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| Author(s) | Ito, Masanori; Watanabe, Yutaka W.; Shigemitsu, Masahito; Tanaka, Shinichi S.; Nishioka, Jun |
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1 **Application of chemical tracers to estimate benthic denitrification in the Okhotsk Sea**

2 *MASANORI ITO^{1*}, YUTAKA W. WATANABE^{1, 2}, MASAHITO SHIGEMITSU², SHINICHI S.*
3 *TANAKA³, JUN NISHIOKA⁴*

4 *¹Graduate School of Environmental Science, Hokkaido University, Sapporo, Japan.*

5 *²Faculty of Earth Environmental Science, Hokkaido University, Sapporo, Japan.*

6 *³Earthquake Research Institute, The University of Tokyo, Tokyo, Japan.*

7 *⁴Institute of Low Temperature Science, Hokkaido University, Sapporo, Japan.*

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9 Keywords: benthic denitrification, the marginal sea, the Okhotsk Sea, chemical tracer, multiple
10 regression analysis

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12 *Corresponding author. E-mail: masaito@ees.hokudai.ac.jp, Telephone and Fax:
13 +81-11-706-2246

14

15 **Abstract**

16 To estimate the benthic denitrification in a marginal sea, we assessed the usefulness of N_2^*
17 using dissolved nitrogen (N_2) and argon (Ar) with N^* in the intermediate layer (26.6-27.4 σ_θ) of
18 the Okhotsk Sea. The examined parameters capable of affecting N_2^* are denitrification, air
19 injection and rapid cooling. We investigated the extents of these effects on N_2^* using multiple
20 linear regression analysis. The best model included two examined parameters of denitrification
21 and air injection based on the Akaike Information Criterion as a measure of the model fit to data.
22 More than 80% of N_2^* was derived from the denitrification, followed in order by air injection.
23 Denitrification over the Okhotsk Sea shelf region was estimated to be $5.6 \pm 2.4 \mu\text{mol kg}^{-1}$. The
24 distribution of N_2^* was correlated with potential temperature (θ) between 26.6-27.4 σ_θ ($r = -$
25 0.55). Therefore, we concluded that N_2^* and N^* as a complement can act as a quasi-conservative

26 tracer of benthic denitrification in the Okhotsk Sea. Our findings suggest that N_2^* in
27 combination with N^* is a useful chemical tracer to estimate benthic denitrification in a marginal
28 sea.

29

30 **1. Introduction**

31 The marine fixed nitrogen (N) cycle is highly dynamic due to large input-output rates and a
32 shorter turnover time compared to the cycles of other biologically important nutrients. Despite
33 recent developments in our understanding of the processes and magnitude of the pathways in
34 the marine N cycle, a number of open questions remain (Brandes *et al.*, 2007). Denitrification
35 in the water column and sediments, including canonical denitrification and anaerobic
36 ammonium oxidation (anammox), is the primary sink of N from fixed forms to molecular
37 nitrogen (N_2) (Brandes *et al.*, 2007; Eugster and Gruber, 2012). Conversely, N_2 fixation by
38 diazotrophs is the major oceanic source of biologically available N in the open ocean
39 (Galloway *et al.*, 2004). In particular, there is controversial disagreement as to how close the
40 marine N cycle is balanced between N_2 fixation and denitrification (Codispoti, 2007; Codispoti
41 *et al.*, 2001; Codispoti and Christensen, 1985; Gruber, 2004; Gruber and Sarmiento, 1997).

42 Denitrification occurs primarily in continental marginal sediments (Christensen *et al.*,
43 1987; Middelburg *et al.*, 1996) and in the three major oxygen deficient zones (ODZs) that
44 experience oxygen concentrations less than 4 $\mu\text{mol/kg}$: the eastern tropical North and South
45 Pacific and the Arabian Sea (Codispoti *et al.*, 2001; Codispoti and Christensen, 1985; Codispoti
46 *et al.*, 2005). While reports related to nitrogen sources and sinks have increased over the past 20
47 years, estimation of benthic denitrification are the most poorly constrained (Codispoti, 2007;
48 DeVries *et al.*, 2012; Liu and Kaplan, 1984; Middelburg *et al.*, 1996).

