Title	Growth of nitrite-oxidizing Nitrospira and ammonia-oxidizing Nitrosomonas in marine recirculating trickling biofilter reactors
Author(s)	Oshiki, Mamoru; Netsu, Hirotoshi; Kuroda, Kyohei; Narihiro, Takashi; Fujii, Naoki; Kindaichi, Tomonori; Suzuki, Yoshiyuki; Watari, Takahiro; Hatamoto, Masashi; Yamaguchi, Takashi; Araki, Nobuo; Okabe, Satoshi
Citation	Environmental microbiology, 24(8), 3735-3750 https://doi.org/10.1111/1462-2920.16085
Issue Date	2022-08
Doc URL	http://hdl.handle.net/2115/90133
Rights	This is the peer reviewed version of the following article: Oshiki, M., Netsu, H., Kuroda, K., Narihiro, T., Fujii, N., Kindaichi, T., Suzuki, Y., Watari, T., Hatamoto, M., Yamaguchi, T., Araki, N. and Okabe, S. (2022), Growth of nitrite-oxidizing Nitrospira and ammonia-oxidizing Nitrosomonas in marine recirculating trickling biofilter reactors. Environ Microbiol, 24: 3735-3750, which has been published in final form at https://doi.org/10.1111/1462-2920.16085. This article may be used for non-commercial purposes in accordance with Wiley Terms and Conditions for Use of Self-Archived Versions. This article may not be enhanced, enriched or otherwise transformed into a derivative work, without express permission from Wiley or by statutory rights under applicable legislation. Copyright notices must not be removed, obscured or modified. The article must be linked to Wiley 's version of record on Wiley Online Library and any embedding, framing or otherwise making available the article or pages thereof by third parties from platforms, services and websites other than Wiley Online Library must be prohibited.
Туре	article (author version)
Additional Information	There are other files related to this item in HUSCAP. Check the above URL.
File Information	cleaned_Manuscript_file_DHS_MO.pdf



- 1 For resubmission to *Environmental microbiology* (EMI-2022-0180)
- 2 Growth of nitrite-oxidizing Nitrospira and ammonia-oxidizing
- 3 Nitrosomonas in marine recirculating trickling biofilter reactors
- 4 Mamoru Oshiki^{1, 2*}, Hirotoshi Netsu^{2, 3}, Kyohei Kuroda⁴, Takashi Narihiro⁴,
- 5 Naoki Fujii⁵, Tomonori Kindaichi⁵, Yoshiyuki Suzuki², Takashiro Watari³,
- 6 Masashi Hatamoto³, Takashi Yamaguchi⁶, Nobuo Araki² & Satoshi Okabe¹
- 7 Division of Environmental Engineering, Faculty of Engineering, Hokkaido University,
- 8 North 13, West 8, Kita-ku, Sapporo, Hokkaido 060-8628, Japan
- Department of Civil Engineering, National Institute of Technology, Nagaoka College, 888
 Nishikatakaimachi, Nagaoka, Niigata 940-8532, Japan.
- Department of Environmental Systems Engineering, Nagaoka University of Technology,
 1603-1 Kamitomioka, Nagaoka, Niigata 940-2188, Japan.
- ⁴Bioproduction Research Institute, National Institute of Advanced Industrial Science and
 Technology (AIST), 2-17-2-1 Tsukisamu-Higashi, Toyohira-ku, Sapporo, Hokkaido,
- 15 062-8517 Japan
- ⁵ Department of Civil and Environmental Engineering, Graduate School of Engineering,
- 17 Hiroshima University, 1-4-1 Kagamiyama, Higashihiroshima, Hiroshima 739-8527,
- 18 Japan

21

- ⁶ Department of Science of Technology Innovation, Nagaoka University of Technology,
- 20 1603-1 Kamitomioka, Nagaoka, Niigata 940-2188, Japan.
- 22 These authors contributed equally: Mamoru Oshiki, Hirotoshi Netsu, Kyouhei Kuroda
- 23 Article type: Research article
- 24 Running title: Growth of nitrifiers in trickling filter reactors
- 25 **Declarations of interest:** none
- ***Corresponding author:**
- 27 Mamoru Oshiki (Ph.D.)
- 28 E-mail; oshiki@eng.hokudai.ac.jp
- 29 Tel/Fax; +81-11-706-7597/7162

Originality-significance statement

Aerobic nitrite oxidation (NO₂⁻ to NO₃⁻) yields much less Gibbs free energy than aerobic ammonia oxidation (NH₃ to NO₂⁻), while dominance of nitrite-oxidizing bacteria (NOB) over ammonia-oxidizing bacteria (AOB) has been found in marine recirculating trickling biofilter reactors. Specific mechanism responsible for the formation of this puzzling microbial community has not been explored in detail, and the present study shed light on this subject. The present study shows novel ecological aspects of nitrite-oxidizing *Nitrospira* and ammonia-oxidizing *Nitrosomonas* proliferated in trickling biofilter reactors.

Summary (190 words)

39

41

51

40 Aerobic ammonia and nitrite oxidation reactions are fundamental biogeochemical reactions contributing to the global nitrogen cycle. Although aerobic nitrite oxidation yields 4.8-folds 42 less Gibbs free energy (ΔG_r) than aerobic ammonia oxidation in the NH₄⁺-feeding marine recirculating trickling biofilter rectors operated in the present study, nitrite-oxidizing and not 43 ammonia-oxidizing Nitrospira (sublineage IV) outnumbered ammonia-oxidizing 44 Nitrosomonas (relative abundance; 53.8% and 7.59%, respectively). CO₂ assimilation 45 efficiencies during ammonia or nitrite oxidation were 0.077 µmol-¹⁴CO₂/µmol-NH₃ and 0.053 46 47 to 0.054 µmol-¹⁴CO₂/µmol-NO₂, respectively, and the difference between ammonia and 48 nitrite oxidation was much smaller than the difference of ΔG_r . Free-energy efficiency of nitrite oxidation was higher than ammonia oxidation (31-32% and 13%, respectively), and 49 high CO₂ assimilation and free-energy efficiencies were a determinant for the dominance of 50 Nitrospira over Nitrosomonas. Washout of Nitrospira and Nitrosomonas from the trickling 52 biofilter reactors was also examined by quantitative PCR assay. Normalized copy numbers of 53 Nitrosomonas amoA was 1.5- to 1.7-folds greater than Nitrospira nxrB and 16S rRNA gene in 54 the reactor effluents. Nitrosomonas was more susceptible for washout than Nitrospira in the 55 trickling biofilter reactors, which was another determinant for the dominance of *Nitrospira* in 56 the trickling biofilter reactors.

Introduction

57

58

59

60

61

62

63

64

65

66

67

68

69

70

71

72

73

74

75

Nitrification is a key microbial process in the global nitrogen cycle and also for biological nitrogen removal from wastewater. In the nitrification process, ammonia is aerobically oxidized to nitrite by aerobic ammonia-oxidizing bacteria and archaea (AOB and AOA, respectively), and the formed nitrite is subsequently oxidized to nitrate by aerobic nitriteoxidizing bacteria (NOB). Phylogenetically diverse NOB such as Nitrospira (phylum Nitrospirota), Nitrospina (Nitrospinota), Nitrobacter, Nitrotoga, Nitrococcus (Proteobacteria), Nitrolancea (Chloroflexota) have been identified by culture-dependent and -independent techniques (Daims et al., 2016). The genus Nitrospira consists of phylogenetically diverse members (i.e., at least six phylogenetic sublineages) (Lebedeva et al., 2011), and Nitrospira population has been found from wide range of man-made and natural ecosystems (Daims et al., 2016). Additionally, the members of the Nitrospira sublineage II has a unique metabolic capability, complete ammonia oxidation (commamox), and commamox Nitrospira oxidize ammonia to nitrate via nitrite in a single cell (Daims et al., 2015; van Kessel et al., 2015). Population of commamox Nitrospira has been found in freshwater and groundwater ecosystems, but rarely found in marine environments (Xia et al., 2018).

