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1 **Brown carbon in atmospheric outflow from the Indo-Gangetic Plain:**  
2 **Mass absorption efficiency and temporal variability**

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36 **Abstract**

37 The simultaneous measurements of brown carbon (BrC) and elemental carbon (EC) are made  
38 in ambient aerosols (PM<sub>2.5</sub>), collected from a site in north-east India during November'09-  
39 March'10, representing the atmospheric outflow from the Indo-Gangetic Plain (IGP) to the  
40 Bay of Bengal (BoB). The absorption coefficient of BrC ( $b_{abs}$ ), assessed from water-soluble  
41 organic carbon (WSOC) at 365 nm, varies from 2 to 21 Mm<sup>-1</sup> and exhibits significant linear  
42 relationship ( $P < 0.05$ ) with WSOC concentration (3 – 29  $\mu\text{g m}^{-3}$ ). The angstrom exponent ( $\alpha$ :  
43  $8.3 \pm 2.6$ , where  $b_{abs} \approx \lambda^{-\alpha}$ ) is consistent with that reported for humic-like substances  
44 (HULIS) from biomass burning emissions (BBE). The impact of BBE is also discernible  
45 from mass ratios of nss-K<sup>+</sup>/EC (0.2 – 1.4) and OC/EC (3.4 – 11.5). The mass fraction of  
46 WSOC (10 – 23 %) in PM<sub>2.5</sub> and mass absorption efficiency of BrC ( $\sigma_{abs-BrC}$ : 0.5 – 1.2 m<sup>2</sup> g<sup>-1</sup>)  
47 bring to focus the significance of brown carbon in atmospheric radiative forcing due to  
48 anthropogenic aerosols over the Indo-Gangetic Plain.

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50 **Words: 161**

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## 52 1. Introduction

53 The light absorbing species of atmospheric particulate matter are gaining considerable  
54 interest in recent years owing to their significant role in regional as well as global climate  
55 change [Fuzzi *et al.*, 2006]. Although several studies have evaluated their impact on the  
56 atmospheric environment, uncertainties associated with the regional scenario are still large  
57 and demand further detailed assessment. One of the possible sources of uncertainty could be  
58 attributed to poor characterization of organic aerosols in the atmospheric particulate matter  
59 [Huebert and Charlson, 2000]. In this context, detailed information on sources, size-  
60 distribution and compositional changes during transport of carbonaceous aerosols is essential  
61 for assessing their atmospheric radiative forcing.

62 Among the carbonaceous species, two distinct forms of carbon [elemental or black  
63 carbon (EC or BC) and brown carbon (BrC)] are of particular interest due to their light  
64 absorbing properties. The EC absorbs solar radiation in the visible region [Bond, 2001; Bond  
65 *et al.*, 2013], whereas BrC shows prominent absorption in the near UV-region [Alexander *et al.*,  
66 2008; Andreae and Gelencsér, 2006; Hecobian *et al.*, 2010; Lack *et al.*, 2012; Liu *et al.*,  
67 2014; Lukács *et al.*, 2007; Yang *et al.*, 2009]. However, real time data on absorption  
68 properties of BrC are rather limited. The omnipresence of BrC in rural, urban and remote  
69 environments has been emphasized by Graber and Rudich, [2006] suggesting the need for its  
70 adequate representation in climate model simulations.

71 The presence of brown carbon is documented based on the absorption spectra of  
72 aqueous extracts of ambient aerosols [Havers *et al.*, 1998; Kirchstetter *et al.*, 2004; Zhang *et al.*,  
73 2013]. Furthermore, its abundance has been studied using aerosol light absorption  
74 measurements near to the specific combustion sources [Bond, 2001]. A significant overlap in  
75 the spectral properties of BrC (assessed from water-soluble organic carbon, WSOC) and  
76 humic-like substances (HULIS), derived from the biomass burning emissions, has been  
77 reported during the LBA-SMOCC (Large scale Biosphere atmosphere experiment in  
78 Amazonia – SMOke aerosols, Clouds, rainfall and Climate) Experiment [Hoffer *et al.*, 2006].  
79 The emission from biomass burning is recognized as a primary source of HULIS and of  
80 brown carbon [Andreae and Gelencsér, 2006; Park *et al.*, 2010]. It has been also suggested  
81 that tar balls from smoldering combustion of bio-fuels (or biomass) are a significant source of  
82 atmospheric brown carbon [Chakrabarty *et al.*, 2010]. In addition to emissions from specific  
83 sources, formation of brown carbon through heterogeneous reactions of secondary organic  
84 aerosols (emitted from biogenic and anthropogenic precursors like terpenes) with ammonia is  
85 also documented by Updyke *et al.*, [2012].

86           Uncertainties in atmospheric radiative forcing estimates continue to cause major  
87 debate (IPCC-2007), largely arising from the poor representation of the organic carbon  
88 fraction in atmospheric aerosols [Forster, 2007]. More recently, Feng et al., [2013] have  
89 emphasized the importance of brown carbon absorption in aerosol radiative forcing ( $\sim 0.25$   
90  $\text{W.m}^{-2}$ ) using a general circulation model coupled to a chemical transport model. Their  
91 results suggest that atmospheric brown carbon could contribute nearly 19 % of the total  
92 absorption by anthropogenic aerosols; whereas 71 % is attributable to that from BC (or EC)  
93 and  $\sim 9$  % is from sulphates and coatings of non absorbing organic compounds on soot  
94 carbon [Feng et al., 2013]. Furthermore, their study also highlights an overall mismatch  
95 between observations and model results for the simulated aerosol radiative forcing and  
96 suggests the need to incorporate absorption due to brown carbon in the global models. In this  
97 study, we have made simultaneous measurements of BrC and EC in ambient aerosols ( $\text{PM}_{2.5}$ )  
98 collected from a downwind sampling site in the Indo-Gangetic Plain, representing the  
99 atmospheric outflow. We have also assessed the mass absorption efficiency of BrC from  
100 water-extracts of aerosols.

## 101 **2. Materials and Methods**

### 102 *2.1. Site description and meteorology*

103           The Indo-Gangetic Plain (IGP), situated in the northern part of the Indian peninsula,  
104 generates a host of airborne pollutants. Fossil-fuel combustion, biomass burning (mainly  
105 agricultural crop-residue) and bio-fuel (wood) are some of the characteristic sources of  
106 pollutants. The impact of anthropogenic aerosols on oceanic regions located downwind of  
107 pollution sources in the Indo-Gangetic Plain has been well documented through field  
108 experiments such as INDOEX (Ramanathan et al., 2001; Lelieveld et al, 2001; Mayol-  
109 Bracero et al., 2002) and ICARB (Sudheer and Sarin, 2008; Sarin et al., 2011; Kumar et al.,  
110 2008; Srinivas and Sarin, 2012; Srinivas and Sarin, 2013). Under favourable meteorological  
111 conditions (shallow boundary height and north-easterly/westerly winds), the downwind  
112 transport of pollutants from the IGP to the Bay of Bengal is a conspicuous feature during the  
113 wintertime (from December to March).