49 There are several ways to constrain the biologically available marine N budget in the global
50 ocean (Codispoti, 2007; Gruber and Sarmiento, 1997). For example, Gruber and Sarmiento

51 (1997) proposed a quasi-conservative tracer N^* ($= ([NO_3^-] - 16[PO_4^{3-}] + 2.9) \times 0.87$) as an
52 index of N_2 fixation-denitrification using the observed nutrient data. In general, N^* increases
53 with N_2 fixation because diazotrophs N_2 into the internal nitrogen cycle in the ocean, and
54 decreases with denitrification because microbes consume nitrate. However, the distribution of
55 N^* may reflect not only N_2 fixation and denitrification, but also other important processes. Such
56 processes include variable phytoplankton stoichiometry (Weber and Deutsch, 2010),
57 atmospheric deposition (Hansell *et al.*, 2007) and different remineralization rates for total
58 organic phosphorus and nitrogen (Coles and Hood, 2007; Landolfi *et al.*, 2008; Monteiro and
59 Follows, 2012; Yoshikawa *et al.*, 2013; Zamora *et al.*, 2009). These processes are capable of
60 reducing the accuracy of N^* .

61 Another way to estimate the content of denitrification derived from excess N_2 has been
62 developed based on comparison of the dissolved N_2/Ar ratio in the water mass between in and
63 out of the denitrification regions (Codispoti, 2007; Devol *et al.*, 2006; Shigemitsu *et al.*, 2013a).
64 The observations of N_2/Ar have already been used in the three ODZs to estimate water-column
65 denitrification at depths above ~1000 m (Chang *et al.*, 2010; Devol *et al.*, 2006; DeVries *et al.*,
66 2012). However, the dissolved N_2/Ar is sensitive to not only denitrification process but also
67 physical processes, e.g., lower atmospheric pressure, rapid cooling, and air injection by bubbles
68 (Hamme and Severinghaus, 2007). Hamme and Emerson (2013) reported that both net
69 denitrification and bubble injection are the most likely effects capable of causing large changes
70 in the dissolved N_2/Ar ratio, because the N_2/Ar ratio in the air is almost double the ratio in
71 seawater. Regarding constraint of excess N_2 derived from denitrification, Shigemitsu *et al.*
72 (2013a) proposed a new tracer N_2^* based on N_2 and Ar in the denitrification region. However,
73 since this method is difficult to distinguish excess N_2 produced during denitrification from that
74 caused by physical processes, only this method could not estimate the content of
75 denitrification in the continental marginal sea.

76 The Okhotsk Sea (OS) is a subpolar marginal sea in the western North Pacific and the
77 southernmost sea ice production area in the northern hemisphere. The densest water ventilating
78 in the North Pacific region originates over the northwestern continental shelf of the OS (Kitani,
79 1973). This water is called the Dense Shelf Water (DSW, $\sigma_\theta = 26.7-27.0$, where σ_θ is the
80 potential density, Nakatsuka *et al.*, 2004; Nakatsuka *et al.*, 2002), produced by brine rejection
81 during sea ice formation (Kitani, 1973). The DSW penetrates into the intermediate depths to
82 join with the Okhotsk Sea Intermediate Water (OSIW, $\sigma_\theta = 26.8-27.4$, Yamamoto *et al.*, 2002;
83 Yamamoto-Kawai *et al.*, 2004). The OSIW is transported into the North Pacific Intermediate
84 Water (NPIW). The NPIW is widely distributed over the North Pacific, which influences the
85 climate of the North Pacific. This water contains a large amount of anthropogenic CO₂ (e.g.,
86 Sabine *et al.*, 2004; Wakita *et al.*, 2005; Watanabe *et al.*, 2001). However, the density surface of
87 26.8 σ_θ does not outcrop to the atmosphere within the open North Pacific, even in winter
88 (Yasuda, 1997). In addition, Yasuda (1997) found a pycnostad (referred to as the Okhotsk Sea
89 Mode Water) in the 26.6-27.0 σ_θ density range.