Thermodynamically, aerobic ammonia oxidation yields larger Gibbs free energy than aerobic nitrite oxidation; therefore, numerical dominance of AOB and/or AOA over NOB is

expected where nitrification proceeds. Indeed, stoichiometric and thermodynamic calculation of nitrification processes has enabled to approximate population size and growth yields of AOA and NOB in ocean (Zakem et al., 2018; Zhang et al., 2020). On the other hand, numerical dominance of Nitrospira over AOA/AOB has been described in the recirculating trickling filter reactors operated for marine aquacultures (Foesel et al., 2008; Keuter et al., 2011; 2017; Brown et al., 2013). Another example is our previous study (Oshiki et al., 2020), where *Nitrospira* outnumbered AOB and AOA in NH₄⁺-feeding marine recirculating trickling filter reactors (down-hanging sponge, DHS, reactors); 55%, 10% and <0.1% of total biomass, respectively. Numerical dominance of Nitrospira over AOB/AOA in the DHS reactors was somewhat surprising because the ΔG_r of aerobic ammonia oxidation was 4.8-folds higher in the NH₄⁺-feeding DHS reactor than that of aerobic nitrite oxidation (**Supplementary text 1**). However, the mechanism(s) responsible for the dominance of Nitrospira over AOB and AOA in marine trickling filter reactors has not been explored in detail.

76

77

78

79

80

81

82

83

84

85

86

87

88

89

90

91

92

93

94

Consequently, the present study aimed to examine how *Nitrospira* outnumbered the population of AOB (*i.e.*, *Nitrosomonas*) in the DHS reactors. The two DHS reactors were operated with feeding of the inorganic seawater media containing NH₄⁺ or NO₂⁻ (designated as NH₄⁺- and NO₂⁻-feeding DHS reactors, respectively) at 20°C, and metagenomic analyses using the biomass retained in the reactors were performed to examine metabolic potentials of *Nitrospira* and *Nitrosomonas*. Because genomic data only suggested metabolic potential, a

series of batch incubations were performed to examine CO₂ assimilation efficiencies, free-energy efficiencies, and H₂ oxidation activity. CO₂ assimilation efficiencies were determined by examining ¹⁴CO₂ incorporation into the biomass, and this approach has been used for determining the yields of carbon fixation by nitrifiers (Glover *et al.*, 1985; Tsai and Tuovinen 1986; Bayer *et al.*, 2022) and for the biomass yields of anaerobic ammonia oxidizing bacteria (Ali *et al.*, 2015; Awata *et al.*, 2015). Apart from the above metagenomic and physiological experiments, washout of *Nitrosomonas* and *Nitrospira* in the NH₄⁺-feeding DHS reactor was examined by determining the copy number of AOB *amoA*, *Nitrospira nxrB*, and *Nitrospira* 16S rRNA gene in the reactor effluents by quantitative PCR (qPCR) assay.

Results

105

106

107

108

109

110

111

112

113

114

115

116

117

118

119

120

121

122

123

Metagenomic analysis of NH₄⁺- and NO₂-enriched biomass

NH₄⁺- and NO₂⁻-feeding DHS reactors were operated continuously for more than 1 y without disturbances, and Nitrospira proliferated as a dominant population in both the operated reactors as examined by fluorescence in-situ hybridization (Fig. 1). Metagenomic analyses using the biomass collected from the NH₄⁺- or NO₂⁻-feeding DHS reactors (designated as the NH₄⁺- or NO₂⁻-enriched biomass, respectively) were performed, and the 35.4 and 52.8 M reads of 200-bp paired-end reads corresponding to 14.2 and 21.1 Gb were obtained from the NH₄⁺- and NO₂⁻-enriched biomass, respectively. Those sequence reads were assembled into 30 bacterial bins, which contained the 6 Nitrospira (NPIRA01 to NPIRA06 bins) and 2 Nitrosomonas (NMNS01 and NMNS02 bins) bins (Table 1). Relative abundances and phylogeny of the obtained 30 bins were shown in Table 1 and Fig. S1, respectively. The relative abundances of the *Nitrospira* (especially, the NPIRA01, NPIRA02, NPIRA04 bins) and Nitrosomonas bins (the NMNS02 bin) were much higher than the other bins, indicating that Nitrospira and Nitrosomonas were the predominant bacteria in the NH₄⁺- and/or NO₂⁻-enriched biomass. Especially, the sum of the relative abundances of the NPIRA bins were 53.8% and 72.2% in the NH₄⁺- and NO₂⁻-enriched biomass, indicating Nitrospira was highly abundant in both the biomass. As for Nitrosomonas

bins, the sum of the relative abundances of the NMSN bins were 7.59% and 0.62% in the

NH₄⁺- and NO₂⁻-enriched biomass (**Table 1**), indicating *Nitrospira* outnumbered

Nitrosomonas in both the DHS reactors.

124

125

126

127

128

129

130

131

132

133

134

135

136

137

138

139

140

141

Phylogeny and metabolic potential of NPIRA bins

The NPIRA bins were affiliated into the *Nitrospira* sublineage IV (Lebedeva et al., 2011) (Fig. 2). The average nucleotide identity (ANI) values among the NPIRA bins were 71–89% (Table S1), indicating each NPIRA bins represented different Nitrospira species (Richter and Rosselló-Móra, 2006). The NPIRA03 and NPIRA06 bins were affiliated into the Nitrospira marina clade including Nitrospira marina, a mesophilic and halophilic nitrite oxidizing bacterium. Other NPIRA bins were affiliated into a phylogenetically-different clade in which closely-relating *Nitrospira* genome was not available in public database (accessed on Jan. 2021). Phylogenetic affiliation of this clade was examined using the 16S rRNA gene sequence located in the NPIRA02 bin (the NPIRA02 r00020 gene). The NPIRA02 r00020 gene showed the 97.5% identity with the partial 16S rRNA gene sequences of Candidatus Nitrospira salsa clone Cb18 (accession number KC706459.1) (Fig. S2), and this clade was tentatively designated as the Nitrospira salsa clade in the present study. The NPIRA01, NPIRA02, and NPIRA04 bins affiliated into the Nitrospira salsa clade were dominant Nitrospira (i.e., >10% of relative abundance in a biomass) both in the NH₄⁺- or NO₂⁻-feeding DHS reactors (Table 1).

Metabolic potential of the NPIRA bins was investigated by examining presence and absence of functional genes (Table S2). The genes required for nitrite oxidation (nitrite:nitrate oxidoreductase, nxr), energy conservation (cytochrome bd-like heme-copper terminal oxidase), NAD(P)H generation (complex I), and CO₂ fixation through the rTCA cycle (2-oxoglutarate:ferredoxin oxidoreductase, OGOR, five- or four-subunit pyruvate:ferredoxin oxidoreductase, POR, and ATP-citrate lyase, ACL) were conserved in the NPIRA bins (a full description is available as **supplemental text 2**). On the other hand, orthologues of commamox Nitrospira amoCAB (threshold e-value of blastp search 10⁻¹⁵) were not found in the NPIRA bins and also in the other known Nitrospira sublineage IV genomes. The orthologue of commamox Nitrospira hao was also missing in the NPIRA bins and the Nitrospira sublineage IV genomes, whereas the orthologue encoding octaheme cytochrome c (a phylogenetically-relevant protein with Hao) (Bergmann et al., 2005) found in the canonical NO₂-oxidizing Nitrospira genomes (Nitrospira moscoviensis and Nitrospira japonica genomes) was located in the NPIRA2 and NPIRA3 bins (supplemental text 2 and Table S3). Nitrospira bacteria are metabolically versatile bacteria, and hydrogenotrophic growth of Nitrospira moscoviensis (Koch et al.; 2014) has been demonstrated. The genes required for H₂ oxidation and also degradations of carbohydrate and protein were conserved in the NPIRA bins. A hyd gene cluster encoding a Group 3b NiFe hydrogenase and accessory proteins required for the maturation of the NiFe hydrogenase was conserved in the NPIRA03 and