114           Ambient aerosols ( $\text{PM}_{2.5} \approx$  particulate matter whose aerodynamic diameter is less than  
115  $2.5 \mu\text{m}$ ) were collected during November'09 - March'10 from a downwind site (Kharagpur:  
116  $22.3^\circ\text{N}$ ,  $87.3^\circ\text{E}$ ) in the Indo-Gangetic Plain (IGP). During the wintertime, the sampling site  
117 is influenced by long-range transport of pollutants from upwind sources in the IGP. Surface  
118 level meteorological parameters were obtained from NCEP (National Centre for

119 Environmental Predictions)-NCAR reanalysis data sets. The winds were predominantly  
120 north-easterly ( $0.5$  to  $3.8 \text{ m s}^{-1}$ ), and relative humidity and surface temperature varied from 38  
121 to 58 % and  $21.5$  to  $30.4 \text{ }^\circ\text{C}$ , respectively. The air mass back trajectories (7-day AMBTs),  
122 computed from the NOAA website using hybrid single particle Lagrangian integrated  
123 trajectory model (HYSPLIT, version 4.0; [Draxler, 2002]), suggest transport of pollutants  
124 from the upwind source regions.

## 125 2.2. Methodology

126 Aerosol samples ( $\text{PM}_{2.5}$ ,  $N = 46$ ) were collected on pre-combusted tisuquartz filters  
127 (PALLFLEX<sup>®</sup><sup>TM</sup>) using a high-volume ( $\sim 1.13 \text{ m}^3 \text{ min}^{-1}$ ) air sampler (HVS- $\text{PM}_{2.5}$ , Thermo-  
128 Anderson Inc.). Most of the samples ( $N = 42$ ) were collected over a period of  $\sim 22$  hrs. After  
129 collection, all samples were stored in a deep freezer at  $-19^\circ\text{C}$  until the time of their chemical  
130 analysis. For all chemical analyses, sample filters were handled under a clean laminar flow  
131 bench (Class – 1000). The absorption spectra of aqueous extracts of aerosols were measured  
132 on a UV-Vis Spectrophotometer (Model: USB-4000) coupled to a 2 m long waveguide  
133 capillary column. Deuterium and tungsten halogen lamps (DT-Mini-2, Ocean Optics) are  
134 used as a light source. Liquid samples were injected via capillary injector into Liquid-core  
135 Waveguide Capillary Cell (LWCC from World Precision Instrument, Sarasota, FL), with an  
136 internal volume of  $250 \mu\text{L}$ . Absorption spectra were recorded over a wavelength range of 300  
137 to 800 nm with an Ocean Optics Spectra-Suite data acquisition software system (Ocean  
138 Optics, Dunedin, FL). Simultaneously, concentrations of organic and elemental carbon (OC  
139 and EC) were also measured by thermo-optical transmittance method using Sunset-Lab EC-  
140 OC Analyzer. Water-soluble organic carbon (WSOC) was measured on total organic carbon  
141 analyzer (model: Shimadzu, TOC-5000a). Along with the samples, filter (and field) blanks  
142 were also analyzed for OC and WSOC. The contribution from blank signals was found to  
143 vary from 4 to 29 % and 0.1 to 14 % of the maximum and minimum signals measured for OC  
144 and WSOC, respectively. Based on the repeat measurements, the overall analytical  
145 reproducibility was better than 5 % for OC and WSOC, whereas it was less than 10 % for EC.  
146 For further details regarding the experimental protocol and method detection limits for OC,  
147 EC and WSOC, reference is made to our earlier publications [Ram *et al.*, 2010; Rengarajan *et*  
148 *al.*, 2007].

## 149 2.3. Absorption coefficient

150 In this study, absorption spectra of water extracts of aerosols (representing bulk of the  
151 water-soluble organic carbon) have been used to assess the absorption coefficient ( $b_{abs}$ )  
152 similar to that described by Hecobian et al., [2010] and is expressed as:

$$153 \quad b_{abs} = (A_{365} - A_{700}) \times (V_{ext} \times 8) \times \ln(10) / (V_{aero} \times L)$$

154 In this equation,  $A_{365}$  and  $A_{700}$  correspond to measured absorbance at 365 and 700 nm,  
155 respectively.  $V_{ext}$  refers to volume of the aqueous extract ( $\sim 50$  ml) in which  $1/8^{th}$  portion of  
156 aerosol filter is extracted and factor '8' is used to estimate the absorption signal for the full  
157 filter.  $V_{aero}$  corresponds to volume of air filtered ( $\sim 1400 \text{ m}^3 = 1.4 \times 10^9$  ml) through quartz  
158 substrates and  $L$  is the path length of the cell (i.e.,  $\sim 2$  m). We have used absorbance at 365 nm  
159 to estimate the absorption coefficient ( $b_{abs}$ ) of light absorbing water-soluble organic carbon  
160 (also referred as BrC). It is relevant to state that light absorption by OC in solvent extracts is  
161 underestimated by a factor of two than that of particulate OC [Liu et al., 2013]. Earlier  
162 studies have investigated the association of brown carbon with humic like substances in  
163 ambient aerosols [Lukács et al., 2007 and references therein]; however, separation of these  
164 compounds from the aqueous filter extracts is rather complex and experimentally tedious.  
165 Based on the replicate analyses of samples ( $N = 15$ ), reproducibility of the absorbance signal  
166 was ascertained to be within 5 %. The contribution from filter blank to sample signal varied  
167 from 0.13 to 1.5 % of the maximum and minimum signal measured on LWCC. The error  
168 propagation, involving sample collection, extraction and measurement of WSOC, yield an  
169 analytical uncertainty of no more than 19 % in the mass absorption efficiency of light  
170 absorbing WSOC.

### 171 **3. Results and Discussion**

#### 172 *3.1. Angstrom exponent ( $\alpha$ ) and Mass Absorption efficiency (MAE or $\sigma_{abs}$ )*

173 The absorption coefficient of an aerosol in the ambient atmosphere is a function of  
174 wavelength of incident light, and is described by a power law. The power exponent of  
175 wavelength is referred as Angstrom exponent ( $\alpha$ ) of a particular species and its magnitude  
176 depends on aerosol size and composition. Using a similar analogy, Hecobian et al., [2010]  
177 have described absorption coefficient of a light absorbing species in the aqueous extracts  
178 which is dependent on wavelength, and is given by the following relation.

$$179 \quad b_{abs} \sim \lambda^{-\alpha}$$

$$180 \quad b_{abs} \approx K \cdot \lambda^{-\alpha}; K = \text{constant}$$

181 Here  $b_{abs}$  is expressed in units of  $\text{M m}^{-1}$  (or  $10^{-6} \text{ m}^{-1}$ ) and  $\alpha$  denotes Angstrom Exponent of  
182 light absorbing component of water-soluble organic matter, referred here as Brown Carbon

183 (BrC). A value of  $\sim 7$  for  $\alpha$  has been reported for humic like substances extracted from  
184 aerosols, sampled from the Amazonian forest fires [Hoffer *et al.*, 2006]. Likewise, smoke  
185 from smouldering of various bio-fuels has a typical value for  $\alpha$  between 7 and 16 [Chen and  
186 Bond, 2010]. Likewise, in an earlier study by Bones *et al.*, [2010], have estimated  $\alpha$  as  $\sim 7$   
187 for freshly formed secondary organic aerosols (SOAs) compared to that observed for aged  
188 SOAs ( $\sim 4.7$ ). More recently, Hecobian *et al.*, [2010] have reported that  $\alpha$  ranges between  $\sim$   
189 6 and 8 for biomass burning aerosols. However, their study also indicates that significant  
190 differences are observed between the biomass burning and non-burning periods.

