



Title	Tuning of the Optoelectronic Properties for Transparent Oxide Semiconductor ASnO(3) by Modulating the Size of A-Ions
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## Supporting information

### **Tuning of the optoelectronic properties for transparent oxide semiconductor $ASnO_3$ by modulating the size of A-ions**

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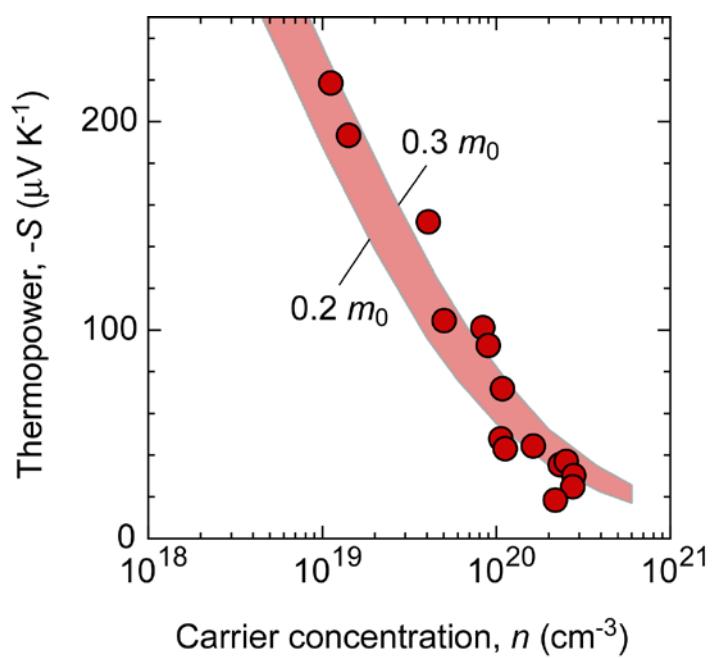
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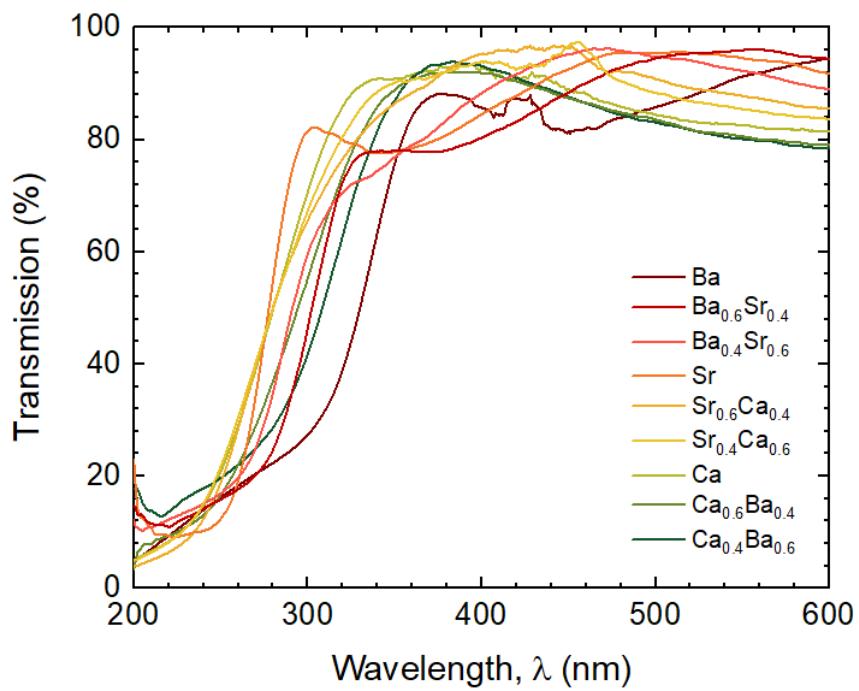
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**Figure S1.** Thermopower ( $-S$ ) of the resultant films as a function of carrier concentration ( $n$ ) at room temperature. The carrier effective mass ( $m^*$ ) of the films are in the range from 0.2 to  $0.3 m_0$ .



**Figure S2.** Optical transmission spectra of the resultant films (film thickness: 100nm).

Table S1. Crystallographic analyses of the resultant films.