90 Primary production in the northwestern shelf region in the OS is extremely high throughout
91 the year, except the ice-covered season (Saitoh *et al.*, 1996; Sorokin and Sorokin, 1999).
92 Because of the strong tidal mixing and extremely low temperature on this region, the particulate
93 organic matter produced on the shelf is exported efficiently to the pelagic intermediate water
94 (Nakatsuka *et al.*, 2002). Therefore, benthic flux of organic matter is possible to be high in this
95 shelf region. Yoshikawa *et al.* (2006) found that there was possibility of large benthic
96 denitrification in the continental region of the OS based on N^* ($= [NO_3^-] + [NO_2^-] + [NH_4^+] -$
97 $16[PO_4^{3-}] + 2.9) \times 0.87$) and the nitrogen isotopic ratio of nitrate. Furthermore, Shigemitsu *et al.*
98 (2013b) indicated that Fe is also reduced in such sub-oxic sediments in addition to nitrate
99 based on high Fe/Al ratio in the same region. However, the quantification of benthic
100 denitrification has not been evaluated.

101 We here try to estimate the extent of benthic denitrification quantitatively in the continental
102 shelf region. In this study, we also used the N^* ($[\text{NO}_3^-] + [\text{NO}_2^-] + [\text{NH}_4^+] - 16[\text{PO}_4^{3-}] + 2.9$)
103 $\times 0.87$) defined in Yoshikawa et al. (2006). Because the N^* include nitrite and ammonium, the
104 N^* would be affected by the anammox and would not be affected by dissimilatory nitrate
105 reduction to ammonium (DNRA). To estimate the extent of benthic denitrification in the
106 marginal sea shelf region, we focused on the density layer ($\sigma_\theta = 26.6\text{-}27.4$) in the OS producing
107 DSW and OSIW and attempted to assess the usefulness of N_2^* with N^* as the complement for
108 detecting benthic denitrification. The DSW produced from sea-ice formation on the
109 northwestern shelf in the OS flows out to intermediate depths along the Sakhalin coast to the
110 southward and joins with the OSIW (Matsuda *et al.*, 2009). Yoshikawa et al. (2006) showed
111 that N^* acts as a quasi-conservative tracer of denitrification during the OSIW formation process.
112 No study that detects a sedimentary denitrification originating in the shelf region using N_2 and
113 Ar exists. The OS is one of the best regions to evaluate the usefulness of this method.

114

115 **2. Materials and Method**

116 This investigation involved sampling and analyzing 28 sites to measure concentrations of
117 dissolved gas using gas chromatography (Figure 1). The selected sites were located from the
118 northwestern continental shelf in the OS to the Bussol Strait. Samples were collected in 24-liter
119 X-Niskin bottles equipped with a CTD on the R/Vs Professor Khromov and Hokko-maru
120 during August to early in September 2007 and May 2008, covering a formation area of the
121 intermediate water in the OS (49-55°N, 142-148°E) and in the western North Pacific (40-42°N,
122 145-146°E), respectively.

123 Samples at each station were obtained at approximately 25 layers above 2000 m depth. The
124 collected seawater for N_2 and Ar was directly transferred from the Niskin bottle to a 60 ml glass
125 vial vessel. After opening the vent of the Niskin bottle, the vessel was washed twice and

126 overflowed with three times the volume of the vessel to avoid air contamination during the
127 transfer procedure. For the final filling, we added 50 μl of saturated HgCl_2 solution to prevent
128 biological activity: the vial vessel was covered with a butyl rubber cap and aluminum seal,
129 paying particular attention to assure that no air bubble contamination occurred. We preserved
130 these vial vessels in the dark and in a cool seawater bath. The N_2 and Ar concentrations were
131 determined by gas-chromatographic system with thermal conductivity detection (Tanaka and
132 Watanabe, 2007) in our laboratory. Dissolved oxygen was also determined on board using the
133 Winkler titration method (Carpenter, 1965). The analytical precisions for replicate
134 measurements of gas concentrations were within 0.04% for N_2 and within 0.05% for Ar.