191 In this study, absorption spectra of aqueous extracts were recorded between 300 nm  
192 and 800 nm for each sample. The absorbance at 365 nm (in near UV region) relative to 700  
193 nm was obtained for all samples in order to estimate  $b_{abs}$ . Earlier studies have documented  
194 strong UV-absorption of water-soluble BrC at 350 to 370 nm [Hecobian *et al.*, 2010 and  
195 references therein]. Therefore, we attribute the prominent absorption at this wavelength range  
196 to the presence of brown carbon (BrC), in order to estimate the absorption coefficient of  
197 water-soluble organic carbon ( $b_{abs-365}$ ). It is noteworthy that a significant linear relationship  
198 (slope = 0.70;  $R^2 = 0.54$ ; P-value < 0.05) is observed between ( $b_{abs-365}$ ) and WSOC (Fig.1);  
199 validating dominant absorption due to BrC in the water-extracts. Furthermore, co-variability  
200 in the temporal trend between concentration of WSOC and non-sea-salt-potassium ion (nss-  
201  $K^+$ ), suggests their common source from biomass burning emissions (Fig.2a). A significant  
202 correlation between WSOC and OC (P-value < 0.05) with an average WSOC/OC ratio of  
203  $0.52 \pm 0.10$  has been reported during the study period [Srinivas and Sarin, 2013b]. It is, thus,  
204 inferred that BrC contribute significantly to the mass concentration of particulate organic  
205 carbon in the atmospheric outflow from the IGP.

206 Further, we have calculated  $b_{abs}$  at varying wavelength (from 300 to 700 nm)  
207 relative to 700 nm. The  $b_{abs}$  shows wavelength dependency as  $\lambda^{-\alpha}$ , where  $\alpha$  refers to Angstrom  
208 exponent (see supporting figure, Fig. S1). It is evident that the absorption signal of WSOC  
209 shows a sharp increase with decrease in wavelength (i.e.,  $b_{abs} \sim \lambda^{-6}$ ; See Table S1 for  
210 goodness of fit parameters for the power relation) and, thus, confirming the presence of  
211 brown carbon in aqueous extracts. This is consistent with earlier observations demonstrating  
212 the similar spectral absorption characteristics of the ambient particulate matter [Andreae and  
213 Gelencsér, 2006; Cheng *et al.*, 2011; Hecobian *et al.*, 2010; Lukacs *et al.*, 2007]. The  
214 angstrom exponent ( $\alpha$ ) of light absorbing WSOC in the IGP-outflow varied from 4.5 to 9.9  
215 (Av:  $6.0 \pm 1.1$ ). However, for most of the sampling days,  $\alpha$  values are greater than 6 (Fig.2b).



216 The impact of biomass burning emissions is also evident through other diagnostic ratios (high  
217 OC/EC:  $7.0 \pm 2.0$  and nss-K<sup>+</sup>/EC:  $0.49 \pm 0.21$ ) in the IGP-outflow [*Srinivas and Sarin,*  
218 2013b]. As stated above, for HULIS type compounds from biomass burning emissions and  
219 bio-fuel emissions, the angstrom exponent is reported to be greater than 6. Recently, Cheng  
220 et al., [2011] have estimated  $\text{\AA}_p$  value of  $\sim 7$  in the aqueous extracts of aerosols from the  
221 Beijing outflow and attributed it to the presence of Brown Carbon. Therefore, the major  
222 source of BrC over the Indo-Gangetic Plain is attributed to biomass and bio-fuel burning  
223 during the study period. A more recent study by Zhang et al., [2013] had demonstrated the  
224 significant differences in the angstrom exponent of light absorbing WSOC between the  
225 offline filter-based aqueous ( $\sim 7.6 \pm 0.5$ ) and methanol ( $\sim 4.8 \pm 0.5$ ) extracts with those  
226 obtained through online measurements using PILS ( $\sim 3.2 \pm 1.2$ ). However, the angstrom  
227 exponent of light absorbing WSOC measured at 365 nm (this study) is consistent with that  
228 reported for biomass burning emissions (See Table 1).

229 We have also estimated the mass absorption efficiency of light absorbing water-  
230 soluble organics as follows.

231 
$$\text{MAE of BrC} = \sigma_{\text{abs-BrC}} (\text{m}^2 \text{g}^{-1}) = (b_{\text{abs-365}})/\text{WSOC}$$

232 In this study, the  $\sigma_{\text{abs-BrC}}$  varied between 0.21 and 1.46 (Av:  $0.78 \pm 0.24$ )  $\text{m}^2 \text{g}^{-1}$  in the  
233 atmospheric outflow from Indo-Gangetic Plain (Fig.2b). The slope of regression line ( $0.70$   
234  $\text{m}^2 \text{g}^{-1}$ ) in Fig. 1 also provides a robust estimate of mass absorption efficiency of BrC ( $\sigma_{\text{abs-}}$   
235  $\text{BrC}$ ) in the atmospheric outflow from the Indo-Gangetic Plain. Lack et al., (2012) have  
236 reported a  $\sigma_{\text{abs-BrC}}$  of  $0.83 \pm 0.42 \text{ m}^2 \text{g}^{-1}$  (measured at 404 nm using multi-wavelength Photo  
237 Acoustic measurements), in aerosols collected from the intense biomass burning emissions.  
238 Likewise, Hecobian et al., (2010) had reported a  $\sigma_{\text{abs-BrC}}$  of  $\sim 0.60$  and  $0.58 \text{ m}^2 \text{g}^{-1}$  for urban  
239 and rural sites, respectively (those characterized by high concentrations of levoglucosan). It  
240 is noteworthy that the mass absorption efficiency of BrC ( $\sigma_{\text{abs-BrC}}$ ) documented for the IGP-  
241 outflow is consistent with that for biomass burning emissions reported in the literature  
242 [*Hoffer et al., 2006; Lack et al., 2012; Yang et al., 2009*]; as summarized in Table 1.

### 243 3.2. Source apportionment

244 The light absorbing organics in the atmosphere can originate either from primary or  
245 secondary processes. Incomplete combustion of biomass/bio-fuel burning and smoldering  
246 combustion processes are suggested as significant primary sources of brown carbon  
247 [*Chakrabarty et al., 2010; Chakrabarty et al., 2013; Cheng et al., 2011; Hecobian et al.,*  
248 2010; *Hoffer et al., 2006; Kirchstetter and Thatcher, 2012; Lukacs et al., 2007*]. However,

249 recent studies have documented the possible formation of atmospheric brown carbon through  
250 secondary processes such as heterogeneous reactions of isoprene in the presence of sulphuric  
251 acid vapour [Limbeck *et al.*, 2003]. Also, through multiphase chemistry of lignin type of  
252 compounds in the cloud water [Gelencser *et al.*, 2003; Gelencser and Varga, 2005; Nguyen *et al.*,  
253 *et al.*, 2010; Nguyen *et al.*, 2012], low temperature combustion of lignin pyrolysis products  
254 [Sareen *et al.*, 2010] and reaction of secondary organic aerosols with NH<sub>3</sub> [Nguyen *et al.*,  
255 2013; Updyke *et al.*, 2012]. Therefore, in order to assess the sources of atmospheric brown  
256 carbon, primary and secondary organic carbon fractions were estimated for the sampling days  
257 using EC-tracer method [Ram and Sarin, 2010; 2011].

