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Contribution of Ryugu-like material to Earth's volatile inventory by Cu and Zn isotopic analysis

Marine Paquet^{1*}, Frederic Moynier^{1*}, Tetsuya Yokoyama², Wei Dai¹, Yan Hu¹, Yoshinari Abe³, Jérôme Aléon⁴, Conel M. O'D. Alexander⁵, Sachiko Amari⁶, Yuri Amelin⁷, Ken-ichi Bajo⁸, Martin Bizzarro^{1,9}, Audrey Bouvier¹⁰, Richard W. Carlson⁵, Marc Chaussidon¹, Byeon-Gak Choi¹¹, Nicolas Dauphas¹², Andrew M. Davis¹², Tommaso Di Rocco¹³, Wataru Fujiya¹⁴, Ryota Fukai¹⁵, Ikshu Gautam², Makiko K. Haba², Yuki Hibiya¹⁶, Hiroshi Hidaka¹⁷, Hisashi Homma¹⁸, Peter Hoppe¹⁹, Gary R. Huss²⁰, Kiyohiro Ichida²¹, Tsuyoshi Iizuka²², Trevor R. Ireland²³, Akira Ishikawa², Motoo Ito²⁴, Shoichi Itoh²⁵, Noriyuki Kawasaki⁸, Noriko T. Kita²⁶, Kouki Kitajima²⁶, Thorsten Kleine²⁷, Shintaro Komatani²¹, Alexander N. Krot²⁰, Ming-Chang Liu²⁸, Yuki Masuda², Kevin D. McKeegan²⁸, Mayu Morita²¹, Kazuko Motomura²⁹, Izumi Nakai²⁹, Kazuhide Nagashima²⁰, David Nesvorný³⁰, Ann N. Nguyen³¹, Larry Nittler⁵, Morihiko Onose²¹, Andreas Pack¹³, Changkun Park³², Laurette Piani³³, Liping Qin³⁴, Sara S. Russell³⁵, Naoya Sakamoto³⁶, Maria Schönbächler³⁷, Lauren Tafla²⁸, Haolan Tang²⁸, Kentaro Terada³⁸, Yasuko Terada³⁹, Tomohiro Usui¹⁵, Sohei Wada⁸, Meenakshi Wadhwa⁴⁰, Richard J. Walker⁴¹, Katsuyuki Yamashita⁴², Qing-Zhu Yin⁴³, Shigekazu Yoneda⁴⁴, Edward D. Young²⁸, Hiroharu Yui⁴⁵, Ai-Cheng Zhang⁴⁶, Tomoki Nakamura⁴⁷, Hiroshi Naraoka⁴⁸, Takaaki Noguchi²⁴, Ryuji Okazaki⁴⁸, Kanako Sakamoto¹⁵, Hikaru Yabuta⁴⁹, Masanao Abe¹⁵, Akiko Miyazaki¹⁵, Aiko Nakato¹⁵, Masahiro Nishimura¹⁵, Tatsuaki Okada¹⁵, Toru Yada¹⁵, Kasumi Yogata¹⁵, Satoru Nakazawa¹⁵, Takanao Saiki¹⁵, Satoshi Tanaka¹⁵, Fuyuto Terui⁵⁰, Yuichi Tsuda¹⁵, Sei-ichiro Watanabe¹⁷, Makoto Yoshikawa¹⁵, Shogo Tachibana⁵¹, Hisayoshi Yurimoto⁸

Affiliations:

¹Université Paris Cité, Institut de physique du globe de Paris, CNRS; 75005 Paris, France

²Department of Earth and Planetary Sciences, Tokyo Institute of Technology; Tokyo 152-8551, Japan.

³Graduate School of Engineering Materials Science and Engineering, Tokyo Denki University; Tokyo 120-8551, Japan.

⁴Institut de Minéralogie, de Physique des Matériaux et de Cosmochimie, Sorbonne Université, Museum National d'Histoire Naturelle, CNRS UMR 7590, IRD; 75005 Paris, France.

⁵Earth and Planets Laboratory, Carnegie Institution for Science; Washington, DC, 20015, USA.

⁶McDonnell Center for the Space Sciences and Physics Department, Washington University; St. Louis, MO 63130, USA.

⁷Guangzhou Institute of Geochemistry, Chinese Academy of Sciences; Guangzhou, GD 510640, China.

⁸Natural History Sciences, IIL, Hokkaido University; Sapporo 001-0021, Japan.

⁹Centre for Star and Planet Formation, GLOBE Institute, University of Copenhagen; Copenhagen, K 1350, Denmark.

- 40 ¹⁰Bayerisches Geoinstitut, Universität Bayreuth; Bayreuth 95447, Germany.
- 41 ¹¹Department of Earth Science Education, Seoul National University; Seoul 08826, Republic
42 of Korea.
- 43 ¹²Department of the Geophysical Sciences and Enrico Fermi Institute, The University of
44 Chicago, 5734 South Ellis Avenue, Chicago 60637, USA.
- 45 ¹³Faculty of Geosciences and Geography, University of Göttingen; Göttingen, D-37077,
46 Germany.
- 47 ¹⁴Faculty of Science, Ibaraki University; Mito 310-8512, Japan.
- 48 ¹⁵ISAS/JSEC, JAXA; Sagami-hara 252-5210, Japan.
- 49 ¹⁶General Systems Studies, The University of Tokyo; Tokyo 153-0041, Japan.
- 50 ¹⁷Earth and Planetary Sciences, Nagoya University; Nagoya 464-8601, Japan.
- 51 ¹⁸Osaka Application Laboratory, SBUWDX, Rigaku Corporation; Osaka 569-1146, Japan.
- 52 ¹⁹Max Planck Institute for Chemistry; Mainz 55128, Germany.
- 53 ²⁰Hawai‘i Institute of Geophysics and Planetology, University of Hawai‘i at Mānoa;
54 Honolulu, HI 96822, USA.
- 55 ²¹Analytical Technology, Horiba Techno Service Co., Ltd.; Kyoto 601-8125, Japan.
- 56 ²²Earth and Planetary Science, The University of Tokyo; Tokyo 113-0033, Japan.
- 57 ²³School of Earth and Environmental Sciences, The University of Queensland; St Lucia QLD
58 4072, Australia.
- 59 ²⁴Kochi Institute for Core Sample Research, JAMSTEC; Kochi 783-8502, Japan.
- 60 ²⁵Earth and Planetary Sciences, Kyoto University; Kyoto 606-8502, Japan.
- 61 ²⁶Geoscience, University of Wisconsin-Madison; Madison, WI 53706, USA.
- 62 ²⁷Max Planck Institute for Solar System Research; 37077 Göttingen, Germany.
- 63 ²⁸Earth, Planetary, and Space Sciences, UCLA; Los Angeles, CA 90095, USA.
- 64 ²⁹Thermal Analysis, Rigaku Corporation; Tokyo 196-8666, Japan.
- 65 ³⁰Department of Space Studies, Southwest Research Institute, Boulder, CO 80302, USA.
- 66 ³¹Astromaterials Research and Exploration Science, NASA Johnson Space Center; Houston,
67 TX 77058, USA.
- 68 ³²Earth-System Sciences, Korea Polar Research Institute; Incheon 21990, Korea.
- 69 ³³Centre de Recherches Péetrographiques et Géo-chimiques, CNRS - Université de Lorraine;
70 54500 Nancy, France.
- 71 ³⁴CAS Key Laboratory of Crust-Mantle Materials and Environments, University of Science
72 and Technology of China, School of Earth and Space Sciences; Anhui 230026, China.
- 73 ³⁵Department of Earth Sciences, Natural History Museum; London, SW7 5BD, UK.
- 74 ³⁶IIL, Hokkaido University; Sapporo 001-0021, Japan.