135 We have found systematic offsets in the data obtained below 1750m depth between the
136 stations near the Bussol Strait (A1 and A2) and station KNOT (data from Hamme and Emerson
137 (2002), <http://web.uvic.ca/~rhamme/download.html>). The average differences of temperature,
138 salinity and density between the Bussol Strait and the station KNOT are 0.7 $^{\circ}\text{C}$, 0.2, 0.2 σ_{θ} ,
139 respectively. The oxygen concentrations in both regions were more than 60 $\mu\text{mol kg}^{-1}$. Since
140 the above results indicated that there were not any significant processes affecting Ar and N_2
141 concentrations, we have applied the following corrections to the N_2 and Ar concentrations in
142 the OS and the western North Pacific. We thus made corrections to the N_2 and Ar data in the OS
143 using the following relations: $[\text{N}_2]_{\text{meas}} = [\text{N}_2]_{\text{meas}_0} + 16.97 \mu\text{mol kg}^{-1}$ and $[\text{Ar}]_{\text{meas}} = [\text{Ar}]_{\text{meas}_0} +$
144 $0.87 \mu\text{mol kg}^{-1}$, where meas and meas₀ represented the measured concentrations after the offset
145 corrections and raw values of our observations. In this study, we used only dissolved gas
146 concentrations near the center of DSW and OSIW layers between 26.6 and 27.4 σ_{θ} to estimate
147 the extent of benthic denitrification that occurred in the northwestern shelf of the OS (Kitani,
148 1973; Wakita *et al.*, 2005; Wakita *et al.*, 2003; Yoshikawa *et al.*, 2006).

149

150 ***3. Approach: Concepts for estimating the benthic denitrification in the Okhotsk Sea***

151 In this study, we used the chemical tracers N_2^{ex} and N_2^* to represent the excess N_2 with N^* .
 152 One approach to estimate denitrification involves the direct measurement of the denitrification
 153 end product, i.e., the excess N_2 above the background value. There is no significant sink of N_2
 154 gas and the only source is denitrification in the interior of the ocean. Therefore, excess N_2 can
 155 be assumed to represent the amount of denitrification.

156 To estimate excess N_2 , we employed a method analogous to Devol et al. (2006) that uses
 157 N_2/Ar ratios normalized to atmospheric equilibrium ratios:

158

$$159 \quad N_2^{ex} = \left[\left(N_2/Ar \right)_{norm} - \left(N_2/Ar \right)_{back} \right] \times [N_2]_{sat} \times 2, \quad (1)$$

160

161 where $(N_2/Ar)_{norm}$ ($\equiv \delta N_2 / \delta Ar = ([N_2]_{meas} / [N_2]_{sat}) / ([Ar]_{meas} / [Ar]_{sat})$) is the ratio normalized to
 162 the atmospheric equilibrium ratio within the denitrifying waters of the ODZs, $(N_2/Ar)_{back}$ is
 163 the normalized ratio predicted for a parcel of water outside the denitrifying zone with the
 164 same density, $[N_2]_{sat}$ is the atmospheric equilibrium saturation of N_2 predicted from θ and S ,
 165 and the factor of 2 converts to units of μM for monoatomic N.

166 Another way of estimating excess N_2 uses N_2^* , which has been developed recently
 167 (Shigemitsu *et al.*, 2013a) as follows:

168

$$169 \quad N_2^* = [N_2]_{meas} - \left([N_2]_{sat} / [Ar]_{sat} \right) \times [Ar]_{meas}, \quad (2)$$

170

171 As mentioned above, since N_2^{ex} largely depended on the $(N_2/Ar)_{back}$ value in equation (1),
 172 we also used N_2^* with the assistance of multiple linear regression analysis (MLRA) to clarify
 173 the influence of denitrification and physical processes on N_2^* in this region. We considered
 174 the following factors to affect N_2^* : denitrification (J_{den}), bubble injection (J_{air}), change of

175 solubility by rapid cooling (J_{cool}) and change of atmospheric air pressure (J_{pres}). In this study,
176 because N_2 and Ar are similarly affected by J_{pres} , the influence of J_{pres} could be ignored.

177 We assumed that the observed N_2^* can be categorized into the fractions affected by J_{den} ,
178 J_{cool} and J_{air} , and used MLRA to fit the data to the following equation:

179

$$180 \quad N_2^* = a_0 + a_1 \cdot J_{\text{den}} + a_2 \cdot J_{\text{air}} + a_3 \cdot J_{\text{cool}}, \quad (3)$$