142

143

144

145

146

147

148

149

150

151

152

153

154

155

156

157

158

159

NPIRA04 bins (Table S2). On the other hand, the genes encoding putative Group 2a NiFe hydrogenase (HupSL) were located in the NPIRA01 and NPIRA04 bins; therefore, the NPIRA04 bin had both the Group 3b and Group 2a NiFe hydrogenase as previously found in Ca. Nitrospira alkalitolerans (Daebeler et al., 2020). The NPIRA03 and NPIRA06 bins had the gene encoding formate dehydrogenase (Fdh) involved in formate oxidation. Nitrospira marina cells can grow chemoorganotrophically on formate even without nitrite (Bayer et al., 2021), and the NPIRA03 20570 and NPIRA06 25420 proteins showed 91.44 and 91.59% identities with Nitrospira marina Fdh (the NMARINA v1 1399 protein). On the other hand, the fdh was not found in the NPIRA bins affiliated to Nitrospira salsa clade. The genes encoding the enzymes involved in the degradation of carbohydrate (glycoside hydrolase, β glucosidase A, and alpha-amylase) and protein (secreted peptidases) were conserved in the NPIRA03, NPIRA04, and NPIRA06 bins. As for the uptake of carbohydrate, amino acid and peptides, ABC transporters for amino acid, oligo- and dipeptides were found in the NPIRA bins whereas the sugar transporters were only found in specific NPIRA bins. The genes encoding sugar ABC transporter were located in the NPIRA04 bin, whereas putative carbohydrate-selective porin was found from NPIRA01, NPIRA02 and NPIRA03 bins. Phylogeny and metabolic potential of NMNS bins The NMNS01 and NMNS02 bins were affiliated into the Nitrosomonas sp. Nm143 and

161

162

163

164

165

166

167

168

169

170

171

172

173

174

175

176

177

178

179

Nitrosomonas aestuarii/marina clades, respectively (Fig. 3), and the members of

Nitrosomonas sp. Nm143 and Nitrosomonas aestuarii were previously found in recirculating marine aquaculture systems (Itoi et al., 2006; Foesel et al., 2008). The genes required for aerobic ammonia oxidation and energy conservation (amo, hao, complex III and terminal oxidase) and for NAD(P)H generation (i.e., reverse electron transport) were generally conserved in the NMNS01 and NMNS02 bins (See Supplemental text 2 for details.). Orthologue of amo was not found from the obtained bins other than NMNS01 and NMNS02 bins. The NMNS bins had the genes encoding ribulose-1,5-bisphosphate carboxylase (RuBisCO) and ribuose-5-phosphate kinase (Table S4), suggesting that those Nitrosomonas fixed CO₂ using the Calvin-Benson cycle. It was previously shown that *Nitrosomonas* was capable of H₂ oxidation coupled with nitrite reduction (Bock et al., 1995), and the NMNS bins had the genes encoding putative NiFe hydrogenase and accessory proteins (i.e., Group 3b and Group 3d NiFe hydrogenase for the NMNS01 and NMNS02 bins, respectively) (Table S4). Those genes were not located as a single gene cluster but found as multiple gene clusters as previously found in the Nitrosomonas oligotropha genome (Sedlacek et al., 2019). Cultivation-dependent characterization of NH₄⁺- and NO₂⁻-enriched biomass CO₂ assimilation and free-energy efficiencies during aerobic ammonia or nitrite oxidation ¹⁴CO₂ assimilation into the NH₄⁺- and NO₂⁻-enriched biomass during aerobic ammonia and nitrite oxidation was examined by incubating the biomass with the addition of 0.5 mM NH₄⁺or NO₂⁻ and ¹⁴C-labeled sodium bicarbonate. The NH₄⁺-enriched biomass consumed

180

181

182

183

184

185

186

187

188

189

190

191

192

193

194

195

196

197

both NH₄⁺and NO₂⁻, and produced NO₃⁻ stoichiometrically (**Fig. 4, left**). During the ammonia and nitrite oxidation, ¹⁴CO₂ was assimilated into the NH₄⁺-enriched biomass, and the CO₂ assimilation efficiencies were determined to be 0.13 ± 0.019 (mean \pm standard deviation derived from triplicate incubation) μmol-CO₂/μmol-NH₃ and 0.053 ± 0.013 μmol-CO₂/μmol-NO₂, respectively (**Table 2**). It is notable that autotrophic bacteria release a part of fixed CO₂ as dissolved organic carbon (DOC) (Oshiki et al., 2011), and the determined ¹⁴CO₂ assimilation does not include the fraction of DOC; e.g., approx. 6-8% and 12% of fixed CO₂ were released as DOC in the culture of marine Nitrosomonas and Nitrospira marina Nb-295, respectively (Bayer et al., 2022). No ¹⁴CO₂ assimilation was detected from the pasteurized biomass and the biomass incubated without the addition of NH₄⁺ or NO₂⁻. The CO₂ assimilation efficiency of the ammonia oxidation reaction could not be determined directly because the formed NO₂ was subsequently oxidized to NO₃ without the accumulation of NO₂ (Fig. 4, left). Therefore, the CO₂ assimilation efficiency of the ammonia oxidation was approximated by subtracting the CO₂ assimilation efficiency of ammonia oxidation reaction with that of nitrite oxidation reaction (i.e., 0.13 μmol-CO₂/μmol-NH₃ and 0.053 μmol-CO₂/µmol-NO₂-, respectively), which was 0.077 µmol-CO₂/ µmol-NH₃. As for the NO₂⁻-enriched biomass, the biomass consumed NO₂⁻ and stoichiometrically produced NO₃⁻ (Fig. 4, right). On the other hand, the NO₂⁻-enriched

199

200

201

202

203

204

205

206

207

208

209

210

211

212

213

214

215

216

217

biomass did not consume NH₄⁺, and no ¹⁴CO₂ assimilation was found. The CO₂ assimilation

efficiency of the nitrite oxidation reaction was determined to be 0.054 ± 0.019 µmol-CO₂/µmol-NO₂⁻, which was the same with that determined using the NH₄⁺-enriched biomass

 $(i.e., 0.053 \pm 0.013 \, \mu mol - CO_2/\mu mol - NO_2^-).$

Free-energy efficiencies during the ammonia and nitrite oxidation were calculated from the obtained CO₂ assimilation efficiencies (see the section Experimental procedures for the calculation of the efficiencies), and compared with those previously determined using AOB and NOB cultures (**Table 2**). The CO₂ assimilation and free-energy efficiencies of the ammonia oxidation reaction obtained in the present study (*i.e.*, 0.077 μmol-CO₂/μmol-NH₃ and 13%, respectively) were comparable with those previously determined from AOB cultures. As for the CO₂ assimilation and free-energy efficiencies of the nitrite oxidation reaction, those obtained in the present study (*i.e.*, 0.053 to 0.054 μmol-CO₂/μmol-NO₂ and 31 to 32%, respectively) were greater than those previously determined from NOB including *Nitrospira marina* Nb-295 (Bayer *et al.*, 2022). The 31 - 32% of the free-energy efficiencies of the nitrite oxidation was >2.3-fold higher than that of ammonia oxidation reaction (*i.e.*, 13%).

Activities of H_2 oxidation

The above metagenomic analysis suggested that *Nitrospira* sp. NPIRA03, NPIRA01 and NPIRA04 and *Nitrosomonas* sp. NSMS01 and NSMS02 were capable of H₂ oxidation, and the activities of the H₂ oxidation of NH₄⁺- and NO₂⁻-enriched biomass were examined by

performing batch incubations. It should be noted that the genes encoding putative NiFe hydrogenase were also found from the bins other than NPIRA and NSMS bins (*i.e.*, DHS20C07, DHS20C08, DHS20C16, DHS20C18, and DHS20C20 bins). Involvement of those bacteria to H₂ oxidation could not be ruled out here, while the abundance of those bins was much less than *Nitrospira* bins (**Table 1**). Although both the NH₄⁺- and NO₂⁻-enriched biomass showed the activities of H₂ oxidation, the activities appeared after 4 d of incubation (**Fig. 5**). Occurrence of the lag phase indicated that H₂ oxidation was not an active metabolic pathway in the NH₄⁺- and NO₂⁻-enriched biomass at least in the operated DHS reactors. This conclusion agreed with the observation obtained from the batch incubation in which the NH₄⁺- and NO₂⁻-enriched biomass were incubated with the addition of ¹⁴CO₂ and H₂. The NH₄⁺- and NO₂⁻-enriched biomass were incubated for 18 h (*i.e.*, within the above lag phase), and no ¹⁴CO₂ assimilation was found in both the biomass during the 18 h of incubation.

