0.88 (t, J=7.10 Hz, 6 H; CH₃); F1 (ppm) 127.73 127.27 125.15 120.12 (Ar) 76.50 (C-1) 74.40 (C-3) 73.34 (C-4) 68.18 (FmocCH₂) 48.78 (C-2) 45.93 (FmocCH) 36.77-21.88 (CH₂) 14.19 (CH₃)

(2*S*,3*S*,4*R*)-1-aminooxy-2-[(octadecanoyl)amino]-octadecane-3,4-diol (15)

Compound **26** (2.69 g, 3.34 mmol) was dissolved in CH_2Cl_2 (40 mL) and MeOH (10 mL) and piperidine (2.55 mL, 10 eq) was added to the solution at room temperature. After 6 h, solvent was removed by air-drying and white solid was suspended in MeOH. Filtration gave compound **15** (1.02 g, 66%) as a white solid.

¹H NMR (500 MHz, CHLOROFORM-*d*) δ ppm 6.19 (d, *J*=8.43 Hz, 1 H; N*H*) 5.55 (br. s, 2 H; ON*H*₂) 4.26 - 4.33 (m, 1 H; *H*-2) 4.00 (dd, *J*=11.73, 4.76 Hz, 1 H; *H*-1a) 3.87 (dd, *J*=11.73, 4.03 Hz, 1 H; *H*-2b) 3.59 - 3.65 (m, 1 H; *H*-4) 3.53 - 3.59 (m, 1 H; *H*-3) 2.32 (br. s, 1 H; O*H*) 2.18 - 2.23 (m, 2 H; C*H*₂) 1.19 - 1.75 (m, 56 H; C*H*₂) 0.88 (t, *J*=6.96 Hz, 6 H; C*H*₃); F1 (ppm) 75.18 (C-3) 74.67 (C-1) 73.07 (C-4) 50.92 (C-2) 36.77-21.65 (CH₂) 14.11 (CH₃)

(2*S*,3*S*,4*R*)-1-methyliminooxy-2-[(octadecanoyl)amino]-octadecane-3,4-diol (27)

Compound **15** (1.08 g, 1.80 mmol) was dissolved in CH_2Cl_2 (10 mL) and MeOH (10 mL). Formaldehyde solution (37% in MeOH, 1.0 mL) and AcOH (400 µL) were added to the solution at room temperature and the solution was stirred overnight. Then, solvent was removed by air-drying and white solid was suspended in MeOH. Filtration gave compound **27** (894 mg, 81%) as a white solid.

¹H NMR (500 MHz, CHLOROFORM-*d*) δ ppm 7.06 (d, *J*=7.82 Hz, 1 H; *H*₂C=NOa) 6.50 (d, *J*=7.82 Hz, 1 H; *H*₂C=NOb) 6.05 (br. s, 1 H; N*H*) 4.46 (dd, *J*=12.70, 5.86 Hz, 1 H; *H*-1a) 4.23 - 4.32 (m, 2 H; *H*-1b, *H*-2) 3.54 - 3.66 (m, 2 H; *H*-3, *H*-4) 3.37 - 3.42 (m, 1 H; O*H*-3) 2.12 - 2.24 (m, 3 H; O*H*-4, NHCOC*H*₂) 1.75-1.18 (m, 56 H) 0.87 (t, *J*=6.84 Hz, 6 H, C*H*₃); F1 (ppm) 139.00 (H₂C=NO) 74.44 (C-3) 73.62 (C-4) 73.17 (C-1) 51.68 (C-2) 36.50-21.88 (CH₂) 14.10 (CH₃)

(2S,3S,4R)-1-methylaminooxy-2-[(octadecanoyl)amino]-octadecane-3,4-diol (16)

To a suspension containing compound **27** (122 mg, 200 μ mol) in THF (2 mL), NaBH₃CN (38 mg, 600 μ mol) and AcOH (8 mL) were added at room temperature. The reaction mixture was stirred for 1 h and neutralized with 4 N NaOH aq.. The solution was diluted with EtOAc and washed with sat. NaHCO₃ 3 times and brine. The organic phase was dried over MgSO₄ and concentrated under reduced pressure. Purification by flash column chromatography on silica gel (Hexane-EtOAc, 2:1-1:1, DCM-THF, 3:1-1:1) yielded **16** (98 mg, 80%) as a white amorphous solid.

¹H NMR (500 MHz, CHLOROFORM-*d*) δ ppm 6.31 - 6.37 (m, 1 H; N*H*) 4.24 - 4.32 (m, 1 H; *H*-2) 3.99 - 4.05 (m, 1 H; *H*-1a) 3.84 - 3.90 (m, 1 H; *H*-1b) 3.55 - 3.62 (m, 1 H; *H*-4) 3.50 - 3.54 (m, 1 H; *H*-3) 2.73 (s, 3 H; *Me*NHO) 2.17 - 2.23 (m, 2 H; NHCOC*H*₂) 1.76 - 1.26 (m, 56 H; C*H*₂) 0.88 (m, 6 H; C*H*₃); F1 (ppm) 75.74 (*C*-3) 74.08 (*C*-4) 72.42 (*C*-1) 51.35 (*C*-2) 39.07 (*Me*NHO) 37.25-21.98 (*C*H₂) 14.18 (*C*H₃)

Experimental Procedure of Chemical Synthesis of H2NNH-PS-derivatives 17

HOOC-PS-Fmoc-Isop 28

To a solution of compound **8** (3.45 g, 5.95 mmol) and NaHCO₃ (1.01 g, 12.1 mmol) in CH₂Cl₂ (40 mL) and H₂O (30 mL), TEMPO (50 mg, 302 μ mol), KBr (80 mg, 604 μ mol), and Na₂CO₃ (30 mg, 302 μ mol) were added at 0°C. The solution was stirred vigorously at 0°C for 5 min then added dropwise *t*-BuOCl (1 mL, 9.06 mmol). After 19 hours, the reaction was quenched by MeOH, diluted with CHCl₃, and washed with brine. The organic phase was dried over MgSO₄ and concentrated under reduced pressure. Purification of the crude product was performed by flash column chromatography on silica gel (hexane-EtOAc, 7:2, CHCl₃-MeOH, 10:1) gave compound **28** (1.89 g, 53%) as a white solid;

¹H NMR (500 MHz, DMSO- d_6 , 328 K) δ ppm 7.86 (d, *J*=7.57 Hz, 2 H; Fmoc-Ar) 7.75 (br. s, 1 H; N*H*) 7.66 - 7.70 (m, 2 H; Fmoc-Ar) 7.40 (t, *J*=7.57 Hz, 2 H; Fmoc-Ar) 7.28 - 7.33 (m, 2 H; Fmoc-Ar) 4.30 (dd, *J*=10.01, 6.84 Hz, 1 H; Fmoc-CH₂a) 4.16 - 4.26 (m, 3 H; Fmoc-CH, Fmoc-CH₂b, *H*-3) 4.08 - 4.14 (m, 1 H; *H*-4) 4.01 - 4.08 (m, 1 H; *H*-2) 1.51-1.05 (m, 26 H; CH₂) 1.37 (s, 3 H; (RO)₂CCH₃a) 1.25 (s, 3 H; (RO)₂CCH₃b) 0.85 (t, *J*=7.08 Hz, 3 H; CH₃); F1 (ppm) 127.54 126.82 124.93 119.95 (Ar) 76.91 (C-3) 75.56 (C-4) 65.46 (Fmoc-CH₂) 54.37 (C-2) 46.39 (Fmoc-CH) 31.78-21.49 (CH₂) 27.12 ((RO)₂CCH₃a) 25.15 ((RO)₂CCH₃b) 14.10 (CH₃)

Lactone-Fmoc 29

To a suspension of compound **28** (310 mg, 522 μ mol) in CH₂Cl₂/H₂O (4:1, 2.5 mL) was added TFA (7.5 mL) and the mixture was stirred at room temperature for 3 hours. The reaction mixture was diluted with EtOAc and washed with H₂O (2 times), sat. NaHCO₃ aq. (3 times) and brine. The organic phase was dried over MgSO₄, and

concentrated under reduced pressure. Purification of the crude product by flash column chromatography on silica gel (hexane-EtOAc, 5:1~3:1) yielded **29** (270 mg, 96%) as a white solid;

¹H NMR (500 MHz, DMSO-*d*₆) δ ppm 7.87 (d, *J*=7.34 Hz, 2 H; Fmoc-Ar) 7.75 (br. s, 2 H; Fmoc-Ar) 7.41 (t, *J*=7.03 Hz, 2 H; Fmoc-Ar) 7.29 - 7.36 (m, 2 H; Fmoc-Ar) 5.67 (br. s, 1 H; N*H*) 4.58 (br. s, 1 H; *H*-2) 4.21 - 4.38 (m, 4 H; Fmoc-C*H*₂, Fmoc-C*H*, *H*-4) 4.08 (br. s, 1 H; *H*-3) 1.62 (br. s, 2 H; C*H*₂) 1.18 - 1.43 (m, 56 H; C*H*₂) 0.85 (t, *J*=6.73 Hz, 3 H; C*H*₃); F1 (ppm) 127.50 127.05 125.13 120.00 (Ar) 86.34 (C-4) 70.03 (C-3) 65.72 (Fmoc-C*H*₂) 53.64 (C-2) 46.75 (Fmoc-CH) 31.77-20.48(C*H*₂) 13.71 (C*H*₃)

Lactone-Stearoyl 30

Compound **29** (162 mg, 300 µmol) was dissolved in DMF (5 mL) and piperidine (150 µL, 1.5 mmol) was added to the solution at room temperature. After 30 minutes, solvent was evaporated with toluene and the crude product was used for the next reaction without further purification. The product and StearoylOSu (300 mg) were suspended in THF (5 mL) and the reaction mixture was stirred at room temperature for 1 day. The solution was concentrated under reduced pressure. The desired product was precipitated in MeOH and filtration gave compound **30** as a white solid (105 mg, 60%, 2 steps). ¹H NMR (500 MHz, pyridine) δ ppm 8.96 (br. s, 1 H; NH) 5.61 - 5.65 (m, 1 H; H-2) 4.72 - 4.77 (m, 1 H; H-4) 4.69 (d, *J*=5.29 Hz, 1 H; H-3) 2.62 - 1.21 (m, 58 H; CH₂) 0.91

(t, *J*=6.53 Hz, 6 H; *CH*₃); F1 (ppm) 87.55 (*C*-4) 71.78 (*C*-3) 52.83 (*C*-2) 36.38-22.51 (*C*H₂) 13.99 (*C*H₃)

H2NNH-PS-Stearic 17

To a solution of compound 20 (320 mg, 552 μ mol) in CHCl₃/MeOH (2:1, 30 mL), hydrazine monohydrated (1.2 mL) was added. The reaction solution was stirred at room temperature for 1.5 hours and then concentrated by air-drying. The desired product was precipitated in MeOH and filtration gave compound 20 (307 mg, 91%) as a white solid. ¹H NMR (500 MHz, CHLOROFORM-*d*, 328 K) δ ppm 7.90 (br. s, 1 H; H₂NN*H*) 6.51 (d, *J*=7.57 Hz, 1 H; N*H*) 4.64 - 4.71 (m, 1 H; *H*-2) 3.87 (br. s, 2 H; *H*₂NNH) 3.67 (br. s, 1 H; *H*O-3) 3.51 - 3.61 (m, 2 H; *H*-2) 2.37 (br. s, 1 H; *H*O-4) 2.23 (t, *J*=7.33 Hz, 2 H; *CH*₂) 1.82-1.07 (m, 56 H; C*H*₂) 0.88 (t, *J*=6.35 Hz, 6 H; C*H*₃); F1 (ppm) 76.24 (*C*-3) 73.24 (*C*-4) 51.83 (*C*-2) 36.07-23.09 (*C*H₂) 13.69 (*C*H₃)

ExperimentalProcedureofChemicalSynthesisofH2NO-/MeHNO-Ser-derivatives 18, 19

Carbamic acid, *N*-[(1*S*)-1-(hydroxymethyl)-2-oxo-2-(tetradecylamino)ethyl]-, phenylmethyl ester 32

WSC (4.32 g, 22.5 mmol) was added to a DCM solution (200 mL) containing Z-Serine **31** (4.50 g, 18.8 mmol), tetradecylamine (4.80 g, 22.5 mmol) and HOBt[·] H₂O (3.00 g, 19,6 mmol) at room temperature. The reaction mixture was stirred vigorously for 2 days and then diluted with CHCl₃. The solution was washed with 1 N aqueous HCl, sat. NaHCO₃ aq. and brine. Another CHCl₃ solution was used for extraction from aqueous phases two times. Organic phases were combined, dried over MgSO₄ and concentrated under reduced pressure. Purification by recrystallization (hexane-EtOAc) gave compound **32** as a white solid (6.95 g). Filtrate was evaporated and purification by flash column chromatography on silica gel (Hexane-EtOAc, 1:1) yielded **32** (315 mg) as a white solid. Totally, 7.27 g of compound **32** was obtained (89%). ; ¹H NMR (500 MHz, CDCl₃): δ 7.36 (m, 5H; Ar-*H*), 6.52 (br.s, 1H; N*H*-Alkyl), 5.79 (br.s, 1H; N*H*-Z), 5.14 (s, 2H; PhC*H*₂), 4.14 (m, 2H; Ser- α *H*, Ser- β *H*a), 3.66 (br.s, 1H; Ser- β *H*b), 3.23 (m, 2H; NHC*H*₂), 2.93 (br.s, 1H; O*H*), 1.47 (m, 2H; NHCH₂C*H*₂), 1.25 (m, 22H; C*H*₂), 0.88 (t, *J* = 6.60 Hz, 3H; C*H*₃);

Carbamic acid, *N*-[(1*S*)-1-(*tert*-butyldimethylsilyloxymethyl)-2-oxo-2-(tetradecylamino)ethyl]-, phenylmethyl ester 33

To a mixture containing compound **32** (6.95 g, 16.0 mmol) and imidazole (3.27 g, 48.0 mmol) in CH₂Cl₂ (150 mL), TBSCl (3.61 g, 24 mmol) was added at room temperature. The reaction mixture was stirred vigorously overnight and then diluted with EtOAc. The solution was washed with sat. NaHCO₃ aq. and then CHCl₃ solution was used for extraction from aqueous phases. Organic phases were combined, dried over MgSO₄ and concentrated under reduced pressure. Purification by flash column chromatography on silica gel (Hexane-EtOAc, 20:1-5:1) yielded **33** quantitatively as a white solid. ; ¹H NMR (500 MHz, CDCl₃): δ 7.36 (m, 5H; Ar-*H*), 6.44 (br.s, 1H; N*H*-Alkyl), 5.68 (br.s, 1H; N*H*-Z), 5.12 (m, 2H; PhC*H*₂), 4.14 (br.s, 1H; Ser- α *H*), 4.04 (dd, *J* = 9.65, 3.51 Hz, 1H; Ser- β *H*a), 3.61 (br.s, 1H; Ser- β *H*b), 3.24 (m, 2H; NHC*H*₂), 1.47 (m, 2H; NHCH₂C*H*₂), 1.25 (m, 22H; C*H*₂), 0.89 (m, 12H; C*H*₃, *tBu*), 0.09 (m, 6H; Si-C*H*₃);

Propanamide,

3-tert-butyldimethylsilyloxy-N-tetradecyl-2-[(triphenylmethyl)amino]-, (2S)- (34)

To a THF solution (50 mL) containing compound **33** (9.02 g, 16.4 mmol), 10% Pd/C (3.0 g) was added at room temperature. The reaction mixture was stirred vigorously for 5 hours under a 0.25 MPa of H_2 atmosphere and then diluted with CHCl₃. The solid reagent was removed by celite filter and filtrate was evaporated under reduced pressure. Then, the mixture was dissolved in toluene and evaporated under reduced pressure again. This crude product was used directly for the next reaction without any purification steps.

The crude product was dissolved in CH₂Cl₂ (150 mL) and DIEA (5.65 mL). Then TrCl (5.50 g, 19.7 mmol) was added to a solution and the reaction mixture was stirred vigorously for 11 hours at room temperature. EtOAc was used for dilution and the solution was washed with sat. NaHCO₃ aq. and brine. Organic phase was dried over MgSO₄ and concentrated under reduced pressure. Purification by flash column chromatography on silica gel (Hexane-EtOAc, 25:1-5:1) yielded **34** (9.34 g, 86%) as a yellow oil.; ¹H NMR (500 MHz, CDCl₃): δ 7.43 (m, 6H; Ar-*H*), 7.26 (m, 7H; N*H*-Alkyl, Ar-*H*), 7.20 (m, 3H; Ar-*H*), 3.62 (dd, *J* = 9.52, 2.98 Hz, 1H; Ser- β Ha), 3.24 (br.s, 1H; Ser- α H), 3.20 (m, 1H; N*H*-Tr), 3.15 (m, 2H; NHCH₂), 2.49 (dd, *J* = 9.52, 4.46 Hz, 1H; Ser- β Hb), 1.47 (m, 2H; NHCH₂CH₂), 1.26 (m, 22H; CH₂), 0.88 (t, *J* = 7.14, 3H; CH₃), 0.81 (s, 1H, *tBu*), -0.09 (m, 3H; Si-CH₃a), -0.11 (m, 3H; Si-CH₃a);

Propanamide, 3-hydroloxy-N-tetradecyl-2-[(triphenylmethyl)amino]-, (2S)- (35)

To a THF solution (100 mL) containing compound **34** (9.34 g, 14.2 mmol), TBAF solution (1 M in THF, 14.9 mL, 14.9 mmol) was added at room temperature. The solution was stirred vigorously for 45 minutes and evaporated under reduced pressure. Purification by flash column chromatography on silica gel (Toluene-EtOAc, 20:1,

Hexane-EtOAc, 2:1) yielded **35** (6.84 g, 89%) as a yellow oil.; ¹H NMR (500 MHz, CDCl₃): δ 7.41 (m, 6H; Ar-*H*), 7.31 (m, 1H; N*H*-Alkyl), 7.26 (m, 6H; Ar-*H*), 7.20 (m, 3H; Ar-*H*), 3.52 (dd, *J* = 10.64, 3.55 Hz, 1H; Ser- β *H*a), 3.25 (br.s, 1H; Ser- α *H*), 3.16 (m, 1H; NHC*H*₂a), 3.05 (m, 1H; NHC*H*₂b), 2.99 (m, 1H; N*H*-Tr), 2.66 (dd, *J* = 10.64, 5.02 Hz, 1H; Ser- β *H*b), 2.48 (br.s, 1H; O*H*), 1.43 (m, 2H; NHCH₂C*H*₂), 1.29 (m, 22H; C*H*₂), 0.88 (t, *J* = 6.80, 3H; C*H*₃);

Propanamide, 3-phthalimidyloxy-*N*-tetradecyl-2-[(triphenylmethyl)amino]-, (2S)-(36)

Into a THF solution (100 mL) containing compound **35** (6.37 g, 11.7 mmol), Ph₃P (3.69 g, 14.1 mmol) and *N*-hydroxyphthalimide (2.30 g, 14.1 mmol), DIAD (2.78 mL, 14.1 mmol) was dropped at 0 degree. After 15 min, the reaction mixture was stirred at 50 degree for 6 hours. Then, the solution was evaporated under reduced pressure. Purification by flash column chromatography on silica gel (Hexane-EtOAc, 20:1-6:1) yielded **36** (6.77 g, 84%) as a noncolored viscous oil.; ¹H NMR (500 MHz, CDCl₃): δ 7.74 (m, 2H; Phth-Ar-*H*), 7.66 (m, 2H; Phth-Ar-*H*), 7.61 (m, 1H; N*H*-Alkyl), 7.47 (d, *J* = 8.15, 6H; Tr-Ar-*H*), 7.16 (t, *J* = 7.85, 6H; Tr-Ar-*H*), 7.07 (t, *J* = 7.25, 3H; Tr-Ar-*H*), 4.33 (d, *J* = 9.36 Hz, 1H; Ser-β*H*a), 3.78 (d, *J* = 6.95 Hz, 1H; N*H*-Tr), 3.37 (m, 2H; Ser-α*H*, Ser-β*H*b), 3.17 (m, 1H; NHC*H*₂a), 3.02 (m, 1H; NHC*H*₂b), 1.47 (m, 2H; NHCH₂C*H*₂), 1.29 (m, 22H; C*H*₂), 0.89 (t, *J* = 6.80, 3H; C*H*₃); (¹³C NMR (125 MHz (HSQC), CDCl₃): δ 171.35, 163.22, 145.44, 134.31, 128.27, 128.24, 127.54, 126.25, 123.17, 79.10, 71.38, 55.45, 39.20, 31.58, 29.35, 29.34, 29.32, 29.31, 29.25, 29.22, 29.06, 29.01, 28.94, 26.53, 22.35, 13.83;

Propanamide, 3-aminooxy-N-tetradecyl-2-[(triphenylmethyl)amino]-, (2S)- (37)

To a THF solution (85 mL) containing compound **36** (5.88 g, 8.55 mmol), MeNH₂ (40% in MeOH, 3.32 mL, 30.0 mmol) was added at room temperature and the reaction mixture was stirred for 4 hours. Then, the solution was diluted with toluene and evaporated under reduced pressure. Purification by flash column chromatography on silica gel (Hexane-EtOAc, 3:1-2:1-1:1) yielded **37** (4.45 g, 95%) as a noncolored oil.; ¹H NMR (500 MHz, CDCl₃): δ 7.42 (dd, *J* = 8.25, 0.88, 6H; Ar-*H*), 7.28 (t, *J* = 7.07, 6H; Ar-*H*), 7.21 (m, 4H; N*H*-Alkyl, Ar-*H*), 5.19 (br.s, 2H; ON*H*₂), 3.79 (dd, *J* = 10.31, 3.83 Hz, 1H; Ser- β Ha), 3.36 (m, 1H; N*H*-Tr), 3.18 (m, 1H; Ser- α H), 3.09 (m, 2H; NHC*H*₂), 2.72 (dd, *J* = 10.31, 4.12, 1H; Ser- β Hb), 1.43 (m, 2H; NHCH₂C*H*₂), 1.29 (m, 22H; C*H*₂), 0.88 (t, *J* = 6.92, 3H; C*H*₃); (¹³C NMR (125 MHz (HSQC), CDCl₃): δ 172.84, 145.70, 128.69, 127.99, 126.81, 76.18, 71.56, 57.66, 39.30, 31.88, 29.65, 29.63, 29.62, 29.61, 29.57, 29.51, 29.31, 26.99, 22.64, 14.08;

Carbamic acid, N-

3-[(2S)-1-oxo-1-tetradecylamino-2-(triphenylmethyl)amino]-propanoxy-,

9H-fluoren-9-ylmethyl ester (38)

Compound **37** (4.97 g, 8.91 mmol) and Fmoc-OSu (3.61 g, 10.7 mmol) were dissolved in THF (60 mL) and the reaction mixture was stirred at room temperature for 2 hours. Then, the solution was evaporated under reduced pressure. Purification by flash column chromatography on silica gel (Hexane-EtOAc, 10:1-5:1) yielded **38** (5.97 g, 86%) as a white amorphous solid.; ¹H NMR (500 MHz, CDCl₃): δ 7.77 (d, *J* = 7.57, 2H; Fmoc-Ar-*H*), 7.52 (d, *J* = 7.57, 2H; Fmoc-Ar-*H*), 7.42 (m, 9H; N*H*-Alkyl, Fmoc-Ar-*H*, Tr-Ar-*H*), 7.31 (m, 3H; ON*H*, Fmoc-Ar-*H*), 7.23 (t, *J* = 7.57, 6H; Tr-Ar-*H*), 7.17 (m, 3H; Tr-Ar-*H*), 4.43 (dd, J = 10.01, 6.84 Hz, 1H; Fmoc-C*H*₂a), 4.30 (m, 1H; Fmoc-C*H*₂b), 4.15 (t, J = 7.33 Hz, 1H; Fmoc-C*H*), 3.97 (dd, J = 10.50, 2.44 Hz, 1H; Ser- β *H*a), 3.62 (br.s, 1H; N*H*-Tr), 3.36 (m, 1H; Ser- α *H*), 3.16 (m, 1H; NHC*H*₂a), 3.03 (m, 1H; NHC*H*₂b), 2.95 (dd, J = 10.50, 5.37, 1H; Ser- β *H*b), 1.43 (m, 2H; NHCH₂C*H*₂), 1.26 (m, 22H; C*H*₂), 0.88 (t, J = 6.72, 3H; C*H*₃); (¹³C NMR (125 MHz (HSQC), CDCl₃): δ 157.43, 145.73, 143.39, 141.24, 128.71, 127.94, 127.08, 126.68, 124.99, 120.00, 77.94, 71.64, 67.68, 56.59, 46.78, 39.45, 31.88, 29.59, 29.31, 26.97, 22.67, 14.07;

Carbamic acid, N-3-[(2S)-2-amino-1-oxo-1-tetradecylamino]-propanoxy-,

9H-fluoren-9-ylmethyl ester (39)

To a solution of compound **38** (780 mg, 1.0 mmol) in CH₂Cl₂ (8 mL) and MeOH (1.5 mL), TFA (500 mL) was added and the reaction mixture was stirred at room temperature for 1 hour. Then, the solution was neutralized by sat. NaHCO₃ aq. and extracted with CHCl₃ three times. Organic phase was dried over MgSO₄ and evaporated under reduced pressure. Purification by flash column chromatography on silica gel (Hexane-EtOAc, 3:1, CHCl₃:*i*-PrOH, 40:1) yielded **39** (501 mg, 93%) as a white solid.; ¹H NMR (500 MHz, CDCl₃): δ 7.73 (d, *J* = 7.34, 2H; Ar-*H*), 7.68 (br.s, 1H; N*H*-Alkyl), 7.56 (d, *J* = 7.34, 2H; Ar-*H*), 7.37 (t, *J* = 7.64, 2H; Ar-*H*), 7.28 (t, *J* = 7.34, 2H; Ar-*H*), 4.46 (d, *J* = 6.72 Hz, 2H; Fmoc-CH₂), 4.19 (t, *J* = 6.72 Hz, 1H; Fmoc-CH), 4.03 (m, 1H; Ser- β Ha), 3.94 (m, 1H; Ser- β Hb), 3.53 (m, 1H; Ser- α H), 3.21 (m, 2H; NHCH₂), 1.89 (br.s, 2H; NH₂), 1.47 (m, 2H; NHCH₂CH₂), 1.26 (m, 22H; CH₂), 0.88 (t, *J* = 7.03, 3H; CH₃); (13C NMR (125 MHz (HSQC), CDCl3): δ 172.16, 157.74, 143.34, 143.33, 141.17, 127.70,