- 75 ³⁷Institute for Geochemistry and Petrology, Department of Earth Sciences, ETH Zurich,
76 Zurich, Switzerland.
- 77 ³⁸Earth and Space Science, Osaka University; Osaka 560-0043, Japan.
- 78 ³⁹Spectroscopy and Imaging, Japan Synchrotron Radiation Research Institute; Hyogo 679-
79 5198 Japan.
- 80 ⁴⁰School of Earth and Space Exploration, Arizona State University; Tempe, AZ 85281, USA.
- 81 ⁴¹Geology, University of Maryland, College Park, MD 20742, USA.
- 82 ⁴²Graduate School of Natural Science and Technology, Okayama University; Okayama 700-
83 8530, Japan.
- 84 ⁴³Earth and Planetary Sciences, University of California; Davis, CA 95616, USA.
- 85 ⁴⁴Science and Engineering, National Museum of Nature and Science; Tsukuba 305-0005,
86 Japan.
- 87 ⁴⁵Chemistry, Tokyo University of Science; Tokyo 162-8601, Japan.
- 88 ⁴⁶School of Earth Sciences and Engineering, Nanjing University; Nanjing 210023, China.
- 89 ⁴⁷Department of Earth Science, Tohoku University; Sendai, 980-8578, Japan.
- 90 ⁴⁸Department of Earth and Planetary Sciences, Kyushu University; Fukuoka 819-0395, Japan.
- 91 ⁴⁹Earth and Planetary Systems Science Program, Hiroshima University; Higashi-Hiroshima,
92 739-8526, Japan.
- 93 ⁵⁰Kanagawa Institute of Technology; Atsugi 243-0292, Japan.
- 94 ⁵¹UTokyo Organization for Planetary and Space Science, University of Tokyo; Tokyo 113-
95 0033, Japan.

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(*corresponding author: paquet@ipgp.fr, moynier@ipgp.fr)

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106 **Abstract** (162 words)

107

108 **Abstract (162 words)**

109 Initial analyses showed that asteroid Ryugu's composition is close to CI (Ivuna-like)
110 carbonaceous chondrites – the chemically most primitive meteorites, characterized by near-solar
111 abundances for most elements. However, some isotopic signatures (e.g., Ti, Cr) overlap with other
112 carbonaceous chondrite (CC) groups, so the details of the link between Ryugu and the CI
113 chondrites are not fully clear yet. Here we show that Ryugu and CI chondrites have the same zinc
114 and copper isotopic composition. As the various chondrite groups have very distinct Zn and Cu
115 isotopic signatures, our results point at a common genetic heritage between Ryugu and CI
116 chondrites, ruling out any affinity with other CC groups. Since Ryugu's pristine samples match
117 the solar elemental composition for many elements, their Zn and Cu isotopic compositions likely
118 represent the best estimates of the solar composition. Earth's mass-independent Zn isotopic
119 composition is intermediate between Ryugu/CC and non-carbonaceous chondrites, suggesting a
120 contribution of Ryugu-like material to Earth's budgets of Zn and other moderately volatile
121 elements.

122

123 **Main text**

124 **Introduction.** Ivuna-type (CI) carbonaceous chondrites (CCs) have elemental abundances that
125 are the closest to the composition of the solar photosphere (e.g., [1]) (the exceptions being H, C,
126 N, O, Li and the noble gases). Thus, the CIs are key reference samples for investigating how early
127 Solar System processes shaped the compositions of the planets and their building blocks. The
128 return of the Hayabusa2 spacecraft in December 2020, after two successful touchdown and
129 sampling events on the Cb-type asteroid (162173) Ryugu^{2,3}, offers the unprecedented opportunity
130 to study volatile element fractionation processes using samples unaffected by terrestrial alteration,
131 in particular water incorporation. Initial studies on bulk chemical and isotopic compositions
132 revealed similarities between Ryugu and CIs⁴⁻⁷. However, Ryugu samples exhibit slightly higher
133 $\Delta^{17}\text{O}$ than the average from other CI samples, Orgueil and Ivuna, which is interpreted in terms of
134 original heterogeneity between small samples, or contamination of the meteorites by terrestrial
135 water incorporated into the structure of the alteration minerals (e.g., phyllosilicates, sulfates, iron
136 oxides and hydroxides), not adsorbed to the surfaces⁵. Similarly, although Ti and Cr isotope
137 compositions show that asteroid Ryugu formed in the CC reservoir, it was not possible to establish
138 a clear genetic link to just one of the CC groups because the Cr and Ti isotopic compositions of
139 Ryugu overlap not only with Ivuna-like (CI) but also with the Bencubbin-like (CB)⁵, Renazzo-like
140 (CR)⁵ and High-iron (CH) groups. However, the low volatile contents of these three groups of
141 meteorites, as well as the metal-rich nature of the CB and CH chondrites, argue against any affinity
142 with Ryugu^{5,6}. Thus, because CI chondrites and Ryugu samples share the same Ti and Cr
143 nucleosynthetic signatures, as well as similar mineralogical and elemental compositions^{5,6}, it has
144 been proposed that they formed contemporaneously from the same outer Solar System reservoir^{3,5-}
145 ⁷.

146

147 Material akin to carbonaceous chondrites such as Ryugu and the parent body of the CIs could
148 have delivered significant fractions of the moderately and highly volatile elements present in inner
149 Solar System planets (e.g., 8-13). Because Ryugu samples have been handled carefully to avoid
150 possible contamination, they are ideally suited to estimate the solar composition and assess the
151 contribution of these outer Solar System objects to the inventory of volatile elements in the
152 terrestrial planets. Highly volatile and moderately volatile elements are defined as elements with
153 50% condensation temperatures (T_c) <665 K and 665–1135 K, respectively, under canonical
154 nebular gas conditions at 10^{-4} bar (e.g., 14). Zinc and Cu are ideal elements to investigate volatility-
155 related processes, such as volatile element loss, during planetary accretion¹⁵, and are classified as
156 moderately volatile elements (MVE) (with T_c of 726 K and 1037 K, respectively)^{14,16}.
157 Carbonaceous chondrite groups display distinct Zn and Cu isotopic mass fractionation effects (e.g.,
158 17-21), defining a trend from CIs to CKs (CK = Karoonda-like), with the CIs being the most
159 volatile-rich and isotopically heaviest (for both Zn and Cu) of the CC groups. We have measured
160 the Zn and Cu isotopic compositions of Ryugu samples to (i) verify the link between Ryugu and
161 CI chondrites for moderately volatile elements, and (ii) assess the contribution of Ryugu-like
162 material to the inventory of moderately volatile elements in Earth.