Washout of Nitrospira and Nitrosomonas from the NH₄+-feeding DHS reactor

Washout of *Nitrospira* and *Nitrosomonas* cells from the NH₄⁺-feeding DHS reactor were examined by determining the copy number of AOB amoA, *Nitrospira nxrB* and *Nitrospira* 16S rRNA gene in reactor effluents. For the purpose, the qPCR assays of AOB amoA, *Nitrospira nxrB* and *Nitrospira* 16S rRNA gene were carried out. There was a linear relationship between the log copy number of standard DNAs and threshold cycle values (Ct value) ($R^2 > 0.995$), and efficiencies of PCR amplification were 1.83 to 2.02. As shown in **Fig.**

6, all the AOB *amoA*, *Nitrospira nxrB* and *Nitrospira* 16S rRNA gene were detected from the effluents, indicating both *Nitrosomonas* and *Nitrospira* were detached and washed out from the NH₄⁺-feeding DHS reactor. The ratio of the copy numbers of AOB *amoA* to *Nitrospira nxrB* and *Nitrospira* 16S rRNA gene increased in the effluents. Especially, *Nitrospira* 16S rRNA gene was less abundant than AOB *amoA* in the effluents, indicating larger amounts of *Nitrosomonas* population was washed out from the NH₄⁺-feeding DHS reactor.

Discussion

262

263

264

265

266

267

268

269

270

271

272

273

274

275

276

277

Nitrospira bacteria affiliated into the Nitrospira salsa or Nitrospira marina clade were enriched in the NH₄⁺- and NO₂⁻-feeding DHS reactors, and those *Nitrospira* outnumbered the population of Nitrosomonas both in the DHS reactors. Those Nitrospira enriched in the DHS reactors were most likely canonical nitrite-oxidizing Nitrospira because 1) the NPIRA bins and the known Nitrospira sublineage IV genomes did not have the orthologue of amoCAB and hao and 2) the Nitrospira of the NO2-enriched biomass did not show the activity of aerobic ammonia oxidation (Fig. 4) although they (i.e., the NPIRA01 to NPIRA06) were commonly found in the NH₄⁺- and NO₂⁻-enriched biomass. Additionally, commamox Nitrospira (affiliated into the Nitrospira sublineage II) has been found from freshwater and groundwater ecosystems (Xia et al., 2018), whereas no commamox Nitrospira has been recognized from the Nitrospira sublineage IV often found in saline environments (Daims et al., 2016; Park et al., 2020). It is obvious to raise the question of how Nitrospira sp. NPIRA02 and NPIRA04 outnumbered Nitrosomonas sp. NMNS01 and NMNS02 in the NH₄⁺-feeding DHS reactor. Not only the present study, the previous studies have also indicated Nitrospira sublineage IV population outnumbered the population of aerobic

ammonia oxidizers in marine aquaculture systems (**Table S5**) (Foesel *et al.*, 2008; Keuter *et al.*, 2011; 2017; Brown *et al.*, 2013).

278

279

280

281

282

283

284

285

286

287

288

289

290

291

292

293

294

295

296

Aerobic ammonia oxidation reaction yields 4.8-times higher free energy than NO₂⁻ oxidation reaction in the NH₄⁺-feeding DHS reactor. However, the free energy recovered from the ammonia oxidation reaction must be much lower than the ΔG_r due to the following reasons. First, there is no evidence that the first reaction of aerobic ammonia oxidation to NH₂OH involves translocation of H⁺ and formation of proton motive force (Costa *et al.*, 2006; Simon and Klotz, 2013). The reaction of aerobic ammonia oxidation to NH₂OH is a monooxygenation reaction catalyzed by Amo (Lancaster et al., 2018), and the free energy released during monooxygenation reactions (more specifically, O2 reduction reaction) are not conserved and dissipated (VanBriesen, 2001). Dissipation of the free energy during the monooxygenation reaction of CH₄ (Yuan and VanBriesen, 2002) and NH₃ (Hollocher et al., 1982) has been described, and both the reactions are catalyzed phylogenetically relevant monooxygenase (i.e., Pmo and Amo, respectively). Indeed, Nitrosomonas europaea cells showed nearly same effective H⁺/O ratios during aerobic ammonia and NH₂OH oxidation (i.e., 4.1 and 3.9 of effective H⁺/O ratios, respectively) (Hollocher et al., 1982), indicating the contribution of the reaction of ammonia oxidation to NH₂OH to the translocation of H⁺ is minor. The ΔG_r 'o for the reaction of aerobic ammonia oxidation to NH₂OH was -170.5 kJ mol-NH₃⁻¹ (Supplementary text 1), and this free energy (accounting for more than half of

NH₂OH to NO₂⁻ releases 4 e^- , whereas the amounts of the e^- available for the respiration are less than 2 e due to the following reasons; 1) the 2 e out of the produced 4 e is consumed to oxidize 1 mol NH₃ to NH₂OH by Amo (Whittaker et al., 2000), and 2) a part of the produced 4 e⁻ is consumed to generate NAD(P)H by reverse electron transport. In Nitrosomonas europaea cells, the 0.35 e⁻ enters reverse electron flow (Whittaker et al., 2000). Additionally, the biochemistry of NH₂OH oxidation to NO₂ by AOB is still controversial because the Nitrosomonas europaea Hao oxidized NH2OH to NO but not further to NO2-, and specific mechanisms of NO oxidation to NO₂ has not been elucidated (Carantoa and Lancaster, 2017; Lancaster et al., 2018). NO oxidation to NO₂ releases 1 e⁻, and the involvement of the released e^- in the respiration of AOB needs to be investigated in other studies. The above discussion pointed out that the amounts of the free energy recovered from aerobic ammonia oxidation were much lower than that calculated as ΔG_r of the reaction. The reduction of the free energy recovered from aerobic ammonia oxidation would result in the low free-energy efficiencies of the ammonia oxidation reaction shown in **Table 2** (*i.e.*, 7 to 13%). The ΔG_r of aerobic nitrite oxidation was much smaller than that of aerobic ammonia oxidation (-83.5 kJ mol-NO₂⁻¹ and -283.3 kJ mol-NH₃⁻¹ at the batch incubation,

 ΔG_r of aerobic ammonia oxidation) will be dissipated. Secondly, the oxidation of 1 mol

297

298

299

300

301

302

303

304

305

306

307

308

309

310

311

312

313

314

315

respectively,) (Supplementary text 1), while the CO2 assimilation efficiency during aerobic

nitrite oxidation were close to that of the ammonia oxidation reaction; i.e., 0.053 µmol-

CO₂/µmol-NO₂ (**Table 2**). Such high CO₂ assimilation and free-energy efficiencies (31-32%) were rarely observed from axenic cultures of NOB (e.g., Nitrospira marina Nb-295; 0.032 μmol-CO₂/μmol-NO₂ and 22%, respectively), but high biomass yield of marine NOB (Nitrospinae) was previously found in environmental samples (Kitzinger et al., 2020) where the NOB coexisted with other microbes. Low free-energy efficiencies of aerobic nitrite oxidation found in the axenic cultures were reasonable because NO2 oxidation reaction can not couple with the reduction of quinone molecules (ubiquinone_{ox/red}, ΔE° = +0.11 V) directly due to the high redox potential of NO_2^-/NO_3^- oxidation rection ($\Delta E^{*o} = +0.42 \text{ V}$) (Madigan et al., 2019). Additionally, the electrons released from the NO₂ oxidation reaction enter to terminal oxidase bypassing a cytochrome bc1 complex (Lücker et al., 2010; Simon and Klotz et al., 2013), resulting in the decrease of the number of translocated H⁺ during the respiration: therefore, the free-energy efficiencies of nitrite oxidation reaction were expected to be low. On the other hand, Nitrospira in the NH₄⁺- and NO₂⁻-feeding DHS reactors showed high CO₂ assimilation and free energy efficiencies. As compared with Nitrobacter and Nitrococcus, the following bioenergetic advantages of the nitrite oxidation by Nitrospira were often introduced in literatures; 1) nitrite oxidation occurs in the periplasmic spaces (Spieck et al., 1998; Lücker et al., 2010), which directly contributes to the generation of proton motive force across a membrane (i.e., $1 \text{ NO}_2^- + 1 \text{ H}_2\text{O} \rightarrow 1 \text{ NO}_3^- + 2 \text{ H}^+_{\text{periplasm}} + 2 \text{ e}^-$), and 2) Nitrospira used an energetically more efficient rTCA cycle for CO₂ fixation as compared with Nitrobacter and

