Copper(I)-Catalyzed Enantioselective Boryl Substitution of Allyl Acylals: An Efficient Approach for Enantioenriched \( \alpha \)-Chiral \( \gamma \)-Acetoxyallylboronates

Y. Takenouchi  
R. Kojima  
R. Momma  
H. Ito

Division of Applied Chemistry and Frontier Chemistry Center,  
Faculty of Engineering, Hokkaido University, Sapporo,  
Hokkaido, 060-8628, Japan.  
E-mail: hajito@eng.hokudai.ac.jp

Click here to insert a dedication.

Abstract A novel approach has been developed for the enantioselective synthesis of \( \alpha \)-chiral \( \gamma \)-acetoxyallylboronates via the copper(I)-catalyzed \( \gamma \)-boryl-substitution of allyl acylals. This reaction proceeded with high \( E/Z \) selectivity and enantioselectivity (\( E/Z = >99:1 \), up to 80% yield, up to 99% ee). The subsequent allylation of aldehyde with the allylboronate afforded the mono-protected anti-1,2-diol derivative with high stereoselectivity.

Key words boron, enantioselectivity, copper, catalysis, allylation

The asymmetric allylation of aldehydes with allylboronates is a useful transformation in organic synthesis because of the high synthetic utility of the 1,2-diol products. Allylboronates bearing a substituent at their \( \gamma \)-position relative to the boron atom are especially important organometallic reagents for the construction of consecutive chiral centers via C–C bond forming reactions because they can react with aldehydes in a highly stereospecific manner through a six-membered transition state. In particular, optically active \( \gamma \)-alkoxyallylboronates have been widely used for the preparation of chiral 1,2-diol moieties, which can be found in a wide range of natural products and synthetic drugs. However, the synthetic methods used for the construction of these boronates typically require a boron source bearing stoichiometric chiral auxiliary.

We previously reported the first catalytic synthesis of \( \alpha \)-chiral linear or carbocyclic \( \gamma \)-alkoxyallylboronates via the copper(I)-catalyzed \( \gamma \)-boryl substitution of allyl acetals. Although our previous reaction showed high enantioselectivity and broad substrate scope in terms of its functional group compatibility, it was not amenable to sterically hindered substrates because they exhibited poor reactivity toward the boryl copper nucleophile. In addition, this reaction required harsh reaction conditions to allow for the removal of the benzyl groups from the mono-protected 1,2-diols, which were obtained by the allylation of aldehydes with the corresponding \( \gamma \)-alkoxyallylboronates. Furthermore, the route required for the synthesis of the dibenzyl acetal substrates showed limited substrate scope, as well as being a laborious and time-consuming procedure.

To address these issues, we focused on allyl acylals as alternative substrates for the copper-catalyzed boryl substitution reaction. Allyl acylals have been shown to be well suited to nucleophilic substitution reactions, such as palladium-catalyzed asymmetric alkylations or Lewis acid-catalyzed cyanation. We therefore expected that allyl acylals would be more reactive than allyl acetals toward nucleophilic boryl substitution reactions because the acetoxy group in the former is more electron withdrawing than the ether group in the latter, making the LUMO of the allyl acylal substrate lower in energy and more reactive toward a nucleophilic boryl copper intermediate.

![Scheme 1 Copper(I)-Catalyzed Enantioselective Boryl Substitution of Allyl Acylals](image-url)
Furthermore, acetyl groups can be removed under milder conditions than those required to remove ether groups, making this process more efficient than our previous method. Notably, a facile synthetic method has been reported for the direct construction of allyl acylals from aldehydes and acetanhydride using an acid catalyst.10

Herein, we report the enantioselective synthesis of α-chiral γ-acetoxyallylboronates using a chiral copper catalyst and bis(pinacolato)diboron \([\text{B}_2\text{(pin)}_2]\) as a boron source. Notably, this reaction was successfully applied to a wide range of allyl acylal substrates, including sterically hindered compounds, to give the desired products in good yields.