163

164 **Results.** The Zn and Cu isotope compositions for four Ryugu samples (see **Methods section**),
165 together with six CC samples [Alais (CI), Allende A and B (CV), Murchison (CM), Orgueil (CI),
166 Tagish Lake (C2-ungrouped) and Tarda (C2-ungrouped)], were determined following the same
167 analytical protocol, and on the same samples as in [5] (**Table 1**). Most of these CC samples have
168 previously been characterized for their Zn and Cu isotopic composition^{17-19,21}, except Tarda. The

169 isotopic compositions are given as the permil deviations from the JMC-Lyon Zn and NIST
 170 SRM976 Cu standards:

$$171 \quad \delta^x\text{Zn} = \left[\frac{\left(\frac{x\text{Zn}}{64\text{Zn}}\right)_{\text{Sample}}}{\left(\frac{x\text{Zn}}{64\text{Zn}}\right)_{\text{JMC-Lyon}}} - 1 \right] \times 1000 \quad (1)$$

172 where x = 66, 67 and 68.

$$173 \quad \delta^{65}\text{Cu} = \left[\frac{\left(\frac{65\text{Cu}}{63\text{Cu}}\right)_{\text{Sample}}}{\left(\frac{65\text{Cu}}{63\text{Cu}}\right)_{\text{SRM976}}} - 1 \right] \times 1000 \quad (2).$$

174 During the course of this study, the two standards gave $\delta^{66}\text{Zn}$ of 0.00 ± 0.005 ‰ (2SE; n = 163;
 175 JMC-Lyon) and $\delta^{65}\text{Cu}$ of 0.00 ± 0.02 ‰ (2SE; n = 54; NIST SRM976). Zinc isotope measurements
 176 are also corrected for mass-dependent fractionation using the exponential law²², with the
 177 normalizing ratio of $^{68}\text{Zn}/^{64}\text{Zn}$ of 0.3856²³. Zinc isotopic anomalies are quantified using the epsilon
 178 notation relative to the JMC Lyon standard, as follows:

$$179 \quad \varepsilon^{66}\text{Zn}_c = \left[\frac{\left(\frac{66\text{Zn}}{64\text{Zn}}\right)_{\text{Sample}}}{\left(\frac{66\text{Zn}}{64\text{Zn}}\right)_{\text{JMC-Lyon}}} - 1 \right] \times 10^4 \quad (3)$$

180 where c is the normalizing ratio $^{68}\text{Zn}/^{64}\text{Zn}$.

181

182 The Ryugu samples span a very limited range of mass-dependent Zn isotopic compositions
 183 with $\delta^{66}\text{Zn}$ from $+0.41 \pm 0.06$ ‰ to $+0.45 \pm 0.02$ ‰ (2SD), with an average value of $+0.43 \pm 0.05$
 184 ‰ (2SD, n = 48) (**Fig. 1a and Table 1**). The $\delta^{65}\text{Cu}$ values for Ryugu samples range from $0.00 \pm$
 185 0.08 ‰ to $+0.09 \pm 0.05$ ‰, (average value of $+0.04 \pm 0.11$ ‰, n = 8, 2SD) (**Fig. 1b and Table 1**).
 186 Zinc and Cu abundances also span limited ranges from 338 ± 4 ppm to 383 ± 6 ppm (average 361
 187 ± 40 ppm, n = 4, 2SD), and from 133 ± 2 ppm to 168 ± 1 ppm (average 147 ± 37 ppm, n = 4, 2SD),
 188 respectively (**Table 2**). These values are higher than the abundances reported for any CI chondrite,

189 consistent with other element abundances for Ryugu samples relative to CIs (**Fig. 1, Table 2**),
190 although Ryugu samples have lower H₂O contents than CIs⁵. It is worth noting that the samples
191 from both landing sites show identical $\delta^{66}\text{Zn}$ values (**Fig. 1a**). In addition, the soluble organic
192 matter (SOM) extractions of samples C0107 and A0106, which was done prior to purification of
193 Zn and Cu, do not seem to have affected the Zn and Cu isotope compositions of the Ryugu samples
194 (see **Methods section**). All the CCs measured in this study have $\delta^{66}\text{Zn}$ and $\delta^{65}\text{Cu}$ values, as well
195 as Zn and Cu abundances, that are consistent with previous studies (e.g., [17-21]), except for the
196 Cu isotopic composition for Allende A, which is more similar to CO chondrites (**Figs. 1b and 2**).
197 We note, however, that until now only one other measurement of Allende has been reported in the
198 literature¹⁹, and so the difference could represent heterogeneity in the different analyzed fractions
199 of Allende in this study and in [19]. Similar sample heterogeneities for Cu isotopic compositions
200 have been reported for several fragments of Orgueil²¹. The Ryugu samples exhibit Zn and Cu
201 isotopic compositions that are similar to the Alais and Orgueil samples analyzed in this study (**Figs.**
202 **1 and 2**). This is consistent with previous work on the bulk elemental, isotopic, and mineralogical
203 properties of these samples, which reveal a genetic link between the Ryugu samples and CI
204 chondrites, implying formation from the same outer Solar System reservoir⁴⁻⁶.

205

206 **Discussion.** Earlier work has shown that bulk CC chondrites define negative correlations in
207 plots of $\delta^{66}\text{Zn}$ versus $1/\text{Zn}$ ^{17,18} and $\delta^{65}\text{Cu}$ versus $1/\text{Cu}$ ¹⁹. The variable degree of volatile element
208 depletion among the different CC groups reflect mixing of chemically and isotopically distinct
209 reservoirs during their accretion (e.g., 17-19,24,25). The CI chondrites, along with Ryugu, are the
210 least volatile depleted and isotopically heaviest (for both Zn and Cu) of the CC groups, while the
211 most volatile depleted chondrites, the CVs, are the isotopically lightest (this study; [17-21]) (**Figs.**

212 **1 and 2**). The CC trend is interpreted as the result of mixing of volatile-rich material enriched in
213 heavy Zn and Cu isotopes and volatile-poor material enriched in light Zn and Cu isotopes (e.g.,
214 [17-19]). Similar correlations are observed for other moderately volatile elements in CCs, such as
215 Te^{25} and $\text{Rb}^{26,27}$ and their associated isotope compositions, which is interpreted as mixing between
216 matrix (volatile-rich) and chondrules (volatile-poor) (e.g., [17-19,24,26]). Such mixing between
217 distinct CC reservoirs is also observed in the relationships between $\delta^{66}\text{Zn}$ and nucleosynthetic
218 isotope anomalies, such as $\epsilon^{54}\text{Cr}$ (parts per ten thousand mass-independent variations of the
219 $^{54}\text{Cr}/^{52}\text{Cr}$ ratio relative to a terrestrial standard) (**Fig. 3**) (e.g., [18]). Ryugu and the CIs have similar
220 Cu and Zn mass-dependent isotopic compositions and differ markedly from the CBs and CHs
221 (**Figs. 1, 2 and 3**). We can, therefore, exclude any genetic relationship to the CB or CH groups for
222 the Ryugu samples. A shared nucleosynthetic heritage between Ryugu and CI chondrites has been
223 established based on their identical Ti and Cr nucleosynthetic isotope anomalies^{5,6}. Our Zn and Cu
224 results show that this parentage extends to mass-dependent fractionation of moderately volatile
225 elements, strengthening the link between CI chondrites and Ryugu (**Figs. 1, 2 and 3**). The near
226 solar Zn and Cu relative abundances of the Ryugu samples, which are free of the potential
227 ambiguities of terrestrial alteration, suggests that the Zn and Cu isotopic compositions measured
228 for Ryugu and the CI chondrites most likely preserved the proto-Sun's composition¹ (**Fig. 1c**).