316

317

318

319

320

321

322

323

324

325

326

327

328

329

330

331

332

333

Nitrosococcus who used the Calvin-Benson cycle. However, the difference of CO₂ fixation pathway might not result in a drastic change of CO2 assimilation efficiency although the Calvin-Benson cycle requires more ATPs investment for CO₂ fixation and involves a wasteful oxygenase side reaction of ribulose-1,5-bisphosphate carboxylase/oxygenase (Berg 2011; Bayer et al., 2022). For the fixation of 3 mol CO₂ (i.e., HCO₃⁻) to 1 mol of phosphoglyceraldehyde which is the simplest sugar and a precursor for the biomass synthesis (White et al., 2012), the rTCA cycle requires 5 and 6 mol of ATP and NADPH equivalents (such as NAD(P)H, ferredoxin and FADH₂), whereas the Calvin-Benson cycle requires 9 and 6 mol of ATP and NADPH equivalents (Bar-Even et al., 2010). Therefore, the amounts of NADPH equivalents are the same between the two pathways. To generate 6 mol of NAD(P)H from NAD(P)⁺ by the reverse electron transport, Nitrosomonas europaea and Nitrobacter winogradskyi consumed 30 mol of ATP (Aleem, 1966; Sewell and Aleem, 1969); therefore, the energy cost for reverse electron flow is >3 folds higher than that for CO₂ fixation. The ATP cost of the rTCA cycle was 4 mol-ATP/mol-phosphoglyceraldehyde fewer than that of the Calvin-Benson cycle, while this energy conservation is much smaller than the energy consumption for reverse electron transport. Indeed, the CO₂ assimilation efficiencies previously determined from marine Nitrococcus with Calvin-Benson cycle (Nitrococcus mobilis) were comparable with that of marine Nitrospira with rTCA cycle (Nitrospira marina

335

336

337

338

339

340

341

342

343

344

345

346

347

348

349

350

351

Nb-295); *i.e.*, 0.014-0.031 and 0.032 μmol-CO₂/μmol-NO₂-, respectively (Glover 1985; Bayer *et al.*, 2022) (**Table 2**).

353

354

355

356

357

358

359

360

361

362

363

364

365

366

367

368

369

370

371

It is notable that NOB can receive some micronutrients produced in microbial community (Mee et al., 2014; Kim et al., 2021), which would raise CO₂ assimilation efficiencies of *Nitrospira*. One example is the vitamin-B₁₂ auxotrophy of *Nitrospira marina*. The Nitrospira marina genome lacked a couple of genes required for the biosynthesis of vitamin B₁₂, and their growth ceased in vitamin-B₁₂ deficient media (Bayer et al., 2021). However, the growth of Nitrospira marina was found in a mixed culture even when vitamin B₁₂ was not supplied into the media (Park et al., 2020). Additionally, Nitrospira cells can incorporate organic matters available in the culture. The addition of undefined organic matters such as tryptone increased apparent growth yields of *Nitrospira marina* (Watson et al., 1986; Bayer et al., 2021), and formate utilization by Nitrospira bacteria has been also demonstrated (Gruber-Dorninger et al., 2015; Koch et al., 2015; Lawson et al., 2021). Availability of organic matters in the operated DHS reactors fed with inorganic media was suggested from the growth of heterotrophic bacteria in the reactors. In the NH₄⁺- and NO₂⁻-feeding DHS reactors, heterotrophs (i.e., the DHS20C01 to DHS20C21 bins) accounted for ca. 10% of total biomass (Table 1), and the presence of those heterotrophs indirectly indicated that organic matters were available in the operated DHS reactors (likely in the form of extracellular polymeric substances, soluble microbial products and cell debris). The obtained Nitrospira

bins (and also NMNS bins, **Table S6**) had a couple of (di)peptide transporters and ABC transporters, and scavenging micronutrients might contribute to the increase of apparent CO₂ assimilation efficiencies. In addition to the possible interactions of micronutrients, oxidative stress in mixed communities would be less because coexisting microbes can scavenge O₂ and reactive oxygen species. Less oxidative stress would reduce energy demands of the rTCA cycle of *Nitrospira* where oxygen-sensitive enzymes (*e.g.*, four-subunit pyruvate:ferredoxin oxidoreductase) are involved (Berg *et al.*, 2011; Bayer *et al.*, 2022).

Although the above discussion provided a bioenergetic insight in the CO₂ assimilation of AOB and NOB, the growth yields of AOB were still higher than those of NOB including *Nitrospira* (**Table 2**). How did *Nitrospira* with lower CO₂ assimilation efficiencies outnumbered *Nitrosomonas* in the NH₄⁺-feeding DHS reactor? Notably, the DHS reactor and the recirculating marine aquaculture systems in the previous studies were operated as trickling filter reactors, and *Nitrospira* and AOB proliferated in the form of biofilm on the biomass carriers (**Table S5**). The biomass carriers in those trickling filter reactors were exposed to permanent shear forces, and shear forces changed microbial diversity and composition of the developed biofilm (Rickard *et al.*, 2004) and promoted the proliferation of auto-aggregating bacteria (Rochex *et al.*, 2008). In the nitrifying biofilm, AOB were preferentially localized on the surface of the biofilm (*i.e.*, aerobic zone) whereas *Nitrospira* cells were easy to form cell aggregates (Spieck *et al.*, 2006; Ushiki *et al.*, 2013) and heterologously distributed in the

biofilm and abundant in microaerobic zone (i.e., inner part of biofilm) (Okabe et al., 1999; Schramm et al., 2000). Such spatial distribution of AOB and NOB might occur in the sponge carrier. Additionally, Nitrospira spp. in activated sludge tended to form physically stronger cell aggregates than Nitrosomonas oligotropha (Larsen et al., 2008), suggesting Nitrosomonas in the NH₄⁺-feeding DHS reactor was likely to be more susceptible for the detachment from biofilm. Occurrence of the washout of AOB and NOB from trickling filter reactors has not been well investigated so far (Keuter et al., 2011), which was examined in the present study by determining the gene copy numbers of AOB amoA, Nitrospira nxrB, and *Nitrospira* 16S rRNA genes (**Fig. 2**) in the NH₄⁺-enriched biomass and the reactor effluents. The NMNS01 bin had one copy of amoCAB gene cluster as well as the closed relative Nitrosomonas genomes (Nitrosomonas sp. Nm143, Nitrosomonas sp. UBA8640, and Nitrosomonas aestuarii). As for Nitrospira, three nxrAB gene clusters and one 16S rRNA gene were found in the Nitrospira marina genome. This ratio of the copy number of nxrAB to 16S rRNA gene (i.e., 3) agreed with the ratio of the copy numbers of Nitrospira nxrB to Nitrospira 16S rRNA gene found in the NH₄+-enriched biomass and the reactor effluents (i.e., 3.0 to 3.4) (Fig. 6). Assuming one amoA gene copy per Nitrosomonas genome and three nxrB and one 16S rRNA gene copy per Nitrospira genome, the abundance of Nitrosomonas was 1.5- and 1.7-folds greater (based on the normalized copy number of nxrB and 16S rRNA gene, respectively) than Nitrospira in the effluents of NH₄⁺-feeding DHS reactor although

391

392

393

394

395

396

397

398

399

400

401

402

403

404

405

406

407

408

Nitrospira was 1.89- and 1.86-folds greater than Nitrosomonas in NH₄⁺-enriched biomass.

The greater abundance of Nitrosomonas in the effluents indicated that Nitrosomonas tended to be washed out more frequently from the NH₄⁺-feeding DHS reactor. This washout of AOB was another determinant for the dominance of Nitrospira over Nitrosomonas in the NH₄⁺-feeding DHS reactors.

