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1 **Title**

2 High throughput electrochemically driven metal microprinting with multicapillary droplet cell

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1 **Abstract**

2 A high speed and low-cost electrochemical additive manufacturing method for parallel
3 microprinting of metals is proposed. Multicapillary 3D printed solution flow type microdroplet
4 cells (Sf-MDC) with large capillary diameters (600 μm) are used to electrodeposit high-quality
5 Ni microstructures. Three Ni lines each having a width of around 900 μm and a thickness of
6 around 24.6 μm , 20.7 μm and 15.2 μm are simultaneously printed on Cu substrate. The
7 influence of temperature and scanning speed on the height of printed Ni lines is studied.
8 Independent control of solutions inside each capillary enables the simultaneous formation of
9 Ni-Cu-Ni and Cu-Ni-Cu microdots without any cross-mixing of solutions. By replacing 3D
10 printed inner capillaries (around 600 μm in diameter) with silica capillaries (around 100 μm in
11 diameter) inside the Sf-MDC, free standing 3D structures such as Ni microrods are successfully
12 fabricated. Localized electrodeposition using the Sf-MDC allows parallel process 3D printing
13 of metal microstructures for a wide range of applications.

14 **Keywords:** Electrodeposition; 3D printing; Droplet cell; Ni microrods; Electrochemistry

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1 **1. Introduction**

2 Robust, reproducible and cost-effective fabrication of metals at micrometer scale would enable
3 rapid development of advance electronic devices, microelectromechanical systems (MEMS)
4 and biomedical devices [1–6]. Conventionally, mask-based lithography is the main standard of
5 industries to produce micro- or nano- scale patterns. However, production of expensive masks,
6 time-consuming pre- and post-processing steps and difficulty in fabricating freestanding 3D
7 structures are some limitations of this technique [7,8]. One-step fabrication techniques such as
8 focused ion beam-induced deposition (FIB-ID), electrohydrodynamic printing (EHD), direct
9 ink-writing (DIW), laser induced forward transfer (LIFT), local electrophoretic deposition
10 (EPD), and laser induced photoreduction (LIPR) have also been used to manufacture metal
11 microstructures for various applications [9–13]. Organic material contamination and high
12 carbon content in FIB-ID processes may hinder the properties of printed structures. EHD and
13 DIW methods often need sintering, and removal of binders and solvent, which may result in
14 porous structures and reduce the mechanical strength of the printed metals [14]. Besides the
15 above mentioned techniques, robocasting [15], hybrid metal additive manufacturing [16],
16 metal jetting [17] and laser engineering net shaping [18] techniques have also been utilized for
17 2D and 3D fabrication of metal microstructures. However, high equipment costs and risk
18 factors associated with the use of high-power lasers and heat treatments limit the use of these
19 techniques.

20 In recent years, additive manufacturing of metal microstructures using electrochemical
21 microdroplet cells is gaining attention [19–30]. In this technique, micro- or nano-sized
22 capillaries are used for localized electrodeposition of metal ions from the solution. Template
23 free fabrication of complex structures without the use of thermal treatments makes this
24 technique simpler, safer, and cost-effective as compared to previously mentioned techniques.
25 Recently, Chen et al. has reported a low-cost electrochemical metal 3D printer capable of

1 printing highly conductive copper microstructures of 400 μm feature size [31]. Lei et al. has
2 reported a meniscus-confined electrodeposition approach for microprinting of copper lines on
3 gold, silicon and glass-substrates [32]. Lin et al. has also reported a nanopipette based metal
4 microprinting technique capable of fabricating high aspect ratio and high density 3D structures
5 with nanoscale precision [33]. Most of these techniques are limited to micro- or nano- scale
6 resolution and cannot be scaled up. Slow deposition rates (0.008 to $20.4 \mu\text{m}^3\text{s}^{-1}$) and printing
7 speeds (100 to 180 nms^{-1}) are the major challenges to overcome [31]. Research on the
8 fabrication of metal microstructures using large sized nozzles or capillaries is still in its early
9 stages. Technological advances in terms material diversity, high-throughput and large spatial
10 resolution are required for the industrial implementation of electrochemical droplet cell based
11 additive manufacturing.

12 High-throughput fabrication of metal microstructures can be achieved by combining multiple
13 capillaries in a parallel manner that are capable of simultaneous electrodeposition. The
14 deposition rate or printing speed can be increased by the use of large sized capillaries [34]. Few
15 studies have reported the applications of parallel, multi-capillary 3D printing for fast
16 fabrication of metal microstructures [29, 31–33]. These studies are limited to fabricating metal
17 microstructures in the range of 1 - $10 \mu\text{m}$ and offer slow deposition rates. The droplet cells are
18 fabricated manually using a series of complex processes. Problems such as capillary clogging
19 due to small diameter and the need to replace the cell after several hours of usage require
20 fabrication of new cells which consumes time. A single-step method to simultaneously
21 fabricate electrochemical droplet cells with multiple capillaries for high-throughput
22 electrodeposition would be important for a broad range of patterning applications [38].