229

230 Our study also provides evidence for mass-independent Zn isotope variations ($\epsilon^{66}\text{Zn}$) in
231 Ryugu samples (**Fig. 4 and Table 1**). These Zn isotopic anomalies are consistent with previous
232 observations^{28,29}. While non-carbonaceous chondrites (NCs) display negative $\epsilon^{66}\text{Zn}$ (ordinary
233 chondrites: -0.21 ± 0.04 ‰, 2SE, $n = 12$, [28,29]; enstatite chondrites: -0.19 ± 0.08 ‰, $n = 8$,
234 [28,29]), the Ryugu samples and all CCs exhibit identical positive $\epsilon^{66}\text{Zn}$ within error ($+0.33 \pm 0.04$

235 ‰, 2SE, with $n = 7$ for Ryugu (**Table 1**) and $+0.39 \pm 0.07$ ‰, $n = 7$ for CCs, respectively) with
236 the value previously reported for CC of $+0.28 \pm 0.04$ ‰ (2SE, $n = 11$) [28,29]. It is worth noting
237 that the first replicate of sample C0108 (measured at 100 ppb Zn) has an $\epsilon^{66}\text{Zn}$ of -0.21 ± 0.17
238 ‰, whereas the second C0108 replicate (measured at 250 ppb Zn) has an $\epsilon^{66}\text{Zn}$ of $+0.35 \pm 0.10$
239 ‰ similar to all other Ryugu samples (see **Methods section**): the first replicate is thus considered
240 an outlier as it was analyzed at the lower concentration of 100 ppb and was excluded from the
241 mean value reported here. The reference geological material BHVO-2 and the Zn standard solution
242 IRMM 3702 measured during the first and second sessions have $\epsilon^{66}\text{Zn}$ (-0.07 ± 0.15 ‰, 2SE, n
243 $= 12$; $+0.02 \pm 0.11$ ‰, 2SE, $n = 7$, respectively), consistent within error with estimates for bulk
244 Earth ($+0.015 \pm 0.075$ ‰, 2SE, $n = 4$ [28] and -0.07 ± 0.013 ‰, 2SE, $n = 3$ [29]). There are no
245 known terrestrial processes which can mass-independently fractionate Zn isotopes. The positive
246 $\epsilon^{66}\text{Zn}$ values in the Ryugu samples, therefore, reinforce their genetic link with the CCs (**Fig. 4**).
247 Thus, the difference between the CCs and NCs, originally identified O and Cr isotope
248 compositions³⁰, and later with Ti, Ni and Mo anomalies³¹⁻³⁶, appears to also hold for Zn isotopes.

249

250 Because meteorites show a large variability of isotope anomalies³⁷ and planetary accretion
251 is stochastic³⁸⁻³⁹, it is likely that Earth's composition does not reflect accretion from a single type
252 of material, both in terms of isotopic and elemental compositions. Although enstatite chondrites
253 are isotopically closest to the Earth⁴⁰, their chemical signatures are extreme and deviate
254 substantially from the bulk composition of Earth. Possible mixtures of primitive and thermally
255 processed meteorites or their components (e.g., chondrules; [6, 41-43]) have been proposed to
256 explain the chemical and isotope composition of the Earth^{32,37,44-48}. In particular, the mass-
257 independent isotopic composition of Zn of the Earth appears intermediate between CCs and NCs.

258 Thus, our new data show that CC-like materials, potentially akin to Ryugu, have likely contributed
259 to the delivery of Zn and more generally the volatile elements to the Earth. Thus, following the
260 same approach as in [28,29], and using the average $\epsilon^{66}\text{Zn} = +0.33 \pm 0.04 \text{‰}$ for Ryugu, $+0.35 \pm$
261 0.13‰ for CI [this study, 28, 29], $-0.20 \pm 0.04 \text{‰}$ (2SE, $n = 20$) for NC (ordinary, enstatite from
262 [28,29]) and $-0.02 \pm 0.04 \text{‰}$ for the BSE (2SE, $n = 7$, [28,29]), the mass fraction of Ryugu- or
263 CI-derived Zn in the BSE is estimated 33.5% or 32.2%, respectively. Thus, we find that ~30% of
264 the terrestrial Zn derives from outer Solar System material, while the NC reservoir contributes to
265 ~70% to the terrestrial Zn. Then, to account for the Zn abundances of the accreting materials by
266 Earth, we estimate the mass fractions of NC and Ryugu-like or CI-like bodies accreted by Earth
267 using the Zn abundance of the BSE of $53.5 \pm 2.7 \text{ ppm}$ [49], and the $[\text{Zn}]_{\text{Ryugu}}$ of $361 \pm 40 \text{ ppm}$ [this
268 study] and $[\text{Zn}]_{\text{CI}}$ of $309 \pm 43.8 \text{ ppm}$ [this study, 17,18,21]. Thus, up to ~5% of Ryugu-like material
269 (or ~6% of CI-like material) might be needed to account for Earth's Zn isotopic composition,
270 consistent with estimations from previous studies on Zn isotopic anomalies^{28,29} and representing a
271 substantial contribution to the terrestrial budget of moderately volatile elements^{11,28,29}.

272

273 **Methods**

274 **Major and trace elements.** Zinc and copper isotopic compositions were measured in four
275 samples from the asteroid (162173) Ryugu [C0108, C0107, A0106-A0107 and A0106]. Fragments
276 A0106-A0107 and A0106 are coming from the first touchdown site, and C0108 and C0107 from
277 the second touchdown site⁴⁻⁶. Samples A0106-A0107 and C0108 were pristine samples, whereas
278 A0106 and C0107 were treated for Soluble Organic Matter (SOM) extraction before chemical
279 purification (see **Supplementary Table 1**). In addition, six CCs [Alais, Allende A, Allende B,
280 Murchison, Orgueil, Tagish Lake and Tarda] were processed following the exact same protocol as
281 the Ryugu samples and were analyzed as controls. For each sample, ~25 mg of powder of all the

282 samples were dissolved at Tokyo Institute of Technology. Elemental abundances were determined
283 using Inductively-Coupled-Plasma Mass-Spectrometry (ICP-MS): major and trace elements for
284 A0106-A0107 and C0108 samples are from [5]. After chemical analysis, the same sample
285 solutions were used to determine Zn and Cu isotopic compositions: Zn fractions were pre-
286 separated, as well as 3% of the bulk rock dissolution for Cu purification which represent about 80-
287 100 ng of Cu.

288
289 **Zinc and copper purification.** All the CC meteorites (except Tarda) have previously been
290 measured for Cu and Zn isotopic compositions and were analyzed as controls. Further chemical
291 purifications of Zn and Cu on the same sample aliquots were conducted at the Institut de Physique
292 du Globe de Paris, using the procedure described by [50] for Zn, and by [51,52] for Cu. For Zn,
293 samples were loaded in 1.5 mol.L^{-1} HBr on 50 μL of AG1-X8 (200-400 mesh) anion exchange
294 resin in home-made PTFE columns. Matrix elements were washed by further addition of 2 mL of
295 1.5 mol.L^{-1} HBr, and Zn was eluted using 2 mL of 0.5 mol.L^{-1} HNO_3 . The collected samples were
296 then evaporated to dryness. For Cu, samples were loaded in 1 mL of 7 mol.L^{-1} HCl on home-made
297 PTFE columns filled with 1.6 mL of AG-MP1 resin. After washing the resin with 8 mL of 7 mol.L^{-1}
298 HCl , the Cu was collected with 16 mL of 7 mol.L^{-1} HCl. Both procedures were repeated twice
299 to ensure clean Zn and Cu fraction. Procedural blank is $< 0.3 \text{ ng}$ of Zn, and 0.6 ng of Cu which is
300 negligible relative to the amount of Zn and Cu in the sample mass analyzed for the Ryugu samples
301 and the CCs.