Initial optimization studies focused on the E/Z selectivity and enantioselectivity of the copper(I)-catalyzed boryl substitution of an allyl acylal to give the corresponding allylboronate. The reaction of acylal (–E)–1a with \(\text{B}_2\text{(pin)}_2\) as a base (1 equiv.) in THF or toluene afforded mixtures of the corresponding (E)- and (Z)-products (Table 1, entries 1 and 2).11 In our previous study involving the borylation of allyl acetals, we only ever observed the formation of the (E)-isomer as a single product, which we attributed to the substrate undergoing an \(\text{SN}_2'\) reaction mechanism with a fixed conformation because of the 1,3-allylic strain of the substrate (see Electronic Supplementary Information).5,12

| Table 1 Optimization of the Reaction Conditions for the Copper(I)-Catalyzed Enantioselective Boryl Substitution of Allyl Acylal (–Z)-1a

<table>
<thead>
<tr>
<th>Entry</th>
<th>Solvent</th>
<th>Ligand</th>
<th>Time (h)</th>
<th>E/Z</th>
<th>Yield (%)</th>
<th>ee (%)</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>THF</td>
<td>(R,R)-BenzP*</td>
<td>30</td>
<td>82:18</td>
<td>78</td>
<td>93</td>
</tr>
<tr>
<td>2</td>
<td>toluene</td>
<td>(R,R)-BenzP*</td>
<td>48</td>
<td>76:24</td>
<td>74</td>
<td>92</td>
</tr>
<tr>
<td>3</td>
<td>DMI</td>
<td>(R,R)-QuinoxP*</td>
<td>45</td>
<td>98:2</td>
<td>73</td>
<td>89</td>
</tr>
<tr>
<td>4*</td>
<td>DMI</td>
<td>(R,R)-QuinoxP*</td>
<td>24</td>
<td>90:10</td>
<td>30</td>
<td>–</td>
</tr>
<tr>
<td>5*</td>
<td>DMI</td>
<td>(R,R)-Segphos</td>
<td>24</td>
<td>87:13</td>
<td>23</td>
<td>–</td>
</tr>
<tr>
<td>6*</td>
<td>DMI</td>
<td>(R,R)-Me-Duphos</td>
<td>24</td>
<td>79:21</td>
<td>30</td>
<td>–</td>
</tr>
<tr>
<td>7*</td>
<td>DMI</td>
<td>(R,R)-BenzP*</td>
<td>24</td>
<td>trace</td>
<td>–</td>
<td>–</td>
</tr>
<tr>
<td>8*</td>
<td>DMI</td>
<td>(R,R)-BenzP*</td>
<td>28</td>
<td>&gt;90:1</td>
<td>79</td>
<td>95</td>
</tr>
</tbody>
</table>

*Reagents and conditions: CuCl (0.01 mmol), ligand (0.01 mmol), (Z)-1a (0.2 mmol), bis(pinacolato)diboron (0.85 mmol), and K(O-t-Bu) (0.05 mmol) in DMI (0.5 mL) at 0 °C. The E/Z selectivity was determined by GC. NMR yield. The ee values of the products were determined by HPLC analysis. The ee value of the major product was difficult to determine using HPLC analysis because both SiO₂ and chiral column chromatography resulted in an insufficient separation of the major product and the unconsumed substrate. 10 mol % of K(O-t-Bu) was used. 2.0 equiv of B₂(pin), and 1.5 equiv of K(O-t-Bu) were used. 0.5 mmol scale.

The use of 1,3-dimethyl-2-imidazolidinone (DMI) as a solvent provided the (E)-product with high E/Z selectivity and excellent enantioselectivity (73% yield, E/Z = 98:2, 89% ee; Table 1, entry 3). Several other chiral ligands, including (R,R)-QuinoxP*, (R)-Segphos and (R,R)-Me-Duphos, were also tested, but resulted in poor yields and E/Z selectivities (Table 1, entries 4–6). The amounts of base and B₂(pin): added to the reaction also had a considerable impact in the reactivity. For example, the use of a catalytic amount of K(O-t-Bu) (10 mol %) yielded a trace amount of the desired product, whereas the use of small excesses of K(O-t-Bu) (1.5 equiv.) and B₂(pin): (2.0 equiv.) resulted in high yield with excellent E/Z- and enantioselectivity (79% yield, E/Z = >99:1, 95% ee; Table 1, entry 8).13