258 A notable feature of the data is seen as co-variability in the temporal trends of both  
259 primary and secondary organic carbon fractions with WSOC (See Supporting Fig. S2). The  
260 fractional contribution of OC<sub>sec</sub> in total organic carbon (OC) varied from 11 to 70 % (Av: 50  
261 ± 15 %). Although analytical uncertainty is large in the assessment of secondary organic  
262 carbon (OC<sub>sec</sub>) based on EC-tracer method, it can be inferred that OC<sub>sec</sub> contributes  
263 significantly to the total OC in the IGP-outflow during November'09-March'10. It is  
264 noteworthy that the mass absorption coefficient of light absorbing WSOC ( $b_{\text{abs-365}}$ ) shows a  
265 positive correlation with estimated abundance of secondary organic carbon (OC<sub>sec</sub>) and nss-  
266 K<sup>+</sup> (Fig.3). As stated earlier, OC<sub>sec</sub> can be sourced from the atmospheric reactions of  
267 precursor VOCs produced from either biomass burning emissions or from fossil-fuel  
268 combustion sources. A recent study by Zhang *et al.*, [2011] had shown significant differences  
269 in the light absorption properties of water-soluble organic carbon in aerosols derived from  
270 fossil-fuel combustion (wherein absorption signal measured at 365 nm is relatively 4 to 6  
271 times higher) and that from biomass burning emissions collected over Los Angeles and  
272 Georgia, respectively. Furthermore, their study highlighted the enhancement in light  
273 absorption by secondary aerosols of nitro aromatics from fossil-fuel combustion sources over  
274 Los Angeles. In the IGP-outflow, temporal variability in the mass absorption efficiency of  
275 light absorbing WSOC (measured at 365 nm) is not significant during the study period  
276 (November'09- March'10) and is consistent with that reported for biomass burning emission  
277 (Table 1). Based on these observations, a likely explanation could be that biogenic secondary  
278 organic aerosols contribute significantly to light absorbing organics over the Indo-Gangetic  
279 Plain.

280 In a laboratory study, Nguyen *et al.*, [2013] have documented the formation of light  
281 absorbing organic species based on reaction of ketolimononaldehyde (C<sub>9</sub>H<sub>14</sub>O<sub>3</sub>), a SOA  
282 formed by the ozonolysis of limonene (C<sub>10</sub>H<sub>16</sub>), with amino acids (e.g. glycine) and NH<sub>4</sub><sup>+</sup>. A

283 similar study by Saleh et al., [2013] had provided the first direct evidence for the formation of  
284 light absorbing organics through secondary processes in the aged biomass/bio-fuel burning  
285 aerosols. In this study, we have not investigated the atmospheric reactions of SOAs at the  
286 molecular level. Nevertheless, in view of significant linear relationship among  $OC_{sec}$ , mass  
287 absorption efficiency of WSOC ( $\sigma_{abs-WSOC}$  at 365 nm) and  $nss-K^+$  (a proxy for biomass  
288 burning emissions), it can be inferred that secondary aerosol formation contribute  
289 significantly to atmospheric brown carbon.

290 Long ago, it was suggested by Andr ea [1983] that the association of potassium with  
291 soot carbon (or EC) in ambient aerosols signifies the importance of biomass/bio-fuel burning  
292 emissions. Since then, the mass ratio of  $nss-K^+/EC$  in fine mode aerosols has been used as a  
293 proxy for assessing the qualitative contribution of biomass burning emissions [*Andreae et al.*,  
294 1984; *Andreae and Merlet*, 2001; *Flament et al.*, 2011; *Guazzotti et al.*, 2003; *Ram and Sarin*,  
295 2010; *Srinivas et al.*, 2011; *Wang et al.*, 2005]. It is noteworthy that the  $nss-K^+/EC$  ratio in  
296 the IGP-outflow during November'09-March'10 (Figure 4) is consistent with that reported  
297 for biomass/bio-fuel burning emissions. Based on radiocarbon data, Gustaffson et al., [2009]  
298 highlighted the significance of residential bio-fuel and agricultural crop-residue burning  
299 emissions from the IGP as a major source of carbonaceous aerosols over the south Asia  
300 (particular in the Northern India). In addition, it has also been suggested that biomass burning  
301 emissions contribute significantly to atmospheric water-soluble (primary and secondary)  
302 organics [*Kawamura et al.*, 2013; *Sciare et al.*, 2008; *Timonen et al.*, 2012]. However, the  
303 volatile organic compounds (VOCs) emitted from the biomass/bio-fuel burning also  
304 contribute to atmospheric water-soluble organics through photochemical aging during the  
305 long-range atmospheric transport. In this regard, several studies have documented the high  
306 WSOC/OC and  $OC_{sec}/OC$  ratios from the biomass burning emissions [*Agarwal et al.*, 2010;  
307 *Favez et al.*, 2009; *Sciare et al.*, 2008].

308 As stated earlier, the study site is influenced by long-range transport of bio-mass/bio-  
309 fuel burning emissions from the upwind source regions in the IGP. We have investigated the  
310 temporal variability of WSOC/OC,  $nss-K^+/EC$ ,  $\sigma_{abs-BrC}$  and  $OC_{sec}/OC$  (Fig.4). No significant  
311 ( $P > 0.05$ ) differences are observed for the diagnostic mass ratio of  $nss-K^+/EC$  during the  
312 sampling period (See Table S2). A large spread in the  $nss-K^+/EC$  ratio for biomass burning  
313 emissions has been reported [*Mayol-Bracero et al.*, 2002; *Novakov et al.*, 2000] compared to  
314 that for fossil-fuel combustion. The near constancy of monthly-mean  $nss-K^+/EC$  ratio  
315 overlaps with the reported range for BBEs. It is also noteworthy that near constancy of mass

316 absorption efficiency of light absorbing WSOC ( $\sigma_{\text{abs-WSOC}}$ ) is comparable with that  
317 documented for biomass burning emissions [Cheng *et al.*, 2011; Hecobian *et al.*, 2010;  
318 Kirillova *et al.*, 2014].

319 A near constancy of nss-K<sup>+</sup>/EC and mass absorption efficiency of light absorbing  
320 water-soluble carbon ( $\sigma_{\text{abs-WSOC}}$ ) in the IGP-outflow to the Bay of Bengal (Supporting  
321 information, one-way ANOVA results given in Table S2) indicate biomass burning emissions  
322 as a significant source of brown carbon over the study site. Several studies have suggested  
323 that biomass burning emissions contribute significantly to atmospheric water-soluble  
324 organics [Falkovich *et al.*, 2005; Graham *et al.*, 2002; Mayol-Bracero *et al.*, 2002; Saarikoski  
325 *et al.*, 2007; Sciare *et al.*, 2008; Timonen *et al.*, 2012]. Although WSOC/OC ratio exhibits  
326 small variability during the winter months (December-February;  $\text{Av} \pm \text{Sd}$ :  $0.50 \pm 0.13$ ;  $P >$   
327  $0.05$ ); lower ( $0.41 \pm 0.04$ ) and higher ( $0.70 \pm 0.18$ ) contribution of WSOC to total OC is  
328 noteworthy in early and late sampling days in November and March. The relatively high  
329 mass ratio of WSOC/OC in the early spring-intermonsoon (in March) can be explained by the  
330 relative increase in solar radiation enhancing the photochemical aging of secondary organic  
331 aerosols over the IGP. Similar to WSOC/OC, the fractional contribution of secondary organic  
332 carbon in total OC during winter months show small variability compared to that in preceding  
333 sampling days (in March, Figure 4). The relative decrease in percentage contribution of  $\text{OC}_{\text{sec}}$   
334 to total OC mass in late winter is due to decrease in source strength of biomass burning  
335 emissions relative to that from the fossil- fuel combustion in the upwind source regions of  
336 IGP and its subsequent long-range atmospheric to the sampling site.

### 337 3.3. Comparison of mass absorption efficiencies of WSOC and EC

338 The absorption of solar radiation by the EC or light absorbing WSOC can be  
339 represented by the following equation.