181

182 where a_0 , a_1 , a_2 and a_3 are the partial regression coefficients. We here used the observed N^*
183 values ($\mu\text{mol kg}^{-1}$) as the proxy for J_{den} , the combinations of the difference between observed
184 and saturated Ar values (ΔAr , $[\text{Ar}]_{\text{meas}} - [\text{Ar}]_{\text{sat}}$) and the atmospheric mixing ratio of N_2 to Ar
185 ($\chi_c = 78.084 / 0.934$) for J_{air} , and the difference between the potential temperature (θ) and
186 reference temperature (T_{ref}) defined as freezing temperature at the surface for J_{cool} , namely, we
187 have

188

$$189 \quad J_{\text{den}} = N^*, \quad (4)$$

190

$$191 \quad J_{\text{air}} = \Delta\text{Ar} \times \chi_c, \quad (5)$$

192

193 and

194

$$195 \quad J_{\text{cool}} = \theta - T_{\text{ref}}. \quad (6)$$

196

197 In the MLRA analysis, we selected the candidate models using all possible combination of
198 the three explanatory variables, i.e., 7 models. We selected the best model using the Akaike

224 interface and thus gas exchange and cooling alters both Ar and N₂ concentrations. Therefore,
225 we used N₂^{*} which is more useful to discuss the effects of denitrification, air injection and
226 rapid cooling on excess N₂ because the estimate of N₂^{ex} largely depended on the (N₂/Ar)_{back}
227 value.

228

229 4.2. Model selection by AIC

230 In the case that a correlation among the explanatory variables in a multiple regression
231 model is high, the multicollinearity problem may occur (Graham, 2003). Confounding in the
232 regression is typically represented as a multicollinearity, of which deleterious effects on the
233 interpretation of the regression coefficients are well known (e.g., Van Sickle, 2013). The
234 explanatory variables were screened for multicollinearity as one of the assumptions in the
235 MLRA. Multicollinearity in MLRA was studied by examining the tolerance, which is a
236 statistic used to determine how much the independent variables are linearly related to one
237 another. The tolerance is calculated as $1-R^2$ (R^2 is the multiple coefficient of determination)
238 for an independent variable when it is predicted by the other independent variables already
239 included in the analysis. The higher the intercorrelation of the independent variables, the
240 closer the tolerance is to 0. Generally, the tolerance less than 0.2 indicates a multicollinearity
241 problem and provides an increase in the standard error of the regression coefficients (Hart and
242 Sailor, 2009; Rawlings *et al.*, 1998). To avoid this problem, we verified the combinations of
243 the explanatory variables having a high correlation. The correlation of J_{air} and J_{cool} (0.66) was
244 slightly high compared to the other combinations (Table 1). However, the tolerance values for
245 all variables in our MLRA were greater than 0.2. Therefore, there were no multicollinearity
246 problems between our explanatory variables. In our study, we used all possible explanatory
247 variables, J_{den} , J_{air} and J_{cool} in the MLRA.

248 To clarify the amount of denitrification occurred in the northwestern shelf of the OS, we

249 compared the results of 7 models (Table 2) and evaluated N_2^* between 26.6-27.4 σ_θ , using
250 Eqns. (3) to (7). The best model was Model 1 due to having the smallest *AIC* for J_{den} and J_{air}
251 (Table 2), ($R = 0.66$, $n = 45$, $p < 0.0001$; simple linear regression). Both denitrification and air
252 injection primarily contributed to N_2^* in this region. Because our explanatory variables had
253 different units, we evaluated which have larger effect on N_2^* by using each standardized
254 dataset. We found that the largest effect was for J_{den} , followed by J_{air} based on the standard
255 partial regression coefficient (a_i' , $i = 0$ to 3) and that a_1' and a_2' were -0.51 and -0.27 ,
256 respectively.

257

258 4.3. Evaluation of combination of N_2^* with N^* as a quasi-conservative tracer for benthic 259 denitrification on the continental shelf

260 In order to evaluate the amount of denitrification by using N_2^* and N^* , it is necessary to
261 validate the extent of N_2^* dynamics as a conservative property. In the relationship between N_2^*
262 and potential temperature of the OSIW, we found a moderate linear relationship between the
263 N_2^* and θ for 26.6-27.4 σ_θ in this region ($r = -0.55$, $n = 42$, $p < 0.0001$; simple linear
264 regression), indicating that N_2^* may act as a quasi-conservative tracer for the OSIW formation
265 process.