23 Herein, a single step fabrication method is applied to fabricate a multicapillary 3D printed
24 solution flow type microdroplet cell (Sf-MDC) with capillary diameters of $600 \mu\text{m}$. Unlike
25 manually made droplet cells, multiple Sf-MDC are fabricated at once with minimum resin

1 consumption and shorter fabrication time. An electrochemical additive manufacturing system
2 is developed for the simultaneous electrodeposition of Ni microstrips on Cu substrate. The
3 influence of temperature and scanning speed on the volume of deposit is studied. An attempt
4 is made to simultaneously fabricate three separate Ni, Cu, Ni and Cu, Ni, Cu microdots on the
5 specimen. Free standing Ni microrods are simultaneously fabricated using the Sf-MDC.

6 **2. Experimental**

7 **Design and Fabrication of Sf-MDC**

8 Fig. 1 depicts the design and fabrication of Sf-MDC used for simultaneous electrodeposition.
9 Blender 2.79 CAD software was used to design the Sf-MDC. Fig. 1 (a) shows the top view of
10 Sf-MDC. The Sf-MDC consists of three solution inlets capable of separately supplying the
11 solution to the inner capillaries. The type of solution, its flowrate and concentration inside each
12 capillary can be independently controlled. The outer capillaries are used for the continuous
13 removal of excess solution through a common solution outlet. The diameter of inner and outer
14 capillaries is 0.6 mm and 1.2 mm, respectively. Due to the complexity of the design and the
15 limited resolution of 3D printer (Formlabs, Form 2), further decrease in the diameter of
16 capillaries resulted in the blockage of capillary heads inside the Sf-MDC. A photopolymer
17 resin (FLGPCL04, Formlabs Form 2) was used for fabricating the Sf-MDC. Fig. 1 (b) shows
18 the image of 3D printed cells attached to the build platform. Multiple cells can be printed at
19 once using a single-step fabrication method with minimum usage of polymer resin. The single
20 cell consumes only about 5 mL of resin and can be used several times. An important aspect of
21 this technique is its simplicity, high-throughput capability and low-cost, especially when
22 compared with manually made droplet cells and micropipette-based techniques.

23 **Materials**

24 Cu sheets (0.1x10x30 mm) were used as a substrate (working electrode). The sheets were
25 cleaned with ethanol and highly purified water. 0.31 M NiSO₄·6H₂O/ 0.4 M H₃BO₃ solution

1 was prepared for Ni electrodeposition. 1 M $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$ / 0.9 M H_2SO_4 was prepared for Cu
2 electrodeposition. The chemicals were purchased from Kanto chemicals, Japan. Pt wires (50
3 μm in diameter) were inserted inside the inner capillaries of Sf-MDC and connected with a
4 common Pt wire (500 μm in diameter) to use as anode. Cu sheets and Pt wires were purchased
5 from Nilaco Corporation, Japan.

6 **Procedures**

7 Fig. 2 schematically depicts the procedure of simultaneous electrodeposition using the Sf-MDC.
8 The substrate is placed on a computer controlled XYZ moving stage. The inlets of the Sf-MDC
9 are connected with three peristaltic pumps. The solution was supplied to the inner capillaries
10 at a flowrate of $1.5 \mu\text{m}^3$ to form droplets at the three capillary tips. The solution droplets were
11 allowed to come in contact with the specimen. Separation distance of around 50 μm was
12 maintained between the specimen and the inner capillary tips. A vacuum pressure of 85 KPa
13 was set to remove the solution from a common outlet with a vacuum pump. Continuous
14 removal of the used solution ensured uniform solution composition at the electrode-electrolyte
15 interface and gave Sf-MDC the ability to control the droplet size by controlling the vacuum
16 pressure. By following these parameters, reproducible meniscus could be formed under each
17 capillary during the metal microprinting process. A constant current of 500 μA is applied to
18 print Ni microstrips and Ni-Cu-Ni microdots on the Cu substrate. The moving speed and
19 temperature of the XYZ stage were varied to examine the influence of these parameters on
20 volume and height of the deposit. Free-standing Ni microrods were simultaneously fabricated
21 by inserting silica capillaries (diameter = 0.1 mm) instead of using 3D printed inner capillaries
22 (diameter = 0.6 mm) to better control the structure of the deposit and prevent the capillary
23 blockage issues. A constant current of 300 μA was applied and the stage was kept stationary
24 for first 30 min of electrodeposition process. The stage was then moved along z-axis at constant
25 speed ($0.25 \mu\text{m s}^{-1}$) to fabricate the microrods.

1 **Observations**

2 The surfaces and cross-sections of the specimens were observed with an optical microscope
3 (Wraycam, NOA630, Tokyo, Japan) and a scanning electron microscope equipped with an
4 energy-dispersive X-ray spectroscopy (EDS) analyzer (SEM, JEOL, JSL6510-LA, Tokyo,
5 Japan). For cross-sectional measurements, the specimens were cut using a cutting machine
6 (RC-120, Ritoku Co. Ltd., Japan) and ultrasonically cleaned with ethanol. The cross-sections
7 of the specimen were embedded inside the epoxy resin and allowed to cure for two days. The
8 exposed surface of the specimens was mechanically abraded with a series of SiC abrasive
9 papers under running water. The abraded specimens were washed with ethanol and highly
10 purified water before SEM observations.