410

411

412

413

414

415

416

417

418

419

420

421

422

423

424

425

426

427

428

In summary, CO₂ assimilation efficiencies of *Nitrosomonas* and NO₂-oxidizing Nitrospira were determined, and the difference of the CO₂ assimilation efficiencies between Nitrosomonas and Nitrospira was much smaller (0.077 µmol-CO₂/ µmol-NH₃ and 0.053- $0.054 \,\mu\text{mol-CO}_2/\mu\text{mol-NO}_2$, respectively) as compared with the difference of ΔG_r . Such small difference in the CO₂ assimilation efficiencies was likely due to that large parts of free energies during aerobic ammonia oxidation are dissipated and not conserved. The dissipation of free energy (i.e., efficiency of energy conservation) can not be expected from the value of ΔG_r , and more bioenergetic studies, especially for *Nitrospira*, are required. *Nitrospira* use a novel cytochrome bd-like heme-copper oxidase as a terminal oxidase and the nitrite oxidation occurs in periplasmic spaces, which were not common with that of canonical NO₂-oxidizing Nitrobacter. It will be interesting to investigate the bioenergetic traits of Nitrospira. Apart from the bioenergetics, washout of nitrifying population was another factor driving the dominance of Nitrospira over Nitrosomonas in the DHS reactor; i.e., Nitrosomonas was more susceptible for washout than Nitrospira. Detachment and washout of particular nitrifiers has

- been little explored, and the correlation between physicochemical parameters (e.g., cell
- 430 surface hydrophobicity) remains to be explored in other studies.

Experimental procedures

431

432

Operation of the DHS reactors fed with NH₄⁺ or NO₂⁻

433 The 10-L DHS reactors (0.7 m in height and 0.17 m in width) were operated at 20°C in dark. 434 Details of the operated DHS reactors were previously described by the authors (Oshiki et al., 435 2020). Briefly, the DHS reactors contained polyurethane-sponge media (183 pieces of cubic sponge, 33 mm × 33 mm, set in a polypropylene tube, 32 mm diameter and 32 mm 436 437 long) as a biomass carrier. The sponge media had a 97% void ratio (i.e., a percentage of the volume of sponge pores), 256 m² m⁻³ of specific surface, and 0.63 mm of average pore size. 438 Artificial seawater media (pH 8.0, salinity 33%) containing 0.297 g L⁻¹ NH₄Cl or 0.4 g L⁻¹ 439 NaNO₂-, 1 g L⁻¹ NaHCO₃, 34 g L⁻¹ artificial seawater powder (Marin Art, Tomita 440 441 Pharmaceutical, Naruto, Japan) was supplied to the top of the DHS reactor at the flow rate of 9.62 L d⁻¹ corresponding to 0.39 d of hydraulic retention time (HRT) and 200 mg-N L-sponge 442 media⁻¹ d⁻¹ of total ammonia nitrogen loading rate. The sponge media was exposed to the 443 444 atmosphere, and oxygen naturally dissolved into the media and aerobic condition was 445 maintained without external aeration. The filtrates were collected in a settling tank (volume; 0.9 L) located at the bottom of the DHS reactor, which were recirculated using a magnetic 446 pump at the flow rate of 4 L min⁻¹. The NH₄⁺- or NO₂⁻-feeding DHS reactors has been 447 448 operated for more than 1 y stably, and typical concentrations of NH₄⁺, NO₂⁻, NO₃⁻ and pH

values in the NH₄⁺-feeding DHS reactors were 7.9 μ M, 9.3 μ M, 4.9 mM, and pH 7.3, respectively.

Biomass retained in the sponge media was collected by squeezing the sponge media in the above artificial seawater media without NH₄Cl and NaNO₂-, and subjected to the following DNA extraction and batch incubations. As for the reactor effluents, two liters of the effluents was daily collected from the NH₄+-feeding reactor for 4 d, and the collected effluents were filtered on a 0.2 μm-pore-size PVDF membrane (Advantec, Tokyo, Japan) separately. The filtered membranes were subjected to the DNA extraction.

DNA extraction and determination of DNA concentration

Genomic DNA was extracted from the biomass and the filtered membranes using a DNeasy PowerSoil kit (Qiagen K.K., Tokyo, Japan) as following the instruction manual supplied by manufactures. The concentrations of the extracted DNAs were determined using the Qubit dsDNA BR assay kit and a Qubit 3.0 fluorospectrometer (Thermo Fisher Scientific, Waltham, MD, USA).

Metagenomic analysis

A shotgun sequence library was prepared using a MGIEasy FS DNA Library Prep Set,

MGIEasy Circularization kit, and DNBSEQ-G400RS High-throughput sequencing set (MGI

Tech Japan, Tokyo, Japan). The 200-bp paired-end sequencing was performed using a

DNBSEQ-G400 sequencer. The paired-end sequence reads were trimmed using Trimmomatic

0.39 (SLIDINGWINDOW:6:30 MINLEN:100) (Bolger et al., 2014). Digital normalization of trimmed sequences was performed using bbnorm.sh of BBtools version 38.18 (target=100, min=5) (https://jgi.doe.gov/data-and-tools/bbtools/). Assembled contigs were obtained from NH₄⁺- and NO₂⁻-enriched biomass samples (co-assembly) by Megahit v1.2.9 (--k-min 27 --kmax 141 --k-step 12) (Li et al., 2015). Reads of each sample were mapped to assembled contigs using bbmap.sh of BBtools. Obtained contigs of short length (< 2,500 bp) were removed before binning. The multiple software of MaxBin2 version 2.15 (-markerset 40) (Wu et al., 2016), Metabat2 version 2.2.7 (Kang et al., 2019), MyCC (MyCC 2017.ova) (Lin and Liao, 2016) were used for binning from the contigs. To refine the obtained bins, we used Binning refiner version 1.4.0 with default parameters (Song and Thomas, 2017). The quality of refined bins was checked using CheckM version 1.0.7 (Parks et al., 2014). The relative abundance of obtained bins was calculated using CoverM version 0.6.1 (https://github.com/wwood/CoverM#installation). Phylogenetic position of each bin is estimated using GTDBtk v1.3.0 (release95) (Chaumeil et al., 2019). Average nucleotide identity (ANI) values of the obtained Nitrospira bins and the genomes in Nitrospira sublineage IV were calculated using pyani version 0.2.11 (-m ANIb) (Pritchard et al., 2016). Gene prediction and annotation was performed via the D-FAST pipeline (Tanizawa et al., 2018), and the MetaGeneAnnotator and Glimmer version 2.10, tRNAScan-SE version 1.23,

468

469

470

471

472

473

474

475

476

477

478

479

480

481

482

483

484

and blastn software applications were used for prediction of gene-coding sequences (CDSs),
 tRNA, and rRNA, respectively.

488

489

490

491

492

493

494

495

496

497

498

499

500

501

502

503

504

Genomic tree was constructed using 120 concatenated phylogenetic marker genes of obtained bins and representatives of genus Nitrospira or Nitrosomonas in the release95 of GTDBtk version 1.3.0. For the multiple sequence alignment of Nitrospira spp., we included the genomes of Nitrospira marina Nb-295T, Nitrospirales bacterium isolate MH-Patall autometa 1-10 (WLXC01000001), MAG-Baikal-G1, MAG-Baikal-deep-G158, MAG-Baikal-deep-G159, MAG-ZH-13may13-77, MAG-cas150m-170, MAG-cas50m-175 with Nitrospira genomes in the release95 of GTDBtk. Conserved marker genes were identified using "gtdbtk identify" with default parameters and aligned to reference genomes using "gtdbtk align" with taxonomic filters for phylogenies of Nitrospira (--taxa fileter c Nitrospira), Nitrosomonas (--taxa fileter f Nitrosomonadaceae) or all metagenomic bins (--taxa fileter f UBA8639,g Nitrosomonas,f UBA11606,o ARS69,f Saprospiraceae,g SZUA-3,g GCA-2699125,g CR02bin9,f Nitrospinaceae,f SM1A02,g UBA1845,f B15-G4,g Hyphobacterium,g Marinicaulis,g UBA5701,g Minwuia,f Methyloligellaceae,f Rhodomicrobiaceae,g UBA9145,o UBA10353,g UBA7359,o Xanthomonadales). Phylogenetic tree was constructed using IQ-TREE version 2.0.3 (-B 1000) with automatically optimized substitution models (Nitrospira: LG+F+R7 and Nitrosomonas: JTT+F+R5) (Minh et al., 2020) and with the Nitrosospira lacus (GCF_000355765.4) and Thermodesulfovibrio yellowstonii (GCF_000020985.1) genomes for the Nitrospira and Nitrosomonas trees (Fig. 2 and 3, respectively) as an outgroup.