127.01, 124.90, 119.88, 78.74, 67.14, 53.52, 46.92, 39.23, 31.81, 29.59, 29.57, 29.55, 29.50, 29.46, 29.37, 29.25, 29.21, 26.84, 22.58, 14.02;

Carbamic acid, N-

3-[(2S)-2-octadecanoylamino-1-oxo-1-tetradecylamino]-propanoxy-,

9H-fluoren-9-ylmethyl ester (40)

Compound 39 (1.78 g, 3.31 mmol) and StearicOSu (1.52 g, 3.97 mmol) were dissolved in THF (30 mL) and the reaction mixture was stirred at room temperature for 3 days. Then, precipitated white solid was collected by filtration (1.78 g) and the filtrate was evaporated under reduced pressure. Purification by flash column chromatography on silica gel (Hexane-EtOAc, 10:1-3.5:1-2:1) yielded 40 (510 mg) as a white solid. Totally, 2.29 g of compound **40** was obtained (86%). ; ¹H NMR (500 MHz, CDCl₃): δ 8.78 (s, 1H; ONH), 7.75 (d, J = 7.79, 2H; Ar-H), 7.62 (br.s, 1H; NH-Alkyl), 7.56 (dd, J = 7.45, 1.02, 2H; Ar-H), 7.39 (dd, J = 7.45, 2.37, 2H; Ar-H), 7.30 (t, J = 7.45, 2H; Ar-H), 7.03 (br.s, 1H; Ser-NH), 4.66 (m, 1H; Ser- α H), 4.44 (m, 2H; Fmoc-CH₂), 4.21 (t, J = 7.11 Hz, 1H; Fmoc-CH), 4.16 (dd, J = 11.51, 4.74, 1H; Ser- β Ha), 3.94 (dd, J = 11.17, 7.45, 1H; Ser- β Hb), 3.24 (m, 2H; NHCH₂), 2.24 (t, J = 7.62, 2H; COCH₂), 1.63 (m, 2H; $COCH_2CH_2$), 1.47 (m, 2H; NHCH₂CH₂), 1.23 (m, 50H; CH₂), 0.88 (t, J = 6.94, 6H; CH₃); (13C NMR (125 MHz (HSQC), CDCl₃): δ 173.75, 169.05, 158.37, 143.29, 143.22, 141.25, 141.24, 127.87, 127.15, 127.12, 124.99, 120.04, 76.19, 67.94, 50.98, 46.86, 39.83, 36.55, 31.90, 29.68, 29.64, 29.62, 29.60, 29.54, 29.49, 29.34, 29.25, 26.84, 25.58, 22.67, 14.09;

Octadecanamide, [1-(aminooxymethyl)-2-oxo-2-(tetradecylamino)ethyl]-, (S) (18)

To a flask containing compound **40** (2.69 g, 3.34 mmol), THF (40 mL) and CHCl₃/MeOH (2:1, 10 mL) was added. Although the starting material was not dissolved completely, piperidine (3.30 mL, 33.4 mmol) was added to the reaction mixture and the suspension was stirred at room temperature for 2 hours. Then, the solvent was reduced by air-drying and Et₂O was added for precipitation. Filtration gave compound **18** (1.76 g, 91%) as a white solid. ; ¹H NMR (500 MHz, CDCl₃, 50 °C): δ 6.44 (d, *J* = 7.35, 1H; Ser-N*H*), 6.31 (br.s, 1H; N*H*-Alkyl), 5.59 (br.s, 2H; ON*H*₂), 4.70 (dd, *J* = 13.00, 5.94, 1H; Ser- α *H*), 3.89 (m, 1H; Ser- β *H*a), 3.81 (m, 1H; Ser- β *H*b), 3.24 (m, 2H; NHC*H*₂), 2.23 (t, *J* = 7.63, 2H; COC*H*₂), 1.63 (m, 2H; COCH₂C*H*₂), 1.49 (m, 2H; NHCH₂C*H*₂), 1.27 (m, 50H; C*H*₂), 0.88 (t, *J* = 6.50, 6H; C*H*₃); (13C NMR (125 MHz (HSQC), CDCl₃, 50 °C): δ 173.50, 169.68, 75.42, 52.25, 39.74, 36.69, 31.92, 29.69, 29.65, 29.62, 29.59, 29.55, 29.50, 29.33, 29.27, 26.89, 25.64, 22.66, 14.03;

Octadecanamide, [1-(methyliminooxymethyl)-2-oxo-2-(tetradecylamino)ethyl]-, (S) ¹H NMR (500 MHz, CDCl₃, 50 °C): δ 7.05 (d, J = 7.82, 1H; ON=CH₂a), 6.48 (d, J = 7.82, 1H; ON=CH₂b), 6.32 (d, J = 7.57, 1H; Ser-NH), 6.16 (br.s, 1H; NH-Alkyl), 4.66 (dd, J = 12.70, 5.86, 1H; Ser-αH), 4.40 (m, 1H; Ser-βHa), 4.24 (m, 1H; Ser-βHb), 3.25 (m, 2H; NHCH₂), 2.23 (t, J = 7.33, 2H; COCH₂), 1.63 (m, 2H; COCH₂CH₂), 1.48 (m, 2H; NHCH₂CH₂), 1.27 (m, 50H; CH₂), 0.88 (t, J = 6.84, 6H; CH₃);

Octadecanamide, [1-(methylaminooxymethyl)-2-oxo-2-(tetradecylamino)ethyl]-, (S) (19) ¹H NMR (500 MHz, CDCl₃, 50 °C): δ 6.51 (d, J = 7.07, 1H; Ser-N*H*), 6.31 (br.s, 1H; N*H*-Alkyl), 4.64 (dd, J = 12.73, 5.94, 1H; Ser-α*H*), 3.92 (m, 4.74, 1H; Ser-β*H*a), 3.82 (m, 1H; Ser-β*H*b), 3.25 (dd, J = 13.01, 6.79, 2H; NHC*H*₂), 2.71 (s, 3H; NH*Me*), 2.24 (t, J = 7.64, 2H; COC*H*₂), 1.64 (m, 2H; COCH₂C*H*₂), 1.49 (m, 2H; NHCH₂C*H*₂), 1.27 (m, 50H; C*H*₂), 0.88 (t, J = 6.79, 6H; C*H*₃);

2-5 References

- 1. W. Curatolo, "Glycolipid function", Biochim. Biophys. Acta, 1987, 906, 137-160.
- 2. A. H. Merrill, Jr., "Sphingolipid and Glycosphingolipid Metabolic Pathways in the Era of Sphingolipidomics", *Chem. Rev.*, **2011**, 111, 6387-6422
- T. Wennekes, R. J. B. H. N. van den Berg, R. G. Boot, , G. A. van der Marel, H. S. Overkleeft, and J. M. F. G. Aerts, "Glycosphingolipids--nature, function, and pharmacological modulation", *Angew. Chem. Int. Ed.*, 2009, 48, 8848-8869
- D. Zhou, J. Mattner, C. Cantu III, N. Schrantz, N. Yin, Y. Gao, Y. Sagiv, K. Hudspeth, Y.-P. Wu, T. Yamashita, S. Teneberg, D. Wang, R. L. Proia, S. B. Levery, P. B. Savage, L. Teyton, and A. Bendelac, "Lysosomal Glycosphingolipid Recognition by NKT Cells", *Science*, 2004, 306, 1786-1789
- D. Wu, M. Fujio, and C.-H. Wong, "Glycolipids as immunostimulating agents", *Bioorg. Med. Chem.*, 2008, 16, 1073-1083
- R. D. Goff, S. Singh, and J. S. Thorson, "Glycosyloxyamine neoglycosylation: a model study using calicheamicin", *ChemMedChem*, 2011, 6, 774-776
- Y. Liu, T. Feizi, M. A. Campanero-Rhodes, R. A. Childs, Y. Zhang, B. Mulloy, P. G. Evans, H. M. I. Osborn, D. Otto, P. R. Crocker, and W. Chai, "Neoglycolipid probes prepared via oxime ligation for microarray analysis of oligosaccharide-protein interactions", *Chemistry & Biology*, 2007, 14, 847-859
- Y. D. Vankara and R. R. Schmidt, "Chemistry of glycosphingolipids—carbohydrate molecules of biological significance", *Chem. Soc. Rev.*, 2000, 29, 201-216

- J. A. Morales-Serna, O. Boutureira, Y. Diaz, M. I. Matheu and S. Castillon, "Recent advances in the glycosylation of sphingosines and ceramides", *Carbohydrate Research*, 2007, 342, 1595-1612
- S.-I. Nishimura, K. Yamada, "Polymer-Assisted Enzymatic Synthesis of Ganglioside GM3 Water-Solible Glycopolymer Having Cermide Mimeic Linker as an Efficient Polymer Support", J. Am. Chem. Soc., 1997, 119, 10555-10556
- K. Yamada, E. Fujita, and S.-I. Nishimura, "High performance polymer supports for enzyme-assisted synthesis of glycoconjugates", *Carbohydr. Res.*, 1997, 305, 443-461
- S. C. Timmons and J. S. Thorson, "Increasing carbohydrate diversity via amine oxidation: aminosugar, hydroxyaminosugar, nitrososugar, and nitrosugar biosynthesis in bacteria Shannon", *Curr. Opin. Chem. Biol.*, 2008, 12, 297-305
- F. Peri, P. Dumy, and M. Mutter, "Chemo- and Stereoselective Glycosylation of Hydroxylamino Derivatives : A Versatile Approach to Glycoconjugates", *Tetrahedron*, **1998**, *54*, 12269-12278
- 14. R. D. Goff and J. S. Thorson, "Assessment of chemoselective neoglycosylation methods using chlorambucil as a model", *J. Med. Chem.*, **2010**, 53, 8129-8139
- L. M. Obeid, Y. Okamoto, and C. Mao, "Yeast sphingolipids: metabolism and biology", *Biochim. Biophys. Acta*, 2002, 1585, 163-171
- J.-C. Jung, Y. Lee, S. Moon, J. H. Ryu, and S. Oh, "Phytoceramide Shows Neuroprotection and Ameliorates Scopolamine-Induced Memory Impairment", *Molecule*, **2011**, 16, 9090-9100
- 17. E.A. Sweeney, C. Sakakura, T. Shirahama, A. Masamune, H. Ohta, S. Hakomori, and Y. Igarashi, "Sphingosine and its methylated derivative

N,*N*-dimethylsphingosine (DMS) induce apoptosis in a variety of human cancer cell lines", *Int. J. Cancer*, **1996**, 66, 358-366

- I. Kuroda, M. Musman, I.I. Ohtani, T. Ichiba, J. Tanaka, D.G. Gravalos, and T. Higa, "Pachastrissamine, a Cytotoxic Anhydrophytosphingosine from a Marine Sponge, Pachastrissa sp.", *J. Nat. Prod.*, **2002**, 65, 1505-1506
- M. Morita, K. Motoki, , K. Akimoto, T. Natori, , T. Sakai, , E. Sawa, K. Yamaji, Y. Koezuka, E. Kobayashi, and H. Fukushima, "Structure-Activity Relationship of α-Galactosylceramides against B16-Bearing Mice" *J. Med. Chem.*, **1995**, 38, 2176-2187
- 20. J. M. Langenhan, N. R. Peters, I. A. Guzei, F. M. Hoffmann, and J. S. Thorson, "Enhancing the anticancer properties of cardiac glycosides by neoglycorandomization", *Proc. Natl. Acad. Sci. USA*, **2005**, 102, 12305-12310
- 21. M. R. Carrasco, M. J. Nguyen, Burnell, D. R., M. D. Maclaren, and S. M. Hengel, "Synthesis of neoglycopeptides by chemoselective reaction of carbohydrates with peptides containing a novel N^c-methyl-aminooxy amino acid", *Tetrahedron Lett.*, 2002, 43, 5727-5729
- 22. M. R. Carrasco and R. T. Brown, "A versatile set of aminooxy amino acids for the synthesis of neoglycopeptides" *J. Org. Chem.*, **2003**, 68, 8853-8858.
- 23. F. Peri, "Extending chemoselective ligation to sugar chemistry: convergent assembly of bioactive neoglycoconjugates", *Mini reviews in medicinal chemistry*, 2003, 3, 651-658
- M. Yamaoka, A. Nakazaki, and S. Kobayashi," Total synthesis of fomitellic acid B", *Tetrahedr. Lett.*, 2009, 50, 6764-6768

- 25. S. Kumar, R. Sharma, M. Garcia, J. Kamel, C. McCarthy, A. Muth, and O. Phanstiel, IV, "Chemoselective Amide Formation Using O-(4-Nitrophenyl)hydroxylamines and Pyruvic Acid Derivatives", *J. Org. Chem.*, **2012**, 77, 10835-10845
- 26. M. R. Carrasco, R. T. Brown, V. H. Doan, S. M. Kandel, and F. C. Lee,
 "2-(N-Fmoc)-3-(N-Boc-N- Methoxy)-Diaminopropanoic Acid, an Amino Acid for the Synthesis of Mimics of *O*-Linked Glycopeptides", *Petide Science*, 2006, 84, 414-420
- 27. S. Kim, S. Song, T. Lee, S. Jung, and D. Kim, "Practical Synthesis of KRN7000 from Phytosphingosine", *Synthesis*, **2004**, 6, 847-850
- 28. K. Fukunaga, M. Yoshida, F. Nakajima, R. Uematsu, M. Hara, S. Inoue, H. Kondoc, and S.-I. Nishimura "Design, Synthesis, and Evaluation of β-Galactosylceramide Mimics Promoting β-Glucocerebrosidase Activity in Keratinocytes", *Bioorg. Med. Chem. Lett.*, **2003**, 13, 813-815
- 29. G. Chen, J. Schmieg, M. Tsuji, and R. W. Franck, "Efficient Synthesis of α-C-Galactosyl Ceramide Immunostimulants: Use of Ethylene-Promoted Olefin Cross-Metathesis", Org. Lett., 2004, 6, 4077-4080
- 30. S. Peluso and B. Imperiali, "Asparagine surrogates for the assembly of *N*-linked glycopeptide mimetics by chemoselective ligation", *Tetrahedr. Lett.*, 2001, 42, 2085-2087

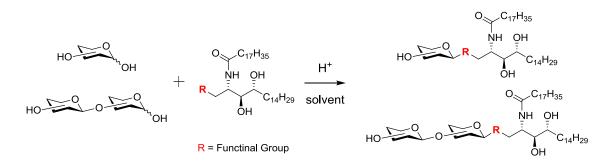
Chapter 3

Glycoblotting Reaction for Construction of Neoglycolipid Library

3-1 Introduction

Glycoblotting reaction^[1-3], also called neoglycosylation, has been used broadly due to its simplicity to conjugate sugars with aglycons. This method focuses on aldehyde group at reducing end of glycan. Aminooxy or hydrazide group can easily capture aldehyde group in acidic condition and the stability of oxime and hydrazone is enough to keep the linkage even though in aqueous solution^[4]. Our laboratory has used this reaction to simply pick up only sugars from biological samples, which include protein, lipid, and so on^[5]. That has enabled to perform simultaneous glycomics rapidly^[6-8]. This technique has been used to immobilize target glycans on the surface of microarray^[9-11] or nanoparticles^[12,13], conjugate glycans with biomolecules^[14], and construct glycoconjugate libraries^[15-20]. In a view of librarization, however, glycosphingolipids have not been a target compound yet^[21, 22].

In this chapter, I tried to construct a neoglycolipid library using functional ceramide to capture free sugars (Scheme 3-1).

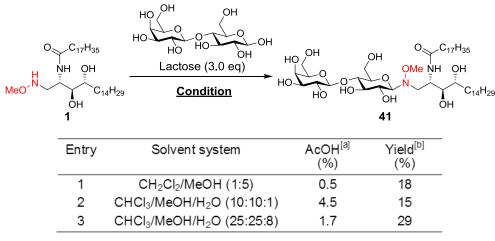


Scheme 3-1. Construction of neoglycolipid library via glycoblotting reaction

3-2 Results and Discussions

3-2-1 Glycoblotting Reaction Using Methoxyamino-Derivative 1

At first, we tried glycoblotting reaction using methoxyamino derivative **1** and lactose as shown in table 3-2-1. Although the yields were not so high, all reactions afforded the same conjugated product.



[a] AcOH percentage in total volume

[b] Yields of isolated products after silica gel chromatography

 Table 3-2-1.
 Non-natural lactosylceramide 41 (N-LacCer) obtained by one step

glycoblotting reaction of ceramide mimic 1

The structure of the purified product was analyzed by NMR measurement, which included ¹H, ¹³C, COSY, TOCSY, ¹H- ¹³C HSQC and HMBC experiments (Fig. 3-2-1). All peaks of HSQC measurement were assigned and the coupling constant of anomeric position at reducing end showed bigger value (8.8 Hz) which was enough to be confirmed as β -glycoside.

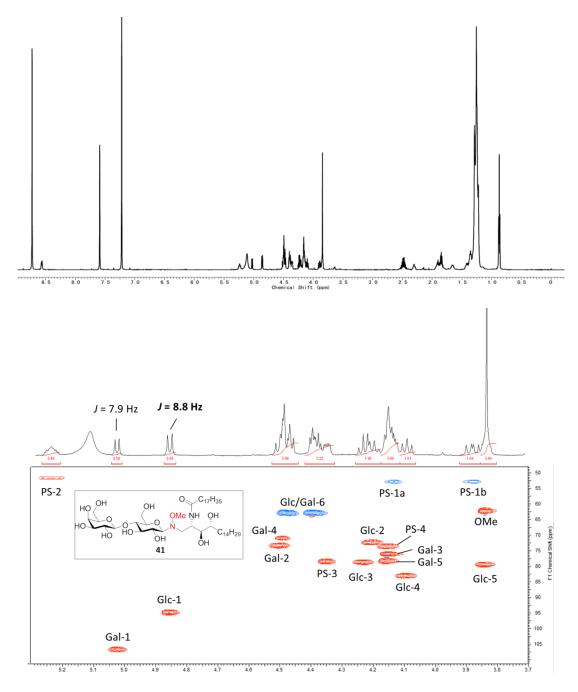


Fig. 3-2-1. NMR spectra of the conjugated product 41

upper: ¹H spectrum, under: ¹H- ¹³C HSQC spectrum

(d5-pyridine, 300 K)

Then, glycoblotting reaction was performed using other 9 free sugars shown below (Fig. 3-2-2). Di-/tri-saccharides were used under the optimized condition for lactose and monosaccharides under another condition including less H₂O. Though all reactions proceeded, two major products were detected on TLC analysis in case of some monosaccharides.

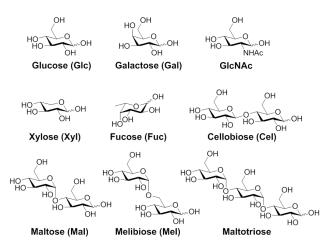


Fig. 3-2-2. Free sugars used for construction of library

To determine the structural difference of two compounds, NMR measurements were performed after acetylation using galactose derivatives. From the results, one of them was confirmed to be β -product as expected (Fig. 3-2-3)

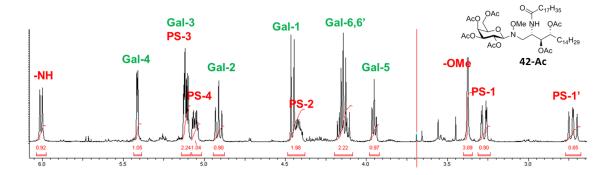


Fig. 3-2-3. ¹H NMR spectrum of acetylated β -product after neoglycosylation using galactose

```
(CDCl<sub>3</sub>, 300 K)
```

Another one was predicted α -product before the measurement because this type of functional group cannot form ring-open structure. However, the ¹H spectrum suggested it was not α -configurated compound (Fig. 3-2-4). Then, we conducted thorough analysis (¹H, ¹³C, COSY, TOCSY, ¹H- ¹³C HSQC, HMBC and NOESY experiments) which clarified unexpected structure caused by opening sugar ring and sphingosine cyclizing.

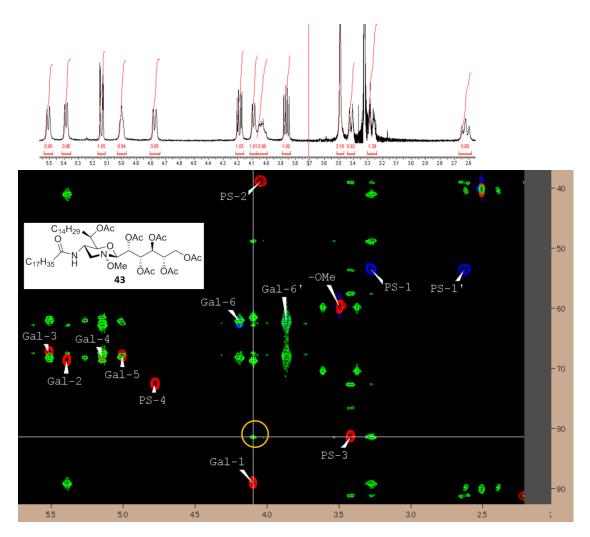


Fig. 3-2-4 HSQC and HMBC spectra of sphingosine-cyclized product 43

HSQC: Red and Blue, HMBC: Green (d₆-DMSO, 300 K)

The mechamism how sphingosine-cyclized product was generated was shown in Fig. 3-2-5. Oxime intermediate cannot exist stably due to positive electric charge on nitrogen atom. The instability is canceled by following attack of hydroxyl group at 5th position of sugar molecule normally (pass A). On the contrary, hydroxyl group at 3rd position of sphingosine is also able to work as a nucleophile by forming sphingosine-cyclized compound **43** (pass B).

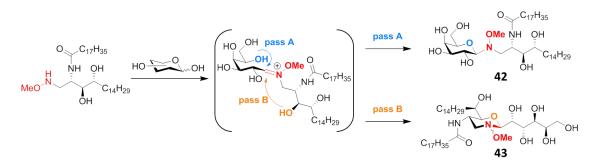


Fig. 3-2-5 Reaction mechanism of generating sphingosine-cyclized product

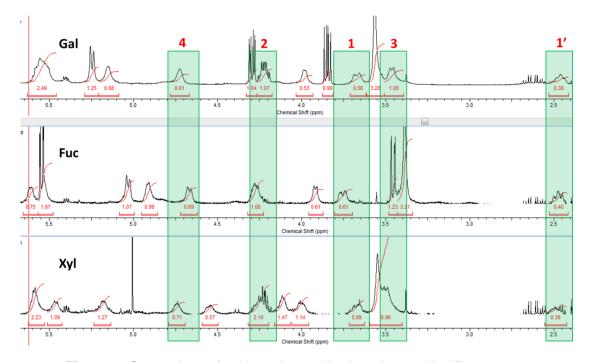


Fig. 3-2-6 Comparison of sphingosine-cyclized products with different sugar

Color bars show ¹H peaks of sphingosine (CDCI₃, 300 K)

In case of other sugars, this type of reaction was also occurred (Fig.3-2-6). The tendency to form cyclized-sphingosine was estimated as Gal > Fuc > Xyl, (Glc) which meant that might relate to the number of axial-configurated group. Additionally, 3 reasons:

- byproduct could orient all substituents toward equatrial direction
- acyl chains in byproduct could co-interact with each other as well as desired product
- The longer reaction time, the more byproduct was generated shown below

suggested that cyclized-sphingosine may be thermodynamic product in this reaction. Although separation of the 2 products was not impossible, laborious purification was far from the aim "simple preparation". That was why we optimized reaction condition to suppress production of the byproduct (Fig. 3-2-7).

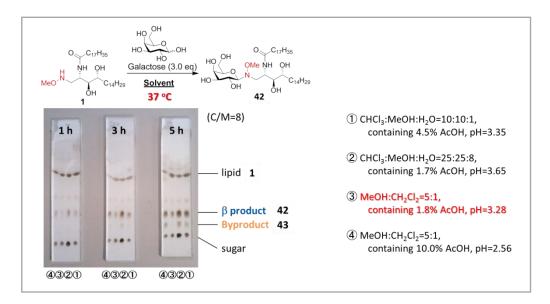


Fig. 3-2-7 Optimization of reaction condition to suppress production of the byproduct

As a result, it was revealed the most suited condition was entry 3 where no sphingosine-cyclized product was detected in 5 hours. That condition was applied to synthesis of other neglycolipids having monosaccharide on their head. As expected, production of the byproduct was suppressed and only β -product was obtained in all reactions. ¹H and ¹³C NMR data of anomeric position were shown in Table 3-2-2. The results indicated tremendous usability of this strategy for construction of neoglycolipids. Additionally, MALDI-TOF MS spectra of conjugated neoglycolipids are shown in Fig. 3-2-8

	Anomeric H1	Anomeric C1	J (Hz)	Yield (%)
	δ (ppm)	δ (ppm)		
Galctoside (42)	4.88	94.81	9.2	41
Glucoside (46)	4.94	93.69	8.3	54
Xyloside (45)	4.79	94.90	8.3	76
Fucoside (44)	4.58	95.18	8.8	50
GlcNAc-oside (47)	4.93	94.43	10.1	24
Lactoside (41)	4.86	93.75	8.8	29
Maltoside (48)	4.85	93.65	8.8	30
Cellbioside (49)	4.88	93.79	8.8	40
Mellibioside (50)	4.87	94.23	8.8	28
Maltotrioside (51)	4.92	93.48	8.8	30

Table 3-2-2 ¹H and ¹³CNMR data of anomeric position at reducing end

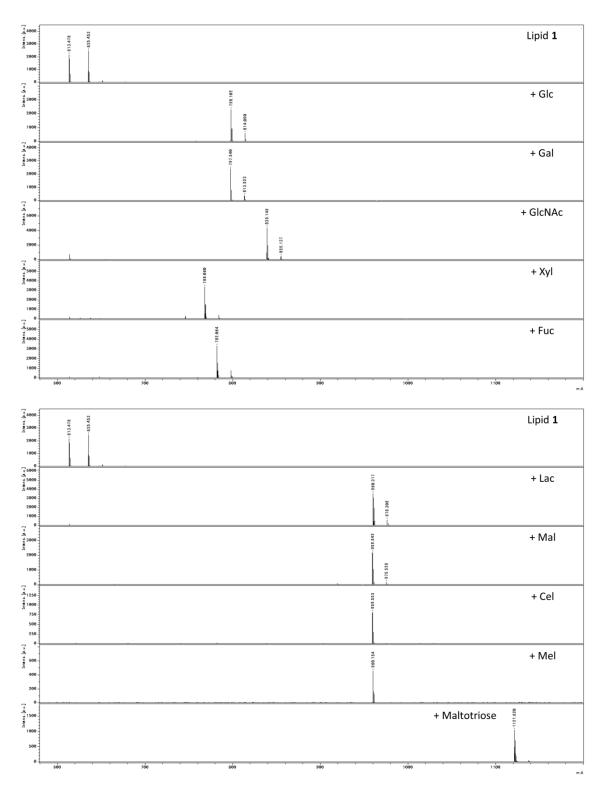
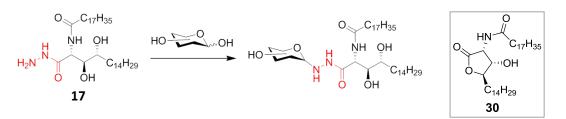


Fig. 3-2-8. MALDI-TOF MS spectra of conjugated products using compound 1

(DHB, positive)

3-2-2 Glycoblotting Reaction Using Other 5 Derivatives

Next, I used hydrazide derivative **17** as a lipid tag for glycoblotting reaction. The reaction first conducted at 50 °C in CHCl₃: MeOH : H_2O : AcOH = 10 : 10 : 1 : 1 system. However, in this condition, lactone compound **30** was major product (Scheme 3-2-3). This result showed hydrazide was decomposed by intramolecular lactonization .