11 **3. Results and discussions**

12 As the electrodeposition process is initiated with Sf-MDC, metal microprinting simultaneously
13 takes place under the droplets-confined microcapillaries. Precise movement of stage in X-, Y-
14 and Z- directions results in fabrication of desired microstructures. This process is similar to
15 fused deposition modeling (FDM) printing process, except the hot end of the extruder filled
16 with melted resin is replaced with droplet-confined microcapillaries, capable of simultaneous
17 electrodeposition. The electrochemical cell (Sf-MDC) enables the control of electrochemical
18 parameters such as current or voltage during the printing process.

19 Fig. 3 shows the optical, SEM and EDS elemental mapping images of Ni microstrips printed
20 on Cu substrate. The optical microscope image of the specimen (Fig. 3 (a)) shows three
21 uniformly sized Ni lines simultaneously printed on Cu substrate at two different scanning
22 speeds ($2 \mu\text{ms}^{-1}$ and $4 \mu\text{ms}^{-1}$). A slight decrease in the width of lines is observed as the scanning
23 speed is increased from $2 \mu\text{ms}^{-1}$ and $4 \mu\text{ms}^{-1}$. This may be attributed to the stability in the
24 movement of droplets at higher scanning speeds which allows the droplets to maintain the
25 surface tension equilibrium. The size of the droplets also depends on the wettability and contact

1 time, diameter of inner and outer capillaries and the distance between the specimen and the
2 capillary tips. The height of the printed Ni lines was estimated from the cross-sectional SEM
3 images. The specimen was cut from the middle of fabricated Ni lines. Fig. 3 (b) shows the EDS
4 mapping images of the printed Ni fabricated at $4 \mu\text{ms}^{-1}$. The deposition rate in the central region
5 is higher than at the edges, resulting in convex shaped cross-sections of the printed Ni lines.
6 This may happen due to higher current density in the central region of the droplets as compared
7 to the edges. The presence of capillary forces and surface tension at the electrode-electrolyte
8 interface allow the droplets to minimize their surface area. Maximum volume of the droplets
9 is thus confined in the central region, resulting in low solution resistance and higher current
10 density. However, very thin layer of droplets spreads outside the diameter of inner capillaries
11 (Fig. S1) and results in high solution resistance. Similar results have been obtained in previous
12 studies [28,32,39]. The shape and height of the simultaneously printed lines are different.
13 Several factors were taken into consideration to investigate the reason of this difference after
14 repeated experiments gave similar results. Since the Sf-MDC is fabricated by stereolithography
15 type 3D printer having a minimum resolution of $100 \mu\text{m}$, a slight variation in the dimensional
16 accuracy of capillaries could lead to difference in the IR drop inside each capillary. The use of
17 high-resolution 3D printers capable of accurately fabricating complex designs having multiple
18 micro- or nano-meter sized capillaries may solve this problem. Factors such as variation in
19 solution removal from each inner capillary through a common outer capillary, unequal distance
20 of droplets from the working electrode and irregular distribution of current through the inner
21 capillaries may also lead to a difference in the height of each printed line. In this study, a single
22 power supply was used to supply the current to the three inner capillaries through a common
23 counter electrode as shown in Fig. 2. The use of separate power supplies for each inner
24 capillary may result in uniform supply of current but with added cost. Fig. 3 (c) shows the
25 cross-sectional SEM images of the specimen. The height of the printed lines is around $24.6 \mu\text{m}$,

1 20.7 μm and 15.2 μm , respectively. The volume of each printed line can be calculated by
2 considering the shape of deposit as a trapezium (Fig. S2). The total volume of simultaneously
3 printed lines is around 0.133 mm^3 ($1.3 \times 10^8 \mu\text{m}^3$). The-deposition rate (printing speed) using
4 this method is around 800 $\mu\text{m}^3\text{s}^{-1}$, and width of deposit ranges from 800 μm to 1100 μm .
5 Simultaneous use of three capillaries with large capillary diameters (600 μm) significantly
6 enhanced the printing speed. It is worth mentioning that the highest printing speed previously
7 achieved using the droplet cell or meniscus confined electrodeposition technique ranges from
8 10^{-3} to $10^2 \mu\text{m}^3\text{s}^{-1}$ at feature size of 10^{-1} to $10^2 \mu\text{m}$, respectively [5,7].