qPCR assay

505

506

507

508

509

510

511

512

513

514

515

516

517

518

519

520

521

522

523

The qPCR assay was conducted using a ABI7500 fast Real-Time PCR System (Thermo Fisher Scientific). The reaction mixture (10 µL) contained KAPA SYBR FAST qPCR master mix (Nippon Genetics, Tokyo, Japan) (5 μL), 0.2 μM each forward and reverse primer, 1 × ROX low dye, and 1 ng of the extracted DNA. Oligonucleotide primers used for PCR amplification were 1) 515f and 806r for prokaryotic 16S rRNA gene, 2) amoA1F and amoA2Rv1 for AOB amoA (Rotthauwe et al., 1997, this study), 3) nxrB169f and nxrB638r (Pester et al., 2014) for Nitrospira nxrB, and 4) Nspra675F and Nspra746R (Graham et al., 2007) for *Nitrospira* 16S rRNA gene, and the nucleotide sequences were shown in **Table S6**. The original amoA2R primer (Rotthauwe et al., 1997) had some mismatch bases against the amoA sequence found in the NMNS01 bin, and the amoA2Rv1 was designed in the present study by adding degenerate bases into the amoA2R primer. As for the amoA1F, nxrB169f, nxrB638r, Nspra675F, and Nspra746R primers, there was no mismatch base between the oligonucleotide primer and target gene found in the above metagenomic analysis. The cycling conditions were the following: 95°C for 3 min; 40 cycles at 95°C for 3 s and 60°C for 20 s; and, finally, 65°C to 95°C in 0.5°C increments for the melting curve analysis. Standard curves (10¹ to 106 copies/μL) were prepared using a dilution series of plasmid DNAs containing PCR products of the above target. Partial sequences of *Escherichia coli* JM109 (SMOBIO technology, Hsinchu, Taiwan) 16S rRNA gene, *Nitrosomonas europaea* (NBRC14298) *amoA*, *Nitrospira inopinata* (JCM31988) 16S rRNA gene were amplified using the above oligonucleotide primer set. *Nitrospira nxrB* was amplified using the genomic DNA extracted from NO₂⁻-enriched biomass. The obtained PCR products were cloned into pUC118 vector using mighty cloning reagent (TakaraBio, Shiga, Japan), and transformed into *E. coli* DH5α cells (SMBIO technology) by heat shock. Plasmids were extracted from the transformants using FastGene Plasmid mini kit (Nippon Genetics), nucleotide sequences of the cloned PCR products were ascertained by performing the Sanger sequencing, and the concentrations of the extracted plasmids were determined fluorometrically as previously described above.

Activity tests

Assimilation of ¹⁴CO₂ into the biomass during aerobic ammonia and nitrite oxidation was examined as previously described (Oshiki *et al.*, 2011). Briefly, the 2.5 mL of biomass suspension (24 and 0.66 μg-protein mL⁻¹ of biomass concentrations for NH₄⁺- and NO₂⁻- enriched biomass, respectively) containing 0.5 mM NH₄⁺ or NO₂⁻ was incubated at 20°C in 10-mL glass vials with shaking at 60 rpm. The ¹⁴C-labeled sodium bicarbonate (Moravek Inc., Brea, CA, USA) was added at a concentration of 10 μCi vial⁻¹, and the vials were sealed with butyl rubber plug and aluminum seal. After 18 h of incubation, the biomass was fixed with

4% paraformaldehyde, washed three times with PBS, and mixed with scintillation cocktail Clear-sol I (Nacalai, Tokyo, Japan). Radioactivity was determined with an ALOKA LSC-6100 liquid scintillation counter. Additional cold incubation with ¹²C-labeled sodium bicarbonate instead of ¹⁴C-labeled one was performed in parallel to determine ammonia and nitrite oxidation rates. Abiotic incorporation of ¹⁴C-labeled sodium bicarbonate was examined by performing the above incubation with the biomass pasteurized at 70°C for 15 min. For the incubation using H₂ as a substrate instead of NH₄⁺ and NO₂⁻, biomass suspension without NH₄⁺ and NO₂⁻ were dispensed into the closed vials, and pure H₂ gas (GL Science, Tokyo, Japan) was injected into headspace at the final concentration of 2% (v/v) using a gas tight syringe.

Chemical analysis

NH₄⁺ concentration was determined fluorometrically using the *o*-phthalaldehyde (OPA) method (Taylor *et al.*, 1974). Liquid samples were mixed with 3.8 mM *o*-phthalaldehyde, and fluorescence intensity was determined at 355 nm of excitation and 460 nm of emission.

NO₂⁻ concentration was determined colorimetrically using the naphthylethylenediamine method (Rice *et al.*, 2012). Liquid samples were mixed with a naphthylethylenediamine-sulfanilamide solution, and absorbance was measured at 540 nm.

NO₃⁻ concentration was determined colorimetrically using the brucine sulfate method (Jenkins and Medsker, 1964). Liquid samples were mixed with 80% (vol/vol) sulfuric

acid and brucine sulfanilic acid solution, and heated at 100°C for 20 min. The absorbance was measured at 410 nm.

Protein concentration was determined by the Lowry method using a DC protein assay kit (Bio-Rad, Hercules, CA, USA). Bovine serum albumin was used as a protein standard.

 $_{\rm H2}$ concentration was determined by gas chromatography as described elsewhere. The 100 μL of gas sample was injected into a gas chromatograph GC-2014 equipped with a thermal conductivity detector and a 2-m stainless column packed with a Molecular Sieve-5A.

Fluorescence in-situ hybridization and microscopy

Fixation of biomass (4% paraformaldehyde) and *in situ* hybridization of oligonucleotide probes were performed as previously described (Kindaichi *et al.*, 2004). The fixed biomass was sonicated at 3 watt for 4 minutes, and hybridized with the oligonucleotide probe Ntspa712 (Daims *et al.*, 2001) for *Nitrospira* or Nsm156 (Mobarry *et al.*, 1996) for *Nitrosomonas*, respectively.

Calculation of CO₂ assimilation and free-energy efficiencies.

CO₂ assimilation efficiencies of NH₄⁺- and NO₂⁻-enriched biomass during aerobic ammonia and nitrite oxidation (unit; μ mol-CO₂/ μ mol-NH₃ or μ mol-NO₂⁻) were calculated by dividing the molar amounts of ¹⁴CO₂ fixed during the incubation with those of the consumed NH₄⁺ and NO₂⁻. The fixed CO₂ includes the carbon fixed into cellular materials and also that fixed as

extracellular polymeric substances (EPS) (Okabe *et al.*, 2005). On the other hand, the fixed CO₂ does not include the carbon released as dissolved organic carbon (DOC); *i.e.*, *ca.* 6-8% and 12% of fixed CO₂ were released as DOC in the culture of marine AOB (*Nitrosomonas marina* C-25 and *Nitrosomonas* sp. C-15) and marine NOB (*Nitrospira marina* Nb-295), respectively (Bayer *et al.*, 2022).

The values of free-energy efficiency were calculated using the following equation (Glover, 1985).

Where, 495; the free energy (kJ/mol) required for converting CO₂ into CH₂O, and ΔG_r ; the free energy (kJ/mol) obtained from aerobic ammonia or nitrite oxidation during the batch incubations (*i.e.*, 283.3 kJ/mol-NH₃ and 83.5 kJ/mol-NO₂-, respectively) (**Supplementary text 1**). For AOB and NOB examined in the previous studies, the 286.7 kJ/mol-NH₃ for AOB, and 73.8 kJ/mol-NO₂- for NOB were used. Those values corresponded to ΔG_r that were the Gibbs free energy change at when the concentrations of all reactants were 1 mM at pH 7 and 25°C (**Supplementary text 1**).

Accession numbers

Raw metagenomic sequence data and the assembled and annotated 30 bins obtained in the present study are available in the DDBJ nucleotide sequence database under the accession number DRA013035 and those in **Table S7**, respectively.