266 Plots for the Kuril Basin and Bussol Strait areas (sta. A1, A2, A4 and B1) were almost
267 distributed below the regression line (open circles in Figure 4) due to the entrainment of low
268 N_2^* waters from upper layer by active tidal mixing. The other data below the regression line
269 were shallow layer ($< 75\text{m}$ depth) except the deepest layer over the northwestern shelf. In this
270 shallower layer, O_2 concentration was higher than oxygen deficient condition ($< 4 \mu\text{mol/kg}$)
271 indicating that denitrification in the water column did not occur. Furthermore, the DSW may
272 flow out of the shelf by the gravity current along the slope and/or a western boundary current
273 (e.g., Nakatsuka *et al.*, 2002; Ohshima *et al.*, 2002). Therefore these plots may include only

274 physical effects and consequently lower than the regression line.

275 On the other hand, plots above the regression line may be due to vertical transport of high
276 N_2^* waters benthic denitrification occurred in the east Sakhalin sediment and northwestern
277 shelf, because the intermediate water in this study includes the layer below the DSW (> 27.0
278 σ_θ). This relationship between N_2^* vs. θ suggests that N_2^* at intermediate depths was not
279 essentially changed, i.e., N_2^* levels were not affected by regeneration and the high N_2^*
280 signature in DSW is not modified by in situ denitrification or regeneration during the
281 formation process of OSIW and its pathways. Moreover, our finding demonstrates that N_2^*
282 acts as a quasi-conservative tracer for the benthic denitrification.

283 Therefore, since both N_2^* along with N^* act as quasi-conservative tracer in the DSW,
284 $a_1 \times J_{den}$ in the MLRA act conservatively in this layer. Thus, we concluded that the
285 combination of N_2^* with N^* is useful to estimate a benthic denitrification in the OS.

286

287 4.4. Estimate of denitrification, air injection and rapid cooling using MLRA and AIC

288 Using N_2^* along with N^* , ΔAr and θ in the MLRA (Eqns. (3)-(6)), we estimated
289 contributions of denitrification, air injection and rapid cooling to excess N_2 for the first time.
290 By multiplying partial regression coefficients (a_i ($i = 0$ to 3) in Table 2) by each explanatory
291 variable value, we can estimate the contribution of each explanatory variables as follows.

292 In the best Model 1, the obtained partial regression coefficient, a_1 for J_{den} was -1.09 (Table
293 2). We estimated the benthic denitrification in this intermediate water to be approximately 5.6
294 $\pm 2.4 \mu\text{mol kg}^{-1}$ (SD) from $a_1 \times J_{den}$ values (Figure 5 a-1). In the same water, observed N^*
295 value was $-5.1 \pm 2.2 \mu\text{mol kg}^{-1}$. The N^* value was approximately $2.6 \pm 1.1 \mu\text{mol kg}^{-1}$
296 converted to N_2 yield during denitrification. These results demonstrate that excess N_2
297 calculated by using N_2^* along with N^* was found to be at least twice N^* alone, which agrees
298 with the difference between N_2^{ex} and N^* in the Arabian Sea (Devol *et al.*, 2006). What cause

299 the difference of denitrification between excess N_2 and N^* in this region?

300 In the Arabian Sea, Devol et al. (2006) deduced from previous studies that there are
301 several possible mechanisms to increase the N_2 yield during denitrification: (1) oxidation of
302 the ammonium regenerated during water-column denitrification to N_2 , (2) contributions of N_2
303 resulting from processes taking place within the sediments in contact with the ODZ waters,
304 (3) high N:P ratio material produced during N_2 fixation or (4) denitrification fueled by
305 non-Redfield organic matter such as preferential degradation of proteins, and (5) other
306 reactions between metals, iodine and various N species that lead to N_2 production,.

307 First, N^* in this study would be affected by anammox and would not be affected by DNRA
308 in common with N_2^* , because the N^* includes nitrite and ammonium. The rate of DNRA is
309 three orders of magnitude lower than denitrification and anammox and is therefore
310 insignificant to nitrogen cycle (Crowe *et al.*, 2012). Second, in the most intermediate water of
311 the OS, oxygen concentration was larger than the oxygen deficient condition ($< 4 \mu\text{mol/kg}$,
312 Codispoti *et al.*, 2005). Third, N_2 fixation does not occur in the high-latitude OS, which is
313 outside the habitat range of nitrogen fixers (Capone *et al.*, 1997). Therefore, since the cases of
314 (1) to (3) can be ignored, although we have no data about the case of (5), (4) could possibly
315 result in most of the increases of the N_2 yield during denitrification in this study.