9 The influence of temperature and scanning speed on the height of deposit and volume of
10 deposition is shown in Fig. 4. For simplicity, a single capillary is used to deposit Ni line on the
11 substrate. As the temperature of the specimen is varied from 20 $^\circ\text{C}$ to 65 $^\circ\text{C}$, the height of
12 deposit increases from 16.5 μm to 27 μm (Fig. 4 (a)). Although the height of deposit is mainly
13 influenced by the applied current (or voltage), increase in temperature may slightly increase
14 the coating thickness due shift in the activation energy of the reaction. This increases the overall
15 reaction kinetics, and more Ni ions are deposited at the cathode [40–42]. However, the quality
16 of electrodeposited lines may start to degrade at higher temperatures due to thermal stresses.
17 Fig. S3 shows the cross-sectional SEM of the electrodeposited lines fabricated at different
18 temperatures. Fig. 4 (b) shows the influence of moving speed on the volume of deposition. As
19 the scanning speed is increased from 2 μms^{-1} to 8 μms^{-1} , the deposition volume decreases from
20 around $4.8 \times 10^{-3} \text{mm}^3$ to $0.9 \times 10^{-3} \text{mm}^3$. Higher scanning speed leads to a decrease in the
21 volume of deposit due to less contact time between the droplet and the specimen. Fig. 5 shows
22 the ability of Sf-MDC to simultaneously use different electrodeposition solutions during the
23 metal microprinting process. Fig. 5 (a) shows the schematic of Sf-MDC during the Ni-Cu-Ni
24 electrodeposition process. Due to the use of separate pumps for each inner capillary of Sf-MDC,
25 the composition of electrodeposition solution and flowrate can be independently controlled.

1 Fig. 5 (b) shows the optical microscope image of the Ni-Cu-Ni microdots simultaneously
2 printed on Cu substrate. The diameter of Ni microdots is about 800 μm while the diameter of
3 Cu microdot is about 1000 μm , although all the inner capillaries have same diameter (600 μm).
4 Similar results were obtained in case of Cu-Ni-Cu microprinting (Fig. S4). This may happen
5 because the current efficiency of Cu is higher than the current efficiency of Ni. Since the Cu
6 substrate was used to deposit Cu from $\text{CuSO}_4\text{H}_2\text{SO}_4$ solution, some dissolution may happen
7 due to the presence of H_2SO_4 in the solution. Fig. 5 (c) shows the EDS elemental mapping
8 images of Ni-Cu-Ni microdots. Pure Ni and Cu microdots are obtained without any
9 contamination or cross-mixing of solutions. Cu substrate and printed Cu microdot could not be
10 distinguished from the EDS image. The printability of Cu microdot was examined from the
11 CSLM image and height profile of Cu deposit (Fig. S5). The microstructure of deposited Cu is
12 very different from the Cu substrate (Fig. S4). The surface SEM images of simultaneously
13 printed Ni and Cu microdots are shown in Fig. 5 (d). The Ni deposit consists of coarse structure
14 while Cu deposit consists of fine structure. The surface roughness and crystallite size increase
15 by raising the current density [41]. Using Sf-MDC, the optimum applied current is 50-100 μA
16 for Ni electrodeposition and 300 μA for Cu electrodeposition [28]. Since a single power supply
17 was used to supply current to multiple capillaries in this study, a current of 300 μA was set for
18 simultaneous electrodeposition. This resulted in higher surface roughness of Ni. Simultaneous
19 electrodeposition using different solutions can be applied for fabrication of resistance-
20 controlled metal micropatterns for electronics applications.

21 An attempt was made to fabricate freestanding 3D structures such as Ni microrods by Sf-MDC
22 based parallel metal microprinting process. Fig. 6 illustrates the possibility of fabricating Ni
23 microrods on Cu substrate using 3D printed Sf-MDC and conventional Sf-MDC. Fig. 6 (a)
24 shows that three microrods having different aspect ratios are simultaneously fabricated on the
25 specimen. Large capillary diameters (600 μm) and variation in the dimensional accuracy of

1 capillaries is one reason behind non-uniform fabrication of microrods. The use of high-
2 resolution 3D printers capable of accurately fabricating micro- or nano-meter sized capillaries
3 may solve this problem. Maintaining level build-up of materials under each capillary is another
4 important issue to be addressed. Use of separate power supplies for each capillary may also
5 solve this issue but with increased cost. Synchronizing the withdrawal speed of Sf-MDC with
6 the growth rate of deposit is critical in maintaining the level build-up of materials under the
7 capillaries. Higher withdrawal speed results in a decrease in the diameter of microrods. As the
8 withdrawal speed is increased further, the deposit breaks its contact with the droplet. On the
9 other hand, too slow withdrawal speed results in capillary clogging. Fig. 6 (b) shows that high
10 aspect ratio and high-quality Ni microrod can be fabricated by using conventional Sf-MDC.
11 Use of silica capillaries (diameter = 100 μm) instead of 3D printed capillaries (diameter = 600
12 μm) results in uniform fabrication of Ni microrod. Optimized withdrawal speed and smaller
13 capillary diameters result in smooth fabrication of 3D structures as compared to large sized
14 capillaries.