340 Absorption  $\propto \lambda^{-\alpha}$ ; where alpha refers to angstrom exponent.

341 Since the mass absorption efficiency of an absorbing component is proportional to the  
342 absorption; the above equation can be rewritten as follows:

343 Mass absorption efficiency (MAE or  $\sigma$ )  $\propto \lambda^{-\alpha}$

344 Here the mass absorption efficiency is expressed in terms of  $\text{m}^2 \text{g}^{-1}$ . It has been suggested that  
345 EC shows little dependency on wavelength with an angstrom exponent of around 1  
346 (Kirchstetter *et al.*, 2004). However, significantly higher alpha ( $\alpha$ ) values were documented  
347 in the literature for particulate organic matter from biomass/bio-fuel burning emissions  
348 (Hoffer *et al.*, 2006; Kirchstetter *et al.*, 2004; Lukacs *et al.*, 2007; Hecobian *et al.*, 2010;

349 Cheng et al., 2011; Kirchstetter and Thatcher., 2012; Chakraborty et al., 2013; Feng et al.,  
350 2013).

351 For comparison, we have also estimated the MAE of EC (similar to the approach  
352 suggested by Ram and Sarin, [2009]) using Sun-set EC-OC analyzer where the attenuation of  
353 light by EC is measured at 678 nm. Since the MAE of EC is inversely proportional to  
354 wavelength (the suggested angstrom exponent value is one for EC by Kirchstetter et al.,  
355 2004), we can estimate the mass absorption efficiency of EC at other wavelengths (i.e., < 678  
356 nm).

$$357 \quad (\sigma_{\text{EC}})_{\lambda_1} \approx \lambda_1^{-1} \text{ and } (\sigma_{\text{EC}})_{\lambda_2} \approx \lambda_2^{-1}$$

358 In this equation, the monthly averaged MAE of EC at 678 nm has been used to estimate the  
359  $\sigma_{\text{EC}}$  at other wavelengths as follows.

$$360 \quad (\sigma_{\text{EC}})_{\lambda_2} = (\sigma_{\text{EC}})_{\lambda_1} * [\lambda_2/\lambda_1]^{-1}$$

361 Likewise, we have used the MAE of light absorbing WSOC measured at 365 nm to obtain  
362 MAE at other wavelengths

$$363 \quad (\sigma_{\text{wsoc}})_{\lambda_2} = (\sigma_{\text{wsoc}})_{\lambda_1} * [\lambda_2/\lambda_1]^{-\alpha}$$

364 Fig.5 depicts the relative contribution of MAE of WSOC to that of EC, assessed based on  
365 monthly averaged  $\sigma_{\text{EC}}$  and  $\sigma_{\text{wsoc}}$  during the atmospheric outflow to the Bay of Bengal during  
366 November'09-March'10. From this figure, it can be inferred that the relative contribution of  
367 MAE of WSOC is maximum during the wintertime (January), compared to remaining  
368 sampling days (Fig 5), when the ratio of mass absorption efficiency of light absorbing WSOC  
369 relative to EC is  $\sim 0.72$ . It is important to note that relatively higher contribution in early  
370 winter samples could be due to dominant contribution from agricultural crop-residue burning  
371 emissions which occur in the upwind source regions of the IGP during October-November  
372 [Rajput et al., 2014]. However, the impact of bio-fuel emissions in the IGP is more  
373 pronounced during the continental outflow to the Bay of Bengal [Kumar et al., 2010; Ram  
374 and Sarin, 2011; Ram et al., 2012; Srinivas et al., 2011; Srinivas and Sarin, 2013c]. The  
375 results reported in this study are somewhat consistent with that documented for residential  
376 bio-fuels by Kirchstetter and Thatcher, [2012]. Their study documented that the fraction of  
377 solar radiation absorbed by particulate organics generated from the bio-fuel emissions show a  
378 peak at 0.7 at 300 nm and decreased to 0.26 at 550 nm.

### 379 3.4. Implications

380 The measurements of optical and chemical properties of atmospheric constituents have  
381 shown the dominant nature of the anthropogenic aerosols over the Bay of Bengal compared  
382 to that over the Arabian Sea [Kedia *et al.*, 2010; Kumar *et al.*, 2008; Srinivas *et al.*, 2011;  
383 Srinivas and Sarin, 2013a; Sudheer and Sarin, 2008; Vinoj *et al.*, 2004]. In this context,  
384 atmospheric outflow from the Indo-Gangetic Plain is responsible for wide spread dispersal of  
385 pollutants over the Bay of Bengal. The atmospheric radiative forcing estimates have  
386 suggested a relative decrease in solar insolation at the surface Bay of Bengal compared to the  
387 Arabian Sea [Kedia *et al.*, 2010; Vinoj *et al.*, 2004]. Furthermore, it is suggested that aerosols  
388 over BoB are of “more absorbing” type compared to that over the ARS [Kedia *et al.*, 2010;  
389 Nair *et al.*, 2008]. Therefore, aerosol direct radiative forcing estimate increases with the  
390 absorbing BC concentration over this oceanic region. Our study demonstrates ubiquitous  
391 presence of BrC in the atmospheric outflow from the Indo-Gangetic Plain. Due to the  
392 dominance of particulate organic matter in the IGP-outflow (OC: ~ 34 % of PM<sub>2.5</sub> mass;  
393 WSOC/OC: 0.52 ± 0.10) compared to EC (Av: ~ 5 %), it is, thus, important to include the  
394 absorption from brown carbon in radiative forcing estimates over the oceanic regions (BoB)  
395 located downwind of the pollution sources. Thus, presence of BrC over Bay of Bengal in  
396 addition to EC [Srinivas and Sarin, 2013a], would lead to further decrease in incoming short  
397 wave (solar) radiation and, therefore, would reduce surface radiative forcing estimates. Any  
398 changes that decrease the solar insolation can influence circulation pattern in the ocean  
399 surface. To sum-up, we suggest that the combined effect of BrC and BC needs reassessment  
400 in model estimates of aerosol radiative forcing over the Northern Indian Ocean.

401 The absorption of solar radiation by atmospheric water-soluble organic carbon  
402 (WSOC), relative to that by elemental carbon (EC), in the atmospheric outflow from the IGP  
403 is estimated by following the approach similar to that suggested by Kirillova *et al.* [2014].  
404 Briefly, the absorption by WSOC is estimated as a product of the solar emission flux and the  
405 attenuation of light by WSOC (integrated over a broad wavelength range between 300 to  
406 2500 nm) and normalized to that of EC. The wavelength dependent solar emission flux ( $I_0(\lambda)$ )  
407 is obtained through the clear sky Air Mass 1 Global Horizontal (AM1GH) solar irradiance  
408 model by Levinson *et al.*, [2010]. The light attenuation in the atmosphere by an absorbing  
409 species (in this case, WSOC and EC) can be estimated from the Beer-Lambert’s law (for  
410 more details, see [Kirillova *et al.*, 2014 and references therein]) as follows:

$$\frac{I_0 - I}{I_0}(\lambda, X) = 1 - e^{-\left(\sigma_X \left[\frac{\lambda_0}{\lambda}\right]^\alpha C_X h_{ABL}\right)}$$

411

412 Here  $\sigma_x$  and  $\alpha$  refers to mass absorption cross section or efficiency (expressed in  $\text{m}^2 \text{g}^{-1}$ ) and  
 413 angstrom exponent, respectively, for the absorbing species X (i.e., WSOC or EC);  $\lambda_0$  is 365  
 414 nm for WSOC and 678 nm for EC (as explained above) whereas  $\lambda$  refer to any wavelength  
 415 between 300 to 2500 nm. Likewise,  $C_x$  and  $h_{ABL}$  correspond to mass concentration of  
 416 absorbing species ( $\text{g m}^{-3}$ ) and atmospheric boundary layer height (1000 m), respectively.