Scheme. 3-2-3. Glycoblotting reaction using hydrazide derivative 17 and decomposed product 30

Then, I decreased reaction temperature below 40 degree which enabled to obtain neoglycosylated hydrazides. MALDI-TOF MS spectra of condugated hydrazides were shown in Fig. 3-2-9

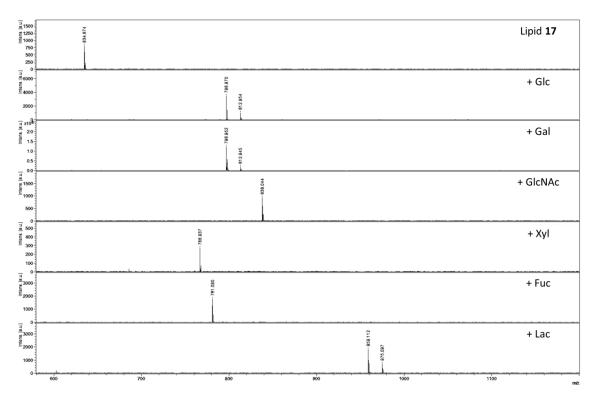
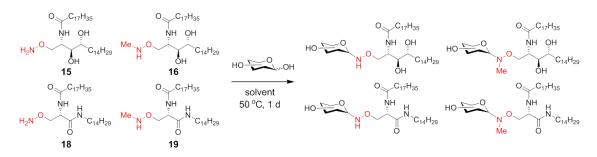


Fig. 3-2-9. MALDI-TOF MS spectra of conjugated products using compound 17

(DHB, positive)

The other ceramide derivatives were also used for glycoblotting (Scheme 3-2-4). These derivatives afforded neoglycosylated compounds almost without decomposition or yielding byproduct. The yields in case of ceramide derivative **15** were shown in Fig. 3-2-10. Results of each compound are shown in Fig. 3-2-11, 3-2-12, and 3-2-13, respectively. Additionally, methyl substituted aminooxy derivatives **16**, **19** mostly afforded β -product as expected (Fig. 3-2-14).



Scheme 3-2-4. Glycoblotting reaction using ceramide derivative 15, 16, 18, and 19

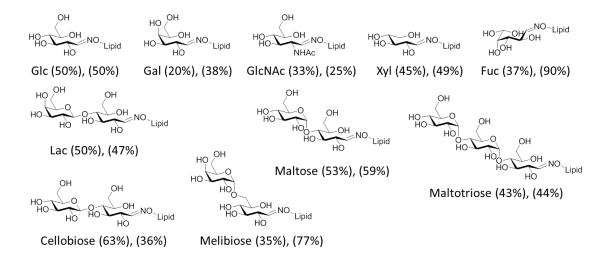


Fig 3-2-10. Yields of neoglycolipids in case of ceramide derivative 15 (left), 16 (right)

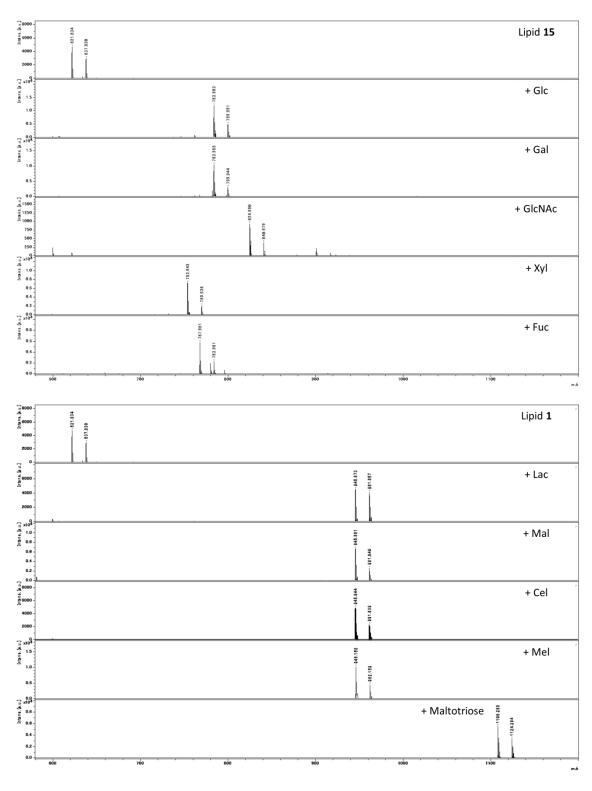


Fig. 3-2-11. MALDI-TOF MS spectra of conjugated products using compound 15

(DHB, positive)

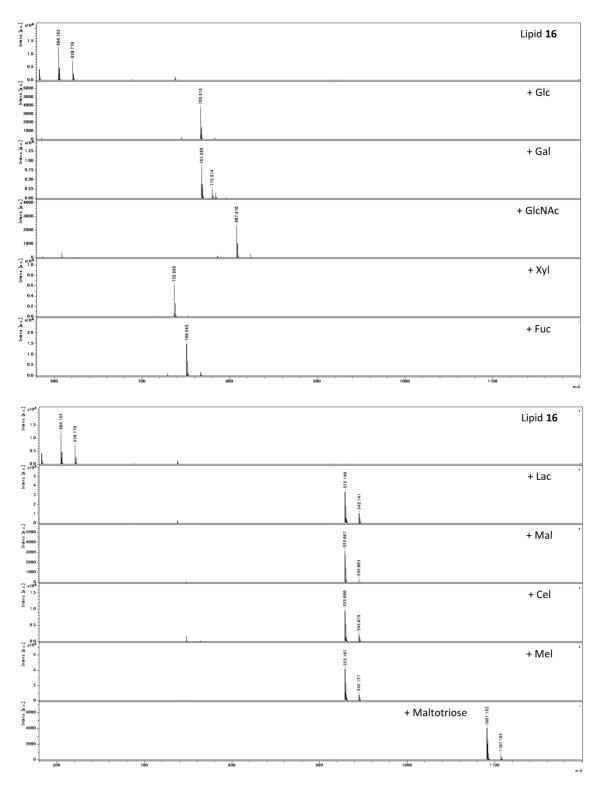


Fig. 3-2-12. MALDI-TOF MS spectra of conjugated products using compound 16

(DHB, positive)

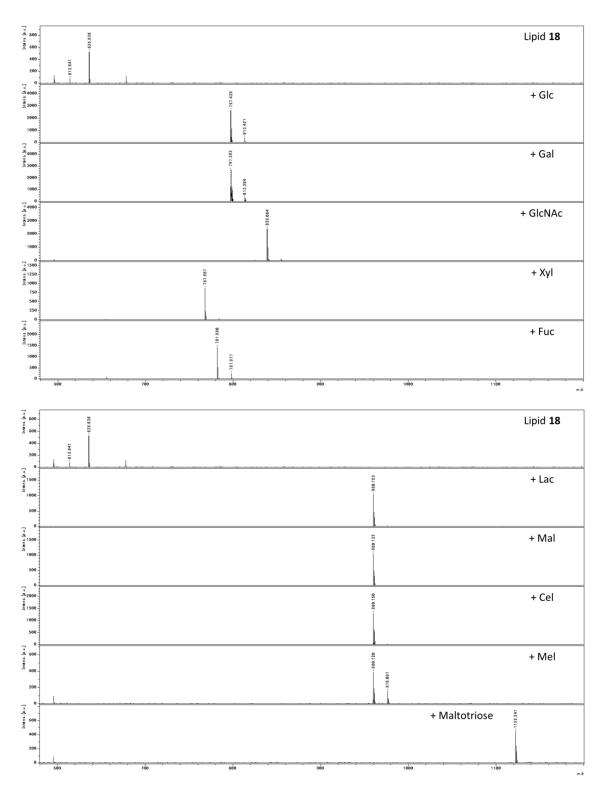


Fig. 3-2-13. MALDI-TOF MS spectra of conjugated products using compound 18

(DHB, positive)

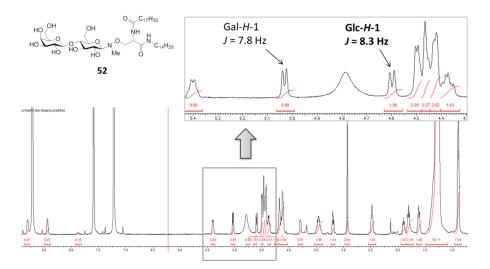


Fig. 3-2-14 ¹H NMR spectrum of β -product 52 (d_5 -pyridine, 300 K)

3-3 Conclusion

In this chapter, I employed glycoblotting reaction to construct neoglycolipid library. In case of methoxyamino derivative **1**, unfortunate side reaction was occurred but optimization of reaction condition enabled to obtain β -product as sole product and construct library. Hydrazide derivative also caused undesired side reaction by self-lactonization. This problem was also overcome by altering condition. Other ceramide derivatives were able to be conjugated with free sugars without any troubles. In this way, neoglycolipid library was successfully constructed by glycoblotting reaction. These results verified the usability of glycoblotting-based method.

3-4 Experimental Section

General Information

All reactions were carried out under a nitrogen atmosphere with dry solvents under anhydrous conditions, unless otherwise noted. Proton and carbon NMR was recorded at 298K with Varian UnityInova 500 MHz (Agilent Inc., USA; ¹H: 500 MHz, ¹³C: 125 MHz) or Bruker AVANCE DRX 600, equipped with a cryoprobe (Bruker BioSpin Co., Germany; ¹H: 600 MHz, ¹³C: 150 MHz). Chemical shifts are given in ppm and referenced to internal TMS ($\delta_{\rm H}$ 0.00 in CDCl₃), CHCl₃ ($\delta_{\rm H}$ 7.26 in CDCl₃), pyridine-*m*-H $(\delta_{\rm H} 7.22 \text{ in } d_5$ -Pyridine), d_5 -Pyridine ($\delta_{\rm C} 123.87$) or CDCl₃ ($\delta_{\rm C} 77.00$). Assignments in ¹H NMR were made by first-order analysis of the spectra by using ACD/NMR processor software (Advanced Chemistry Development, inc.) and were verified by H-H COSY and HSQC experiments. High/low resolution electrospray ionization mass spectra (ESI-MS) were recorded by JMS-700TZ (JEOL, Japan). TLC was performed on Merck pre-coated plates (20 cm \times 20 cm; layer thickness, 0.25 mm; Silica Gel 60F₂₅₄); of spots were visualized by spraying a solution 90:5:5 (v/v/v)MeOH-p-anisaldehyde-concentrated sulfuric acid and heating at 250 °C for ca. 1/2 min, a solution of 95: 5 (v/v) MeOH-concentrated sulfuric acid and heating at 180°C for ca. 1/2 min, and by UV light (256 or 365 nm) when applicable. Column chromatography was performed on Silica Gel N60 (spherical type, particle size 40-50 µm; Kanto Chemical Industry) with the solvent systems specified, and the ratio of solvent systems was given in v/v. The reaction progress of enzymatic hydrolysis was measured by Park and Johnson method using a microplate reader (SpectraMaxM5, Molecular Devices Co.,

Sunnyvale, CA). The following abbreviations were used to explain the multiplicities: s = singlet, d = doublet, dd = double doublet, t = triplet, m = multiplet, br = broad. In addition to those specified above, the following abbreviations, designations and formulas are used throughout the Supporting Information: MeOH = methanol, H_2O = water, EtOAc = ethyl acetate, DCM = dichloromethane, DMF = dimethylformamide, CHCl₃ = Chloroform, Et₃N = triethylamine, NaHCO₃ = sodium bicarbonate, MgSO₄ = magnesium sulfate, aq. = aqueous, sat. = saturated, 1N HCl = 1 normal hydrogen chloride solution

Materials

Solvents and other reagents for the chemical syntheses were purchased from Sigma-Aldrich Co., Tokyo Chemical Industry Co., Ltd., and Wako Pure Chemical Industries, Ltd. (Osaka, Japan) and used without further purification.

Glycoblotting reaction of ceramide mimic 1 with lactose.

To demonstrate the feasibility of the glycoblotting reaction of compound **1** with reducing sugars, some conditions were tested to synthesize *N*-LacCer **41** as indicated in Table 3-4-1.

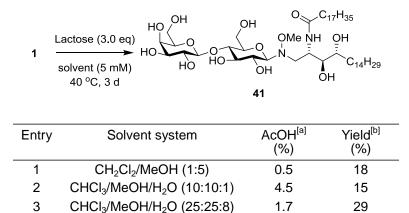


Table 3-4-1. Glycoblotting reaction using ceramide derivative 1 with lactose

[a] AcOH percentage in total volume

[b] Yields of isolated products after silica gel chromatography

(2*S*, 3*S*, 4*R*)-1-{*N*-[β-galactosylpyranosyl-(1→4)-β-glucopyranosyl] -*N*-(methoxy)amino}-2-[*N*-(octadecanoyl)amino]octadecane-3,4-diol (41).

A suspension of compound **1** (61 mg, 100 µmol) and lactose (103 mg, 300 µmol) in various solvent was stirred at 40 °C for 3 days (Table 1). Then, silica gel was added to a reaction solution and the mixture was evapolated completely. Purification of the crude product by flash column chromatography on silica gel (CHCl₃-*i*PrOH, 10:1, CHCl₃-MeOH, 4:1) yielded **14** as a white solid; ¹H NMR (600 MHz, d_5 -Pyridine) δ 8.56 (d, J = 7.45 Hz, 1H; NH), 5.24 (m, 1H; PS-H-2), 5.03 (d, J = 7.89 Hz, 1H; Gal-H-1), 4.86 (d, J = 8.77 Hz, 1H; Glc-H-1), 4.49 (m, 4H; Glc-H-6a, Gal-H-2, Gal-H-4,

Gal-*H*-6b), 4.40 (m, 3H; PS-*H*-3, Glc-*H*-6b, Gal-*H*-6b), 4.35 (dd, J = 7.45, 3.07 Hz, 1H; PS-*H*-3), 4.25 (m, 2H; Glc-*H*-2, Glc-*H*-3), 4.15 (m, 4H; PS-*H*-1a, PS-*H*-4, Gal-*H*-3, Gal-*H*-5), 4.10 (t, J = 9.21 Hz, 1H; Glc-*H*-4), 3.89 (dd, J = 14.03, 9.65 Hz, 1H; PS-*H*-1b), 3.84 (m, 4H; Glc-*H*-5, OMe), 2.49 (m, 2H; NHCOCH₂a,b), 2.30 (m, 1H; PS-*H*-5a), 1.96-1.77 (m, 4H; PS-*H*-5b, PS-*H*-6a, NHCOCH₂CH₂a,b), 1.66 (m, 1H; PS-*H*-6b), 1.46-1.17 (m, 50H; CH₂), 0.87 (t, J = 7.02 Hz, 6H; CH₃); ¹³C NMR (150 MHz (HSQC), d_5 -Pyridine): δ 105.74 (Gal-*C*-1), 93.75 (Glc-*C*-1), 82.10 (Glc-*C*-4), 78.46 (Glc-*C*-5), 77.70 (Glc-*C*-3), 77.51 (PS-*C*-3), 77.37 (Gal-*C*-5), 75.14 (Gal-*C*-3), 72.68 (PS-*C*-4), 72.45 (Gal-*C*-2), 71.45 (Glc-*C*-2), 70.03 (Gal-*C*-4), 62.10 (Glc-*C*-6, Gal-*C*-6), 61.50 (OMe), 52.06 (PS-*C*-1), 50.96 (PS-*C*-2), 36.94 (NHCOCH₂), 34.81 (PS-*C*-5), 32.21-22.65 (CH₂), 14.85 (CH₃); HRMS (ESI) Calcd. for C₄₉H₉₆N₂O₁₄Na [M+Na]⁺ 959.67592, found 959.67192

General procedure of glycoblotting reaction

100 mmol (about 60 mg) of ceramide derivative and 5 eq. of sugar were suspended in solvent and the glass bottle was capped tightly. The reaction mixture was stirred vigorously at prescribed temperature. After several hours, the solution was transferred into flask. Solvent was removed to about half volume and silica gel was added to a reaction solution and the mixture was evapolated completely. Purification of the crude product by flash column chromatography on silica gel (CHCl₃-*i*PrOH, 40:1, CHCl₃-MeOH, 3:1, depending on the compound) afforded product as a white solid.

Compound data of conjugated products using compound 1

Acetylated compound 42

¹H NMR (500 MHz, CHLOROFORM-*d*) δ ppm 6.01 (d, *J*=7.83 Hz, 1 H) 5.41 (dd, *J*=3.35, 0.84 Hz, 1 H) 5.09 - 5.14 (m, 2 H) 5.03 - 5.08 (m, 1 H) 4.91 (t, *J*=9.78 Hz, 1 H) 4.46 (d, *J*=9.78 Hz, 1 H) 4.39 - 4.44 (m, 1 H) 4.09 - 4.19 (m, 2 H) 3.93 - 3.98 (m, 1 H) 3.38 (s, 3 H) 3.27 (dd, *J*=13.98, 3.07 Hz, 1 H) 2.73 (dd, *J*=13.98, 11.74 Hz, 1 H) 2.15 (s, 3 H) 2.09 (s, 6 H) 2.04 (s, 3 H) 2.01 (s, 3 H) 1.54 - 1.70 (m, 3 H) 1.20 - 2.23 (m, 58 H) 0.88 (t, *J*=6.99 Hz, 6 H)

Cyclized-sphingosine 43

¹H NMR (600 MHz, DMSO-*d*₆) δ ppm 7.69 (d, *J*=9.21 Hz, 1 H; N*H*) 5.52 (d, *J*=9.87 Hz, 1 H; Gal-*H*-3) 5.39 (d, *J*=8.55 Hz, 1 H; Gal-*H*-2) 5.15 (d, *J*=9.87 Hz, 1 H; Gal-*H*-4) 5.01 (br. s, 1 H; Gal-*H*-5) 4.78 (d, *J*=10.52 Hz, 1 H; PS-*H*-4) 4.19 (dd, *J*=11.84, 4.60 Hz, 1 H; Gal-*H*-6a) 4.10 (d, *J*=8.55 Hz, 1 H; Gal-*H*-1) 3.99 - 4.07 (m, 1 H; PS-*H*-2) 3.87 (dd, *J*=11.84, 6.58 Hz, 1 H; Gal-*H*-6b) 3.49 (s, 3 H; O*Me*) 3.42 (d, *J*=10.52 Hz, 1 H; PS-*H*-3) 3.24 - 3.30 (m, 1 H; PS-*H*-1) 2.63 (t, *J*=12.50 Hz, 1 H; PS-*H*-1) 2.10 – 1.20 (m, 58 H) 2.05 - 1.94 (m, 18 H; Ac) 0.85 (t, *J*=6.58 Hz, 6 H)

¹³C NMR (150 MHz, Pyr, HSQC) δ ppm 89.29 (Gal-C-1) 81.57 (PS-C-3) 72.77 (PS-C-4) 68.81 (Gal-C-2) 68.27 (Gal-C-4) 68.07 (Gal-C-5) 67.46 (Gal-C-3) 62.65 (Gal-C-6) 59.86 (OMe) 53.74 (PS-C-1) 39.07 (PS-C-2)

Glc (yield: 54%)

¹H NMR (600 MHz, Pyr) δ ppm 8.60 (d, *J*=7.89 Hz, 1 H; N*H*) 5.24 - 5.30 (m, 1 H; PS-*H*-2) 4.94 (d, *J*=8.33 Hz, 1 H; Glc-*H*-1) 4.54 (dd, *J*=11.40, 2.19 Hz, 1 H; Glc-*H*-6a) 4.35 (dd, *J*=7.45, 3.07 Hz, 1 H; PS-*H*-3) 4.21 - 4.30 (m, 4 H; Glc-*H*-2, Glc-*H*-3,

Glc-*H*-6b, PS-*H*-1a) 4.14 - 4.19 (m, 1 H; PS-*H*-4) 4.06 - 4.10 (m, 1 H; Glc-*H*-4) 3.94 - 3.98 (m, 1 H; Glc-*H*-5) 3.91 (dd, *J*=13.59, 9.65 Hz, 1 H; PS-*H*-1b) 3.86 (s, 3 H; OMe) 2.55 - 1.12 (m, 58 H) 0.87 (t, *J*=7.02 Hz, 6 H; CH₃)

¹³C NMR (150 MHz, Pyr, HSQC) δ ppm 93.69 (Glc-*C*-1) 80.54 (Glc-*C*-5) 79.85 (Glc-*C*-3) 77.49 (PS-*C*-3) 72.48 (PS-*C*-4) 71.92 (Glc-*C*-2) 71.78 (Glc-*C*-4) 62.88 (Glc-*C*-6) 61.35 (OMe) 52.04 (PS-*C*-1) 50.93 (PS-*C*-2) 37.12 - 14.29 (Alkylchain)

Gal (yield: 41%)

¹H NMR (500 MHz, pyridine) δ ppm 8.50 (d, *J*=8.56 Hz, 1 H; N*H*) 5.27 (m, 1H; PS-*H*-2) 4.86 (d, *J*=9.22 Hz, 1 H; Gal-*H*-1) 4.60 (t, *J*=8.89 Hz, 1 H; Gal-*H*-2) 4.51 (d, *J*=2.96 Hz, 1 H; Gal-*H*-4) 4.41 - 4.47 (m, 1 H; Gal-*H*-6a) 4.34 - 4.39 (m, 1 H; Gal-*H*-6b) 4.29 - 4.34 (m, 1 H; PS-*H*-3) 4.25 (dd, *J*=13.83, 3.29 Hz, 1 H; PS-*H*-1a) 4.13 - 4.19 (m, 2 H; Gal-*H*-3, ; PS-*H*-4) 4.06 (t, *J*=6.26 Hz, 1 H; Gal-*H*-5) 3.93 (dd, *J*=13.50, 9.88 Hz, 1 H; PS-*H*-1b) 3.83 (s, 3 H; OMe) 2.5 - 1.10 (m, 58 H; CH₂) 0.87 (t, *J*=6.59 Hz, 6 H; CH₃)

GlcNAc (yield: 24%)

¹H NMR (600 MHz, Pyr) δ ppm 9.32 (br. s, 1 H; GlcNAc-NH) 8.24 (br. s, 1 H; PS-NH) 5.22 - 5.29 (m, 1 H; PS-H-2) 4.93 (d, *J*=10.08 Hz, 1 H; GlcNAc-H-1) 4.57 - 4.64 (m, 1 H; GlcNAc-H-2) 4.50 - 4.55 (m, 1 H; GlcNAc-H-6a) 4.20 - 4.30 (m, 3 H; GlcNAc-H-3, GlcNAc-H-6b; PS-H-3) 4.02 - 4.15 (m, 3 H; PS-H-1ab, PS-H-4) 3.99 (t, *J*=9.65 Hz, 1 H; GlcNAc-H-4) 3.88 - 3.93 (m, 1 H; GlcNAc-H-5) 3.62 (s, 3H; OMe) 2.59 - 1.12 (m, 58 H) 0.88 (t, *J*=6.80 Hz, 6 H) ¹³C NMR (150 MHz, Pyr, HSQC) δ ppm 94.43 (GlcNAc-*C*-1) 3.89 80.13

(GlcNAc-C-5) 77.55 (PS-C-3) 77.32 (GlcNAc-C-3) 72.40 (PS-C-4) 72.16 (GlcNAc-C-4) 62.79 (GlcNAc-C-6) 60.91 (OMe) 53.88 (GlcNAc-C-2) 51.07 (PS-C-2) 49.30 (PS-C-1) 36.54-14.27 (Alkylchain) 23.28 (Ac)

Xyl (yield: 76%)

¹H NMR (600 MHz, Pyr) δ ppm 8.54 (d, *J*=7.89 Hz, 1 H; N*H*) 5.26 - 5.32 (m, 1 H; PS-*H*-2) 4.79 (d, *J*=8.33 Hz, 1 H; Xyl-*H*-1) 4.38 (dd, *J*=7.45, 3.51 Hz, 1 H; PS-*H*-3) 4.31 (dd, *J*=11.40, 5.26 Hz, 1 H; Xyl-*H*-5a) 4.16 - 4.25 (m, 2 H; PS-*H*-1a, PS-*H*-4) 4.06 - 4.16 (m, 3 H; ; Xyl-*H*-2, Xyl-*H*-3, Xyl-*H*-4) 3.84 - 3.91 (m, 4 H; PS-*H*-1b, OMe) 3.64 (t, *J*=10.96 Hz, 1 H; Xyl-*H*-5b) 2.06 - 1.20 (m, 58 H) 0.88 (t, *J*=6.80 Hz, 6 H) ¹³C NMR (150 MHz, Pyr, HSQC) δ ppm 94.90 (Xyl-C-1) 79.66 (Xyl-C-3) 77.32 (PS-C-3) 72.63 (PS-C-4) 71.69 (Xyl-C-2) 70.99 (Xyl-C-4) 69.11 (Xyl-C-5) 61.38 (OMe) 52.01 (PS-C-1) 51.07 (PS-C-2) 37.01 - 14.27 (Alkylchain)

Fuc (yield: 50%)