As shown in Table 2, various α-chiral γ-acetoxyallylboronates were obtained in high yields and enantioselectivities under the optimized reaction conditions. Furthermore, several optically active products bearing an allyl substituent (e.g., R = Me, hexyl, methylcyclopentyl) were obtained in high yields and enantioselectivities ([S,E]-2b, 80% yield, 99% ee; (S,E)-2c, 80% yield, 98% ee; (S,E)-2d, 76% yield, 94% ee). This reaction also showed good functional group tolerance, as exemplified by the borylation of substrates bearing a silyl ether or acetoxy group, which proceeded in high yield and excellent enantioselectivity without any degradation of the functional groups ([S,E]-2e, 77% yield, 93% ee; (S,E)-2f, 60% yield, 93% ee; (S,E)-2g, 62% yield, 95% ee). α-Branch allyl acylals ([Z]-1h and [Z]-1i), which have steric congestion around their C=C bond, also reacted smoothly to afford the corresponding borylated products (58 and 42% yield, respectively), but the enantiopurities of these products were unfortunately low (59 and 55% ee, respectively), compared with 2b and 2c. The borylation of the (E)-substrate ([E]-1j) (E/Z = 95:5) proceeded with poor enantioselectivity to give the corresponding product with the opposite absolute configuration for the boron atom ([R,E]-2j, 81% yield, 74% ee, E/Z = 91:9).

| Table 2 Substrate Scope of the Copper(I)-Catalyzed Enantioselective Boryl Substitution of Allyl Acylal (–Z)-1a

<table>
<thead>
<tr>
<th>Entry</th>
<th>Solvent</th>
<th>Ligand</th>
<th>Time (h)</th>
<th>E/Z</th>
<th>Yield (%)</th>
<th>ee (%)</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>THF</td>
<td>(R,R)-BenzP*</td>
<td>30</td>
<td>82:18</td>
<td>78</td>
<td>93</td>
</tr>
<tr>
<td>2</td>
<td>toluene</td>
<td>(R,R)-BenzP*</td>
<td>48</td>
<td>76:24</td>
<td>74</td>
<td>92</td>
</tr>
<tr>
<td>3</td>
<td>DMI</td>
<td>(R,R)-QuinoxP*</td>
<td>45</td>
<td>98:2</td>
<td>73</td>
<td>89</td>
</tr>
<tr>
<td>4*</td>
<td>DMI</td>
<td>(R,R)-QuinoxP*</td>
<td>24</td>
<td>90:10</td>
<td>30</td>
<td>–</td>
</tr>
<tr>
<td>5*</td>
<td>DMI</td>
<td>(R,R)-Segphos</td>
<td>24</td>
<td>87:13</td>
<td>23</td>
<td>–</td>
</tr>
<tr>
<td>6*</td>
<td>DMI</td>
<td>(R,R)-Me-Duphos*</td>
<td>24</td>
<td>79:21</td>
<td>30</td>
<td>–</td>
</tr>
<tr>
<td>7*</td>
<td>DMI</td>
<td>(R,R)-BenzP*</td>
<td>24</td>
<td>trace</td>
<td>–</td>
<td>–</td>
</tr>
<tr>
<td>8*</td>
<td>DMI</td>
<td>(R,R)-BenzP*</td>
<td>28</td>
<td>&gt;90:1</td>
<td>79</td>
<td>95</td>
</tr>
</tbody>
</table>

*Reagents and conditions: CuCl (0.025 mmol), (R,R)-BenzP* (0.025 mmol), (Z)-1 (0.5 mmol), bis(pinacolato)diboron (0.85 mmol) and K(O-t-Bu) (0.6 mmol) in DMI (1.0 mL) at 0 °C. The E/Z values of the products were determined by HPLC analysis. 1.5 equiv of K(O-t-Bu) and 2.0 equiv of B₂(pin) were used. NMR yield. THF (0.3 mL) and DMI (0.3 mL) were used as a solvent. 10 mol % of CuCl and (R,R)-BenzP* were used. 1.0 mL was used as a solvent. 15 mol % of CuCl and (R,R)-BenzP* were used. 0.2 mmol scale.
We then proceeded to compare the reactivities of the allyl acetal and acylal substrates. Allyl acetal 3 and acylal 1k, which both have a tri-substituted alkene moiety, were selected as model substrates. The boryl substitution of acetal 3 provided only a trace amount of the corresponding borylated product (E)-4 in 4 h. Even after an extended reaction time (>24 h), the allyl acetal remained largely intact. The low conversion of the acetal substrate was attributed to steric hindrance around the C=C double bond of the substrate and the poor leaving group ability of the methyl ether group compared with the acetyl group. In contrast, the acylal substrate 1k reacted much more effectively than the acetal to give the borylated product in 49% yield after 24 h. These results therefore demonstrate that acylal substrates can undergo allyl substitution much more effectively than the corresponding acetals.