417 Using this equation, we have estimated attenuation of solar radiation by WSOC and  
 418 EC in the atmospheric outflow from the IGP. Furthermore, the fractional contribution of solar  
 419 absorption by WSOC relative to EC is estimated as follows (adopted from Kirillova et al.,  
 420 [2014]).

$$f = \frac{\int_{300}^{2500} I_0(\lambda) \left[ \frac{I_0 - I}{I_0}(\lambda, WSOC) \right] d\lambda}{\int_{300}^{2500} I_0(\lambda) \left[ \frac{I_0 - I}{I_0}(\lambda, EC) \right] d\lambda}$$

421

422 We have estimated the fractional contribution of solar absorption by light absorbing WSOC  
 423 relative to that of EC in the atmospheric outflow from the Indo-Gangetic Plain to the Bay of  
 424 Bengal. Fig. 6 depicts the fractional solar absorption of WSOC relative to that of EC in the  
 425 IGP-outflow during the study period. From this figure, it is implicit to infer that the amount  
 426 of solar radiation absorbed by WSOC relative to that by EC varied from 2 – 34 %.

427 Although absorption of solar radiation by WSOC (relative to EC) is estimated using  
 428 a simple approach (this study), caution needs to be exercised while interpreting these results  
 429 Fig. 6). It has been suggested that light absorption of OC in the solvent extracts could be  
 430 underestimated by a factor of two [Liu et al., 2013]. Therefore, the estimated relative  
 431 radiative forcing of light absorbing WSOC relative to EC has inherent uncertainty. However,  
 432 the recent study by Kirillova et al., [2014] suggested that the warming effect caused by  
 433 atmospheric brown carbon (through direct and indirect effects) could offset the net cooling  
 434 effect estimated for projected WSOC concentrations. Based on radiative transfer modelling, it  
 435 is suggested that brown carbon could reduce radiative forcing by  $\sim 20$  %, at top of the  
 436 atmosphere, on a global scale; thus, emphasizing this component as crucial for assessing the  
 437 aerosol direct effect [Liu et al., 2014]. The significant contribution of light absorbing WSOC

438 in the atmospheric outflow from IGP, therefore, suggests a need for reassessment of the  
439 climate impact of this species on a global scale.

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445

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447 **References**

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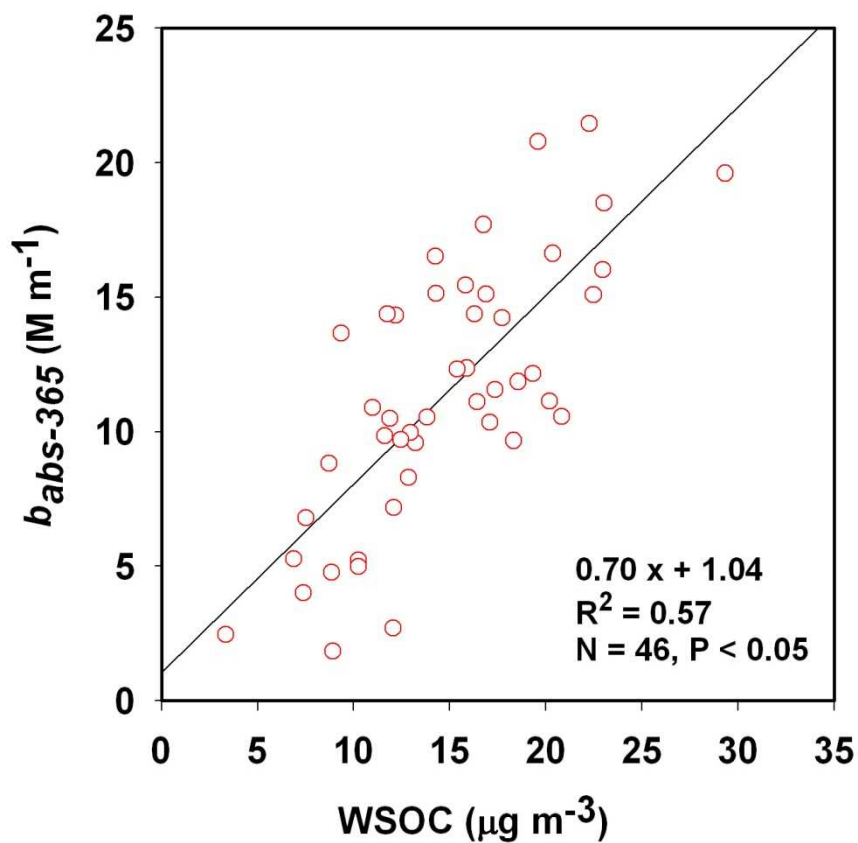
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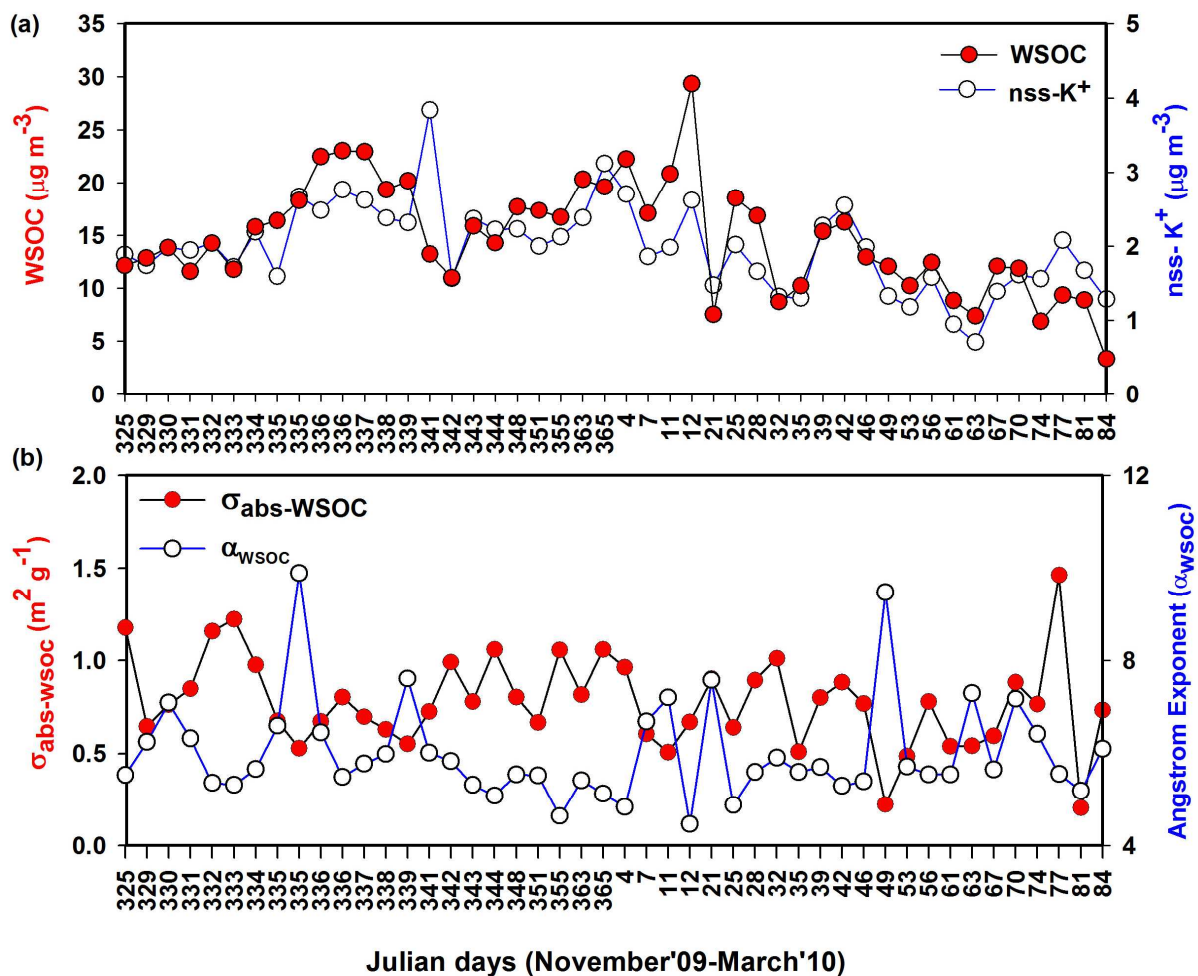
665 *Fig.1.* Scatter plot for mass concentration of water-soluble organic carbon (WSOC) and  
666 absorption coefficient ( $b_{abs}$ ) at 365 nm (where  $M = x \cdot 10^{-6}$ ).