¹H NMR (600 MHz, Pyr) δ ppm 8.67 (d, *J*=8.33 Hz, 1 H; N*H*) 5.21 - 5.28 (m, 1 H; PS-*H*-2) 4.62 (t, *J*=9.21 Hz, 1 H; Fuc-*H*-2) 4.58 (d, *J*=8.77 Hz, 1 H; Fuc-*H*-1) 4.39 (dd, *J*=13.59, 4.82 Hz, 1 H; PS-*H*-1a) 4.31 - 4.36 (m, 1 H; PS-*H*-3) 4.26 - 4.31 (m, 1 H; PS-*H*-4) 4.08 (dd, *J*=8.77, 3.07 Hz, 1 H; Fuc-*H*-3) 4.03 (d, *J*=3.07 Hz, 1 H; Fuc-*H*-4) 3.87 (s, 3 H; OMe) 3.70 - 3.79 (m, 2 H; Fuc-*H*-5, PS-*H*-1b) 2.51 – 1.15 (m, 58 H) 1.49 (d, *J*=6.14 Hz, 3 H; Fuc-*H*-6) 0.88 (t, *J*=6.80 Hz, 6 H; CH₃)

¹³C NMR (150 MHz, Pyr, HSQC) δ ppm 95.18 (Fuc-*C*-1) 77.65 (PS-*C*-3) 76.58 (Fuc-*C*-3) 73.18 (Fuc-*C*-5) 72.86 (Fuc-*C*-4) 72.44 (PS-*C*-4) 68.83 (Fuc-*C*-2) 61.50 (OMe) 53.74 (PS-*C*-1) 50.87 (PS-*C*-2) 37.26 - 14.33 (Alkylchain) 17.77 (Fuc-*C*-6)

Cellobi (yield: 40%)

¹H NMR (600 MHz, Pyr) δ ppm 8.62 (d, *J*=7.67 Hz, 1 H; N*H*) 5.22 - 5.27 (m, 1 H; PS-*H*-2) 5.12 (d, *J*=7.89 Hz, 1 H; Glc-B-*H*-1) 4.88 (d, *J*=8.77 Hz, 1 H; Glc-A-*H*-1) 4.54 - 4.59 (m, 1 H; Glc-B-*H*-6a) 4.45 - 4.50 (m, 1 H; Glc-A-*H*-6a) 4.39 - 4.44 (m, 1 H; Glc-A-*H*-6b) 4.34 (dd, *J*=7.24, 3.51 Hz, 1 H; PS-*H*-3) 4.30 (dd, *J*=11.62, 6.36 Hz, 1 H; Glc-B-*H*-6b) 4.11 - 4.26 (m, 7 H; Glc-A-*H*-2, Glc-A-*H*-3, Glc-A-*H*-4, Glc-B-*H*-3, Glc-B-*H*-4, PS-*H*-1a, PS-*H*-4) 4.06 - 4.10 (m, 1 H; Glc-B-*H*-2) 4.01 - 4.06 (m, 1 H; Glc-B-*H*-5) 3.87 - 3.93 (m, 2 H; Glc-A-*H*-5, PS-*H*-1b) 3.85 (s, 3 H; OMe) 2.55-1.15 (m, 58 H; CH₂) 0.87 (t, *J*=6.80 Hz, 6 H; CH₃)

¹³C NMR (150 MHz, Pyr, HSQC) δ ppm 105.00 (Glc-B-C-1) 93.79 (Glc-A-C-1) 81.40 (Glc-A-C-4) 78.52 (Glc-A-C-5, Glc-B-C-5) 78.22 (Glc-B-C-3) 77.89 (Glc-A-C-3) 77.50 (PS-C-3) 74.76 (Glc-B-C-2) 72.58 (PS-C-4) 71.84 (Glc-B-C-4) 71.47 (Glc-A-C-2) 62.56 (Glc-B-C-6) 62.05 (Glc-A-C-6) 61.37 (OMe) 52.04 (PS-C-1) 51.02 (PS-C-2) 37.32 - 14.53 (Alkyl chain)

Malto (yield: 30%)

¹H NMR (600 MHz, Pyr) δ ppm 8.57 (br. s, 1 H; N*H*) 5.84 (s, 1 H; Glc-B-*H*-1) 5.24 (br. s, 1 H; PS-*H*-2) 4.85 (d, *J*=8.77 Hz, 1 H; Glc-A-*H*-1) 4.44 - 4.61 (m, 4 H; Glc-A-*H*-6a, Glc-B-*H*-3, Glc-B-*H*-4, Glc-B-*H*-6a) 4.27 - 4.40 (m, 4 H; Glc-A-*H*-3, Glc-A-*H*-6b, Glc-B-*H*-6b, PS-*H*-3) 4.09 - 4.26 (m, 6 H; Glc-A-*H*-2, Glc-A-*H*-4, Glc-B-*H*-2,

Glc-B-*H*-5, PS-*H*-1a, PS-*H*-4) 3.79 - 3.91 (m, 4 H; PS-*H*-1b, OMe) 3.75 (br. s, 1 H Glc-A-*H*-5) 2.54-1.07 (m, 56 H; CH₂) 0.82 - 0.93 (m, 6 H; CH₃)

¹³C NMR (150 MHz, Pyr, HSQC) δ ppm 103.18 (Glc-B-*C*-1) 93.65 (Glc-A-*C*-1) 81.63 (Glc-A-*C*-4) 79.18 (Glc-A-*C*-3) 78.69 (Glc-A-*C*-5) 77.55 (PS-*C*-3) 75.38 (Glc-B-*C*-3) 75.22 (Glc-B-*C*-4) 74.39 (Glc-B-*C*-2) 72.56 (PS-*C*-4) 71.70 (Glc-B-*C*-5) 71.22 (Glc-A-*C*-2) 62.68 (Glc-B-*C*-6) 62.09 (Glc-A-*C*-6) 61.44 (OMe) 52.11 (PS-*C*-1) 50.87 (PS-*C*-2) 36.82-14.35 (Alkyl chain)

Melibiose (yield: 28%)

¹H NMR (500 MHz, pyridine) δ ppm 5.42 - 5.48 (d, *J*=3.13 Hz, 1 H; Gal-*H*-1) 5.17 - 5.28 (m, 1 H; PS-*H*-2) 4.83 (d, *J*=8.77 Hz, 1 H; Glc-*H*-1) 4.48 - 4.71 (m, 4 H; Gal-*H*-2, Gal-*H*-3, Gal-*H*-4, Gal-*H*-5) 4.33 - 4.47 (m, 4 H; Glc-*H*-6a, Gal-*H*-6, PS-*H*-3) 4.13 - 4.33 (m, 6 H; Glc-*H*-2, Glc-*H*-3, Glc-*H*-4, Glc-*H*-6b, PS-*H*-1a, PS-*H*-4) 3.90 - 4.07 (m, 2 H; Glc-*H*-5, PS-*H*-1b) 3.84 (s, 3 H; OMe) 2.60-1.04 (m, 58 H; CH₂) 0.81 - 0.94 (m, 6 H; CH₃)

F1 (ppm) 100.52 (Gal-C-1) 94.23 (Glc-C-1) 79.65 (Glc-C-3) 77.81 (Glc-C-5) 77.42 (PS-C-3) 72.78 (Glc-C-4) 72.73 (PS-C-4) 72.50 (Gal-C-5) 71.50 (Glc-C-2) 71.35 (Gal-C-3) 70.91 (Gal-C-4) 70.60 (Gal-C-2) 68.47 (Glc-C-6) 62.60 (Gal-C-6) 61.67 (OMe) 52.67 (PS-C-1) 51.00 (PS-C-2) 36.91-14.34 (Alkyl chain)

Maltotri (yield: 30%)

¹H NMR (600 MHz, Pyr) δ ppm 8.61 (d, *J*=7.89 Hz, 1 H; N*H*) 5.93 (d, *J*=3.95 Hz, 1 H; Glc-B-*H*-1) 5.73 (d, *J*=3.51 Hz, 1 H; Glc-C-*H*-1) 5.24 - 5.29 (m, 1 H; PS-*H*-2) 4.92 (d, *J*=8.77 Hz, 1 H; Glc-A-*H*-1) 4.65 (t, *J*=9.21 Hz, 1 H; Glc-B-*H*-3) 4.61 (t, *J*=9.21 Hz, 1 H; Glc-C-*H*-3) 4.52 - 4.58 (m, 2 H; Glc-C-*H*-4, Glc-C-*H*-6a) 4.39 - 4.47 (m, 2 H; Glc-B-*H*-6) 4.29 - 4.39 (m, 5 H; Glc-A-*H*-3, Glc-A-*H*-6a, Glc-B-*H*-5, Glc-C-*H*-6b, PS-*H*-3) 4.22 - 4.29 (m, 3 H; Glc-A-*H*-2, Glc-A-*H*-6b, Glc-B-*H*-4) 4.13 - 4.22 (m, 4 H; Glc-C-*H*-2, Glc-C-*H*-5, PS-*H*-1a, PS-*H*-4) 4.07 - 4.13 (m, 2 H; Glc-A-*H*-4, Glc-B-*H*-2) 3.85 - 3.91 (m, 1 H; PS-*H*-1b) 3.79 - 3.85 (m, 4 H; Glc-A-*H*-5, OMe) 2.54-1.19 (m, 58 H; CH₂) 0.87 (t, *J*=6.80 Hz, 6 H; CH₃)

¹³C NMR (150 MHz, Pyr, HSQC) δ ppm 103.04 (Glc-C-C-1) 103.00 (Glc-B-C-1) 93.48
(Glc-A-C-1) 82.14 (Glc-A-C-4) 81.35 (Glc-B-C-4) 79.10 (Glc-A-C-3) 78.68
(Glc-A-C-5) 77.55 (PS-C-3) 75.43 (Glc-C-C-3) 75.22 (Glc-C-C-4) 74.80 (Glc-B-C-3)
74.51 (Glc-C-C-2) 73.88 (Glc-B-C-2) 73.38 (Glc-B-C-5) 72.51 (PS-C-4) 71.80
(Glc-C-C-5) 71.17 (Glc-A-C-2) 62.74 (Glc-C-C-6) 61.99 (Glc-A-C-6) 61.90
(Glc-B-C-6) 61.40 (OMe) 52.14 (PS-C-1) 50.81 (PS-C-2) 36.93 - 14.26 (Alkyl chain)

Compound data of conjugated products using compound 16

Glc (yield: 68%)

¹H NMR (500 MHz, pyridine) δ ppm 8.85 (d, *J*=8.58 Hz, 1 H; N*H*) 5.26 (br. s, 1 H; PS-*H*-2) 4.69 - 4.78 (m, 1 H; PS-*H*-1a) 4.71 (d, *J*=8.91 Hz, 1 H; Glc-*H*-1) 4.61 (dd, *J*=10.89, 7.26 Hz, 1 H; PS-*H*-1b) 4.51 (d, *J*=10.56 Hz, 1 H; Glc-*H*-6a) 4.29 - 4.39 (m, 2 H; Glc-*H*-6b, PS-*H*-3) 4.10 - 4.24 (m, 4 H; Glc-*H*-2, Glc-*H*-3, Glc-*H*-4, PS-*H*-4) 3.88 (br. s, 1 H; Glc-*H*-5) 2.96 (s, 3 H; *Me*NO) 2.55-1.07 (m, 58 H; *C*H₂) 0.87 (t, *J*=6.43 Hz, 6 H; *C*H₃)

F1 (ppm) 95.17 (Glc-C-1) 80.07 (Glc-C-5) 79.64 (Glc-C-3) 76.50 (PS-C-3) 72.67 (PS-C-4) 72.13 (PS-C-1) 71.78 (Glc-C-2) 71.60 (Glc-C-4) 62.81 (Glc-C-6) 51.82 (PS-C-2) 38.10 (*Me*NO) 36.82-14.11 (Alkyl chain)

Gal (yield: 22%)

¹H NMR (500 MHz, pyridine) δ ppm 8.81 (d, *J*=8.55 Hz, 1 H; N*H*) 5.25 (m, 1 H; PS-*H*-2) 4.69 - 4.74 (m, 2 H; Gal-*H*-1, PS-*H*-1a) 4.59 - 4.66 (m, 1 H; PS-*H*-1b) 4.56 (d, *J*=2.63 Hz, 1 H; Gal-*H*-4) 4.46 - 4.52 (m, 1 H; Gal-*H*-2) 4.40 - 4.44 (m, 1 H; Gal-*H*-6) 4.30 - 4.39 (m, 1 H; PS-*H*-3) 4.18 - 4.25 (m, 1 H; PS-*H*-4) 4.14 (dd, *J*=9.21, 3.29 Hz, 1 H; Gal-*H*-3) 4.02 (t, *J*=5.92 Hz, 1 H; Gal-*H*-5) 2.99 (s, 3 H; *Me*NO) 2.57-1.16 (m, 58 H; *C*H₂) 0.87 (t, *J*=6.74 Hz, 6 H; *C*H₃)

F1 (ppm) 95.52 (Gal-C-1) 78.54 (Gal-C-5) 76.45 (Gal-C-3) 76.32 (PS-C-3) 72.85 (PS-C-4) 72.10 (PS-C-1) 70.05 (Gal-C-4) 69.39 (Gal-C-2) 62.23 (Gal-C-6) 51.77 (PS-C-2) 37.85 (*Me*NO) 36.87-14.24 (Alkyl chain)

GlcNAc (yield: 37%)

¹H NMR (500 MHz, pyridine) δ ppm 9.02 (br. s, 1 H; GlcNAc-N*H*) 8.86 (br. s, 1 H; PS-N*H*) 5.15 (m, 1 H; PS-*H*-2) 4.82 - 4.99 (m, 1 H; GlcNAc-*H*-1) 4.43 - 4.75 (m, 4 H; GlcNAc-*H*-2, GlcNAc-*H*-6a, PS-*H*-1) 4.25 - 4.42 (m, 3 H; GlcNAc-*H*-3, GlcNAc-*H*-6b, PS-*H*-3) 4.04 - 4.24 (m, 2 H; GlcNAc-*H*-4, PS-*H*-4) 3.85 (br. s, 1 H; GlcNAc-*H*-5) 3.00 (s, 3 H; *Me*NO) 2.59-1.02 (m, 58 H; *C*H₂) 2.14 (s, 3 H; *Ac*) 0.76 - 0.97 (m, 6 H; *C*H₃) F1 (ppm) 93.55 (GlcNAc-*C*-1) 80.03 (GlcNAc-*C*-5) 77.47 (PS-*C*-3) 76.65 (GlcNAc-*C*-3) 72.63 (PS-*C*-4) 72.47 (GlcNAc-*C*-4) 71.72 (PS-*C*-1) 62.76

(GlcNAc-C-6) 54.60 (GlcNAc-C-2) 51.86 (PS-C-2) 38.26 (*Me*NO) 36.84-13.97 (Alkyl chain) 23.49 (Ac)

Xyl (yield: 55%)

¹H NMR (500 MHz, pyridine) δ ppm 8.87 (d, *J*=8.59 Hz, 1 H; N*H*) 5.22 - 5.28 (m, 1 H PS-*H*-2) 4.71 - 4.78 (m, 1 H; PS-*H*-1a) 4.62 (d, *J*=8.92 Hz, 1 H; Xyl-*H*-1) 4.58 - 4.65 (m, 1 H; PS-*H*-1b) 4.30 - 4.39 (m, 2 H; Xyl-*H*-5a, PS-*H*-3) 4.18 - 4.26 (m, 1 H; PS-*H*-4) 4.05 - 4.17 (m, 3 H; Xyl-*H*-2, Xyl-*H*-3, Xyl-*H*-4) 3.61 (t, *J*=10.24 Hz, 1 H; Xyl-*H*-5b) 2.96 (s, 3 H; *Me*NO) 2.55-1.16 (m, 58 H) 0.87 (t, *J*=6.60 Hz, 6 H; *C*H₃)

F1 (ppm) 95.76 (Xyl-C-1) 79.77 (Xyl-C-3) 76.34 (PS-C-3) 72.58 (PS-C-4) 72.03 (PS-C-1) 71.65 (Xyl-C-2) 70.88 (Xyl-C-4) 68.71 (Xyl-C-5) 51.68 (PS-C-2) 37.93 (*Me*NO) 36.84-14.06 (Alkyl chain)

Fuc (yield: 49%)

¹H NMR (500 MHz, pyridine) δ ppm 8.77 (d, *J*=8.43 Hz, 1 H; N*H*) 5.06 (m, 1 H; PS-*H*-2) 4.67 (d, *J*=9.16 Hz, 1 H; Fuc-*H*-1) 4.56 - 4.65 (m, 2 H; PS-*H*-1) 4.34 - 4.44 (m, 2 H; Fuc-*H*-2, PS-*H*-3) 4.18 - 4.25 (m, 1 H; PS-*H*-4) 4.06 - 4.11 (m, 1 H; Fuc-*H*-3) 4.01 - 4.05 (m, 1 H; Fuc-*H*-4) 3.74 - 3.80 (m, 1 H; Fuc-*H*-5) 2.95 (s, 3 H; *Me*NO) 2.53-1.17 (m, 58 H; *C*H₂) 1.52 (d, *J*=6.23 Hz, 3 H; Fuc-*H*-6) 0.87 (t, *J*=6.78 Hz, 6 H; *C*H₃) F1 (ppm) 95.74 (Fuc-*C*-1) 76.40 (Fuc-*C*-3) 76.35 (PS-*C*-3) 73.01 (Fuc-*C*-5) 72.83 (PS-*C*-4) 72.75 (Fuc-*C*-4) 72.12 (PS-*C*-1) 69.23 (Fuc-*C*-2) 52.01 (PS-*C*-2) 37.76 (*Me*NO) 36.78-14.32 (Alkyl chain) 17.35 (Fuc-*C*-6);

Lac (yield: 40%)

¹H NMR (500 MHz, pyridine) δ ppm 8.83 (d, *J*=8.30 Hz, 1 H; N*H*) 5.21 - 5.28 (m, 1 H; PS-*H*-2) 5.08 (d, *J*=7.82 Hz, 1 H; Gal-*H*-1) 4.72 - 4.78 (m, 1 H; PS-*H*-1a) 4.60 - 4.69 (m, 2 H; Glc-*H*-1, PS-*H*-1b) 4.47 (m, 6 H; Glc-*H*-6, Gal-*H*-2, Gal-*H*-4, Gal-*H*-6) 4.32 - 4.39 (m, 1 H; PS-*H*-3) 4.13 - 4.27 (m, 6 H; Glc-*H*-2, Glc-*H*-3, Glc-*H*-4, Gal-*H*-3, Gal-*H*-5, PS-*H*-4) 3.77 - 3.85 (m, 1 H; Glc-*H*-5) 2.95 (s, 3 H; *Me*NO) 2.53-1.16 (m, 58 H; *CH*₂) 0.88 (t, *J*=6.47 Hz, 6 H; *CH*₃)

F1 (ppm) 105.85 (Gal-*C*-1) 95.21 (Glc-*C*-1) 81.99 (Glc-*C*-4) 78.18 (Glc-*C*-5) 77.56 (Glc-*C*-3) 77.46 (Gal-*C*-5) 76.56 (PS-*C*-3) 75.36 (Gal-*C*-3) 72.93 (PS-*C*-4) 72.41 (PS-*C*-1) 72.34 (Gal-*C*-2) 71.48 (Glc-*C*-2) 70.18 (Gal-*C*-4) 62.38 (Glc-*C*-6) 62.14 (Gal-*C*-6) 52.22 (PS-*C*-2) 38.19 (*Me*NO) 36.88-14.35 (Alkyl chain)

Mal (yield: 76%)

¹H NMR (500 MHz, pyridine) δ ppm 8.82 (d, *J*=8.79 Hz, 1 H; N*H*) 5.87 (d, *J*=3.66 Hz, 1 H; Glc-B-*H*-1) 5.20 - 5.25 (m, 1 H; PS-*H*-2) 4.72 (dd, *J*=10.99, 2.93 Hz, 1 H; PS-*H*-1a) 4.62 (d, *J*=8.79 Hz, 1 H; Glc-A-*H*-1) 4.50 - 4.66 (m, 4 H; Glc-B-*H*-3, Glc-B-*H*-4, Glc-B-*H*-6a, PS-*H*-1b) 4.37 - 4.46 (m, 2 H; Glc-A-*H*-6) 4.31 - 4.37 (m, 2 H; Glc-B-*H*-6b, PS-*H*-3) 4.20 - 4.30 (m, 2 H; Glc-A-*H*-3, Glc-A-*H*-4) 4.12 - 4.20 (m, 4 H; Glc-A-*H*-2, Glc-B-*H*-2, Glc-B-*H*-5, PS-*H*-4) 3.65 - 3.72 (m, 1 H; Glc-A-*H*-5) 2.92 (s, 3 H; *Me*NO) 2.54-1.17 (m, 58 H; C*H*₂) 0.87 (t, *J*=6.78 Hz, 6 H; C*H*₃)

F1 (ppm) 102.94 (Glc-B-C-1) 94.95 (Glc-A-C-1) 80.98 (Glc-A-C-4) 78.77 (Glc-A-C-3) 78.22 (Glc-A-C-5) 76.26 (PS-C-3) 75.11 (Glc-B-C-4) 75.06 (Glc-B-C-3) 74.29 (Glc-B-C-2) 72.10 (PS-C-4) 72.01 (PS-C-1) 71.77 (Glc-B-C-5) 71.44 (Glc-A-C-2)

91

62.52 (Glc-B-C-6) 61.76 (Glc-A-C-6) 51.76 (PS-C-2) 38.10 (*Me*NO) 36.66-14.09 (Alkyl chain)

Cel (yield: 28%)

¹H NMR (500 MHz, pyridine) δ ppm 8.82 (d, *J*=7.82 Hz, 1 H; N*H*) 5.20 - 5.30 (m, 1 H; PS-*H*-2) 5.17 (d, *J*=7.57 Hz, 1 H; Glc-B-*H*-1) 4.71 - 4.81 (m, 1 H; PS-*H*-1a) 4.65 (m, 2 H; Glc-A-*H*-1, PS-*H*-1b) 4.53 - 4.59 (m, 1 H; Glc-B-*H*-6a) 4.40 - 4.52 (m, 2 H; Glc-A-*H*-6) 4.29 - 4.39 (m, 2 H; Glc-B-*H*-6b, PS-*H*-3) 4.22 (m, 6 H; Glc-A-*H*-2, Glc-A-*H*-3, Glc-A-*H*-4, Glc-B-*H*-3, Glc-B-*H*-4, PS-*H*-4) 4.07 - 4.14 (m, 1 H; Glc-B-*H*-2) 4.00 - 4.07 (m, 1 H; Glc-B-*H*-5) 3.76 - 3.87 (m, 1 H; Glc-A-*H*-5) 2.94 (s, 3 H; *Me*NO) 2.55-1.08 (m, 58 H; C*H*₂) 0.88 (br. s., 6 H; C*H*₃)

F1 (ppm) 105.00 (Glc-B-C-1) 95.03 (Glc-A-C-1) 81.06 (Glc-A-C-4) 78.53 (Glc-B-C-5) 78.11 (Glc-A-C-3, Glc-A-C-5, Glc-B-C-3) 76.69 (PS-C-3) 74.87 (Glc-B-C-2) 72.69 (PS-C-4) 72.32 (PS-C-1) 71.54 (PS-C-4) 71.50 (Glc-A-C-2) 62.45 (Glc-B-C-6) 62.17 (Glc-A-C-6) 52.19 (PS-C-2) 38.23 (*Me*NO) 36.84-14.18 (Alkyl chain)

Melibi (yield: 68%)

¹H NMR (500 MHz, pyridine) δ ppm 8.69 (d, *J*=8.55 Hz, 1 H; N*H*) 5.50 (d, *J*=3.42 Hz, 1 H; Gal-*H*-1) 5.11 - 5.18 (m, 1 H; PS-*H*-2) 4.58 - 4.69 (m, 6 H; Glc-*H*-1, Gal-*H*-2, Gal-*H*-4, Gal-*H*-5, PS-*H*-1) 4.51 (dd, *J*=10.01, 2.69 Hz, 1 H; Gal-*H*-3) 4.38 - 4.49 (m, 3 H; Glc-*H*-6a, Gal-*H*-6) 4.29 - 4.38 (m, 2 H; Glc-*H*-6b, PS-*H*-3) 4.17 - 4.24 (m, 1 H; PS-*H*-4) 4.13 (m, 2 H; Glc-*H*-2, Glc-*H*-3) 4.04 - 4.10 (m, 1 H; Glc-*H*-4) 3.86 - 3.94 (m, 1 H; Glc-*H*-5) 2.95 (s, 3 H; *Me*NO) 2.54-1.18 (m, 58 H; C*H*₂) 0.88 (t, *J*=6.59 Hz, 6 H; C*H*₃) F1 (ppm) 100.96 (Gal-C-1) 95.19 (Glc-C-1) 79.68 (Glc-C-3) 77.94 (Glc-C-5) 76.03 (PS-C-3)72.80 (PS-C-4) 72.04 (PS-C-1) 71.95 (Glc-C-2) 71.69 (Glc-C-4) 71.59 (Gal-C-3) 71.28 (Gal-C-2, Gal-C-4) 71.15 (Gal-C-5) 68.58 (Glc-C-6) 62.58 (Gal-C-6) 51.68 (PS-C-2) 38.25 (*Me*NO) 36.81-14.19 (Alkyl chain)

Maltotriose (yield: 53%)

¹H NMR (500 MHz, pyridine) δ ppm 8.81 (d, *J*=8.30 Hz, 1 H; N*H*) 5.91 (d, *J*=3.18 Hz, 1 H; Glc-C-*H*-1) 5.78 (d, *J*=3.18 Hz, 1 H; Glc-B-*H*-1) 5.19 - 5.27 (m, 1 H; PS-*H*-2) 4.56 - 4.77 (m, 5 H; Glc-A-*H*-1, Glc-B-*H*-3, Glc-C-*H*-3, PS-*H*-1) 4.49 - 4.56 (m, 2 H; Glc-C-*H*-4, Glc-C-*H*-6a) 4.31 - 4.48 (m, 7 H; Glc-A-*H*-6, Glc-B-*H*-5, Glc-B-*H*-6, Glc-C-*H*-6b, PS-*H*-3) 4.23 - 4.31 (m, 2 H; Glc-A-*H*-3, Glc-B-*H*-4) 4.14 - 4.23 (m, 5 H; Glc-A-*H*-2, Glc-C-*H*-4, Glc-C-*H*-2, Glc-C-*H*-5, PS-*H*-4) 4.09 - 4.14 (m, 1 H; Glc-B-*H*-2) 3.71 - 3.80 (m, 1 H; Glc-A-*H*-5) 2.93 (s, 3 H; *Me*NO) 2.54-1.17 (m, 58 H; C*H*₂) 0.88 (t, *J*=6.59 Hz, 6 H; C*H*₃)