302
303 **Zinc and Cu measurements.** Zinc and Cu isotope compositions were determined using a
304 Neptune Plus Multi-Collector Inductively-Coupled-Plasma Mass-Spectrometer (MC-ICP-MS) at

305 IPGP, using sample-standard bracketing for instrumental mass bias correction as in [50] for Zn
306 and [51,52] for Cu. Each replicate was analyzed 6–8 times for Zn and 1–5 times for Cu depending
307 on the amount of Cu available for each sample, and the reported errors are the two standard
308 deviations (2SD) of these repeated measurements. For the Zn measurements, the samples were
309 analyzed in two sessions with different sample solution concentrations: one at 100 ppb of Zn and
310 a second one at 250 ppb of Zn, with an uptake of 100 $\mu\text{L}\cdot\text{min}^{-1}$. For the Cu measurements, the
311 samples were analyzed in one session at the concentration of 30 ppb of Cu, with the same uptake.
312 The high purity of the final Zn fraction is needed to remove isobaric and non-isobaric interferences
313 from the signal. Interference on ^{64}Zn by ^{64}Ni is corrected by measuring the intensity of the ^{62}Ni ,
314 assuming natural abundances of Ni isotopes ($^{62}\text{Ni} = 3.63\%$; $^{64}\text{Ni} = 0.93\%$). No N_2 was used during
315 the measurements, as this results in high background on mass 68 from ArN_2 . No interference on
316 mass 68.5 from Ba^{2+} was detected during the sessions. The reference geological material BHVO-
317 2 and the Zn standard solution IRMM 3702 measured during the first and second sessions give
318 values consistent with the literature (e.g., [18,20]). However, during the first session, the Zn
319 fractions were measured at 100 ppb of Zn. All the Ryugu and CC samples had similar positive
320 $\epsilon^{66}\text{Zn}$, except for the first replicate of sample C0108 which had a negative value. In other words,
321 all the samples plot below the mass-dependent equilibrium fractionation line in a $\delta^{68}\text{Zn}$ against
322 $\delta^{66}\text{Zn}$ plot, whereas the C0108 replicate falls above it (**Supplementary Figure 1a**). This motivated
323 our second session of measurements on replicates at higher concentrations (250 ppb of Zn) to
324 ensure that the observed $\epsilon^{66}\text{Zn}$ were not analytical artifacts. The second replicate of sample C0108,
325 analyzed at 250 ppb of Zn, shows the same isotopic signature as the rest of the Ryugu sample set
326 and plots below the mass-dependent fractionation line with a positive $\epsilon^{66}\text{Zn}$ (**Supplementary**

327 **Figure 1b**). In the discussion and associated figures, only the second replicate of sample C0108 is
328 considered and represented.

329

330 **Data availability**

331 All data referred to in this article can be found in the tables or source data.

332

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340

341 **Author Contributions**

342 F.M., M.P. and T.Y. designed the project. H.Y. and T.Y coordinated the isotopic analyses of the
343 samples among members of the Hayabusa2-initial-analysis chemistry team. M.P. and T.Y.
344 processed the samples and separated the Zn and Cu from the matrix. M.P. measured the Zn and
345 Cu isotopic compositions. M.P. and F.M. wrote the first draft of the manuscript, with contributions
346 from T.Y., W. D., Y. H., Y. A., J. A., C. M. O'D. A., S. A., Y. A., K. B., M. B., A. B., R. W. C.,
347 M. C., B.-G. C., N. D., A. M. D., T. D. R., W. F., R. F., I. G., M. K. H., Y. H., H. Hi., H. Ho., P.
348 H., G. R. H., K. I., T. I., T. R. I., A. I., M. I., S. I., N. K., N. T. K., K. K., T. K., S. K., A. N. K.,
349 M.-C. L., Y. M., K. D. McK., M. M., K. M., I. N., K. N., D. N., A. N. N., L. N., M. O., A. P., C.
350 P, L. P., L. Q., S. S. R., N. S., M. S., L. T., H. T., K. T., Y. T., T. U., S. W, M. W., R. J. W., K. Y.,
351 Q.-Z. Y., S. Y., E. D. Y., H. Y., A.-C. Z., T. N., H. N., T. N., R. O., K. S., H. Y., M. A., A. M., A.
352 N., M. N., T. O., T. Y., K. Y., S. N., T. S., S. Tan., F. T., Y. T., S.-I. W., M. Y., S. Tac. and H. Y.

353

354 **Competing Interests Statement**

355 The authors declare no conflicts of interest.

356 **Tables**

357 Table 1. Zinc and copper stable isotope of Ryugu samples and carbonaceous chondrites.

| Sample | Type | n ^a (Zn) | $\delta^{66}\text{Zn}$ (‰) | 2SD ^b | $\delta^{67}\text{Zn}$ (‰) | 2SD ^b | $\delta^{68}\text{Zn}$ (‰) | 2SD ^b | $\epsilon^{66}\text{Zn}$ | 2SE ^b | n ^a (Cu) | $\delta^{65}\text{Cu}$ (‰) | 2SD ^b |
|--------------------------------|------|---------------------|----------------------------|------------------|----------------------------|------------------|----------------------------|------------------|--------------------------|------------------|---------------------|----------------------------|------------------|
| Ryugu | | | | | | | | | | | | | |
| C0108 ^c | | <u>8</u> | <u>0.44</u> | <u>0.05</u> | <u>0.70</u> | <u>0.10</u> | <u>0.91</u> | <u>0.09</u> | <u>-0.21</u> | <u>0.17</u> | 5 | 0.00 | 0.08 |
| Rpt | | 6 | 0.44 | 0.05 | 0.60 | 0.05 | 0.79 | 0.09 | 0.35 | 0.10 | | | |
| C0107 | | 8 | 0.42 | 0.06 | 0.57 | 0.10 | 0.75 | 0.09 | 0.37 | 0.11 | | | |
| Rpt | | 6 | 0.41 | 0.05 | 0.60 | 0.10 | 0.76 | 0.13 | 0.29 | 0.11 | | | |
| <i>Average site C</i> | | 20 | 0.43 | 0.05 | 0.62 | 0.13 | 0.81 | 0.17 | 0.34 | 0.06 | 5 | 0.00 | 0.08 |
| | | | | | | | | | | | | | |
| A0106-A107 | | 8 | 0.45 | 0.02 | 0.62 | 0.06 | 0.80 | 0.05 | 0.42 | 0.10 | 3 | 0.09 | 0.05 |
| Rpt | | 6 | 0.41 | 0.06 | 0.59 | 0.08 | 0.75 | 0.11 | 0.33 | 0.03 | | | |
| A0106 | | 8 | 0.43 | 0.03 | 0.63 | 0.07 | 0.80 | 0.06 | 0.25 | 0.14 | | | |
| Rpt | | 6 | 0.41 | 0.06 | 0.58 | 0.07 | 0.76 | 0.11 | 0.31 | 0.05 | | | |
| <i>Average site A</i> | | 28 | 0.43 | 0.04 | 0.61 | 0.08 | 0.78 | 0.08 | 0.33 | 0.05 | 3 | 0.09 | 0.05 |
| | | | | | | | | | | | | | |
| <i>Average Ryugu</i> | | 48 | 0.43 | 0.05 | 0.61 | 0.11 | 0.79 | 0.14 | 0.33 | 0.04 | 8 | 0.04 | 0.11 |
| | | | | | | | | | | | | | |
| Carbonaceous chondrites | | | | | | | | | | | | | |
| Orgueil | CI1 | 8 | 0.52 | 0.06 | 0.67 | 0.11 | 0.91 | 0.06 | 0.62 | 0.15 | | | |
| Rpt | | 8 | 0.52 | 0.09 | 0.71 | 0.09 | 0.96 | 0.17 | 0.33 | 0.14 | | | |
| <i>Average Orgueil</i> | | 16 | 0.52 | 0.08 | 0.69 | 0.11 | 0.94 | 0.14 | 0.52 | 0.12 | 2 | 0.06 | 0.11 |
| | | | | | | | | | | | | | |
| Alais | CI1 | 8 | 0.45 | 0.07 | 0.62 | 0.11 | 0.82 | 0.08 | 0.35 | 0.19 | | | |
| Rpt | | 6 | 0.44 | 0.07 | 0.62 | 0.10 | 0.80 | 0.12 | 0.35 | 0.09 | | | |
| <i>Average Alais</i> | | 14 | 0.45 | 0.07 | 0.62 | 0.10 | 0.80 | 0.10 | 0.35 | 0.11 | 2 | 0.17 | 0.003 |