The allylboronates (S,E)-2f prepared using our new method were subsequently applied to the stereoselective allylation of aldehydes (Scheme 3). Octynal was successfully allylated with boronate (S,E)-2f in the presence of ZnBr₂, which was added as a Lewis acid catalyst. We previously found that ZnBr₂ is an efficient catalyst for enhancing the stereoselectivity and accelerating the reaction rate for the allylation of aldehydes.14,15 Pleasingly, this reaction provided the desired product in high stereoselectivity and good E/Z selectivity [(E)-anti-5, 68% yield, 100% ee, E/Z = 94:6].

In summary, we have developed a new method for the asymmetric synthesis of chiral γ-acetoxyallylboronates via the copper(I)-catalyzed boryl substitution of allyl acylals. The resulting allylboronates were used to achieve the highly stereoselective allylation of aldehydes. Furthermore, the acetyl groups of the allylated products were readily removed under basic conditions to give the corresponding 1,2-diols. This reaction therefore represents a useful method for the synthesis of 3-(E)-alkenyl-anti-1,2-diols.

Acknowledgment
This study was financially supported by the MEXT (Japan) program (Strategic Molecular and Materials Chemistry through Innovative CouplingReactions) of Hokkaido University, as well as the JSPS (KAKENHI Grant Numbers 15H03804 and 15K13633).

Supporting Information
YES

Primary Data
NO

References and Notes


a) A synthesis of (Z)-allyl dibenzyl acetal

The allyl acetal substrates were synthesized over several steps (a). The synthesis started from commercially available propargyl diethyl acetal, which was subjected to an acid-catalyzed acetal exchange reaction with benzyl alcohol to give the corresponding dibenzyl acetal. The subsequent deprotonation of the alkyne with an alkyl halide was typically low yielding. The exchange reaction generally proceeded in high yield, the subsequent alkylation of the terminal acetal substrate. Although the exchange reaction generally proceeded in high yield, the subsequent alkylation of the terminal acetal substrate. Despite the exchange reaction with benzyl alcohol to give the corresponding dibenzyl acetal, the subsequent deprotonation of the alkyne with an alkyl halide was typically low yielding.

b) A synthesis of (Z)-allyl acetal

In contrast to the acetal substrates, the acyl substrates were much easier to prepare (b). The formylation of a terminal alkyne, followed by the gem-diacetylation of the resulting carbonyl moiety provided the corresponding propargyl acylals in moderate to high yields. The subsequent Z-selective reduction of the alkyne moiety in these propargyl acylals yielded the desired allylic substrates.

(6) Representative examples of the routes used to synthesize the acetal and acyl substrates

(13) Typical procedure for the enantioselective boron substitution of allyl amines: Cat: (2.6 mg, 0.062 mmol), [(R)-BenzP*](7.2 mg, 0.026 mol), bis(pinacolato)diboron (254.8 mg, 1.00 mmol) and K(O-tert-Bu) (84.3 mg, 0.75 mmol) were placed in a screw-capped test tube in a glove box under an argon atmosphere. After the vial was sealed with a screw cap containing a teflon-coated rubber septum, the test tube was removed from the glove box and connected to a vacuum/nitrogen manifold through a needle. Then, dry DMI (1.0 ml) was added to the mixture via a syringe with stirring at room temperature. After 15–30 min, allyl (Z)-1a (129.5 mg, 0.5 mmol) was added to the reaction mixture with vigorous stirring at 0 °C. After the completion of the reaction was checked by GLC analysis, the mixture was directly filtered through a short silica-gel column with hexane/EtOAc (90:10) as the eluent. After removal of the solvents under reduced pressure, the mixture was subjected to flash chromatography (SiO2, hexane/Et2O = 100:0 to 90:10) as the internal standard. The crude product was purified with flash chromatography (SiO2, hexane/EtOAc = 90:10) to give the corresponding γ-acetoxyallylboronate (S)-2a (84.8 mg, 0.257 mmol, 52% isolated yield).