F1 (ppm) 103.20 (Glc-C-C-1) 103.03 (Glc-B-C-1) 95.28 81.51 (Glc-A-C-4) 81.33 (Glc-B-C-4) 79.13 (Glc-A-C-3) 78.53 (Glc-A-C-5) 76.50 (PS-C-3) 75.28 (Glc-C-C-3) 75.18 (Glc-C-C-4) 74.97 (Glc-B-C-3) 74.53 (Glc-C-C-2) 73.85 (Glc-B-C-2) 73.67 (Glc-B-C-5) 72.50 (PS-C-1) 72.22 (Glc-C-C-5) 72.10 (PS-C-4) 71.41 (Glc-A-C-2) 62.65 (Glc-C-C-6) 62.20 (Glc-A-C-6) 61.92 (Glc-B-C-6) 52.00 (PS-C-2) 38.23 (*Me*NO) 36.81-14.17 (Alkyl chain)

Compound data of conjugated products using compound 19

Glc (yield: 34%)

¹H NMR (500 MHz, pyridine) δ ppm 9.04 (d, *J*=8.06 Hz, 1 H; N*H*) 8.64 (t, *J*=5.62 Hz, 1 H; N*H*) 5.46 - 5.52 (m, 1 H; Ser-*H*-α) 4.72 (d, *J*=9.04 Hz, 1 H; Glc-*H*-1) 4.49 - 4.58 (m, 3 H; Glc-*H*-6a, Ser-*H*-β) 4.37 (dd, *J*=11.72, 5.37 Hz, 1 H; Glc-*H*-6b) 4.13 - 4.25 (m, 3 H; Glc-*H*-2, Glc-*H*-3, Glc-*H*-4) 3.88 - 3.95 (m, 1 H; Glc-*H*-5) 3.40 - 3.55 (m, 2 H; *CH*₂) 2.96 (s, 3 H; *Me*NO) 2.55-1.15 (m, 56 H; *CH*₂) 0.86 - 0.92 (m, 6 H; *CH*₃) F1 (ppm) 95.28 (Glc-*C*-1) 80.14 (Glc-*C*-5) 79.91 (Glc-*C*-3) 72.68 (Ser-*C*-β) 71.81 (Glc-*C*-2, Glc-*C*-4) 62.88 (Glc-*C*-6) 53.92 (Ser-*C*-α) 39.88 (*C*H₂) 38.19 (*Me*NO) 36.57-14.17 (Alkyl chain)

Gal (yield: 25%)

¹H NMR (500 MHz, pyridine) δ ppm 8.93 (d, *J*=7.82 Hz, 1 H; N*H*) 8.56 - 8.61 (m, 1 H; N*H*) 5.49 - 5.55 (m, 1 H; Ser-*H*-α) 4.70 (d, *J*=9.04 Hz, 1 H; Gal-*H*-1) 4.56 - 4.61 (m, 1 H; Gal-*H*-4) 4.41 - 4.56 (m, 5 H; Gal-*H*-2, Gal-*H*-6, Ser-*H*-β) 4.16 (dd, *J*=9.04, 2.69 Hz, 1 H; Gal-*H*-3) 4.04 - 4.09 (m, 1 H; Gal-*H*-5) 3.41 - 3.55 (m, 2 H; C*H*₂) 3.00 (s, 3 H; *Me*NO) 2.54-1.15 (m, 56 H; C*H*₂) 0.89 (t, *J*=6.23 Hz, 6 H; C*H*₃)

F1 (ppm) 95.67 (Gal-*C*-1) 78.65 (Gal-*C*-5) 76.58 (Gal-*C*-3) 72.56 (Ser-*C*-β) 70.24 (Gal-*C*-4) 69.55 (Gal-*C*-2) 62.31 (Gal-*C*-6) 53.74 (Ser-*C*-α) 39.87 (*C*H₂) 38.04 (*Me*NO) 36.52-14.22 (Alkyl chain)

Xyl (yield: 38%)

¹H NMR (500 MHz, pyridine) δ ppm 9.02 (d, *J*=8.06 Hz, 1 H; N*H*) 8.64 (t, *J*=5.62 Hz, 1 H; N*H*) 5.44 - 5.50 (m, 1 H; Ser-*H*-α) 4.62 (d, *J*=8.79 Hz, 1 H; Xyl-*H*-1) 4.53 (m, 2 H; Ser-*H*-β) 4.33 - 4.39 (m, 1 H; Xyl-*H*-5a) 4.11 (m, 3 H; Xyl-*H*-2, Xyl-*H*-3, Xyl-*H*-4) 3.61 - 3.68 (m, 1 H; Xyl-H-5b) 3.40 - 3.54 (m, 2 H; CH₂) 2.95 (s, 3 H; *Me*NO) 2.54 -1.17 (m, 56 H; CH₂) 0.86 - 0.92 (m, 6 H; CH₃)

F1 (ppm) 95.79 (Xyl-C-1) 79.81 (Xyl-C-3) 72.67 (Ser-C-β) 71.76 (Xyl-C-2) 71.10 (Xyl-C-4) 69.04 (Xyl-C-5) 53.73 (Ser-C-α) 39.98 (CH₂) 37.78 (*Me*NO) 36.52-14.26 (Alkyl chain)

Fuc (yield: 47%)

¹H NMR (500 MHz, pyridine) δ ppm 9.04 (d, *J*=8.06 Hz, 1 H; N*H*) 8.50 - 8.57 (m, 1 H; N*H*) 5.37 - 5.45 (m, 1 H; Ser-*H*-α) 4.73 (d, *J*=9.04 Hz, 1 H; Fuc-*H*-1) 4.49 - 4.56 (m, 1 H; Ser-*H*-βa) 4.42 - 4.49 (m, 1 H; Ser-*H*-βb) 4.35 (t, *J*=8.79 Hz, 1 H; Fuc-*H*-2) 4.09 - 4.15 (m, 1 H; Fuc-*H*-3) 4.03 - 4.08 (m, 1 H; Fuc-*H*-4) 3.79 - 3.87 (m, 1 H; Fuc-*H*-5) 3.38 - 3.53 (m, 2 H; C*H*₂) 2.92 (s, 3 H; *Me*NO) 1.54 (d, *J*=6.11 Hz, 3 H; Fuc-*H*-6) 2.55-1.12 (m, 56 H; C*H*₂) 0.89 (t, *J*=6.35 Hz, 6 H; C*H*₃)

F1 (ppm) 95.71 (Fuc-*C*-1) 76.40 (Fuc-*C*-3) 73.02 (Fuc-*C*-5) 72.71 (Ser-*C*-β) 72.68 (Fuc-*C*-4) 69.16 (Fuc-*C*-2) 54.24 (Ser-*C*-α) 39.72 (*C*H₂) 37.07 (*Me*NO) 36.30-14.22 (Alkyl chain) 17.25 (Fuc-*C*-6)

Lactose (yield: 8%)

¹H NMR (500 MHz, pyridine) δ ppm 9.06 (d, *J*=7.82 Hz, 1 H; N*H*) 8.62 - 8.68 (m, 1 H; N*H*) 5.44 - 5.50 (m, 1 H; Ser-*H*-α) 5.09 (d, *J*=7.82 Hz, 1 H; Gal-*H*-1) 4.63 (d, *J*=8.79 Hz, 1 H; Glc-*H*-1) 4.46 - 4.57 (m, 7 H; Glc-*H*-6, Gal-*H*-2, Gal-*H*-4, Gal-*H*-6a, Ser-*H*-β) 4.40 - 4.46 (m, 1 H; Gal-*H*-6b) 4.18 (m, 5 H; Glc-*H*-2, Glc-*H*-3, Glc-*H*-4, Gal-*H*-3, Gal-*H*-5) 3.79 - 3.86 (m, 1 H; Glc-*H*-5) 3.40 - 3.55 (m, 2 H; C*H*₂) 2.93 (s, 3 H; *Me*NO) 2.55-1.16 (m, 56 H; C*H*₂) 0.88 (t, *J*=6.35 Hz, 6 H; C*H*₃)

F1 (ppm) 106.08 (Gal-C-1) 95.08 (Glc-C-1) 81.98 (Glc-C-4) 78.28 (Glc-C-5) 77.99 (Glc-C-3) 77.50 (Gal-C-5) 75.33 (Gal-C-3) 72.95 (Ser-C-β) 71.77 (Gal-C-2) 71.47 (Glc-C-2) 70.14 (Gal-C-4) 62.32 (Glc-C-6) 62.15 (Gal-C-6) 54.08 (Ser-C-α) 39.88 (CH₂) 38.22 (*Me*NO) 36.48-14.40 (Alkyl chain)

Maltose (yield: 22%)

¹H NMR (500 MHz, pyridine) δ ppm 9.03 (d, *J*=8.06 Hz, 1 H; N*H*) 8.64 (t, *J*=5.62 Hz, 1 H; N*H*) 5.92 (d, *J*=3.91 Hz, 1 H; Glc-B-*H*-1) 5.44 - 5.50 (m, 1 H; Ser-*H*-α) 4.61 (d, *J*=9.04 Hz, 1 H; Glc-A-*H*-1) 4.50 - 4.59 (m, 4 H; Glc-B-*H*-3, Glc-B-*H*-5, Glc-B-*H*-6a, Ser-*H*-β) 4.45 (br. s., 2 H; Glc-A-*H*-6) 4.24 - 4.39 (m, 3 H; Glc-A-*H*-3, Glc-A-*H*-4, Glc-B-*H*-6b) 4.13 - 4.22 (m, 3 H; Glc-A-*H*-2, Glc-B-*H*-2, Glc-B-*H*-4) 3.72 - 3.77 (m, 1 H; Glc-A-*H*-5) 3.39 - 3.55 (m, 2 H; C*H*₂) 2.92 (s, 3 H; *Me*NO) 2.54-1.18 (m, 56 H; C*H*₂) 0.88 (t, *J*=6.59 Hz, 6 H; C*H*₃)

F1 (ppm) 103.28 (Glc-B-C-1) 95.16 (Glc-A-C-1) 81.27 (Glc-A-C-4) 79.11 (Glc-A-C-3) 78.60 (Glc-A-C-5) 75.53 (Glc-B-C-3) 75.21 (Glc-B-C-5) 74.42 (Glc-B-C-2) 72.94 72.12 71.58 (Glc-A-C-2)62.72 (Glc-B-C-6) 61.94 (Glc-A-C-6) 53.81 (Ser-C-α) 40.03 (CH₂) 38.25 (*Me*NO) 36.58-14.16 (Alkyl chain)

Melibiose (yield: 21%)

¹H NMR (500 MHz, pyridine) δ ppm 8.96 (d, *J*=7.82 Hz, 1 H; N*H*) 8.54 - 8.61 (m, 1 H; N*H*) 5.51 - 5.57 (m, 1 H; Gal-*H*-1) 5.42 - 5.49 (m, 1 H; Ser-*H*-α) 4.60 - 4.71 (m, 4 H; Glc-*H*-1, Gal-*H*-2, Gal-*H*-4, Gal-*H*-5) 4.40 - 4.56 (m, 6 H; Glc-*H*-6a, Gal-*H*-3, Gal-*H*-6, Ser-*H*-β) 4.31 - 4.38 (m, 1 H; Glc-*H*-6b) 4.07 - 4.20 (m, 3 H; Glc-*H*-2, Glc-*H*-3,

Glc-*H*-4) 3.92 - 4.00 (m, 1 H; Glc-*H*-5) 3.38 - 3.56 (m, 2 H; C*H*₂) 2.95 (s, 3 H; *Me*NO) 2.57-1.13 (m, 56 H; C*H*₂) 0.84 - 0.94 (m, 6 H; C*H*₃)

F1 (ppm) 100.95 (Gal-C-1) 95.13 (Glc-C-1) 79.66 (Glc-C-3) 78.01 (Glc-C-5) 72.74 (Ser-C-β) 72.61 (Gal-C-5) 71.91 (Gal-C-3) 71.85 (Glc-C-4) 71.61 (Glc-C-2) 71.12 (Gal-C-4) 70.76 (Gal-C-2) 68.62 (Glc-C-6) 62.63 (Gal-C-6) 53.80 (Ser-C-α) 39.93 (CH₂) 38.15 (*Me*NO) 36.57-14.18 (Alkyl chain)

3-5 References

- S.-I. Nishimura, K. Niikura, M. Kurogochi, T. Matsushita, M. Fumoto, H. Hinou, R. Kamitani, H. Nakagawa, K. Deguchi, N. Miura, K. Monde, and H. Kondo "High-Throughput Protein Glycomics: Combined Use of Chemoselective Glycoblotting and MALDI-TOF/TOF Mass Spectrometry", *Angew. Chem. Int. Ed.*, 2004, 44, 91-96
- H. Shimaoka, H. Kuramoto, J.-i. Furukawa, Y. Miura, M. Kurogochi, Y. Kita, H. Hinou, Y. Shinohara, and S.-I. Nishimura, "One-Pot Solid-Phase Glycoblotting and Probing by Transoximization for High-Throughput Glycomics and Glycoproteomics", *Chem. Eur. J.*, 2007, 13, 1664-1673
- J. M. Langenhan, B. R. Griffith, and J. S. Thorson, "Neoglycorandomization and Chemoenzymatic Glycorandomization: Two Complementary Tools for Natural Product Diversification", *J. Nat. Prod.*, 2005, 68, 1696-1711
- J. Kalia and R. T. Raines, "Hydrolytic Stability of Hydrazones and Oximes", *Angew. Chem. Int. Ed.*, 2008, 120, 7633–7636
- Y. Miura, M. Hato, Y. Shinohara, H. Kuramoto, J.-i. Furukawa, M. Kurogochi, H. Shimaoka, M. Tada, K. Nakanishi, M. Ozaki, S., and S.-I. Nishimura, "BlotGlycoABCTM, an Integrated Glycoblotting Technique for Rapid and Large Scale Clinical Glycomics", *Molecular & Cellular Proteomics*, 2007, 7, 370-377
- K. Hirose, M. Amano, R. Hashimoto, Y. Chuan Lee, and S.-I. Nishimura, "Insight into Glycan Diversity and Evolutionary Lineage Based on Comparative Avio-N-glycomics and Sialic Acid Analysis of 88 Egg Whites of Galloanserae", *Biochemistry*, 2011, 50, 4757-4774

- S.-I. Nishimura, M. Hato, S. Hyugaji, F. Feng, and M. Amano. "Glycomics for drug discovery: metabolic perturbation in androgen-independent prostate cancer cells induced by unnatural hexosamine mimics", *Angew. Chem. Int. Ed. Engl.*, 2012, 51, 3386-3390
- T. Furukawa, M. Arai, F. Garcia-Martin, M. Amano, H. Hinou, and S.-I. Nishimura, "Glycoblotting-based high throughput protocol for the structural characterization of hyaluronan degradation products during enzymatic fragmentation", *Glycoconj. J.*, 2013, 30, 171-182
- S. Hideshima, H. Hinou, D. Ebihara, R. Sato, S. Kuroiwa, T. Nakanishi, S.-I. Nishimura, and T. Osaka, "Attomolar Detection of Influenza A Virus Hemagglutinin Human H1 and Avian H5 Using Glycan-Blotted Field Effect Transistor Biosensor", *Anal. Chem.*, 2013, 85, 5641–5644
- T. Matsushita, W. Takada, K. Igarashi, K. Naruchi, R. Miyoshi, F. Garcia-Martin, M. Amano, H. Hinou, and S.-I. Nishimura, "A straightforward protocol for the preparation of high performance microarray displaying synthetic MUC1 glycopeptides", *Biochim. Biophys. Acta.*, 2014, 1840, 1105-1116
- N. N. Maolanon, M. Blaise, K. K. Sørensen, M. B. Thygesen, E. Clo, J. T. Sullivan,
 C. W. Ronson, J. Stougaard, O. Blixt, and K. J. Jensen, "Lipochitin Oligosaccharides Immobilized through Oximes in Glycan Microarrays Bind LysM Proteins", *ChemBioChem*, 2014, 15, 1-11
- T. Ohyanagi, N. Nagahori, K. Shimawaki, H. Hinou, T. Yamashita, A. Sasaki, T. Jin,
 T. Iwanaga, M. Kinjo, and S-I. Nishimura, "Importance of sialic acid residues illuminaterd by live animal imaging using phosphorylcholine self-assembled monolayers-coated quantum dots", *J. Am. Chem. Soc.*, **2011**, 133, 12507-12517

- M. B. Thygesen, J. Sauer, and K. J. Jensen, "Chemoselective Capture of Glycans for Analysis on Gold Nanoparticles: Carbohydrate Oxime Tautomers Provide Functional Recognition by Proteins", *Chem. Eur. J.*, **2009**, 15, 1649-1660
- 14. N. Matsubara, K. Oiwa, T. Hohsaka, R. Sadamoto, K. Niikura, N. Fukuhara, A. Takimoto, H. Kondo, and S.-I. Nishimura, "Molecular Design of Glycoprotein Mimetics: Glycoblotting by Engineered Proteins with an Oxylamino-Functionalized Amino Acid Residue", *Chem. Eur. J.*, **2005**, 11, 6974-6981
- 15. S. E. Cervigni, P. Dumy, M. Mutter, "Synthesis of Glycopeptides and Lipopeptides by Chemoselective Ligation", *Angew. Chem. Int. Ed.*, **1996**, 35, 1230-1232
- 16. J. M., Langenhan, N. R. Peters, I. Guzei, F. M. Hoffmann, and J. S. Thorson, "Enhancing the anticancer properties of cardiac glycosides by neoglycorandomization". *Proc. Nat. Acad. Sci. USA*, **2005**, 102, 12305-12310
- B. R. Griffith, C. Krepel, X. Fu, S. Blanchard, A. Ahmed, C. E. Edmiston, J. S. Thorson, "Model for antibiotic optimization via neoglycosylation: synthesis of liponeoglycopeptides active against VRE", *J.Am. Chem.Soc.*, 2007, 129, 8150-8155
- R. D. Goff and J. S. Thorson, "Enhancing the divergent activities of betulinic acid via neoglycosylation", *Org. Lett.*, 2009, 11, 461-464
- 19. R. D. Goff and J. S. Thorson, "Assessment of chemoselective neoglycosylation methods using chlorambucil as a model", *J. Med. Chem.*, **2010**, 53, 8129-8139
- 20. R. D. Goff, S. Singh, and J. S. Thorson, "Glycosyloxyamine neoglycosylation: a model study using calicheamicin", *ChemMedChem*, **2011**, 6, 774-776
- 21. F. Peri, P. Dumy and M. Mutter, "Chemo- and Stereoselective Glycosylation of Hydroxylamino Derivatives : A Versatile Approach to Glycoconjugates", *Tetrahedron*, **1998**, *54*, 12269-12278

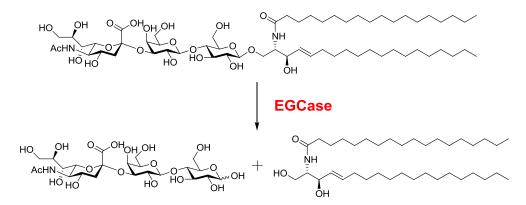
22. Y. Liu, T. Feizi, M. A. Campanero-Rhodes, R. A. Childs, Y. Zhang, B. Mulloy, P. G. Evans, H. M. I. Osborn, D. Otto, P. R. Crocker, and W. Chai, "Neoglycolipid probes prepared via oxime ligation for microarray analysis of oligosaccharide-protein interactions", *Chemistry & Biology*, 2007, 14, 847-859

Chapter 4

Inhibitory Activity of N-LacCer Derivatives for EGCase II

4-1 Introduction

One of my purposes is to determine whether prepared neoglycolipids work as a mimetic compound of natural glycosphingolipids. For that purpose, my interest was directed to the synthesis and biological characteristics of neoglycosphingolipids having a non-natural N(OMe)-glycosidic linkage between carbohydrate and ceramide. Especially, it seems likely that non-natural GSLs bearing N(OMe)-glycosidic linkage might become promising candidates of the inhibitors against endoglycoceramidases (EGCase) and ceramide glycanases (CGase) that digest specifically major GSLs at O-glycosidic linkage between oligosaccharide moiety and ceramide^[1-6].(Scheme 4-1-1)



Scheme 4-1-1 EGCase (and CGase) cleaves O-glycosidic linkage between oligosaccharide moiety and ceramide

Among those enzymes, I focused a recombinant EGCase II from *rhodococcus sp*. This enzyme has been most used broadly because mutated products of that have an activity of synthase, which means to conjugate ceramide and glycan part^[7-9]. The activity was verified by X-ray crystallography measurement^[10]. Additionally, Withers et al. also reported the significance of the nitrogen atom at *exo*-anomeric position of the

disaccharide mimetic compound identified as a potent inhibitor for hydrolysis reaction of EGCase II^[11]. Furthermore, X-ray structural study revealed that Glu-233 which is a general acid/base catalytic residue of this enzyme was an essential amino acid residue to interact with the *exo*-nitrogen atom of these compounds.

In this chapter, we focused on 2 compounds shown below and called *N*-LacCer and *O*-LacCer (Fig. 4-1-1)

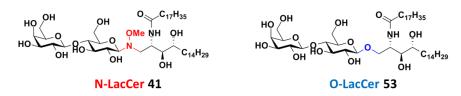
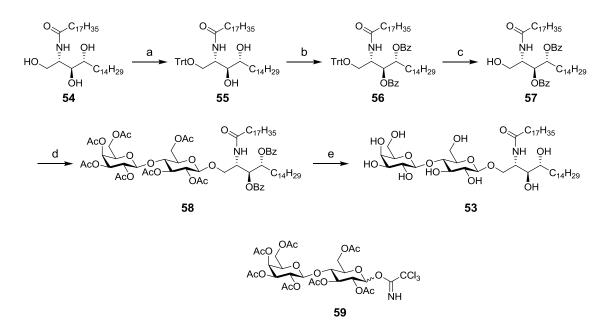


Fig. 4-1-1 Structure of N-LacCer 41 and O-LacCer 53

4-2 Results & Discussions

4-2-1 Synthesis of O-LactosycCeramide **53** for a Positive Control

To evaluate the function of neoglycosides correctly, we prepared O-glycosyllactosylceramide **53**^[12] which was composed by the same length of phytosphingosine and acyl chain as synthesized neoglycolipids (Scheme 4-2-1). This compound was not confirmed as a substrate of EGCase but expected to be hydrolyzed because of O-glycoside.



Scheme 4-2-1. Synthesis of *O*-LacCer **53**. Reagents and conditions: a) TrCl, Pyr, 50 °C, 4 h, 78%; b) BzCl, pyr, CH₂Cl₂, RT, 2 d, 88%; c) HBr-AcOH, CH₂Cl₂, 0 °C, 3 min, 80 %; d) **59**, TMSOTf, CH₂Cl₂, 0 °C, 1 h, 31%; e) NaOMe, CH₂Cl₂, MeOH, RT, 5 h, 95%

4-2-2 Primitive hydrolysis assay

I preliminarily tested the inhibitory effect of *N*-LacCer **41**, methoxyamino derivative having lactoside, on the hydrolysis of *O*-LacCer **53** by a recombinant EGCase II from rhodococcus sp. (Fig. 4-2-1) While *O*-LacCer was hydrolyzed in the presence of the enzyme, *N*-LacCer was remained as the starting form regardless of temperature or the existence of the enzyme. Most interestingly, intact *O*-LacCer was significantly detected even after 8 hours as shown in Entry 5. The result suggested *N*-LacCer worked as an inhibitor to reduce the enzymatic activity. The result indicates an evidence for the inhibitory effect of compound **41** on the hydrolytic activity of this enzyme.

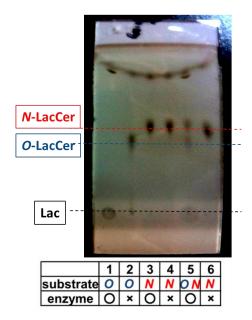


Figure 4-2-1. TLC-based inhibition assay. The spot of each compound and contents of solution were indicated in left figure.

To monitor the inhibitory effect of *N*-LacCer on the hydrolytic activity of recombinant EGCase II, we tested the feasibility of modified Park-Johnson method ^[13] for highly sensitive and quantitative analysis of the reducing power of the generated lactose residue from *O*-LacCer as a substrate during the hydrolysis.

(1)
$$K_3 Fe^{III}(CN)_6$$
 + Reducing Sugar $\xrightarrow{Na_2CO_3} K_4 Fe^{II}(CN)_6$

(2)
$$K_4 Fe^{II}(CN)_6 + 4Fe^{III} NH_4(SO_4)_2$$

 $\longrightarrow Fe^{III}[Fe^{II}(CN_6)]_3 + 6K_2 SO_4 + 2(NH_4)_2 SO_4$

Scheme 4-2-2. A principle of colorimetric reaction of Park and Johnson method

Colorimetric method utilizes the reduction ability of free sugars to convert ferricyanide ions into Prussian blue (ferric ferrocyanide) in alkaline solution as indicated in Scheme 4-2-2. Though it was well known that three reagents were needed commonly in the reaction steps, the revised solutions for more sensitive detection was reported recently (T. Ikuma, K. Takeuchi, Y. Takahashi, K. Sagisaka, and T. Takasawa, High sensitive colorimetric method of reducing sugar using ferric iron reagent. *Res. Bull. Obihiro Univ.* **2001**, *22*, 109-116). I decided to use this improved protocol and the feasibility in the inhibitory assay for EGCase II activity was tested as follows: (A) 50 mM Na₂CO₃, 10 mM KCN aq., (B) 1.5 mM K₃Fe(CN)₆ aq., and (C) 0.15% (w/v) Fe•NH₄(SO₄)₂, 0.2% (w/v) SDS, 0.03 N H₂SO₄ aq. I used lactose as a source of reducing sugar and it was dissolved in the buffer for enzymatic reaction at several concentrations (20~120 ng/mL). 20 μ L of each aliquot was injected to a mixture containing 90 μ L of reagent (A) and same volume of reagent (B) in an eppendorf tube. The solution was divided into three portions of 60 μ L in PCR tubes. The tubes were closed strongly by a weight and incubated at 100°C for 20 min. The heat was cooled down in the refrigerator at -4°C for 3 min and then left at room temperature for 10 min. The tubes were opened and added 135 μ L of reagent (C). After shaken by vortex mixer, 150 μ L of each solution was transferred on the 96-wel plate and absorbance at 690 nm was measured by a plate reader at 15 min from adding reagent (C). The result is shown below as a standard curve of lactose. This data was corrected by subtracting the absorbance value of blank area (containing no lactose). We obtained good linearity and error range was too small to be depicted on the graph (Fig.4-2-3). This result indicated that this modified protocol can be used for the quantification of the released lactose from GSLs in the presence of recombinant EGCase.