| Paquet et al | | | Zn and Cu isotopes in Ryugu | | | | Nature Astronomy | | | | | | |
|----------------------------|------------|----|-----------------------------|------|------|------|------------------|------|-------|------|---|-------|------|
| Tagish Lake | C2- ung | 8 | 0.43 | 0.02 | 0.55 | 0.12 | 0.74 | 0.06 | 0.53 | 0.08 | | | |
| Rpt | | 6 | 0.41 | 0.08 | 0.54 | 0.08 | 0.73 | 0.15 | 0.41 | 0.07 | | | |
| <i>Average Tagish Lake</i> | | 14 | 0.42 | 0.06 | 0.55 | 0.10 | 0.73 | 0.11 | 0.48 | 0.06 | | | |
| Tarda | C2- ung | 8 | 0.46 | 0.06 | 0.58 | 0.09 | 0.80 | 0.06 | 0.50 | 0.12 | | | |
| Rpt | | 6 | 0.46 | 0.09 | 0.64 | 0.14 | 0.83 | 0.15 | 0.41 | 0.08 | | | |
| <i>Average Tarda</i> | | 14 | 0.46 | 0.07 | 0.60 | 0.13 | 0.81 | 0.11 | 0.46 | 0.08 | 1 | -0.40 | |
| Murchison | CM2 | 8 | 0.38 | 0.03 | 0.52 | 0.09 | 0.70 | 0.06 | 0.25 | 0.13 | | | |
| Rpt | | 6 | 0.37 | 0.08 | 0.52 | 0.09 | 0.67 | 0.15 | 0.35 | 0.16 | | | |
| <i>Average Murchison</i> | | 14 | 0.38 | 0.05 | 0.52 | 0.09 | 0.69 | 0.10 | 0.29 | 0.10 | 2 | -0.58 | 0.12 |
| Allende A | CV3 | 5 | 0.21 | 0.03 | 0.26 | 0.09 | 0.37 | 0.08 | 0.28 | 0.17 | | | |
| Rpt | | 6 | 0.22 | 0.10 | 0.28 | 0.16 | 0.35 | 0.20 | 0.47 | 0.15 | | | |
| <i>Average Allende A</i> | | 11 | 0.22 | 0.07 | 0.27 | 0.13 | 0.36 | 0.15 | 0.38 | 0.12 | 2 | -0.80 | 0.08 |
| Allende B | CV3 | 8 | 0.24 | 0.07 | 0.30 | 0.14 | 0.43 | 0.09 | 0.23 | 0.12 | | | |
| Rpt | | 6 | 0.23 | 0.06 | 0.31 | 0.12 | 0.38 | 0.18 | 0.32 | 0.13 | | | |
| <i>Average Allende B</i> | | 14 | 0.24 | 0.06 | 0.31 | 0.13 | 0.41 | 0.14 | 0.27 | 0.09 | | | |
| Reference materials | | | | | | | | | | | | | |
| IRMM3702 | | 7 | 0.22 | 0.04 | 0.26 | 0.05 | 0.43 | 0.06 | 0.02 | 0.11 | | | |
| BHVO2 | | 12 | 0.30 | 0.07 | 0.46 | 0.08 | 0.58 | 0.10 | -0.07 | 0.15 | | | |

^an is the number of measurements

^b2SD is 2 x standard deviation; 2SE is 2 x standard error

^cValue excluded from the averages for Ryugu

Numbers in italic represent averages for Ryugu and the carbonaceous chondrites for the Zn data

358

359 Table 2. Major and trace element compositions of Ryugu samples and carbonaceous chondrites.

| Sample | Type | Zn (ppm) | 2SD | Cu (ppm) | 2SD | Mg (ppm) | 2SD | Mg/Zn | Mg/Cu |
|--------------------------------|--------|------------|-----------|------------|-----------|---------------|--------------|------------|------------|
| Ryugu | | | | | | | | | |
| C0108* | | 352 | 4 | 156 | 2 | 104222 | 1153 | 296 | 668 |
| C0107 | | 383 | 6 | 168 | 1 | 98823 | 2890 | 258 | 589 |
| <i>Average site C</i> | | <i>368</i> | <i>45</i> | <i>162</i> | <i>17</i> | <i>101523</i> | <i>7635</i> | <i>277</i> | <i>628</i> |
| A0106-A107* | | 338 | 4 | 133 | 2 | 106866 | 1250 | 316 | 804 |
| A0106 | | 369 | 5 | 130 | 2 | 112899 | 3191 | 306 | 870 |
| <i>Average site A</i> | | <i>354</i> | <i>45</i> | <i>132</i> | <i>5</i> | <i>109883</i> | <i>8533</i> | <i>311</i> | <i>837</i> |
| <i>Average Ryugu</i> | | <i>361</i> | <i>40</i> | <i>147</i> | <i>37</i> | <i>101509</i> | <i>11700</i> | <i>294</i> | <i>733</i> |
| Carbonaceous chondrites | | | | | | | | | |
| Orgueil | CI1 | 288 | 4 | 131 | 1 | 91158 | 2706 | 3.0 | 696 |
| Alais | CI1 | 298 | 3 | 127 | 1 | 84683 | 2489 | 2.9 | 667 |
| Tagish Lake | C2-ung | 204 | 5 | 105 | 3 | 107210 | 1275 | 1.2 | 1021 |
| Tarda | C2-ung | 201 | 3 | 110 | 2 | 111296 | 2388 | 2.1 | 1012 |
| Murchison | CM2 | 174 | 2 | 117 | 1 | 111130 | 1257 | 1.1 | 950 |
| Allende A | CV3 | 110 | 2 | 97 | 2 | 141784 | 2461 | 1.7 | 1459 |
| Allende B | CV3 | 121 | 1 | 105 | 1 | 157609 | 4882 | 3.1 | 1501 |

*Abundances from [5]

Numbers in italic represent averages for Ryugu, and each of Ryugu sample site

360

361

362 **Figures captions**

363 **Figure 1. Zinc and copper elemental and isotopic compositions for Ryugu and carbonaceous**
364 **chondrites samples.** (a) $\delta^{66}\text{Zn}$ vs Mg/Zn , (b) $\delta^{65}\text{Cu}$ vs Mg/Cu and (c) Zn/Mg vs Cu/Mg that we
365 measured for the Ryugu samples (diamonds) and carbonaceous chondrites (large circles with
366 abbreviations: Or=Orgueil, Als=Alais, Tag=Tagish Lake, Tar=Tarda, Mur=Murchison,
367 All=Allende). Small circles are from the literature [17-21] for Zn and Cu isotope compositions
368 (and references therein for major and trace element compositions). The color identifies the type of
369 chondrite as described in the legend. The purple star in panel c represents the CI chondrite
370 composition from [1]. Data are presented as mean values with 2SD error bars, reported in Table
371 1.

372
373 **Figure 2. $\delta^{66}\text{Zn}$ vs $\delta^{65}\text{Cu}$ for Ryugu samples and carbonaceous chondrites.** Literature data are
374 from [17,19]. Same symbols as in Figure 1 for the samples analyzed in this study. Other chondrite
375 groups from the literature are reported directly on the figure. Data are presented as mean values
376 with 2SD error bars, reported in Table 1. For clarity, only the error bars of our measurements are
377 displayed. Error bars for literature data are not shown.