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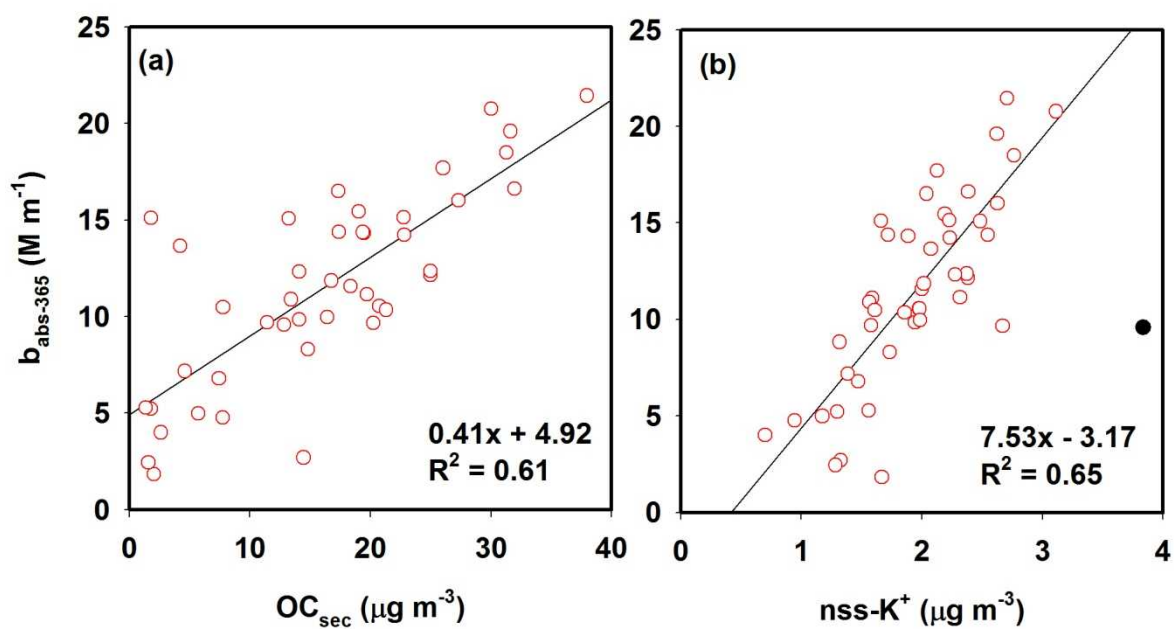
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Fig.2. (a) Temporal variability of WSOC and nss- $\text{K}^+$  concentrations suggest their common source from biomass burning emissions, (b) temporal variability of mass absorption efficiency of light absorbing water-soluble organic carbon ( $\sigma_{\text{abs-wsoc}}$ ) and the Angstrom exponent ( $\alpha_{\text{wsoc}}$ ).

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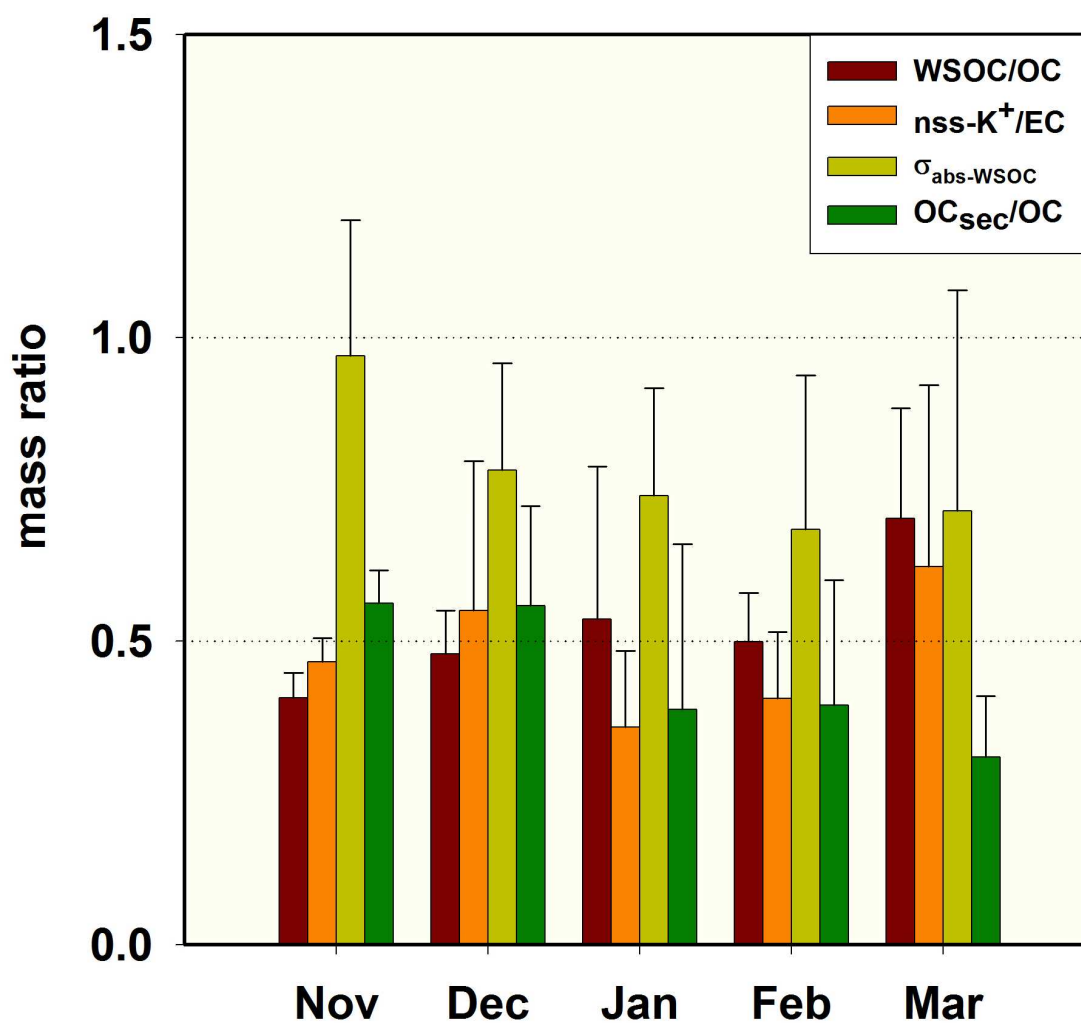


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*Fig. 3.* A strong positive relationship of mass absorption coefficient of water-soluble organic (brown) carbon with the abundance of secondary organic carbon ( $\text{OC}_{\text{sec}}$ ) and  $\text{nss-K}^+$ , suggests the formation of atmospheric brown carbon from biogenic secondary organic aerosols over the IGP.