15 Fig. 7 shows the simultaneous fabrication of Ni microrods with modified Sf-MDC. In this case,
16 silica capillaries are used as inner capillaries. The optical microscope image (Fig. 7 (a)) shows
17 the formation of two Ni microrods with different sizes. The formed micro-rods have sufficient
18 strength to support their own weight. Due to the instabilities that occur during the process, one
19 droplet breaks its contact with the microrod while other continues to grow. Fig. 7 (b) shows
20 the SEM images of the fabricated microrods. The microrod-1 has a non-uniform diameter and
21 breaks its contact with the droplet after 2 h. The microrod-2 has a uniform diameter and
22 continues to grow after microrod-1 breaks its contact with the droplet. The magnified view of
23 SEM images shows that the microrods have different morphologies. Because of the variation
24 in solution removal from each inner capillary through a common outer capillary and irregular
25 distribution of current through the inner capillaries, uniform fabrication of equally sized

1 microrods using this technique is still a challenge. This study provides a step forward towards
2 the development of high speed and low-cost metal 3D printers. By properly optimizing the Sf-
3 MDC and changing the designs of capillaries, an array of more complicated 3D metal
4 microstructures than those presented in Fig. 6 and Fig. 7 can be produced with this parallel
5 printing process to significantly multiply the production.

6 **4. Conclusions**

7 3D printed multicapillary solution flow type microdroplet cells are used to successfully develop
8 a parallel, high speed and low-cost electrochemical metal microprinting system. Three Ni lines
9 having uniform width (around 900 μm) and different thickness (around 24.6 μm , 20.7 μm and
10 15.2 μm , respectively) are simultaneously printed on Cu substrate. The deposition rate or
11 printing speed achieved using this technique is around 800 $\mu\text{m}^3\text{s}^{-1}$, and feature size ranges from
12 800 μm to 1100 μm . An increase in deposition rate is observed at higher temperatures while a
13 decrease in deposition volume is observed at higher scanning speeds. Simultaneous printing of
14 multiple materials (Ni-Cu-Ni and Cu-Ni-Cu microdots) became possible with the Sf-MDC. By
15 using silica capillaries (diameter = 100 μm) as inner capillaries inside the Sf-MDC, two
16 freestanding and straight Ni microrods of different sizes were fabricated. This method can be
17 applied for simultaneous microprinting of resistance-controlled metal microstructures for
18 electronic applications.

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1 Captions

2 Fig.1 Top view of Sf-MDC obtained from CAD software, (a), and image showing multiple Sf-MDC
3 printed at once (b).

4

5 Fig.2 Schematic illustration of Ni microprinting using the Sf-MDC.

6

7 Fig. 3 Optical microscope image of Ni lines fabricated on Cu substrate, (a), EDS elemental mapping
8 images of the specimen cross-sections showing the size of printed Ni lines, (b), and cross-sectional
9 SEM image of the specimen showing thicknesses of the printed Ni lines, (c).

10

11 Fig. 4 Influence of temperature on the height of deposition, (a), and influence of moving speed on
12 volume of deposition, (b).

13

14 Fig. 5 Schematic of simultaneous electrodeposition using different solutions, (a), and corresponding
15 optical microscope image, (b), EDS elemental mapping, (c), and SEM images, (d).

16

17 Fig. 6 Fabrication of Ni microrods with 3D printed Sf-MDC, (a), and conventional Sf-MDC, (b).

18

19 Fig. 7 Electrodeposition with Sf-MDC and the corresponding optical microscope image of specimen,
20 (a), and SEM images of the Ni microrods (b).

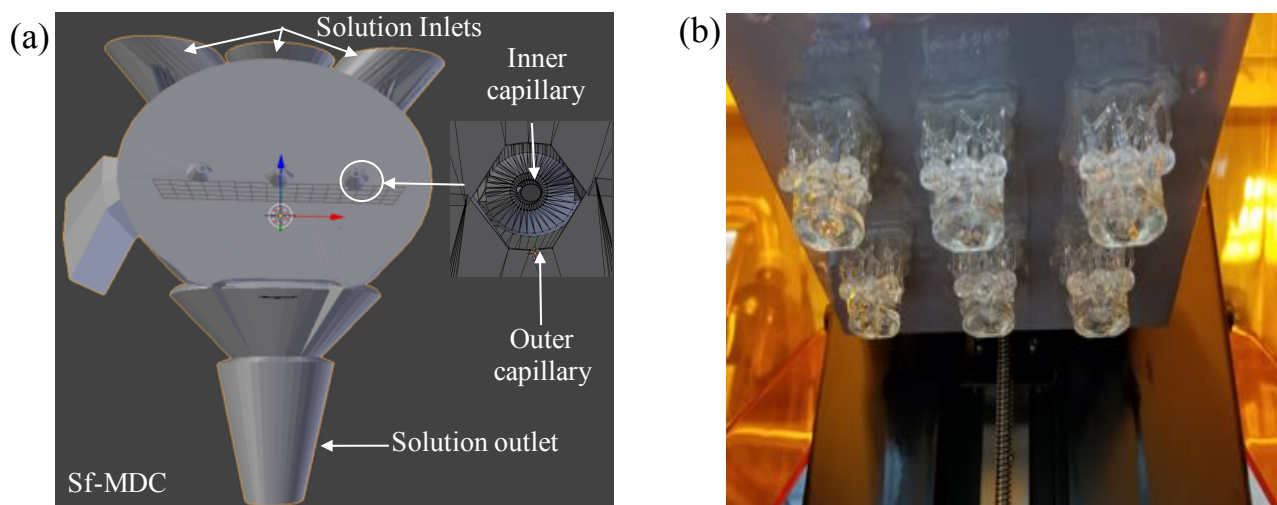
Figures

Fig.1 Top view of Sf-MDC obtained from CAD software, (a), and the image showing multiple Sf-MDC printed at once (b).