378
379 **Figure 3. $\delta^{66}\text{Zn}$ (this study) vs $\epsilon^{54}\text{Cr}$ [5,30,53-55] for Ryugu samples and carbonaceous**
380 **chondrites.** Literature data are from [17,18,20] for Zn isotope compositions, and from [56] for Cr
381 isotope compositions. The dark and light blue shaded areas correspond to the $\epsilon^{54}\text{Cr}$ ranges for
382 site A and site C, respectively, from [6]. Same symbols as in Figure 1 for the samples analyzed in
383 this study. Other chondrite groups from the literature are reported directly on the figure. Data are
384 presented as mean values with 2SD error bars, reported in Table 1.

385
386 **Figure 4. Variations of $\epsilon^{66}\text{Zn}$ among different groups of meteorites.** For comparison purposes,
387 only Ryugu (diamond) and CI (purple circles) samples measured in this study are represented
388 here. Literature data for carbonaceous chondrites ([28] (large symbols), [29] (small symbols)),
389 ordinary chondrites [28,29], enstatite chondrites [28,29], NC and CC iron chondrites [29],
390 ureilites [29] are shown with gray symbols. Bulk Silicate Earth: $+0.015 \pm 0.075 \text{‰}$, 2SE, $n = 4$
391 [28] and $-0.07 \pm 0.013 \text{‰}$, 2SE, $n = 3$ [29]. Data are presented as mean values with 2SE error
392 bars, reported in Table 1.

393 **References**

- 394 1. Lodders, K. (2021). Relative atomic solar system abundances, mass fractions, and atomic masses
395 of the elements and their isotopes, composition of the solar photosphere, and compositions of the
396 major chondritic meteorite groups. *Space Sci. Rev.*, 217(3), 1-33.
- 397 2. Morota, T. et al. (2020). Sample collection from asteroid (162173) Ryugu by Hayabusa2:
398 Implications for surface evolution. *Science*, 368(6491), 654-659.
- 399 3. Tachibana, S. et al. (2022). Pebbles and sand on asteroid (162173) Ryugu: in situ observation and
400 particles returned to Earth. *Science*, 375(6584), 1011-1016.
- 401 4. Yada, T. et al. (2022). Preliminary analysis of the Hayabusa2 samples returned from C-type
402 asteroid Ryugu. *Nat. Astron.*, 6(2), 214-220.
- 403 5. Yokoyama, T. et al. (2022). The first returned samples from a C-type asteroid show kinship to the
404 chemically most primitive meteorites. *Science*, 10.1126/science.abn7850.
- 405 6. Nakamura, E. et al. (2022). On the origin and evolution of the asteroid Ryugu: A comprehensive
406 geochemical perspective, *Proc. Japan Acad., Series B*, 98(6), 227-282.
- 407 7. Ito, M. et al. (2022). A pristine record of outer Solar System materials from asteroid Ryugu's
408 returned sample. *Nat. Astron.*, 1-9.
- 409 8. Wang, Z., & Becker, H. (2013). Ratios of S, Se and Te in the silicate Earth require a volatile-rich
410 late veneer. *Nature*, 499(7458), 328-331.
- 411 9. Savage, P. S. et al. (2015). Copper isotope evidence for large-scale sulphide fractionation during
412 Earth's differentiation. *Geochem. Perspect. Lett.*
- 413 10. Schönbächler, M., Carlson, R. W., Horan, M. F., Mock, T. D., & Hauri, E. H. (2010).
414 Heterogeneous accretion and the moderately volatile element budget of Earth. *Science*, 328(5980),
415 884-887.
- 416 11. Braukmüller, N., Wombacher, F., Funk, C., & Münker, C. (2019). Earth's volatile element
417 depletion pattern inherited from a carbonaceous chondrite-like source. *Nat. Geo.*, 12(7), 564-568.
- 418 12. Varas-Reus, M. I., König, S., Yierpan, A., Lorand, J. P., & Schoenberg, R. (2019). Selenium
419 isotopes as tracers of a late volatile contribution to Earth from the outer Solar System. *Nat.*
420 *Geo.*, 12(9), 779-782.
- 421 13. Kubik, E. et al. (2021). Tracing Earth's volatile delivery with tin. *J. Geophys. Res.: Solid*
422 *Earth*, 126(10), e2021JB022026.
- 423 14. Lodders, K. (2003). Solar system abundances and condensation temperatures of the
424 elements. *ApJ*, 591(2), 1220.
- 425 15. Day, J. M., & Moynier, F. (2014). Evaporative fractionation of volatile stable isotopes and their
426 bearing on the origin of the Moon. *Phil. Trans. R. S. A*, 372(2024), 20130259.
- 427 16. Schaefer, L., & Fegley Jr, B. (2010). Chemistry of atmospheres formed during accretion of the
428 Earth and other terrestrial planets. *Icarus*, 208(1), 438-448.
- 429 17. Luck, J. M., Othman, D. B., & Albarède, F. (2005). Zn and Cu isotopic variations in chondrites and
430 iron meteorites: early solar nebula reservoirs and parent-body processes. *Geochim. Cosmochim.*
431 *Acta*, 69(22), 5351-5363.
- 432 18. Pringle, E. A., Moynier, F., Beck, P., Paniello, R., & Hezel, D. C. (2017). The origin of volatile
433 element depletion in early solar system material: Clues from Zn isotopes in chondrules. *Earth*
434 *Planet. Sci. Lett.*, 468, 62-71.
- 435 19. Luck, J. M., Othman, D. B., Barrat, J. A., & Albarède, F. (2003). Coupled ⁶³Cu and ¹⁶O excesses
436 in chondrites. *Geochim. Cosmochim. Acta*, 67(1), 143-151.
- 437 20. Mahan, B., Moynier, F., Beck, P., Pringle, E. A., & Siebert, J. (2018). A history of violence:
438 Insights into post-accretionary heating in carbonaceous chondrites from volatile element
439 abundances, Zn isotopes and water contents. *Geochim. Cosmochim. Acta*, 220, 19-35.