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691 *Fig.4.* Temporal variability of diagnostic mass ratios (WSOC/OC, nss-K<sup>+</sup>/EC, OC<sub>sec</sub>/OC) and  
692 mass absorption efficiency of light absorbing brown carbon ( $\sigma_{\text{abs-WSOC}}$ ).

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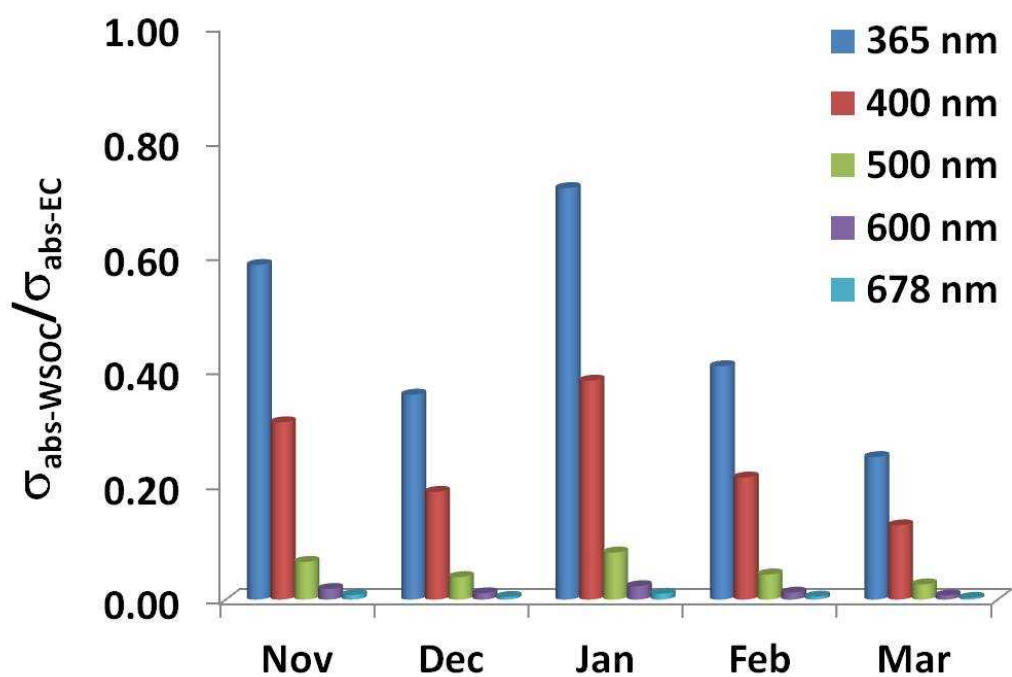
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714 *Fig.5.* Fractional contribution of mass absorption efficiency of light absorbing water-soluble  
715 organic carbon (WSOC) to that of elemental carbon (EC) in the atmospheric outflow to the  
716 Bay of Bengal during November'09 – March'10.

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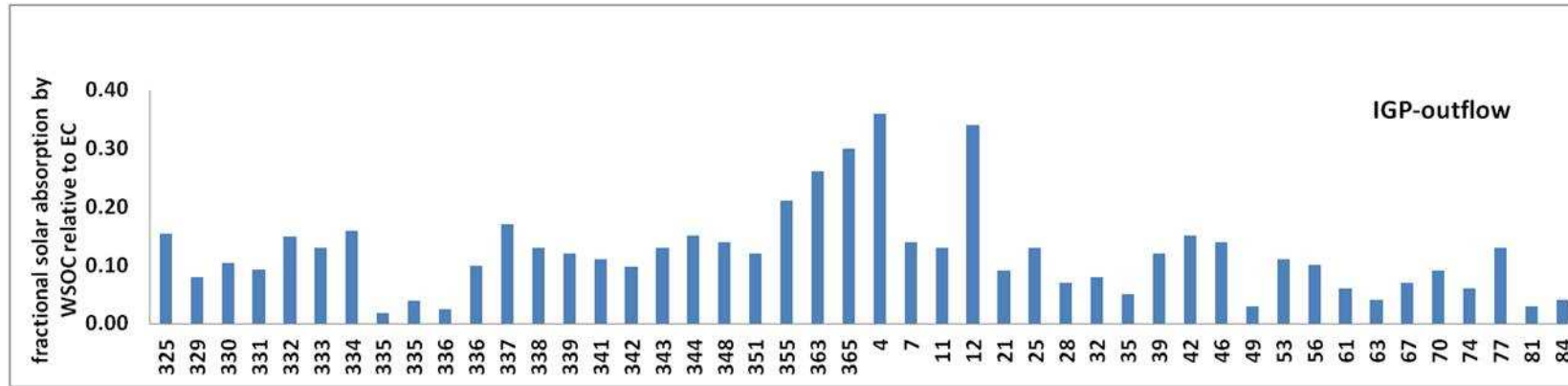
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729 *Fig.6.* Solar absorption by WSOC relative to EC in the atmospheric outflow from the Indo-Gangetic Plain to the Bay of Bengal during  
730 November'09 – March'10.

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734 **Table 1.** Comparison of mass absorption efficiency ( $\sigma_{\text{abs}}$ ) and angstrom exponent ( $\text{\AA}_p$ ) of  
 735 brown carbon (BrC) in the atmospheric outflow from the Indo-Gangetic Plain with other  
 736 literature studies.

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Region	Source	$\lambda(\text{nm})$	$\sigma_{\text{abs-BrC}} (\text{m}^2 \text{g}^{-1})$	$\alpha_{\text{wsoc}}$	Reference
Indo-Gangetic Plain	BB/BF-E	365	$0.78 \pm 0.24$	$6.0 \pm 1.1$	This study
Bay of Bengal (IGP-outflow)	BB/BF-E	365	$0.4 \pm 0.1$	$9.1 \pm 2.5$	Srinivas and Sarin, 2013a
Bay of Bengal (SEA-outflow)	BB/BF-E	365	$0.5 \pm 0.2$	$6.9 \pm 1.9$	Srinivas and Sarin, 2013a
Los-Angeles, USA	BBE	365	0.71	$7.6 \pm 0.5$	Zhang et al., 2013
North America	BBE	404	$0.82 \pm 0.43$	-	Lack et al., 2012
Beijing, China	BBE	550	0.5	-	Yang et al., 2009
Beijing, China	BBE	365	$1.8 \pm 0.2$ (summer)	$7.5 \pm 0.9$	Cheng et al., 2011
Beijing, China	BBE	365	$0.7 \pm 0.2$ (Winter)	$7.0 \pm 0.8$	Cheng et al., 2011
South-eastern US & Atlanta, Georgia	BBE	365	0.64 (urban) & 0.58 (rural)	$7 \pm 1$	Hecobian et al., 2010
Amazon basin	BBE	350 - 400	$\sim 0.5 - 1.5$	$\sim 6 - 7$	Hoffer et al., 2006

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