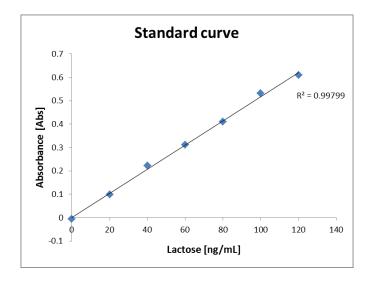
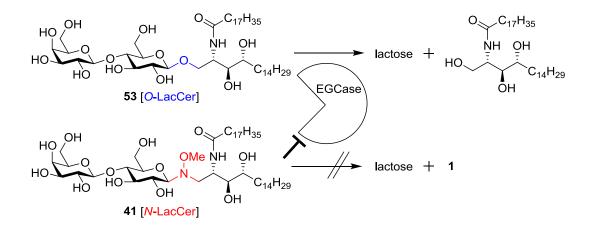


Figure 4-2-3. Standard curve of colorimetric reaction

4-2-4 Inhibition Assay of Recombinant EGCase II



Scheme 4-2-2. A plausible model for the inhibitory effect of N-LacCer 41.

The result of primitive assay suggested that *N*-LacCer inhibited hydrolysis of *O*-LacCer. To determine whether it was true, we first conducted to estimate the km value (Fig. 4-2-4).

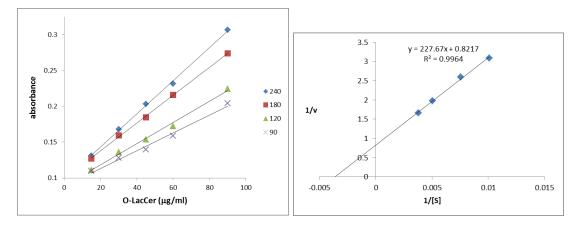


Fig. 4-2-4. Different velocity of hydrolysis dependent on the concentration of *O*-LacCer **53** and estimation of Km value

 K_m value was estimated to be 277 μ M, which was reasonable because the K_m value of asialo GM1 having erythro sphingosine skelton was reported as 500 μ M previously.

Next, It was demonstrated that hydrolysis of *O*-LacCer **53** (0.45 mM) by EGCase II (12 mU/mL) was reduced significantly in the presence of *N*-LacCer **41** (0.27 mM) the release of lactose from *O*-LacCer **53** in the presence of EGCase II (Fig. 4-2-5). This result clearly indicates that *N*-LacCer **41** interacts directly with EGCase II in a similar manner to *O*-LacCer **53** and appeared to act as a competitive inhibitor. Interestingly, Withers et al. reported the significance of the nitrogen atom at exo-anomeric position of the disaccharide mimetic compound identified as a potent inhibitor of EGCase II. Furthermore, X-ray structural study revealed that Glu-233 which is a general acid/base catalytic residue of this enzyme was an essential amino acid residue to interact with the exo-nitrogen atom of this compound. These results may suggest the importance of a nitrogen atom involved in the β -N(OMe)-glycoside bond of N-LacCer **41** for the interaction with Glu-233 residue of EGCase II.

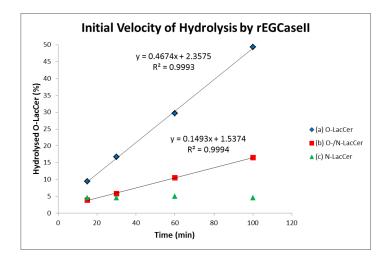


Figure 4-2-5. Inhibitory effect of lactosylceramide mimic **41** (*N*-LacCer) on the hydrolytic activity of recombinant EGCase II. Initial velocity of the hydrolysis of *O*-LacCer **53** was determined by quantitating lactose released from *O*-LacCer using modified Park and Johnson method (Supporting Information). **•**: *O*-LacCer (450 μ M), **•**: *O*-LacCer (450 μ M), and *N*-LacCer (270 μ M), **•**: *N*-LacCer (735 μ M); Reaction condition: 20 mM AcOH buffer (pH 4.7), 0.4% Triton X-100, EGCase II (12 mU/mL), 37°C.

Finally, we tried to estimate the K_i value of N-LacCer **41** for EGCase II. Reactions were performed with different concentration of *O*-LacCer/*N*-LacCer. Results showed reaction velocity decreased depending on inhibitor concentration (Fig. 4-2-6) and K_i value was estimated as 96.5 μ M (Fig. 4-2-7) by dixon plot. Result also suggested inhibition form was competitive.

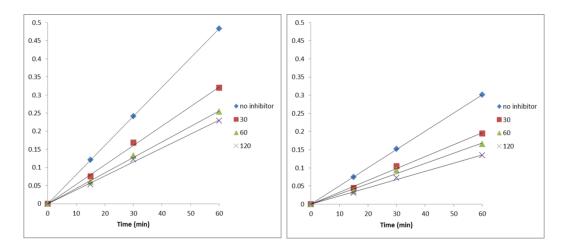


Fig. 4-2-6. Inhibition assay with different concentration

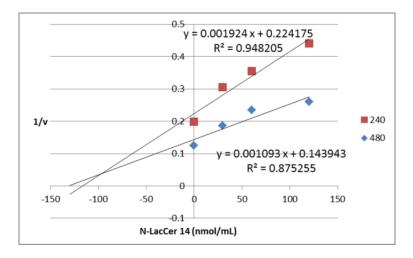


Fig. 4-2-7. Estimation of Ki value

4-3 Conclusion

In this chapter, we evaluate the function of neoglycoside to imitate natural glycoceramide using EGCase II. This enzyme could cleave *O*-glycosidic linkage but not non-natural *N*-glycoside bond. It showed that neoglycolipid was not a substrate of the enzyme. However, when the two compounds, *O*-LacCer and *N*-LacCer, were mixed, the velocity of hydrolysis was apparently reduced compared with only *O*-LacCer. It was significantly important because it indicated that *N*-LacCer could be recognized by the enzyme like natural glycoceramide. This result enhanced the effectiveness of glycoblotting method for preparation of glycolipids.

4-4 Experimental Section

Synthesis of O-LacCer (53).

(2*S*, 3*S*,4*R*)-2-[*N*-(octadecanoyl)amino]-1-(triphenyl-methyloxy)octadecane-3,4-diol (55).

To a solution of compound **54** (1.51 g, 2.59 mmol) in pyridine (20 mL), trityl chloride (1.44 g, 5.17 mmol) was added and the mixture was stirred for 4 h at 50 °C. Reaction was quenched by MeOH and solvent was removed in vacuo completely after silica gel was added to the mixture. The residue was purified by flash column chromatography on silica gel (hexane-EtOAc, 6:1-3:1) to yield the desired compound **55** as a white amorphas solid (1.67 g, 78%); ¹H NMR (600 MHz, CDCl₃): δ 7.41 (d, *J* = 7.52 Hz, 6H; Ar-*H*), 7.31 (d, *J* = 7.61 Hz, 6H; Ar-*H*), 7.25 (m, 3H; Ar-*H*), 6.07 (d, *J* = 8.44 Hz, 1H; N*H*), 4.25 (m, 1H; *H*-2), 3.57 (m, 1H; *H*-3), 3.50 (dd, *J* = 9.90, 3.48 Hz, 1H; *H*-1a), 3.38 (br.s, 1H; OH), 3.34 (dd, *J* = 9.90, 4.58 Hz, 1H; *H*-1b), 3.18 (br.s, 1H; OH), 2.36 (br.s, 1H; OH), 2.14 (t, *J* = 7.61 Hz, 2H; NHCOCH₂a,b), 1.67 (m, 1H; *H*-5a), 1.60 (m, 2H; NHCOCH₂CH₂a,b), 1.44 (m, 2H; *H*-5b, *H*-6a), 1.37-1.18 (m, 51H; CH₂, *H*-6b), 0.88 (t, *J* = 6.97 Hz, 6H; CH₃); ¹³C NMR (150 MHz (HSQC), CDCl₃, 298 K): δ (ppm) = 128.15, 127.79, 127.01 (Ar) 75.36 (C-3), 73.01 (C-4), 62.71 (C-1), 50.40 (C-2), 36.54 (NHCOCH₂), 32.87-21.67 (CH₂), 13.97 (CH₃); HRMS (ESI) Calcd. for C₅₅H₈₇NO₄Na [M+Na]⁺ 848.65328, found 848.64941

(2*S*, 3*S*, 4*R*)-3,4-(di-*O*-Benzoyl)-2-[*N*-(octadecanoyl)amino]-1-(triphenyl-methyloxy)octadecane-3,4-diol (56).

To a solution of compound **55** (826 mg, 1.00 mmol) in pyridine (10 mL), benzoyl chloride (465 μ L, 4.00 mmol) was added dropwise and the mixture was stirred for 2 days at room temperature. It was diluted with EtOAc and washed with 1 N HCl three times, sat. NaHCO₃ aq. and brine. Organic phase was dried over MgSO₄ and concentrated under reduced pressure. Purification of the crude product by flash column chromatography on silica gel (Hexane-EtOAc, 15:1) yielded compound **56** (906 mg, 88%) as colorless oil; ¹H NMR (500 MHz, CDCl₃): 7.96 (d, *J* = 8.18 Hz, 2H; Ar-*H*), 7.89 (d, *J* = 8.18 Hz, 2H; Ar-*H*), 7.53 (m, 2H; Ar-*H*), 7.38 (m, 4H; Ar-*H*), 7.31 (d, *J* = 8.06 Hz, 6H; Ar-*H*), 7.10 (m, 9H; Ar-*H*), 6.10 (d, *J* = 9.53 Hz, 1H; N*H*), 5.83 (dd, *J* = 9.04, 2.69 Hz, 1H; *H*-3) 5.37 (m, 1H; *H*-4), 4.61 (m, 1H; *H*-2), 3.32 (m, 2H; *H*-1a,b), 2.18 (m, 2H; NHCOCH₂a,b), 1.88 (m, 2H; H-5a,b), 1.64 (m, 2H; NHCOCH₂CH₂a,b), 1.45-1.17 (m, 52H; CH₂), 0.87 (t, *J* = 6.96 Hz, 6H; CH₃); ¹³C NMR (150 MHz (HSQC), CDCl₃): δ 132.95, 132.79, 129.66, 128.44, 128.26, 127.67, 126.89 (Ar) 74.08 (*C*-4), 72.61 (*C*-3), 61.62 (*C*-1), 48.51 (*C*-2), 36.77 (NHCOCH₂), 31.85-22.61 (*C*H₂), 14.04 (*C*H₃); HRMS (ESI) Calcd. for C₆₉H₉₅NO₆Na [M+Na]⁺ 1056.70571, found 1056.70205

(2*S*, 3*S*, 4*R*)-3,4-(di-*O*-Benzoyl)-2-[*N*-(octadecanoyl)-amino]octadecane-1,3,4-triol (57).

To a solution of compound **56** (713 mg, 689 μ mol) in CH₂Cl₂ (3 mL), HBr-AcOH (270 μ L, 1.38 mmol) was added dropwise and the mixture was stirred for 3 minutes at 0 °C. It was diluted with EtOAc quickly and washed with brine, sat. NaHCO₃ aq. three times

and brine. Organic phase was dried over MgSO₄ and concentrated under reduced pressure. Purification of the crude product by flash column chromatography on silica gel (Hexane-EtOAc, 10:1-4:1) yielded compound **57** (435 mg, 80%) as colorless oil; ¹H NMR (500 MHz, CDCl₃): δ 8.02 (d, *J* = 8.44 Hz, 2H; Ar-*H*), 7.94 (d, *J* = 8.29 Hz, 2H; Ar-*H*), 7.60 (m, 1H; Ar-*H*), 7.50 (m, 1H; Ar-*H*), 7.46 (t, *J* = 8.13 Hz, 2H; Ar-*H*), 7.35 (t, *J* = 8.13 Hz, 2H; Ar-*H*), 6.78 (d, *J* = 9.38 Hz, 1H; N*H*), 5.52 (dd, *J* = 9.38, 2.50 Hz, 1H; H-3) 5.38 (m, 1H; *H*-4), 4.44 (m, 1H; *H*-2), 3.66 (m, 2H; *H*-1a,b), 2.28 (t, *J* = 7.82 Hz 2H; NHCOC*H*₂a,b), 2.02 (m, 2H; *H*-5a,b), 1.67 (m, 2H; , NHCOC*H*₂C*H*₂a,b), 1.44 (m, 1H; *H*-6a), 1.40-1.17 (m, 51H; C*H*₂, *H*-6b), 0.88 (t, *J* = 6.96 Hz, 6H; C*H*₃); ¹³C NMR (150 MHz (HSQC), CDCl₃): δ 133.55, 132.95, 129.84, 129.59, 128.52, 128.25, (Ar) 74.03 (*C*-4), 73.18 (*C*-3), 61.42 (*C*-1), 49.92 (*C*-2), 36.72 (NHCOC*H*₂), 31.86-22.62 (CH₂), 14.04 (CH₃); HRMS (ESI) Calcd. for C₅₀H₈₁NO₆Na [M+Na]⁺ 814.59616, found 814.59354

(2S,3S,4R)-3,4-(di-*O*-Benzoyl)-2-[*N*-(octadecanoyl)-amino]-1-{*O*-[β -2,3,4,6-tetraace tylgalactosyl-pyranosyl-(1 \rightarrow 4)- β -2,3,6-triacetylglucopyranosyl]}octadecane-1,3,4-tr iol (58).

Compound **57** (276 mg, 379 μ mol) and lactose donor **59** (408 mg, 522 μ mol) were dissolve in CH₂Cl₂ (5 mL). The mixture was stirred at 0 °C and TMSOTf (5 mL, 28 μ mol) was added dropwise. After 1 hour, the solution was diluted with EtOAc and washed with sat. NaHCO₃ aq. and brine. Organic phase was dried over MgSO₄ and concentrated under reduced pressure. Purification of the crude product by flash column chromatography on silica gel (CH₂Cl₂-Et₂O, 15:1, Hexane-EtOAc, 10:1-4:1) yielded

compound **58** (165 mg, 31%); ¹H NMR (500 MHz, CDCl₃): δ 8.01 (d, J = 8.22 Hz, 2H; Ar-H), 7.96 (d, J = 8.22 Hz, 2H; Ar-H), 7.61 (m, 1H; Ar-H), 7.54 (m, 1H; Ar-H), 7.47 (t, *J* = 8.22 Hz, 2H; Ar-*H*), 7.40 (t, *J* = 8.22 Hz, 2H; Ar-*H*), 6.14 (d, *J* = 9.48 Hz, 1H; N*H*), 5.58 (dd, J = 8.22, 3.48 Hz, 1H; PS-H-3) 5.34 (m, 2H; PS-H-4, Gal-H-4), 5.11 (t, J = 9.16 Hz, 1H; Glc-H-3), 5.07 (dd, J = 10.43, 7.90 Hz, 1H; Gal-H-2), 4.93 (dd, J = 10.43, 3.48 Hz, 1H; Gal-*H*-3), 4.80 (dd, *J* = 9.48, 7.58 Hz, 1H; Glc-*H*-2), 4.58 (m, 1H; PS-*H*-2), 4.41 (d, *J* = 7.90 Hz, 2H; Glc-*H*-1, Gal-*H*-1), 4.34 (m, 1H; Glc-*H*-6a), 4.08 (m, 2H; Gal-H-6a,b), 3.85 (m, 3H; PS-H-1a,Glc-H-6b, Gal-H-5), 3.70 (m, 2H; PS-H-1, Glc-*H*-4), 3.51 (ddd, J = 9.80, 5.06, 1.90 Hz, 1H; Glc-*H*-5), 2.24 (s, 2H; NHCOCH₂a,b), 2.14 (s, 3H; Ac), 2.06 (s, 3H; Ac), 2.02 (s, 3H; Ac), 1.99 (s, 3H; Ac), 1.98 (s, 3H; Ac), 1.95 (s, 3H; Ac), 1.94 (s, 3H; Ac), 1.85 (m, 2H; PS-H-5), 1.66 (m, 2H; NHCOCH₂CH₂a,b), 1.41-1.16 (m, 52H; CH₂), 0.88 (t, J = 6.95 Hz, 6H; CH₃); ¹³C NMR (150 MHz (HSQC), CDCl₃): δ 133.24, 132.94, 129.67, 129.62, 128.41, 128.26, (Ar) 100.35 (Glc-C-1, Gal-C-1), 75.80 (Glc-C-4), 73.65 (PS-C-4), 72.44 (Glc-C-5), 72.37 (Glc-C-3), 72.31 (PS-C-3), 71.49 (Glc-C-2), 70.79 (Gal-C-3), 70.54 (Gal-C-5), 68.75 (Gal-C-2), 67.23 (PS-C-1), 66.54 (Gal-C-4), 61.61 (Glc-C-6), 60.55 (Gal-C-6), 47.91 (PS-C-2), 36.73 (NHCOCH₂), 32.03-22.06 (CH₂), 20.76-20.25 (Ac), 14.50 (CH₃); HRMS (ESI) Calcd. for C₇₆H₁₁₅NO₂₃Na [M+Na]⁺ 1432.77575, found 1432.77069

(2S,3S,4R)-2-[*N*-(octadecanoyl)amino]-1-{*O*-[β -galactosylpyranosyl-(1 \rightarrow 4)- β -gluco pyranosyl]}octadecane-1,3,4-triol (53).

Compound **58** (100 mg, 71 μ mol) was dissolved in CH₂Cl₂ (200 mL) and MeOH (1 mL). To the solution, sodium methoxide (1.5 mg, 28 μ mol) was added and the mixture

was stirred 5 hours at room temperature. Solvent was completely removed by air drying and the product was precipitated in MeOH. After filtration, white powder of compound **53** (61 mg, 95%) was obtained; ¹H NMR (600MHz, d_5 -Pyridine): δ 8.57 (d, J = 8.77 Hz, 1H; NH), 7.64 (br.s, 1H; Glc-2-OH), 7.52 (d, J = 3.95 Hz, 1H; Gal-2-OH), 6.95 (br.s, 1H; Gal-3-OH), 6.65 (br.s, 1H; Gal-6-OH), 6.58 (br.s, 1H; Gal-4-OH), 6.53 (d, J = 6.14 Hz, 1H; PS-3-OH), 6.45 (br.s, 1H; Glc-6-OH), 6.21 (s, 1H; Glc-3-OH), 5.92 (d, J = 7.23 Hz, 1H; PS-4-OH), 5.16 (m, 1H; PS-H-2), 5.01 (Gal-H-1, overlapped with H₂O), 4.91 (d, J = 7.89 Hz, 1H; Glc-H-1), 4.79 (dd, J = 10.52, 5.70 Hz, 1H; PS-H-1a), 4.56-4.43 (m, 10.52)5H; Glc-H-6a,b, Gal-H-2, Gal-H-4, Gal-H-6a), 4.39 (m, 3H; PS-H-1b, PS-H-3, Gal-H-6b), 4.25 (m, 3H; PS-H-4, Glc-H-3, Glc-H-4), 4.15 (m, 1H; Gal-H-3, Gal-H-5), 4.04 (t, J = 7.89 Hz, 1H; Glc-H-2), 3.84 (m, 1H; Glc-H-5), 2.45 (t, J = 7.45 Hz, 2H; NHCOCH₂a,b), 2.24 (m, 1H; PS-H-5a), 1.95 (m, 2H; PS-H-5b, PS-H-6a), 1.82 (m, 2H; NHCOCH₂CH₂a,b), 1.70 (m, 1H; PS-H-6b), 1.50-1.19 (m, 50H; CH₂), 0.88 (t, J = 6.58 Hz, 6H; CH₃); ¹³C NMR (150 MHz (HSQC), *d*₅-Pyridine): δ 105.75 (Gal-C-1), 105.22 (Glc-C-1), 81.89 (Glc-C-4), 77.31 (Gal-C-5), 76.72 (Glc-C-3), 76.59 (Glc-C-5), 75.79 (PS-C-3), 75.12 (Gal-C-3), 74.66 (Glc-C-2), 72.67 (PS-C-4), 72.40 (Gal-C-2), 70.55 (PS-C-1), 69.95 (Gal-C-4), 61.95 (Glc-C-6, Gal-C-6), 51.95 (PS-C-2), 37.22 (NHCOCH₂), 33.43-22.66 (CH₂), 14.83 (CH₃); HRMS (ESI) Calcd. for C₄₈H₉₃NO₁₄Na [M+Na]⁺ 930.64937, found 930.64499

A) *TLC-based assay*: 1 mM solutions of the substrate/inhibitor, *O*-LacCer **53** or *N*-LacCer **41**, were prepared by means of the reaction buffer (20 mM AcOH buffer containing 0.4% Triton X-100, pH 4.7). To 98.5 μL of them (*O*-LacCer: Entry 1 and 3, *N*-LacCer: Entry 2, 4, and 6, 1:1 mixture of *O*-LacCer/*N*-LacCer: Entry 5), 2.5 μ L of EGCase solution (Entry 1, 3, and 5) or the buffer (Entry 2, 4, and 6) was added at 37°C (Entry 1-5) or 0°C (Entry 6). Reaction temperature was kept for 8 hours, then TLC analysis was performed (CHCl₃/MeOH, 3:1) to observe the progress of hydrolysis. Spots were visualized by spraying a solution of 95:5 (v/v) MeOH-concentrated sulfuric acid and heating at 180°C for ca. 1/2 min.

- B) Hydrolysis assay using modified Park-Johnson method for calculation of Km value: 240, 180, 120, and 90 μg/mL of *O*-LacCer 53 were incubated with 5 mU/mL of rEGCase at 37°C in the reaction buffer (20 mM AcOH buffer containing 0.4% Triton X-100, pH 4.7) respectively. 20 μL of aliquots were extracted after 15, 30, 60, and 100 min. The solution was mixed with coloring reagents in PCR tube for quenching the reaction and subjected to the monitoring of the reaction progress. In the PCR tube, 90 μL of reagents (A) and (B) described above were mixed respectively. After adding the reaction mixture collected, the solution was divided into three portions in other tubes (n=3) and colorimetric reaction was performed as described. The raw data obtained were corrected by using the blank value
- C) Inhibition assay using modified Park-Johnson method: Before enzymatic the reaction, 900 μ M solution of *O*-LacCer **53** and 750 μ M solution of *N*-LacCer **41** was prepared by dissolving in the reaction buffer (20 mM AcOH buffer containing 0.4% Triton X-100, pH 4.7) and 3 μ L of enzyme solution was diluted to 10 μ L by the same buffer. Solutions of *O*-LacCer, *N*-LacCer and the reaction buffer were

mixed in eppendorf tubes as shown Table 4-4-1. Then, the reaction was started by adding the enzyme solution at 37° C and 20 μ L of aliquots were extracted after 15, 30, 60, and 100 min. The solution was mixed with coloring reagents in PCR tube for quenching the reaction and subjected to the monitoring of the reaction progress.

In the PCR tube, 90 μ L of reagents (A) and (B) described above were mixed respectively. After adding the reaction mixture collected, the solution was divided into three portions in other tubes (n=3) and colorimetric reaction was performed as described. The raw data obtained were corrected by using the blank value and plotted in the graph after the unit of the enzyme activity was converted into percentage of the hydrolysis of whole *O*-LacCer in comparison with the standard curve of lactose as shown in Figure 4-2-5.

Entry	а	b	С
O-LacCer	50	50	0
N-LacCer	0	36	98
Buffer	48	12	0
EGCase II	2	2	2
Total (µL)	100	100	100

Table 4-4-1. Composition of the reaction mixture

4-5 References

- M. Ito and T. Yamagata, "A Novel Glycosphingolipid-degrading Enzyme Cleaves of the Linkage of Neutral and Acidic between the Oligosaccharide and Ceramide Glycosphingolipids", **1986**, 4, 14278-14282
- 2. M. Ito and T. Yamagata, "Purification and characterization of glycosphingolipid-specific endoglycosidases (endoglycoceramidases) from a mutant sp. Evidence for three molecular strain of Rhodococcus species of endoglycoceramidase with different specificities" J. Biol. Chem., 1989, 264, 9510-9519
- 3. Y. Miura, T. Arai, A. Ohtake, M. Ito, K. Yamamoto, and T. Yamagata, "Requirement for a different hydrophobic moiety and reliable chromogenic substrate for endo-type glycosylceramidases", *Glycobiology*, **1999**, 9, 957-960
- Y. Miura, T. Arai, and T. Yamagata, "Synthesis of amphiphilic lactosides that possess a lactosylceramide-mimicking *N*-acyl structure: Alternative universal substrates for endo-type glycosylceramidases", *Carbohydr. Res.*, 1996, 289, 193-199.
- Y. Horibata, N. Okino, S. Ichinose, A. Omori, and M. Ito, "Purification, characterization, and cDNA cloning of a novel acidic endoglycoceramidase from the jellyfish, *Cyanea nozakii*", *J. Biol. Chem.*, 2000, 275, 31297-31304
- Y.-T. Li, C.-W. Chou, S.-C. Li, U. Kobayashi, Y.-h. Ishibashi, and M. Ito, " Preparation of homogenous oligosaccharide chains from glycosphingolipids. *Glycoconj J.*, 2009, 26, 929-933.