- 440 21. Barrat, J. A., Zanda, B., Moynier, F., Bollinger, C., Liorzou, C., & Bayon, G. (2012). Geochemistry
441 of CI chondrites: Major and trace elements, and Cu and Zn isotopes. *Geochim. Cosmochim.*
442 *Acta*, 83, 79-92.
- 443 22. Rosman, K. J. R. (1972). A survey of the isotopic and elemental abundance of zinc. *Geochim.*
444 *Cosmochim. Acta*, 36(7), 801-819.
- 445 23. Russell, W. A., Papanastassiou, D. A., & Tombrello, T. A. (1978). Ca isotope fractionation on the
446 Earth and other solar system materials. *Geochim. Cosmochim. Acta*, 42(8), 1075-1090.
- 447 24. Clayton, R. N., & Mayeda, T. K. (1999). Oxygen isotope studies of carbonaceous
448 chondrites. *Geochim. Cosmochim. Acta*, 63(13-14), 2089-2104.
- 449 25. Hellmann, J. L., Hopp, T., Burkhardt, C., & Kleine, T. (2020). Origin of volatile element depletion
450 among carbonaceous chondrites. *Earth Planet. Sci. Lett.*, 549, 116508.
- 451 26. Pringle, E. A., & Moynier, F. (2017). Rubidium isotopic composition of the Earth, meteorites, and
452 the Moon: Evidence for the origin of volatile loss during planetary accretion. *Earth Planet. Sci.*
453 *Lett.*, 473, 62-70.
- 454 27. Nie, N. X. et al. (2021). Imprint of chondrule formation on the K and Rb isotopic compositions of
455 carbonaceous meteorites. *Sci. Adv.*, 7(49), eabl3929.
- 456 28. Savage, P. S., Moynier, F., & Boyet, M. (2022). Zinc isotope anomalies in primitive meteorites
457 identify the outer solar system as an important source of Earth's volatile inventory. *Icarus*, 386,
458 115172.
- 459 29. Steller, T., Burkhardt, C., Yang, C., & Kleine, T. (2022). Nucleosynthetic zinc isotope anomalies
460 reveal a dual origin of terrestrial volatiles. *Icarus*, 386, 115171.
- 461 30. Trinquier, A., Birck, J. L., & Allègre, C. J. (2007). Widespread ⁵⁴Cr heterogeneity in the inner solar
462 system. *ApJ*, 655(2), 1179.
- 463 31. Trinquier, A., Elliott, T., Ulfbeck, D., Coath, C., Krot, A. N., & Bizzarro, M. (2009). Origin of
464 nucleosynthetic isotope heterogeneity in the solar protoplanetary disk. *Science*, 324(5925), 374-
465 376.
- 466 32. Warren, P. H. (2011). Stable-isotopic anomalies and the accretionary assemblage of the Earth and
467 Mars: A subordinate role for carbonaceous chondrites. *Earth Planet. Sci. Lett.*, 311(1-2), 93-100.
- 468 33. Budde, G., Burkhardt, C., Brennecka, G. A., Fischer-Gödde, M., Kruijer, T. S., & Kleine, T. (2016).
469 Molybdenum isotopic evidence for the origin of chondrules and a distinct genetic heritage of
470 carbonaceous and non-carbonaceous meteorites. *Earth Planet. Sci. Lett.*, 454, 293-303.
- 471 34. Kruijer, T. S., Burkhardt, C., Budde, G., & Kleine, T. (2017). Age of Jupiter inferred from the
472 distinct genetics and formation times of meteorites. *Proc. Natl. Acad. Sci.*, 114(26), 6712-6716.
- 473 35. Burkhardt, C., Dauphas, N., Hans, U., Bourdon, B., & Kleine, T. (2019). Elemental and isotopic
474 variability in solar system materials by mixing and processing of primordial disk
475 reservoirs. *Geochim. Cosmochim. Acta*, 261, 145-170.
- 476 36. Nanne, J. A., Nimmo, F., Cuzzi, J. N., & Kleine, T. (2019). Origin of the non-carbonaceous-
477 carbonaceous meteorite dichotomy. *Earth Planet. Sci. Lett.*, 511, 44-54.
- 478 37. Dauphas, N., Chen, J. H., Zhang, J., Papanastassiou, D. A., Davis, A. M., & Travaglio, C. (2014).
479 Calcium-48 isotopic anomalies in bulk chondrites and achondrites: Evidence for a uniform isotopic
480 reservoir in the inner protoplanetary disk. *Earth Planet. Sci. Lett.*, 407, 96-108.
- 481 38. Chambers, J. E. (2004). Planetary accretion in the inner Solar System. *E Earth Planet. Sci.*
482 *Lett.*, 223(3-4), 241-252.
- 483 39. Walsh, K. J., Morbidelli, A., Raymond, S. N., O'Brien, D. P., & Mandell, A. M. (2011). A low
484 mass for Mars from Jupiter's early gas-driven migration. *Nature*, 475(7355), 206-209.
- 485 40. Javoy, M. et al. (2010). The chemical composition of the Earth: Enstatite chondrite models. *Earth*
486 *Planet. Sci. Lett.*, 293(3-4), 259-268.
- 487 41. Morbidelli, A., Libourel, G., Palme, H., Jacobson, S. A., & Rubie, D. C. (2020). Subsolar Al/Si and
488 Mg/Si ratios of non-carbonaceous chondrites reveal planetesimal formation during early
489 condensation in the protoplanetary disk. *Earth Planet. Sci. Lett.*, 538, 116220.

- 490 42. Frossard, P., Guo, Z., Spencer, M., Boyet, M., & Bouvier, A. (2021). Evidence from achondrites
491 for a temporal change in Nd nucleosynthetic anomalies within the first 1.5 million years of the inner
492 solar system formation. *Earth Planet. Sci. Lett.*, 566, 116968.
- 493 43. Alexander, C. M. D. (2022). An exploration of whether Earth can be built from chondritic
494 components, not bulk chondrites. *Geochim. Cosmochim. Acta*, 318, 428-451.
- 495 44. Lodders, K. (2000). An oxygen isotope mixing model for the accretion and composition of rocky
496 planets. *From dust to terrestrial planets*, 341-354.
- 497 45. Schiller, M., Bizzarro, M., & Fernandes, V. A. (2018). Isotopic evolution of the protoplanetary disk
498 and the building blocks of Earth and the Moon. *Nature*, 555(7697), 507-510.
- 499 46. Schiller, M., Bizzarro, M., & Siebert, J. (2020). Iron isotope evidence for very rapid accretion and
500 differentiation of the proto-Earth. *Sci. Adv.*, 6(7), eaay7604.
- 501 47. Mezger, K., Maltese, A., & Vollstaedt, H. (2021). Accretion and differentiation of early planetary
502 bodies as recorded in the composition of the silicate Earth. *Icarus*, 365, 114497.
- 503 48. Johansen, A. et al. (2021). A pebble accretion model for the formation of the terrestrial planets in
504 the Solar System. *Sci. Adv.*, 7(8), eabc0444.
- 505 49. Sossi, P. A., Nebel, O., O'Neill, H. S. C., & Moynier, F. (2018). Zinc isotope composition of the
506 Earth and its behaviour during planetary accretion. *Chem. Geol.*, 477, 73-84.
- 507 50. van Kooten, E., & Moynier, F. (2019). Zinc isotope analyses of singularly small samples (< 5 ng
508 Zn): investigating chondrule-matrix complementarity in Leoville. *Geochim. Cosmochim.*
509 *Acta*, 261, 248-268.
- 510 51. Moynier, F., Creech, J., Dallas, J., & Le Borgne, M. (2019). Serum and brain natural copper stable
511 isotopes in a mouse model of Alzheimer's disease. *Sci. Rep.*, 9(1), 1-7.
- 512 52. Moynier, F. et al. (2020). Copper and zinc isotopic excursions in the human brain affected by
513 Alzheimer's disease. *Alzheimer's & Dementia: DADM*, 12(1), e12112.
- 514 53. Petitat, M., Birck, J. L., Luu, T. H., & Gounelle, M. (2011). The chromium isotopic composition
515 of the ungrouped carbonaceous chondrite Tagish Lake. *ApJ.*, 736(1), 23.
- 516 54. Schoenberg, R. et al. (2016). The stable Cr isotopic compositions of chondrites and silicate
517 planetary reservoirs. *Geochim. Cosmochim. Acta*, 183, 14-30.
- 518 55. Dey, S., Yin, Q. Z., & Zolensky, M. (2021, March). Exploring the Planetary Genealogy of Tarda—
519 A Unique New Carbonaceous Chondrite. In *52nd Lunar and Planetary Science Conference* (No.
520 2548, p. 2517).
- 521 56. Zhu, K. et al. (2021). Chromium isotopic insights into the origin of chondrite parent bodies and the
522 early terrestrial volatile depletion. *Geochim. Cosmochim. Acta*, 301, 158-186.
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