316 The N^* is a useful tracer for denitrification and N_2 fixation, inputs from terrestrial and
317 atmosphere, but in the OS, quantification of benthic denitrification by N^* alone is possible to
318 underestimate/overestimate. That is, there are possible mechanisms to increase/decrease the
319 N^* in this region: dissolved organic nitrogen (DON) input from the Amur River and phosphate
320 elution from sediments.

321 During early sedimentary diagenesis, particulate organic N is hydrolyzed to DON by
322 bacterial hydrolytic enzymes, which can increase the N^* . A large fraction of DON is
323 ultimately remineralized to ammonium (NH_4^+). In the presence of O_2 , a portion of the

324 regenerated NH_4^+ is oxidized to NO_3^- (nitrification) before it can escape from the sediments.
325 This NO_3^- may, in turn, be used as a terminal electron acceptor by denitrifying bacteria
326 producing gaseous forms of N (coupled nitrification-denitrification) (Lehmann *et al.*, 2007;
327 Thibodeau *et al.*, 2010).

328 Under anaerobic sediment conditions, phosphate may also be eluted from the sediment
329 (e.g., Sundby *et al.*, 1992), which can decrease the N^* . Yoshikawa *et al.* (2006) showed that
330 the DSW reflects not only sedimentary denitrification but also phosphate elution from
331 sediment from relationships between apparent oxygen utilization and phosphate concentration
332 and nitrate concentration.

333 Therefore, the best way to estimate a benthic denitrification in this region is to make up
334 for each other complementarily N_2^* and N^* . The physical processes are disadvantage for N_2^* .
335 The phosphate elution, DON input from the river and remineralization with non-Redfield
336 ratio are disadvantage for N^* . On the other hand, the advantage for N_2^* is independent from
337 Redfield ratio and it can directly estimate end product of denitrification as excess N_2 , if there
338 are unknown biological processes. The advantage for N^* is to be used as index of
339 denitrification including anammox and N_2 fixation.

340 On the other hand, the obtained partial regression coefficient, a_2 for J_{air} was -2.38 (Table 2)
341 and the average air injection value in the intermediate water was $0.8 \pm 1.3 \mu\text{mol kg}^{-1}$ (Figure
342 5 a-2). The maximum air injection (c.a. $2.4 \mu\text{mol kg}^{-1}$) was found at approximately 26.7
343 σ_θ in this region. Note that the air injection values should not be represented strictly only air
344 injection, but also other physical effects: rapid cooling, heating, slow gas exchange and
345 mixing of water masses, because ΔAr used in J_{air} includes effects of all processes resulting in
346 disequilibrium in Ar concentration. However, it is difficult to estimate ΔAr due only to air
347 injection. Therefore, as we mentioned above, we confirmed how much the independent
348 variables are linearly related to one another by examining the tolerance. Since the tolerance

349 values for all variables in our MLRA were greater than 0.2, our analysis may be insensitive to
350 this error. Craig and Weiss (1971) found that the mean N_2 saturation anomaly due to air
351 injection was 1.9% in the Pacific Deep Water. However, compared with the maximum N_2
352 concentration due to air injection predicted by Model 1, approximately $< 1\%$, only half the air
353 injection is predicted with the model. The unexpected low value of air injection may be
354 caused by sea ice formation when DSW originates. When forming the dense shelf water in the
355 wintertime, sea surface water of the OS is covered by ice due to the strong cooling from the
356 air mass of Siberia. Therefore, the air-sea gas exchange may be weakened by the sea ice
357 coverage, and consequently we could observe the low value of the air injection (Tanaka and
358 Watanabe, 2007). The effect of air injection is propagated into the abyssal zone over the
359 northwestern shelf and the surface layer of OS. Its effect disappears below 100 m in open
360 waters (Figure 5 a-2). In the best Model 1, we found that N_2^* was very sensitive to
361 denitrification and air injection and insensitive to rapid cooling ($< 0.5 \mu\text{mol kg}^{-1}$, Figure 5
362 b-4) over the OS.