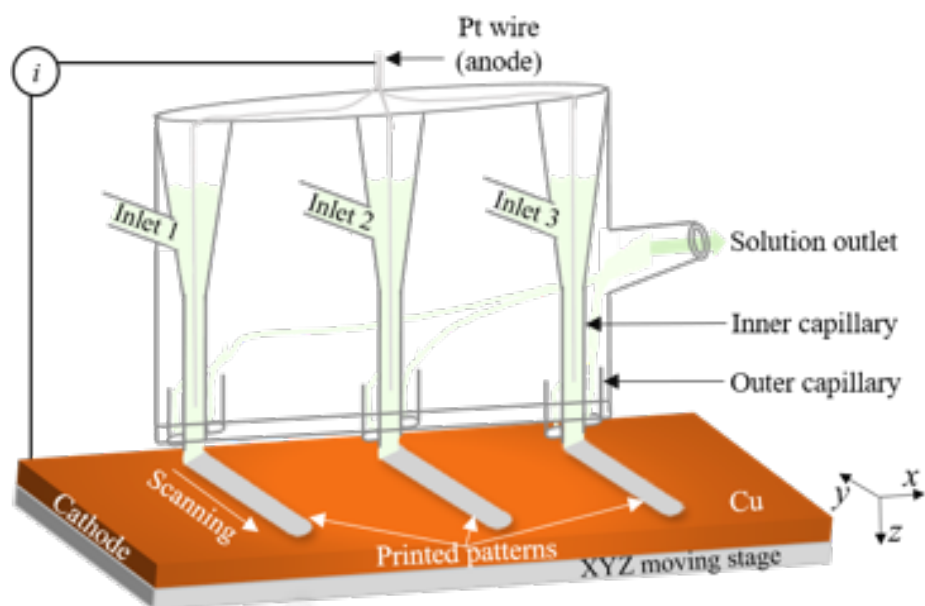


Fig.2 Schematic illustration of Ni microprinting using the Sf-MDC.