- M. D. Vaughan, K. Johnson, S. DeFrees, X. Tang, R. A. J. Warren, and S. G. Withers, "Glycosynthase-Mediated Synthesis of Glycosphingolipids", *J. Am. Chem. Soc.*, 2006, 128, 6300-6301
- S. M. Hancock, J. R. Rich, M. E. C. Caines, N. C. J. Strynadka and S. G Withers, "Designer enzymes for glycosphingolipid synthesis by directed evolution", *Nat. Chem. Biol.*, 2009, 5, 508-514
- J. R. Rich and S. G. Withers, "A Chemoenzymatic Total Synthesis of the Neurogenic Starfish Ganglioside LLG-3 Using an Engineered and Evolved Synthase", *Angew. Chem. Int. Ed.*, 2012, 51, 8640–8643
- M. E. C. Caines, M. D. Vaughan, C. A. Tarling, S. M. Hancock, R. A. J. Warren, S. G. Withers, N. C. J. Strynadka, "Structural and Mechanistic Analyses of endo-Glycoceramidase II, a Membrane-associated Family 5 Glycosidase in the Apo and GM3 Ganglioside-bound Forms", *J. Biol. Chem.*, 2007, 282, 14300-14308
- M. E. C. Caines, S. M. Hancock, C. A. Tarling, T. M. Wrodnigg, R. V. Stick, A. E. Sttz, A. Vasella, S. G. Withers, and N. C. J. Strynadka, "The Structural Basis of Glycosidase Inhibition by Five-Membered Iminocyclitols: The Clan A Glycoside Hydrolase Endoglycoceramidase as a Model System", *Angew. Chem. Int. Ed.*, 2007, 46, 4474-4476.
- C. Xia, Q. Yao, J. Schümann, E. Rossy, W. Chen, L. Zhu, W. Zhang, G. D. Libero and P. G. Wang, "Synthesis and biological evaluation of α-galactosylceramide (KRN7000) and isoglobotrihexosylceramide (iGb3)", *Bioorg. Med. Chem. Lett.*, 2006, 16, 2195-2199
- J. T. Park, and M. J. Johnson, "A Submicrodetermination of Glucose", J. Biol. Chem., 1949, 181, 149-151

Chapter 5

Further Functionalization of Neoglycolipids

5-1 Introduction

Not only for glycolipids, introduction of specific tags or immobilization to research tools are often more useful to discover bioactive compounds or to understand detailed functions than simply synthesizing candidates^[1-8]. (Fig. 5-1-1)

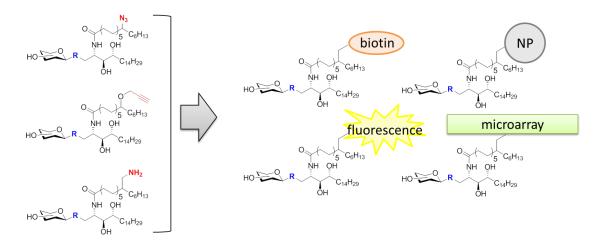


Fig. 5-1-1. Examples of introduction of specific tags and immobilization to research tools

Among many specific reactions, which include glycoblotting reaction of course, I focused on azide-alkyne cyclization^[9, 10] in this chapter. This reaction was reported by R. Huisgen et al^[11]. and enables to easily form triazole structure in the presence of Cu catalyst. Additionally, azide-attached compound may keep the original character because of the smallness of azide group compared with other tags, e.g. fluorescent group or biotin. Beltozzi et al. achieved to detect sialic acid, which had been converted from azide-attached N-acetylmannosamine through biosynthesis, at the terminal of glycoprotein^[12-15]. That is one of the highlights to use azide compound. I tried to apply this approach to analysis of GSLs metabolism^[16, 17]. Understanding the metabolism should be one way to overcome diseases and infections that is why many studies have

been reported. One of the main strategies of glycosphingolipidomics is to use glycolipids labeled by fluorescence^[18, 19]. Fluorescent compounds are incubated with cell culture and detected by combination of HPLC and MS after extraction step. In spite of very high sensitivity, there is a crucial probrem that glycosidic linkage between ceramide and sugar at reducing end opt to be cleaved^[20-23]. That caused the ratio of nonglycosylated ceramide to be big. At that point, my neoglycolipids has possibility to show resistance for hydrolysis as in chapter 4. If the resistance can be indicated, usability of neoglycolipids will increase greatly.

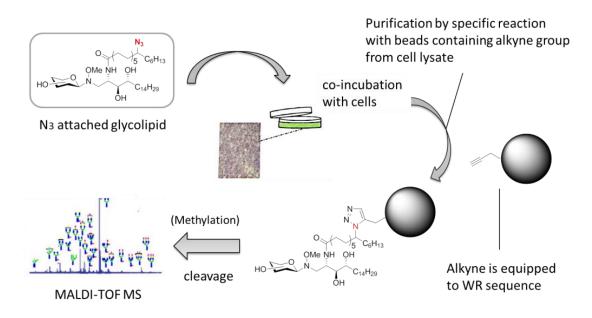


Fig. 5-1-2. Abstract of strategy in this chapter

In chapter 2, several ceramide derivatives were synthesized and the acyl group, which means stearoyl, was attached in the latter step for every derivative. That was because acyl group should be exchanged easily. That has great advantage to introduce azide group to ceramide derivatives briefly.

Abstract of this chapter is shown in Fig.5-1-2. Azide group was introduced to neoglycolipids as a form of acyl chain and the azide-attached compounds were incubated with cell. A functional bead displaying alkyne group was used to extract only neoglycolipids from cell culture and MALDI-TOF MS after cleavage would enable to analyze metabolized glycolipids simultaneously.

5-2 Results & Discussions

5-2-1 Synthesis of azide derivatives and alkyne displayed beads

To conduct analysis of GSLs metabolism with azide-alkyne cyclization, I designed several azide-attached compounds and an alkyne-displayed functional bead. (Fig. 5-2-1) As glycolipids, I focused on LacCer, which is a starting glycolipid to be differentiated, and designed 3 compounds. One of them was natural type which had O-glycosidic linkage between ceramide and glucose residue. The others were neoglycolipids containing oxime linkage and N(-OMe) linkage, respectively. Additionally, simple methyl glucoside was designed as a control. The functional bead was desined general resin for Fmoc solid phase peptide synthesis (SPPS), Rink-Amide-ChemMatrix, and alkyne group was introduced to amine group at the terminal of WRGG sequence. The sequence has been used to amplify sensitivity of MS analysis.

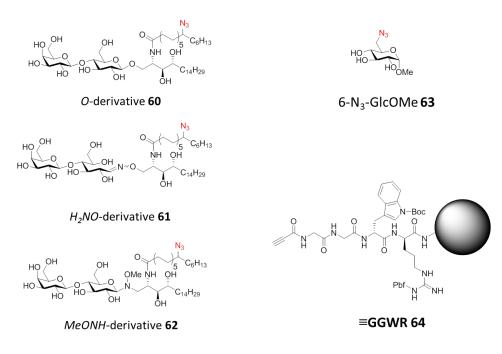
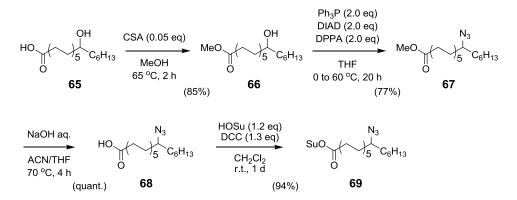


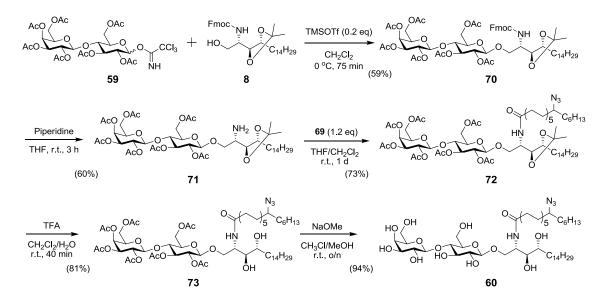
Fig. 5-2-1. Taget compounds

At first, I synthesized activated fatty ester containing azide group (Scheme 5-2-1). As the starting material, racemic 12-hydroxy stearic acid **65** was used and esterified in acidic condition. Mitsunobu reaction enabled azide group to be substituted with hydroxyl group in 1 step using DPPA. After hydrolysis reaction in basic condition, *N*-hydroxysuccinimide was condensed with carboxylic acid using DCC to yield desired activated fatty ester **69**.



Scheme 5-2-1. Synthesis of succinimidyl 12-N₃-octadecanoate 69

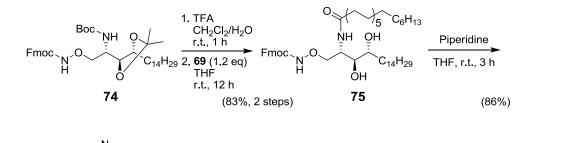
Next, natural type of LacCer containing *O*-glycoside was synthesized (Scheme 5-2-2). Using lactosyl imidate **59** in chapter 4 as donor and monohydroxyllipid **8** in chapter 2 as acceptor, glycosylation was performed and gave β -product **70**. After removal of Fmoc group by piperidine treatment, compound **69** prepared above was used to attach fatty acyl chain containing azide group to amide group of sphingosine. Isopropiliden group was deprotected by TFA treatment and all acetyl groups were removed under basic condition to afford desired LacCer derivative **60**.

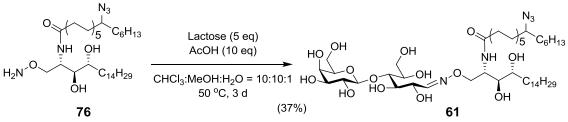


Scheme 5-2-2. Synthesis O-LacCer derivative 60

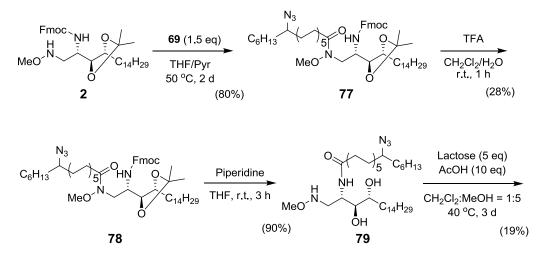
LacCer derivatives using glycoblotting reaction were synthesized as same way in chapter 2 except activated fatty ester (Scheme 5-2-3, 5-2-4). The results indicated that acyl group of ceramide derivatives were exchanged easily as expected.

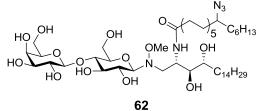
Although the racemic sturucture, 12-N₃-octadecanoyl group, gave a pair of diastereomers after conjugation with sphingosine, the difference was not detected on TLC analysis and NMR spectra in my synthesis.





Scheme 5-2-3. Synthesis H₂NO-LacCer derivative 61

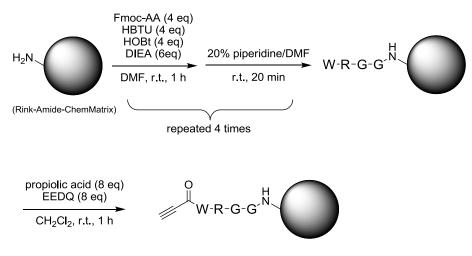




Scheme 5-2-4. Synthesis O-LacCer derivative 62

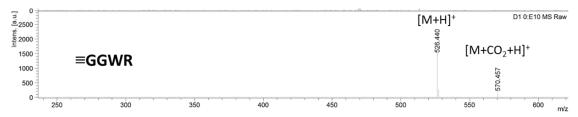
Alkyne-displayed bead was synthesized as below (Scheme 5-2-5).

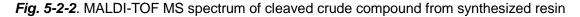
Rink-Amide-ChemMatrix resin was selected and amino acid residues were elongated under general condition of SPPS. At the last step, propiolic acid was condensed with the terminal amino group.



Scheme 5-2-5. Synthesis of alkyne-displayed resin

After synthesis, aliquot of resin was treated by cleavage reagent (TFA : TIS : $H_2O =$ 95 : 2.5 : 2.5) for detection of MALDI-TOF MS. As shown in Fig. 5-2-2, all reactions proceeded completely although the minor peak derived from Boc group attached to Trp residue was also detected.





(DHB, positive mode)

5-2-2 Click reaction on alkyne-displayed bead

Using synthesized compounds, I tried click reaction on alkyne attached resin. First, the reaction was performed in aqueous solution with detergent but LacCer derivatives were not reacted. Because glcose derivative **63** could form triazole structure in contrast, it was obvious the solubility of glycolipids was problematic (data not shown). Then, solvent system was changed to CHCl₃/MeOH/H₂O system and reaction condition was optimized, which enabled LacCer derivatives to react the alkyne group on resin. However, quantitativity was not enough to discuss yet (Fig. 5-2-3). Further optimization of reaction condition or selection resin was required.

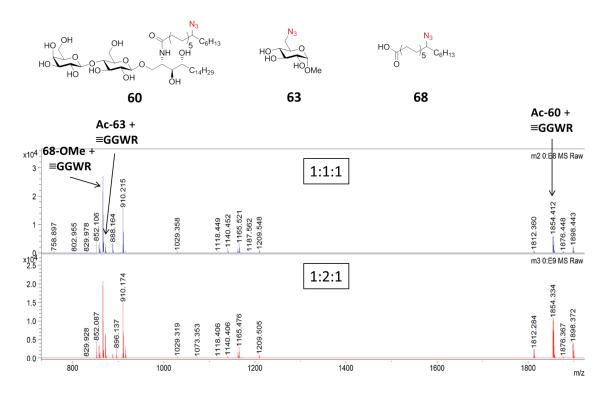


Fig. 5-2-3. MALDI-TOF MS spectra of cleaved crude product from resin after click reaction; Ratios shown in the middle means ratio of concentration of compound **60**, **63**, and **68**

Furthermore, minor peak which was 44 Da bigger than product mass was also detected in a series of experiments (Fig. 5-2-4). Although the peak diminished later, many de-acetylated compounds were detected. This result compelled me to abbreviate acetylation step before cleavage in next experiment.

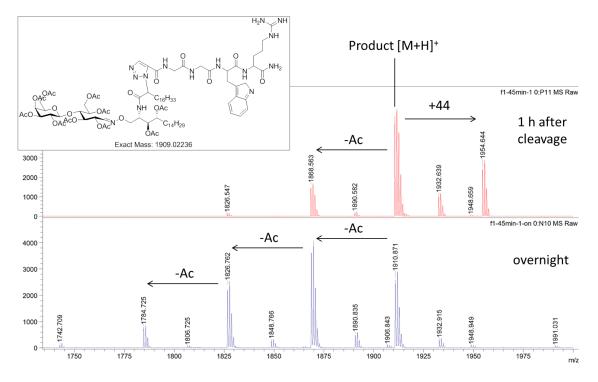


Fig. 5-2-4. Comparison of MALDI-TOF MS spectra after cleavage step

(DHB, positive)

5-2-3 Analysis of GSLs metabolism

Although quantitative analysis had not been developed, I tried to perform glycolipidomics. Synthesized compound **60**, **61**, and **62** were respectively co-incubated with PC-3 cells and compound **63** as a control. After incubation, culture medium was lyophilized. Then, the powder was suspended in CH_2Cl_2 / MeOH solution. Click reaction was performed using clear supernatant liquid under same condition as before except concentration of sodium ascorbate and CuSO₄. After methylation step^[24], MALDI-TOF MS measurement was conducted and the results were shown in Fig. 5-2-5

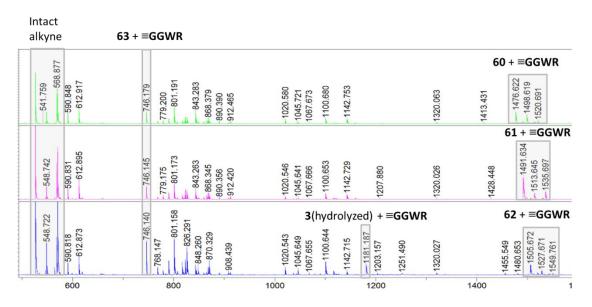


Fig. 5-2-5. MALDI-TOF MS of extracted products (DHB, positive)

Unfortunately, any metabolized product was not detected in each case. To make matters worse, unknown peaks which were certainly derived from resin caused analysis to be difficult and compound **62** was partly hydrolyzed by acid reagent. Those problems will be overcome to choice resin carefully and optimize cleavage condition.

5-3 Conclusion

In this chapter, I tried to conduct further functionalization of neoglycolipids to develop a novel utility value. Azide group was selected as a specific tag and introduced neoglycolipids easily by using fatty acyl group containing azide group. I intended to analyze GSLs metabolism by easy extraction of neoglycolipids through a click reaction. Although metabolized products were not detected in my experiments unfortunately, the strategy of easy extraction itself was indicated to be useful. Metabolized products will be observed to optimize resin, reaction condition for cleavage, and incubation condition.

5-4 Experimental Section

5-4-1 Synthesis of azide derivatives and alkyne displayed beads

Methyl 12-azide-octadecanoate 67

Into a THF solution (80 mL) containing compound **66** (2.57 g, 8.17 mmol), Ph₃P (4.28 g, 16.3 mmol), and DPPA (3.51mL, 16.3 mmol), DIAD (3.21 mL, 16.3 mmol) was dropped at 0 degree. After 20 min, the reaction mixture was stirred at 60 degree for 20 hours. Then, the solution was evaporated under reduced pressure. Purification by flash column chromatography on silica gel (Hexane-EtOAc, 40:1) yielded **67** (2.14 g, 77%) as a colorless oil.;

¹H NMR (500 MHz, CHLOROFORM-*d*) δ ppm 3.65 (s, 3 H; O*Me*) 3.19 - 3.25 (m, 1 H; N₃C*H*) 2.29 (t, *J*=7.57 Hz, 2 H; C*H*₂) 1.65-1.23 (m, 29 H; C*H*₂) 0.89 (t, *J*=6.84 Hz, 3 H; C*H*₃)

12-azide-octadecanoic acid 12-HSA-N₃-OH 68

4 N NaOH aq. (10 mL) was added to a solution of compound **67** (7.63 g, 22.5 mmol) in THF (20 mL) and ACN (20 mL). The reaction mixture was stirred at 70 degree for 4 hours and neutralized with 1 N HCl aq. CHCl₃ was used 2 times to extract desired compound from aqueous solution. The organic phase was dried over MgSO₄ and concentrated under reduced pressure. Purification by flash column chromatography on silica gel (hexane-EtOAc, 10:1~3:1) gave compound **68** as a colorless oil quantitatively.;

¹H NMR (500 MHz, CHLOROFORM-*d*) δ ppm 3.19 - 3.25 (m, 1 H; N₃CH) 2.35 (t, *J*=7.57 Hz, 2 H; CH₂) 1.67-1.24 (m, 29 H; CH₂) 0.89 (t, *J*=6.84 Hz, 3 H; CH₃)

Succinimidyl 12-azide-octadecanoate 12-HSA-N₃-OSu 69

Compound **68** (7.39 g, 22.7 mmol), DCC (6.09 g, 29.5 mmol), and *N*-hydroxysuccinimide (3.14 g, 27.2 mmol) were dissolved in CH_2Cl_2 (100 mL). The reaction mixture was stirred at room temperature for 1 day and diluted with Et_2O . Precipitate was removed by filtration with celite and filtrate was concentrated under reduced pressure. Purification by flash column chromatography on silica gel (hexane-EtOAc, 10:1~3:1) gave compound **69** (9.00 g, 94%) as a white amorphous solid.

¹H NMR (500 MHz, CHLOROFORM-*d*) δ ppm 3.18 - 3.26 (m, 1 H; N₃C*H*) 2.83 (br. s., 4 H; NCOC*H*₂) 2.60 (t, *J*=7.45 Hz, 2 H; C*H*₂) 1.79-1.23 (m, 29 H; C*H*₂) 0.89 (t, *J*=6.59 Hz, 3 H; C*H*₃)

AcLac-O-PS-Fmoc-Isop 70

To a solution of donor **59** (795 mg, 1.02 mmol) and acceptor **8** (393 mg, 679 μ mol) in CH₂Cl₂ (10 mL), TMSOTf (25 μ L, 136 μ mol) was added dropwise at 0 degree. The reaction mixture was stirred at 0 degree for 75 minutes and quenched by sat. NaHCO₃ aq.. CHCl₃ was used 2 times to extract desired compound from aqueous solution. The organic phase was dried over MgSO₄ and concentrated under reduced pressure. Purification by flash column chromatography on silica gel (hexane-EtOAc, 2:1~1.7:1) gave compound **70** (478 mg, 59%) as a white solid

¹H NMR (500 MHz, CHLOROFORM-*d*) δ ppm 7.79 (d, *J*=7.32 Hz, 2 H; Fmoc-*Ar*) 7.56 - 7.62 (m, 2 H; Fmoc-*Ar*) 7.41 (t, *J*=7.32 Hz, 2 H; Fmoc-*Ar*) 7.33 (t, *J*=7.32 Hz, 2 H; Fmoc-*Ar*) 5.35 (d, *J*=3.05 Hz, 1 H; Gal-*H*-4) 5.20 (t, *J*=9.45 Hz, 1 H; Glc-*H*-3) 5.12 (dd, *J*=9.76, 8.23 Hz, 1 H; Gal-*H*-2) 4.97 (dd, *J*=10.37, 3.35 Hz, 1 H; Gal-*H*-3) 4.90 (t, *J*=8.23 Hz, 1 H; Glc-*H*-2) 4.86 (d, *J*=9.15 Hz, 1 H; NH) 4.48 (d, *J*=8.23 Hz, 3 H; Glc-*H*-6a, Gal-*H*-1, Fmoc-CH₂a) 4.40 (m, 2 H; Glc-*H*-1, Fmoc-CH₂b) 4.18 - 4.24 (m, 1 H; Fmoc-CH) 4.05 - 4.17 (m, 5 H; Glc-*H*-6b, Gal-*H*-6ab, PS-*H*-3, PS-*H*-4) 3.97 - 4.04 (m, 1 H; PS-*H*-1a) 3.87 (t, *J*=6.71 Hz, 2 H; Gal-*H*-5, PS-*H*-2) 3.78 (t, *J*=9.45 Hz, 1 H; Glc-*H*-4) 3.53 - 3.61 (m, 2 H; Glc-*H*-5, PS-*H*-1b) 2.16-1.92 (s, 21 H; *Ac*) 1.59-1.15 (m, 26 H) 1.41, 1.32 (s, 3 H; CCH₃) 0.89 (t, *J*=6.86 Hz, 3 H; CH₃)

AcLac-O-PS-NH2-Isop 71

To a THF solution (5 mL) of compound **70** (470 mg, 392 μ mol), piperidine (1 mL) was added and the reaction mixture was stirred at room temperature for 3 hours. The solution was concentrated under reduced pressure and purification by flash column chromatography on silica gel (hexane-EtOAc, 1:1-1:2, EtOAc) gave compound **71** (230 mg, 60%) as a white solid.

¹H NMR (500 MHz, CHLOROFORM-*d*) δ ppm 5.35 (d, *J*=3.39 Hz, 1 H; Gal-*H*-4) 5.20 (t, *J*=9.33 Hz, 1 H; Glc-*H*-3) 5.11 (dd, *J*=10.18, 7.92 Hz, 1 H Gal-*H*-2) 4.90 - 4.98 (m, 2 H; Glc-*H*-2, Gal-*H*-3) 4.47 - 4.53 (m, 3 H; Glc-*H*-1, Glc-*H*-6a, Gal-*H*-1) 4.05 - 4.17 (m, 4 H; Glc-*H*-6b, Gal-*H*-6ab, PS-*H*-4) 3.87 (t, *J*=7.07 Hz, 1 H; Gal-*H*-5) 3.76 - 3.83 (m, 3 H; Glc-*H*-4, PS-*H*-1a, PS-*H*-3) 3.67 - 3.72 (m, 1 H; PS-*H*-1) 3.58 - 3.63 (m, 1 H; Glc-*H*-5) 2.96 - 3.02 (m, 1 H; PS-*H*-2) 2.15-1.97 (s, 21 H, *Ac*) 1.60-1.22 (m, 26 H; C*H*₂) 1.40 , 1.30 (s, 6 H; CC*H*₃) 0.88 (t, *J*=6.93 Hz, 3 H; C*H*₃)

AcLac-O-PS-HSAN₃-Isop 72

Compound **71** (230 mg, 236 μ mol) and succinimidyl ester **69** (199 mg, 471 μ mol) were dissolved in CH₂Cl₂ (10 mL) and the reaction mixture was stirred at room temperature for 2 days. The solution was concentrated under reduced pressure and purification by flash column chromatography on silica gel (hexane-EtOAc, 2:1-3:2-1:1) gave compound **72** (230 mg, 73%) as a white solid

¹H NMR (500 MHz, CHLOROFORM-*d*) δ ppm 5.55 (d, *J*=9.04 Hz, 1 H; N*H*) 5.35 (d, *J*=3.42 Hz, 1 H; Gal-*H*-4) 5.19 (t, *J*=9.28 Hz, 1 H; Glc-*H*-3) 5.08 - 5.13 (m, 1 H; Gal-*H*-2) 4.96 (dd, *J*=10.01, 3.18 Hz, 1 H; Gal-*H*-3) 4.88 (t, *J*=8.30 Hz, 1 H; Glc-*H*-2) 4.45 - 4.51 (m, 3 H; Glc-*H*-1, Glc-*H*-6a, Gal-*H*-1) 4.04 - 4.19 (m, 6 H; Glc-*H*-6b, Gal-*H*-6ab, PS-*H*-2, PS-*H*-3, PS-*H*-4) 4.00 (dd, *J*=10.01, 3.91 Hz, 1 H; PS-*H*-1a) 3.87 (t, *J*=6.84 Hz, 1 H; Gal-*H*-5) 3.79 (t, *J*=9.53 Hz, 1 H; Glc-*H*-4) 3.57 - 3.66 (m, 2 H; Glc-*H*-5, PS-*H*-1b) 3.19 - 3.26 (m, 1 H; N₃C*H*) 2.15-1.97 (s, 21 H; *Ac*) 2.18-1.21 (m, 57 H; C*H*₂) 1.41, 1.31 (s, 6 H; CC*H*₃) 0.86 - 0.92 (m, 6 H; C*H*₃)

F1 (ppm) 101.03 (Glc-*C*-1, Gal-*C*-1) 77.97 (PS-*C*-3) 76.46 (PS-*C*-4) 75.96 (Glc-*C*-4) 72.86 (Glc-*C*-5) 72.67 (Glc-*C*-3) 72.04 (Glc-*C*-2) 71.06 (Gal-*C*-5) 71.00 (Gal-*C*-3) 69.13 (Gal-*C*-2) 69.06 (PS-*C*-1) 66.54 (Gal-*C*-4) 63.00 (N₃*C*H) 61.66 (Glc-*C*-6) 61.15 (Gal-*C*-6) 47.97 (PS-*C*-2) 36.68-14.16 (Alkyl chain) 20.99 (*Ac*)

AcLac-O-PS-HSAN₃73

To a suspension of compound **72** (185 mg, 144 μ mol) in CH₂Cl₂/H₂O (4:1, 500 μ L) was added TFA (8 mL) and the mixture was stirred at room temperature for 40 minutes. The reaction mixture was diluted with EtOAc and washed with Brine, sat. NaHCO₃ aq. (2

times) and brine. The organic phase was dried over $MgSO_4$, and concentrated under reduced pressure. Purification of the crude product by flash column chromatography on silica gel (hexane-EtOAc, 3:2~1:1~2:3) yielded **73** (145 mg, 81%) as a white amorphous solid;

¹H NMR (500 MHz, CHLOROFORM-*d*) δ ppm 6.18 (d, *J*=8.30 Hz, 1 H; N*H*) 5.37 (d, *J*=2.44 Hz, 1 H; Gal-*H*-4) 5.22 (t, *J*=9.28 Hz, 1 H; Glc-*H*-3) 5.13 (t, *J*=9.28 Hz, 1 H; Gal-*H*-2) 5.00 (dd, *J*=10.50, 2.93 Hz, 1 H; Gal-*H*-3) 4.88 (t, *J*=8.55 Hz, 1 H; Glc-*H*-2) 4.55 - 4.60 (m, 1 H; Glc-*H*-6a) 4.49 - 4.55 (m, 2 H; Glc-*H*-1, Gal-*H*-1) 4.18 - 4.24 (m, 1 H; PS-*H*-2) 4.08 - 4.18 (m, 4 H; Glc-*H*-6b, Gal-*H*-6ab, PS-*H*-1a) 3.92 (t, *J*=6.59 Hz, 1 H; Gal-*H*-5) 3.77 - 3.84 (m, 2 H; Glc-*H*-4, PS-*H*-1b) 3.63 - 3.69 (m, 1 H; Glc-*H*-5) 3.56 - 3.62 (m, 1 H; PS-*H*-4) 3.50 - 3.56 (m, 1 H; PS-*H*-3) 3.21 - 3.31 (m, 2 H; N₃CH, OH) 2.54 (br. s, 1 H; OH) 2.22-1.23 (m, 57 H; CH₂) 2.17-1.99 (s, 21 H; *Ac*) 0.87 - 0.94 (m, 6 H; CH₃)

F1 (ppm) 100.88 (Glc-*C*-1, Gal-*C*-1) 76.07 (Glc-*C*-4) 74.98 (PS-*C*-3) 73.01 (Glc-*C*-5) 72.72 (PS-*C*-4) 72.31 (Glc-*C*-3) 71.73 (Glc-*C*-2) 70.86 (Gal-*C*-3, Gal-*C*-5) 69.41 (PS-*C*-1) 69.11 (Gal-*C*-2) 66.72 (Gal-*C*-4) 63.08 (N₃*C*H) 61.63 (Glc-*C*-6) 60.54 (Gal-*C*-6) 50.29 (PS-*C*-2) 36.75-14.15 (Alkyl chain) 20.69 (*Ac*)

Lac-O-PS-HSAN₃ 60

Compound **73** (145 mg, 117 μ mol) and sodium methoxide (3 mg, 56 μ mol) were dissolved in MeOH (5 mL) and CHCl₃ (1 mL) and the reaction mixture was stirred at room temperature for 1 day. The solution was concentrated by air-drying and the desired product was precipitated in MeOH. Filtration gave compound **60** (104 mg, 94%) as a white solid.