363 In the continental shelf region, more than 80% of N_2^* was derived from the denitrification,
364 followed in order by air injection. The N_2^* in the intermediate water in the continental shelf
365 (except A and B stations) was $8.3 \pm 2.4 \mu\text{mol kg}^{-1}$. The benthic denitrification ($a_1 \times J_{\text{den}}$) was
366 $6.8 \pm 2.4 \mu\text{mol kg}^{-1}$. The air injection ($a_2 \times J_{\text{air}}$) was $1.9 \pm 0.4 \mu\text{mol kg}^{-1}$. Therefore the effect
367 of benthic denitrification was about 80% or more than.

368

369 **5. Conclusions**

370 This study indicates the N_2^* along with N^* may be useful tracers to estimate the benthic
371 denitrification in the OS. Most notably, this is the first study to estimate the contributions of
372 denitrification and other factors on N_2^* separately. Our results provide compelling evidence
373 for evaluate the benthic denitrification in the continental marginal sediments where

374 denitrification occurs primarily (Christensen *et al.*, 1987; Middelburg *et al.*, 1996). However,
375 some limitations are worth noting. Although our hypotheses were supported statistically, J_{air}
376 cannot represent only air injection, but also other physical effects and MLRA in this study can
377 be used only when properties in the water mass behave conservatively. Future work should
378 therefore include follow-up work designed to evaluate whether the air injection was
379 represented strictly and also whether N_2^* along with N^* continue to be used to improve
380 understanding of the benthic denitrification.

381

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558

559

560 **Figure Legends**

561 **Figure 1.** Sampling points for dissolved gas and nutrients (black filled circles) in the Okhotsk
562 Sea (49—55°N, 142—148°E) and the western North Pacific (40-42°N, 145-146°E). The
563 dissolved gas samples from station KNOT (44°N, 155°E) and the western North Pacific
564 (KY08 st04, st07 and st11) were used as background values. The solid line represents vertical
565 sections from the stations in Figure 5. We used dissolved gas data from station KNOT from R.
566 C. Hamme's data (<http://web.uvic.ca/~rhamme/download.html>).

567

568 **Figure 2.** The relationship between N_2^{ex} ($\mu\text{mol kg}^{-1}$) and N_2^* ($\mu\text{mol kg}^{-1}$) in the intermediate
569 water (26.6-27.4 σ_θ) in the Okhotsk Sea. The dashed line represents 1:1 line; the solid line
570 shows a simple linear regression based on all our data (open symbols), ($y = 5.68 + 0.47x$; $r =$
571 0.99).

572

573 **Figure 3.** Profiles of $\delta N_2/\delta Ar$ in the Okhotsk Sea and throughout the western North Pacific
574 versus σ_θ . Open circles denote samples near the Bussol Strait (A1, A2, A4 and KC2-5),
575 triangles denote samples from the east of Sakhalin (B1, B4 and C5), pluses denote samples
576 from Line D, and crosses denote samples from the northwestern shelf region. Closed circles
577 denote average values outside the denitrifying zone of the OS (KY08 st4, st7, st11 and
578 KNOT). Also shown is the best fit line for $\delta N_2/\delta Ar$ in waters outside the denitrifying zone
579 $(N_2/Ar)_{back}$. Error bars represent standard errors in the range 0.001-0.004.

580

581 **Figure 4.** Relationship between calculated N_2^* ($\mu\text{mol kg}^{-1}$) and σ_θ between 26.6-27.4 σ_θ . The
582 solid line denotes the regression line ($y = 7.20 - 1.43x$; $r = -0.55$).

583

584 **Figure 5.** Vertical cross sections of N_2^* , denitrification, air injection and rapid cooling from

585 the northwestern shelf (140°E) to the Bussol Strait, east of Sakhalin (153°E) (the solid line in
586 Figure 1) in (a) Model 1 and (b) Model 2; (a-1) and (b-1): N_2^* ; (a-2) and (b-2): denitrification;
587 (a-3) and (b-3): air injection; (b-4): rapid cooling. All units were $\mu\text{mol kg}^{-1}$. White contour
588 lines indicated density surfaces with thin lines (contour interval 0.2 σ_θ). The shading color
589 indicated the calculated values of each property. Ocean Data View was used to draw these
590 figures (Schlitzer, 2001).