Composition	Average ionic radius of A-site ion (Å)	$(a^2c)^{1/3}$ (Å)	$D$ (nm)
CaSnO <sub>3</sub>	1.34	3.949	20.2
Ca <sub>0.8</sub> Sr <sub>0.2</sub> SnO <sub>3</sub>	1.36	3.973	21.1
Ca <sub>0.6</sub> Sr <sub>0.4</sub> SnO <sub>3</sub>	1.38	4.002	23.5
Ca <sub>0.4</sub> Sr <sub>0.6</sub> SnO <sub>3</sub>	1.40	4.014	17.6
Ca <sub>0.2</sub> Sr <sub>0.8</sub> SnO <sub>3</sub>	1.42	4.031	31.1
SrSnO <sub>3</sub>	1.44	4.033	33.6
Sr <sub>0.8</sub> Ba <sub>0.2</sub> SnO <sub>3</sub>	1.47	4.052	22.4
Sr <sub>0.6</sub> Ba <sub>0.4</sub> SnO <sub>3</sub>	1.51	4.092	24.2
Sr <sub>0.4</sub> Ba <sub>0.6</sub> SnO <sub>3</sub>	1.54	4.132	27.7
Sr <sub>0.2</sub> Ba <sub>0.8</sub> SnO <sub>3</sub>	1.58	4.107	31.3
BaSnO <sub>3</sub>	1.61	4.138	24.3
Ba <sub>0.8</sub> Ca <sub>0.2</sub> SnO <sub>3</sub>	1.56	4.129	24.6
Ba <sub>0.6</sub> Ca <sub>0.4</sub> SnO <sub>3</sub>	1.50	4.090	27.0
Ba <sub>0.4</sub> Ca <sub>0.6</sub> SnO <sub>3</sub>	1.45	4.066	39.1
Ba <sub>0.2</sub> Ca <sub>0.8</sub> SnO <sub>3</sub>	1.39	3.998	22.7

$D$ : lateral coherence length

Table S2. Optical transmission in wavelength of 260 nm and bandgap of the resultant films.

Composition	Transmission, $T$ (%)	$E_g$ (eV)
CaSnO <sub>3</sub>	28.9	4.64
Ca <sub>0.8</sub> Sr <sub>0.2</sub> SnO <sub>3</sub>	29.2	4.66
Ca <sub>0.6</sub> Sr <sub>0.4</sub> SnO <sub>3</sub>	30.4	4.66
Ca <sub>0.4</sub> Sr <sub>0.6</sub> SnO <sub>3</sub>	28.1	4.57
Ca <sub>0.2</sub> Sr <sub>0.8</sub> SnO <sub>3</sub>	26.8	4.53
SrSnO <sub>3</sub>	15.1	4.43
Sr <sub>0.8</sub> Ba <sub>0.2</sub> SnO <sub>3</sub>	13.6	4.25
Sr <sub>0.6</sub> Ba <sub>0.4</sub> SnO <sub>3</sub>	18.0	4.17
Sr <sub>0.4</sub> Ba <sub>0.6</sub> SnO <sub>3</sub>	17.0	3.97
Sr <sub>0.2</sub> Ba <sub>0.8</sub> SnO <sub>3</sub>	16.6	3.77
BaSnO <sub>3</sub>	18.3	3.59
Ba <sub>0.8</sub> Ca <sub>0.2</sub> SnO <sub>3</sub>	23.3	3.73
Ba <sub>0.6</sub> Ca <sub>0.4</sub> SnO <sub>3</sub>	22.2	3.99
Ba <sub>0.4</sub> Ca <sub>0.6</sub> SnO <sub>3</sub>	21.8	4.35
Ba <sub>0.2</sub> Ca <sub>0.8</sub> SnO <sub>3</sub>	28.5	4.55

Table S3. Activation energy of the electrical conductivity for the resultant films.

Composition	$E_a$ (meV)
$\text{Ca}_{0.7}\text{Sr}_{0.3}\text{SnO}_3$	22.25
$\text{Ca}_{0.5}\text{Sr}_{0.5}\text{SnO}_3$	3.12
$\text{Ca}_{0.4}\text{Sr}_{0.6}\text{SnO}_3$	-1.49
$\text{Ca}_{0.3}\text{Sr}_{0.7}\text{SnO}_3$	-5.55
$\text{Ca}_{0.2}\text{Sr}_{0.8}\text{SnO}_3$	-5.78
$\text{Ca}_{0.1}\text{Sr}_{0.9}\text{SnO}_3$	-7.58
$\text{SrSnO}_3$	-20.11
$\text{Sr}_{0.8}\text{Ba}_{0.2}\text{SnO}_3$	-2.17
$\text{Sr}_{0.6}\text{Ba}_{0.4}\text{SnO}_3$	-2.65
$\text{Sr}_{0.4}\text{Ba}_{0.6}\text{SnO}_3$	-10.14
$\text{Sr}_{0.2}\text{Ba}_{0.8}\text{SnO}_3$	-11.14
$\text{BaSnO}_3$	-14.48
$\text{Ba}_{0.8}\text{Ca}_{0.2}\text{SnO}_3$	-5.61
$\text{Ba}_{0.6}\text{Ca}_{0.4}\text{SnO}_3$	-0.12
$\text{Ba}_{0.4}\text{Ca}_{0.6}\text{SnO}_3$	13.68