¹H NMR (500 MHz, pyridine) δ ppm 8.60 (d, *J*=8.77 Hz, 1 H; N*H*) 7.63 (br. s, 1 H; Glc-O*H*-2) 7.51 (d, *J*=4.39 Hz, 1 H; Gal-O*H*-2) 6.91 - 6.97 (m, 1 H; Gal-O*H*-3) 6.64 - 6.69 (m, 1 H; Gal-O*H*-6) 6.57 (d, *J*=4.39 Hz, 1 H; Gal-O*H*-4) 6.54 (d, *J*=6.26 Hz, 1 H; PS-O*H*-3) 6.44 - 6.50 (m, 1 H; Glc-O*H*-6) 6.22 (s, 1 H; Glc-O*H*-3) 5.95 (d, *J*=6.89 Hz, 1 H; PS-O*H*-4) 5.16 - 5.20 (m, 1 H; PS-*H*-2) 5.07 (d, *J*=7.83 Hz, 1 H; Gal-*H*-1) 4.90 (d, *J*=7.83 Hz, 1 H; Glc-*H*-1) 4.79 (dd, *J*=10.34, 5.64 Hz, 1 H; PS-*H*-1a) 4.43 - 4.55 (m, 5 H; Glc-*H*-6ab, Gal-*H*-2, Gal-*H*-4, Gal-*H*-6a) 4.36 - 4.42 (m, 3 H; Gal-*H*-6b, PS-*H*-1b, PS-*H*-3) 4.21 - 4.29 (m, 3 H; Glc-*H*-3, Glc-*H*-4, PS-*H*-4) 4.11 - 4.17 (m, 2 H; Gal-*H*-3, Gal-*H*-5) 4.01 - 4.07 (m, 1 H; Glc-*H*-2) 3.81 - 3.86 (m, 1 H; Glc-*H*-5) 3.29 - 3.36 (m, 1 H; N₃C*H*) 2.45-1.18 (m, 57 H; C*H*₂) 0.84 - 0.90 (m, 6 H; C*H*₃)

FmocNHO-PS-HSAN₃75

To a suspension of compound **25** (1.53 g, 2.2 mmol) in CH₂Cl₂/H₂O (4:1, 2.5 mL), TFA (20 mL) was added and the mixture was stirred at room temperature for 1 hour. The reaction mixture was diluted with CHCl₃ and washed with sat. NaHCO₃ aq. 2 times. The organic phase was dried over Na₂SO₄ and concentrated under reduced pressure. The crude product was directly used for succeeding reaction without further purification. The product and compound **69** (1.12 g, 2.64 mmol) were suspended in THF (40 mL) and the mixture was stirred at room temperature for 12 hours. The reaction solution was evaporated under reduced pressure completely. Purification of the crude product by flash column chromatography on silica gel (Toluene-EtOAc, 6:1-2:1) yielded **75** (1.57 g, 83%) as a white amorphous solid; ¹H NMR (500 MHz, CHLOROFORM-*d*) δ ppm 8.18 - 8.26 (m, 1 H; N*HO*) 7.76 (m, 2 H; Fmoc-*Ar*) 7.56 (m, 2 H; Fmoc-*Ar*) 7.41 (m, 2 H; Fmoc-*Ar*) 7.28 - 7.36 (m, 2 H; Fmoc-*Ar*) 6.55 - 6.72 (m, 1 H; N*H*) 4.43 - 4.52 (m, 2 H) 4.21 - 4.30 (m, 2 H) 4.14 - 4.20 (m, 1 H) 4.05 - 4.11 (m, 1 H) 3.65 - 3.70 (m, 1 H) 3.58 - 3.65 (m, 1 H) 3.46 - 3.52 (m, 1 H) 3.17 - 3.26 (m, 1 H; N₃C*H*) 2.25-1.18 (m, 57 H) 0.84 - 0.93 (m, 6 H)

H₂NO-PS-HSAN₃76

Compound **75** (1.23 g, 1.43 mmol) was dissolved in THF (10 mL) and piperidine (2 mL) was added to the solution at room temperature. After 6 h, solvent was removed by evaporation and purification of the crude product by flash column chromatography on silica gel (Toluene-EtOAc, 2:1-1:2, EtOAc) yielded **76** (783 mg, 86%) as a white amorphous solid;

¹H NMR (500 MHz, CHLOROFORM-*d*) δ ppm 6.22 (d, *J*=8.30 Hz, 1 H; N*H*) 4.28 - 4.33 (m, 1 H) 3.97 - 4.01 (m, 1 H) 3.85 - 3.90 (m, 1 H) 3.59 - 3.64 (m, 1 H) 3.54 - 3.58 (m, 1 H) 3.19 - 3.25 (m, 1 H; N₃C*H*) 2.23-1.21 (m, 57 H) 0.86 - 0.91 (m, 6 H)

Lac=NO-PS-12-N₃-octadecane 61

Compound **76** (60 mg, 100 μ mol) and lactose (180 mg, 500 μ mol) was suspended in CHCl₃/MeOH/H₂O/AcOH (25:25:8:1, 5 mL). The reaction mixture was stirred vigorously at 50 degree. After 1 day, the solution was transferred into flask. Solvent was removed to about half volume and silica gel was added to a reaction solution and the mixture was evapolated completely. Purification of the crude product by flash column chromatography on silica gel (CHCl₃-MeOH, 10:1-4:1) afforded product **61** (35 mg, 37%) as a white solid.

MeON-HSAN₃-PS-Fmoc-Isop 77

Compound **2** (1.41 g, 2.32 mmol) and compound **69** (1.47 g, 3.48 mmol) were dissolved in CH_2Cl_2 (30 mL) and the reaction mixture was stirred at room temperature for 3 days. The solution was concentrated under reduced pressure and purification of the crude product by flash column chromatography on silica gel (Hexane-EtOAc, 50:1-9:1, EtOAc) yielded **77** (1.70 g, 80%) as a white solid;

¹H NMR (500 MHz, CHLOROFORM-*d*) δ ppm 7.74 (d, *J*=7.57 Hz, 2 H; Fmoc-*Ar*) 7.58 (t, *J*=6.11 Hz, 2 H; Fmoc-*Ar*) 7.38 (t, *J*=7.45 Hz, 2 H; Fmoc-*Ar*) 7.30 (t, *J*=7.57 Hz, 2 H; Fmoc-*Ar*) 5.26 (d, *J*=7.33 Hz, 1 H; N*H*) 4.25 - 4.32 (m, 2 H; Fmoc-*CH*₂) 4.10 - 4.22 (m, 5 H; *H*-1a, *H*-2, *H*-3, *H*-4, Fmoc-*CH*) 3.68 (s, 3 H; O*Me*) 3.55 - 3.64 (m, 1 H; *H*-1b) 3.20 (m, 1 H; N₃C*H*) 2.80 (br. s, 1 H; O*H*) 2.48-1.15 (m, 57 H; C*H*₂) 1.48, 1.35 (s, 6 H; CC*H*₃) 0.86 - 0.91 (m, 6 H; C*H*₃)

MeONH-PS -HSAN₃ 79

Compound **78** (477 mg, 544 µmol) was dissolved in THF (6 mL) and piperidine (1.5 mL) was added to the solution at room temperature. After 3 h, solvent was removed by air-drying completely and precipitation was performed by MeOH. Precipetated white solid was separated by filtration and purification of the filtercake by flash column chromatography on silica gel (Toluene-EtOAc, 2:1-1:2, EtOAc) yielded **79** (322 mg, 90%) as a white amorphous solid;

¹H NMR (500 MHz, CHLOROFORM-*d*) δ ppm 6.23 (d, *J*=8.06 Hz, 1 H; N*H*) 4.18 - 4.26 (m, 1 H; *H*-2) 3.64 - 3.79 (m, 1 H; O*H*) 3.54 - 3.63 (m, 5 H; *H*-3, *H*-4, O*Me*) 3.18 - 3.29 (m, 3 H; *H*-1ab, N₃C*H*) 2.24-1.19 (m, 57 H; C*H*₂) 0.85 - 0.93 (m, 6 H; C*H*₃)

Lac-MeON-PS-HAS

Compound **79** (60 mg, 100 μ mol) and lactose (180 mg, 500 μ mol) was suspended in CHCl₃/MeOH/H₂O/AcOH (25:25:8:1, 5 mL). The reaction mixture was stirred vigorously at 50 degree. After 1 day, the solution was transferred into flask. Solvent was removed to about half volume and silica gel was added to a reaction solution and the mixture was evapolated completely. Purification of the crude product by flash column chromatography on silica gel (CHCl₃-MeOH, 10:1-4:1) afforded product **62** (19 mg, 19%) as a white solid.

6-N₃-Glc-OMe 63

Into a THF solution (20 mL) containing compound **6-HO-AcGlc-OMe** (640 mg, 2.00 mmol), Ph₃P (629 mg, 2.40 mmol), and DPPA (517 μ L, 16.3 mmol), DIAD (473 μ L, 2.40 mmol) was dropped at 0 degree. After 20 minutes, the reaction mixture was stirred at room temperature for 32 hours. Then, the solution was evaporated under reduced pressure. Purification by flash column chromatography on silica gel (Hexane-EtOAc, 5:1-2:1) yielded azide compound with some mixtures but the product was used for next reaction without further purification. The product and sodium methoxide were dissolved in MeOH and the reaction mixture was stirred at room temperature for 2 hours. DowX was used for neutralization and removed by filtration. The filtrate was concentrated under reduced pressure and purification of the crude product by flash column chromatography on silica gel (CHCl₃-MeOH, 30:1-10:1, EtOAc) yielded **63** (310 mg, 71%, 2 steps) as a white amorphous solid

¹H NMR (500 MHz, CHLOROFORM-*d*) δ ppm 4.79 (d, *J*=2.69 Hz, 1 H) 4.53 (br. s, 1 H) 3.99 (br. s, 1 H) 3.68 - 3.76 (m, 2 H) 3.51 - 3.59 (m, 3 H) 3.41 - 3.50 (m, 5 H)

HCC-WRGG-Resine 64

Rink-Amide-ChemMatrix (100 mg, 52 µmol) resin was swelled in filtertube before reaction and washed with DMF. Into the tube, a cocktail of reagents in DMF (574 µL) containing HOBt (208 µmol), HBTU (208 µmol), DIEA (312 µmol), and Fmoc-AA residue (Fmoc-Gly-OH, Fmoc-Trp(Boc)-OH, and Fmoc-Arg(Pbf)-OH, respectively) was added and the reaction mixture was shaken at room temperature for 1 hour. After washing with DMF, 20% piperidine solution in DMF was added and the solution was shaken at room temperature for 20 minutes. These steps were repeated 4 times and then EEDQ (816 µmol) and propiolic acid (816 µmol) in CH₂Cl₂ were added and the mixture was shaken at room temperature for 1 hour. After reaction completed, the resin was washed with DMF and CH₂Cl₂ and stocked in CH₂Cl₂ at room temperature. Characterization of the product was confirmed by MALDI-TOF MS measurement by using aliquot of the resin which was treated by cleavage solution composed of TFA : H₂O: TIS= 95 : 2.5 : 2.5.

5-4-2 Azide-alkyne reaction

In filter tube, the prepared resin (10 mg, 5.2 μ mol) was swelled in CH₂Cl₂ before reaction. After CH₂Cl₂ was flushed, reaction solution (CHCl₃/MeOH/H₂O, 1:3:1, 400 μ L) containing compound **60** (280 nmol), compound **63** (280 nmol), compound **68** (280 nmol), sodium ascorbate (200 nmol), Copper sulfate pentahydrate (67 nmol), and TBTA (100 nmol). The mixture was shaken at 50 degree for 2 hours and the resin was washed with MeOH, $CH_2Cl_2/MeOH$ (1:1), and CH_2Cl_2 . Then, Ac_2O/Pyr solution was added and shaken at 40 degree for 1.5 hours. After washing with CH_2Cl_2 , CH_2Cl_2 solution of TMS-diazemethane was added and the reaction mixture was shaken at room temperature for 24 hours. After washing with CH_2Cl_2 , Characterization of the product was confirmed by MALDI-TOF MS measurement by using aliquot of the resin which was treated by cleavage solution composed of TFA : H_2O : TIS= 95 : 2.5 : 2.5.

5-4-3 Extraction of azide derivatives from culture medium

Cell Culture

PC-3 cells were grown in RPMI-1640 (Sigma) supplemented with 10% (v/v) heat-inactivated fetal bovine serum at 37° C

Co-incubation of prepared compounds with the Cells

To the medium, DMSO solution of compound 60, 61, or 62 were added respective ly with compound 63 (Final conc. : 250 μ M). Cells were incubated on 60-mm culture dishes for 48 h and scraped with tripsin-EDTA.

Extraction by click reaction

In filter tube, the prepared resin (10 mg, 5.2 μ mol) was swelled in CH₂Cl₂ before reaction. After CH₂Cl₂ was flushed, reaction solution (CHCl₃/MeOH/H₂O, 1:3:1, 400 μ L) containing 20% of total lyophilized product, sodium ascorbate (1000 nmol), Copper sulfate pentahydrate (1000 nmol), and TBTA (100 nmol). The mixture was shaken at 50 degree for 2 hours and the resin was washed with MeOH, CH₂Cl₂/MeOH (1:1), and CH₂Cl₂. Then, Ac2O/Pyr solution was added and shaken at 40 degree for 1.5 hours. After washing with CH_2Cl_2 , CH_2Cl_2 solution of TMS-diazemethane was added and the reaction mixture was shaken at room temperature for 24 hours. After washing with CH_2Cl_2 , Characterization of the product was confirmed by MALDI-TOF MS measurement by using aliquot of the resin which was treated by cleavage solution composed of TFA : H_2O : TIS= 95 : 2.5 : 2.5.

5-5 References

- H. Hinou, R. Miyoshi, Y. Takasu, H. Kai, M. Kurogochi, S. Arioka, X.-D. Gao, N. Miura, N. Fujitani, S. Omoto, T. Yoshinaga, T. Fujiwara, T. Noshi, H. Togame, H. Takemoto, and S.-I. Nishimura, "A Strategy for Neuraminidase Inhibitors Using Mechanism-Based Labeling Information", *Chem. Asian J.*, **2011**, 6, 1048-1056
- H. Kai, H. Hinou, and S.-I. Nishimura, "Aglycone-focused randomization of 2-difluoromethylphenyl-type sialoside suicide substrates for neuraminidases", *Bioorg. Med. Chem.*, 2012, 20, 2739-2746
- J. M. H. Cheng, S. H. Chee, D. A. Knight, H. A.-Orbea, I. F. Hermans, M. S. M. Timmer, B. L. Stocker, "An improved synthesis of dansylated α-galactosylceramide and its use as a fluorescent probe for the monitoring of glycolipid uptake by cells", *Carbohydr. Res.*, 2011, 346, 914-926
- T. Sakai, O. V. Naidenko, H. Iijima, M. Kronenberg, and Y. Koezuka, "Syntheses of Biotinylated α-Galactosylceramides and Their Effects on the Immune System and CD1 Molecules", J. Med. Chem., 1999, 42, 1836-1841
- M. B. Thygesen, J. Sauer, K. J. Jensen, "Chemoselective Capture of Glycans for Analysis on Gold Nanoparticles: Carbohydrate Oxime Tautomers Provide Functional Recognition by Proteins", *Chem. Eur. J.*, 2009, 15, 1649-1660
- P.-H. Liang, M. Imamura, X. Li, D. Wu, M. Fujio, R. T. Guy, B.-C. Wu, M. Tsuji, and C.-H. Wong, "Quantitative Microarray Analysis of Intact Glycolipid-CD1d Interaction and Correlation with Cell-Based Cytokine Production", *J. Am. Chem. Soc.*, 2008, 130, 12348-12354

- T. Ohyanagi, N. Nagahori, K. Shimawaki, H. Hinou, T. Yamashita, A. Sasaki, T. Jin, T. Iwanaga, M. Kinjo, and S-I. Nishimura, "Importance of sialic acid residues illuminaterd by live animal imaging using phosphorylcholine self-assembled monolayers-coated quantum dots", *J. Am. Chem. Soc.*, 2011, 133, 12507-12517
- H. Hoshi, K. Shimawaki, Y. Takegawa, T. Ohyanagi, M. Amano, H. Hinou, and S-I. Nishimura, "Molecular shuttle between extracellular and cytoplasmic space allows for monitoring of GAG biosynthesis in human articular chondrocytes", *Biochim. Biophys. Acta*, 2012, 1820, 1391-1398
- H. C. Kolb, M. G. Finn & K. B. Sharpless, "Click Chemistry: Diverse Chemical Function from a Few Good Reactions" *Angew. Chem. Int. Ed.*, 2001, 40, 2004-2021
- V. V. Rostovtsev, L. G. Green, V. V. Fokin, K. B. Sharpless "A Stepwise Huisgen Cycloaddition Process: Copper(I)-Catalyzed Regioselective Ligation of Azides and Terminal Alkynes.", *Angew. Chem. Int. Ed.*, 2002, 114, 2708-2711
- R. Huisgen, G. Szeimies and L. Moebius, "1,3-Dipolar Cycloadditions. Past and Future", Angew. Chem. Int. Ed., 1963, 2, 565-598
- E. Saxon, S. J. Luchansky, H. C. Hang, C. Yu, S. C. Lee, and C. R. Bertozzi "Investigating Cellular Metabolism of Synthetic Azidosugars with the Staudinger Ligation", J. Am. Chem. Soc., 2002, 124, 14893-14902
- J. M. Baskin, J. A. Prescher, S. T. Laughlin, N. J. Agard, P. V. Chang, I. A. Miller, A. Lo, J. A. Codelli, and C. R. Bertozzi, "Copper-free click chemistry for dynamic in vivo imaging", *Proc. Natl. Acad. Sci. USA*, 2007, 104, 16793-16797
- 14. S. T. Laughlin and C. R. Bertozzi, "Metabolic labeling of glycans with azido sugars and subsequent glycan-profiling and visualization via Staudinger ligation", *Nature Protocols*, 2007, 2, 2930-2944

- 15. S. S. V. Berkel, M. B. van Eldijk, and J. C. M. van Hest, "Staudinger ligation as a method for bioconjugation", *Angew. Chem. Int. Ed.*, **2011**, 50, 8806-8827
- 16. A. H. Merrill, "Sphingolipid and glycosphingolipid metabolic pathways in the era of sphingolipidomics" *Chem. Rev.*, 2011, 111, 6387-6422
- 17. H. Farwanah and T. Kolter, "Lipidomics of Glycosphingolipids", *Metabolites*, 2012, 2, 134-164
- 18. A. Ohtak, S. Daikoku, K. Suzuki, Y. Ito, and O. Kanie, "Analysis of the Cellular Dynamics of Fluorescently Tagged Glycosphingolipids by Using a Nanoliquid Chromatography–Tandem Mass Spectrometry Platform", Anal. Chem., 2013, 85, 8475-8482
- 19. R. B. Keithley, A. S. Rosenthal, D. C. Essaka, H. Tanaka, Y. Yoshimura, M. M. Palcic, O. Hindsgaul and N. J. Dovichi "Capillary electrophoresis with three-colorfluorescence detection for the analysis of glycosphingolipid metabolism", *Analyst*, **2013**, 138, 164-170
- 20. C. D. Whitmore, U. Olsson, E. A. Larsson, O. Hindsgaul, M. M. Palcic, and N. J. Dovichi, "Yoctomole analysis of ganglioside metabolism in PC12 cellular homogenates", *Electrophoresis*, **2007**, 28, 3100-3104
- C. D. Whitmore, O. Hindsgaul, M. M. Palcic, R. L. Schnaar, and N. J. Dovichi, "Metabolic Cytometry. Glycosphingolipid Metabolism in Single Cells", *Anal. Chem.*, 2007, 79, 5139-5142
- 22. D. C. Essaka, J.White, P. Rathod, C. D. Whitmore, O. Hindsgaul, M. M. Palcic, and N. J. Dovichi "Monitoring the Uptake of Glycosphingolipids In Plasmodium falciparum-Infected Erythrocytes Using Both Fluorescence Microscopy and

Capillary Electrophoresis with Laser-Induced Fluorescence Detection", *Anal. Chem.*, **2010**, 82, 9955-9958

- 23. D. C. Essaka, J. Prendergast, R. B. Keithley, M. M. Palcic, O. Hindsgaul, R. L. Schnaar, and N. J. Dovichi. "Metabolic Cytometry: Capillary Electrophoresis with Two-Color Fluorescence Detection for the Simultaneous Study of Two Glycosphingolipid Metabolic Pathways in Single Primary Neurons", *Anal. Chem.*, 2012, 84, 2799-2804
- 24. Y. Miura, Y. Shinohara, J.-i. Furukawa, N. Nagahori, and S.-I. Nishimura, "Rapid and Simple Solid-Phase Esterification of Sialic Acid Residues for Quantitative Glycomics by Mass Spectrometry", *Chem. Eur. J.*, 2007, 13, 4797-4804

Chapter 6

Concluding Remarks

Glycosylation is one of the most important posttranslational modifications and performed by template-independent manner providing diversity structure and function into glycoconjugates such as glycoprotein or glycolipid. To elucidate the functions of glycoconjugates deeply, preparation of natural/mimetic compound is important task but the innovative method to synthesize glycosphingolipids has not been developed up to date. In this study, I demonstrated a novel synthetic strategy based on glycoblotting to construct neoglycolipid library.

In chapter 2, we synthesized 6 ceramide derivatives which were expected to imitate natural glycosphingolipids after glycosylation. In the synthesis of methoxyamino derivative **1**, we have developed an efficient protocol using a specific $N\beta \rightarrow N\alpha$ acyl migration to allow for the selective modification at $N\alpha$ position of the methoxyamino-functionalized phytosphingosine derivatives This discovery will contribute to construct synthetic strategy of methoxyamino derivative.

In chapter 3, we employ glycoblotting reaction to construct neoglycolipid library. In case of methoxyamino derivative, unfortunate side reaction was occurred but optimization of reaction condition enabled to obtain β -product as sole product and construct library. Other ceramide derivatives were able to be conjugated with free sugars, which verified the usability of glycoblotting-based method.

In chapter 4, we evaluate the function of neoglycoside to imitate natural glycoceramide using rEGCase II. This enzyme could cleave *O*-glycosidic linkage but not non-natural *N*-glycoside bond. It showed that neoglycolipid was not a substrate of the enzyme. However, when the two compounds, *O*-LacCer and *N*-LacCer, were mixed, the velocity of hydrolysis was apparently reduced compared with only *O*-LacCer. It was significantly important because it indicated that *N*-LacCer could be recognized by the enzyme like natural glycoceramide. This result enhanced the effectiveness of glycoblotting method for preparation of glycolipids.

In chapter 5, azide group could be attached to functional ceramide derivatives. After co-incubation with cell, ceramide derivatives were easily recovered by specific azide-alkyne cyclization.

This work must accelerate glycolipids/neoglycolipids study.

Acknowledgements

The present dissertation is described author's study from 2009-2014 at Graduate School of Life Science, Hokkaido University.

First of all, I would like to express my sincere gratitude to Professor Shin-Ichiro Nishimura, for helpful suggestions, discussion, and encouragements. This work would not have been successful without him.

I would like to express thanks to Professors Kenji Monde and Nobuyuki Tamaoki of Hokkaido University for reading and suggestions on the dissertation manuscript.

I am deeply indebted to Assistant Prof. Hiroshi Hinou and Dr. Kentaro Naruchi in Hokkaido University for helpful suggestions and comments for this work.

I greatly appreciate kindness of Ms. Yukina Umemoto to perform cell culture and co-incubation with my compounds.

And, I really want to thank to my colleagues in Nishimura's lab. for daily supports and encouragements.

Finally, I express my sincere gratitude to my grandparents, sister, and especially parents, Tetsuya Ishida and Yuko Ishida, for their moral supports and warm encouragements.

154

May your soul rest in peace, Yuya Sakai, Seiji Handa, and Cherry.

I am very sorry that I can't give you this thesis and I can't tell you my feelings face to face.

I have remembered your continuous efforts and bright smiles which have always encouraged me when I faced difficult situations.

I would have not achieved this work without you.

I can never believe the cruel losses.

I can never thank you enough for all the things and kindness you gave. I can never forget great time we stayed together.

Let me offer my sincere condolences and please take a long repose.

2014 Sapporo Junya Ishida