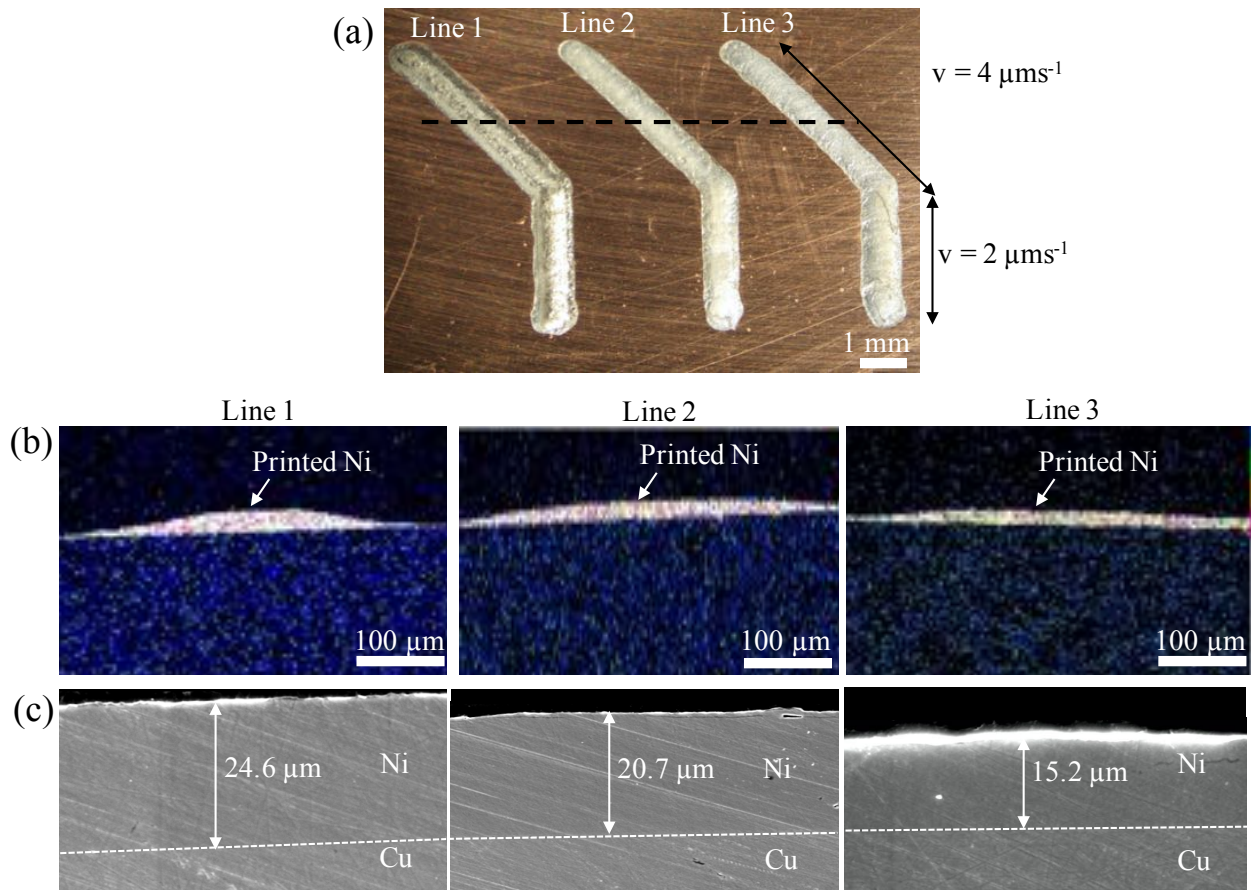


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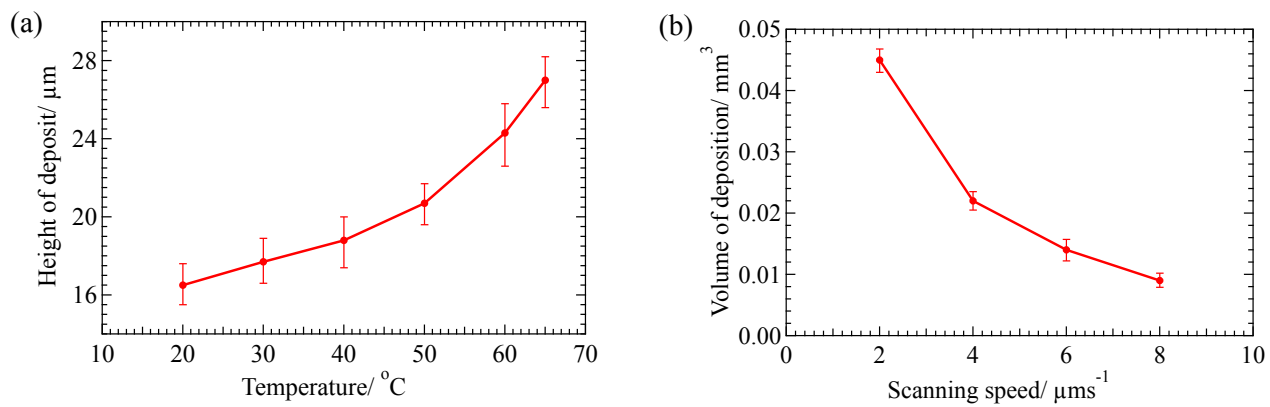


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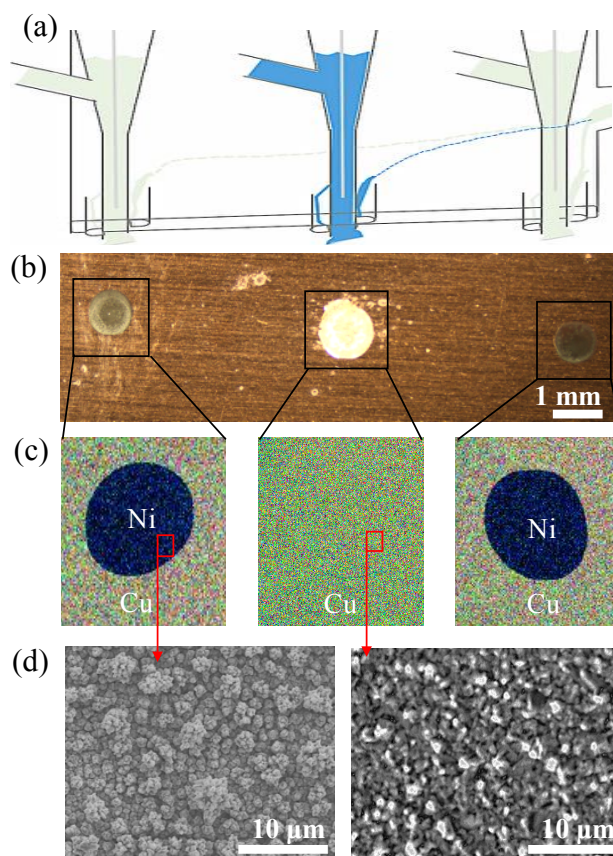


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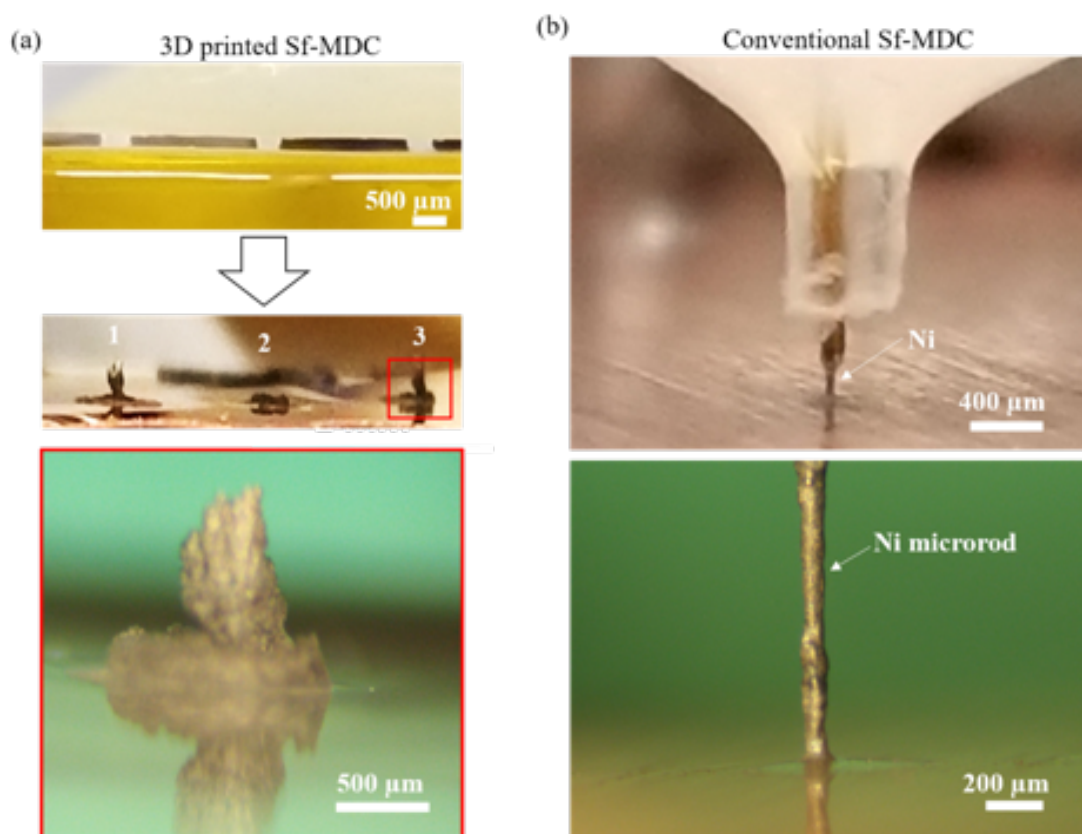


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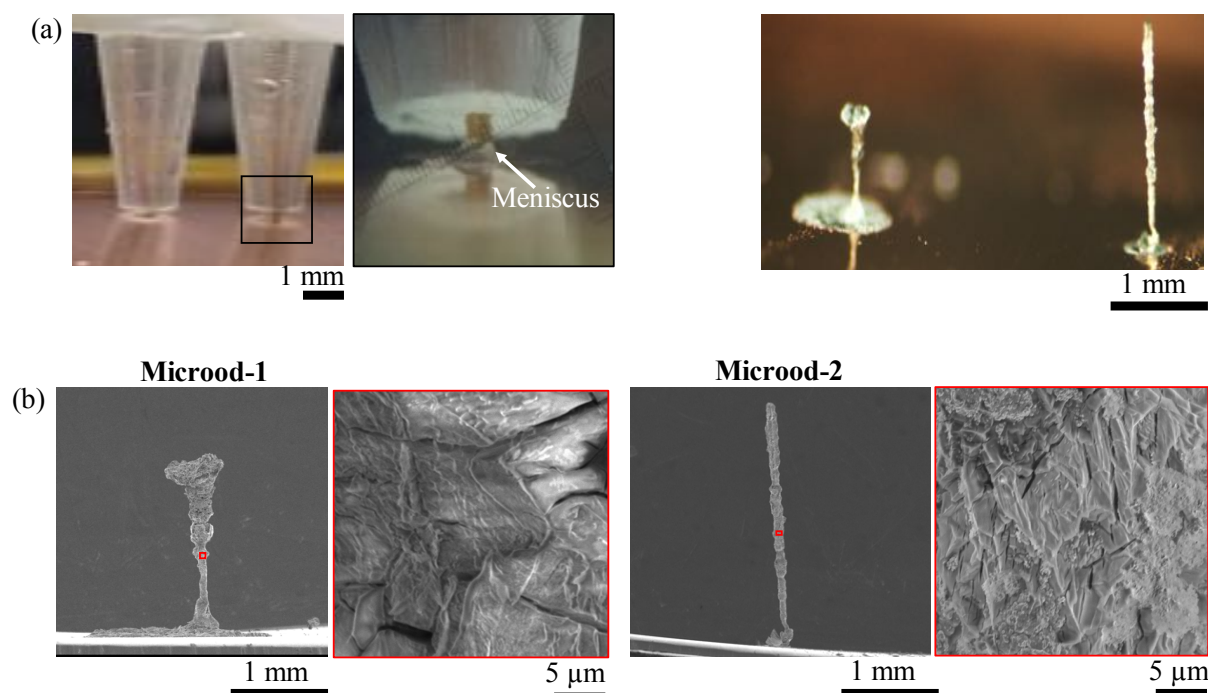


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