Acknowledgements

597

598

599

600

601

610

611

This work was supported by JSPS KAKENHI [grant numbers; 19K05805 for M.O., 602 20H02290 for N.A, 20H00641 for T.Y., 19H00776 for S.O.], JST FOREST Program 603 [JPMJFR216Z for M.O.], and Nagase Science and Technology Foundation granted to M.O.. 604 Computations were partially performed on the NIG supercomputer at ROIS National Institute 605 of Genetics. The authors would like to express our sincere appreciation for Dr. Barbara Bayer 606 (University of Vienna) for sharing a reference dataset of the DIC fixation yields of nitrifiers that are shown in Table 2. Strain JCM31988 and NBRC14298 were provided by Japan 607 608 Collection of Microorganisms, RIKEN BRC and National Institute of Technology and 609 Evaluation, respectively.

Conflict of interest

The authors declare no conflicts of interest associated with this manuscript.

References

612

613 Aleem, M.I.H. (1966) Generation of reducing power in chemosynthesis. II. Energy-linked 614 reduction of pyridine nucleotides in the chemoautotroph, *Nitrosomonas europaea*. 615 Biochim Biophys Acta 113: 216–224. 616 Ali, M., Oshiki, M., Awata, T., Isobe, K., Kimura, Z., Yoshikawa, H., Hira, D., Kindaichi, T., 617 Satoh, H., Fujii, T., Okabe, S., (2015) Physiological characterization of anaerobic 618 ammonium oxidizing bacterium "Candidatus Jettenia caeni." Environ. Microbiol. 17: 619 2172-2189. 620 Awata, T., Kindaichi, T., Ozaki, N., Ohashi, A., 2015. Biomass yield efficiency of the marine 621 anammox bacterium, "Candidatus Scalindua sp.," is affected by salinity. Microb. 622 Environ. 30: 86-91. 623 Bar-Even, A., Noor, E., Lewis N.E., and Milo, R. (2010) Design and analysis of synthetic 624 carbon fixation pathways. Proc Natl Acad Sci USA 107: 8889–8894. 625 Bayer, B., Saito, M.A., McIlvin, M.R., Lücker, S., Moran, D.M., Lankiewicz, T.S., et al. 626 (2021) Metabolic versatility of the nitrite-oxidizing bacterium Nitrospira marina and 627 its proteomic response to oxygen-limited conditions. ISME J 15: 1025-1039. 628 Bayer, B., McBeain, K., Carlson, C.A., Santoro, A.E. (2022) Carbon content, carbon fixation yield and dissolved organic carbon release from diverse marine nitrifiers. bioRxiv 629 630 2022.01.04.474793. https://doi.org/10.1101/2022.01.04.474793

631 Berg, I.A. (2011) Ecological aspects of the distribution of different autotrophic CO₂ fixation 632 pathways. Appl Environ Microbiol 77: 1925–1936. 633 Bergmann, D.J., Hooper, A.B., Klotz, M.G. (2005) Structure and sequence conservation of 634 hao cluster genes of autotrophic ammonia-oxidizing bacteria: evidence for their 635 evolutionary history. Appl. Environ. Microbiol. 71: 5371–5382. 636 Bock, E., Schmidt, I., Stüven, R., Zart, D., (1995) Nitrogen loss caused by denitrifying 637 Nitrosomonas cells using ammonium or hydrogen as electron donors and nitrite as 638 electron acceptor. Arch. Microbiol. 163: 16-20. 639 Bolger, A.M., Lohse, M., and Usadel, B. (2014) Trimmomatic: a flexible trimmer for Illumina 640 sequence data. Bioinformatics 30: 2114–2120. 641 Brown, M.N., Briones, A., Diana, J., and Raskin, L. (2013) Ammonia-oxidizing archaea and 642 nitrite-oxidizing nitrospiras in the biofilter of a shrimp recirculating aquaculture 643 system. FEMS Microbiol Ecol 83: 17-25. 644 Carantoa, J.D., and Lancaster, K.M. (2017) Nitric oxide is an obligate bacterial nitrification 645 intermediate produced by hydroxylamine oxidoreductase. Proc Natl Acad Sci USA 646 **114**: 8217–8222. 647 Chaumeil, P.A., Mussig, A.J., Hugenholtz, P., and Parks, D.H. (2019) GTDB-Tk: a toolkit to 648 classify genomes with the Genome Taxonomy Database. Bioinformatics 36: 1925– 649 1927.

650 Costa, E., Pérez, J., and Kreft, J-U. (2006) Why is metabolic labour divided in nitrification? 651 Trends Microbiol 14: 213–219. 652 Daebeler, A., Kitzinger, K., Koch, H., Herbold, C.W., Steinfeder, M., Schwarz, J., et al. 653 (2020) Exploring the upper pH limits of nitrite oxidation: diversity, ecophysiology, 654 and adaptive traits of haloalkalitolerant *Nitrospira*. *ISME J* **14**: 2967–2979. 655 Daims, H., Nielsen, J.L., Nielsen, P.H., Schleifer, K.-H., Wagner, M. (2001) In situ 656 characterization of Nitrospira-like nitrite-oxidizing bacteria active in wastewater 657 treatment plants. Appl. Environ. Microbiol. 67: 5273-5284. 658 Daims, H., Lebedeva, E.V., Pjevac, P., Han, P., Herbold, C., Albertsen, M., et al. (2015) 659 Complete nitrification by *Nitrospira* bacteria. *Nature* **528**: 504–509. 660 Daims, H., Lücker, S., and Wagner, M. (2016) A new perspective on microbes formerly 661 known as nitrite-oxidizing bacteria. Trends Microbiol 24: 699–712. 662 Foesel, B.U., Gieseke, A., Schwermer, C., Stief, P., Koch, L., Cytryn, E., et al. (2008) 663 Nitrosomonas Nm143-like ammonia oxidizers and Nitrospira marina-like nitrite 664 oxidizers dominate the nitrifier community in a marine aquaculture biofilm. FEMS 665 Microbiol Ecol 63: 192-204. 666 Graham, D.W., Knapp C.W., Van Vleck, E.S., Bloor, K., Lane, T.B., and Graham, C.E. 667 (2007) Experimental demonstration of chaotic instability in biological nitrification.

668

ISME J 1: 385–393.

669 Glover, H.E. (1985) The relationship between inorganic nitrogen oxidation and organic 670 carbon production in batch and chemostat cultures of marine nitrifying bacteria. Arch 671 Microbiol 142: 45-50. 672 Gruber-Dorninger, C., Pester, M., Kitzinger, K., Savio, D.F., Loy, A., Rattei, T., et al. (2015) 673 Functionally relevant diversity of closely related Nitrospira in activated sludge. ISME 674 J 9: 643-655. 675 Hollocher, T.C., Kumar, S., and Nicholas, D.J. (1982) Respiration-dependent proton 676 translocation in Nitrosomonas europaea and its apparent absence in Nitrobacter agilis 677 during inorganic oxidations. J Bacteriol 149: 1013–1020. 678 Itoi, S., Niki, A., and Sugita, H. (2006) Changes in microbial communities associated with the 679 conditioning of filter material in recirculating aquaculture systems of the pufferfish 680 Takifugu rubripes. Aquaculture 256: 287–295. 681 Jenkins, D., and Medsker, L.L. (1964) Brucine method for determination of nitrate in ocean, 682 estuarine, and fresh waters. Anal Chem 36: 610-612. 683 Kang, D.D., Li, F., Kirton, E., Thomas, A., Egan, R., An, H., and Wang, Z. (2019) MetaBAT 684 2: an adaptive binning algorithm for robust and efficient genome reconstruction from 685 metagenome assemblies. PeerJ. 7: e7359.

686 Keuter, S., Kruse, M., Lipski, A., and Spieck, E. (2011) Relevance of *Nitrospira* for nitrite 687 oxidation in a marine recirculation aquaculture system and physiological features of a 688 Nitrospira marina-like isolate. Environ Microbiol 13: 2536–2547. 689 Keuter, S., Beth, S., Quantz, G., Schulz, C., and Spieck, E. (2017) Longterm monitoring of 690 nitrification and nitrifying communities during biofilter activation of two marine 691 recirculation aquaculture systems (RAS). Int J Aquacult Fish Sci 3: 051–061. 692 Kim, S., Kang, I., Lee, J-W., Jeon, C.O., Giovannoni, S.J., and Cho, J-C. (2021) Heme 693 auxotrophy in abundant aquatic microbial lineages. Proc. Natl. Acad. Sci. U. S. A. 118: 694 e2102750118. 695 Kindaichi, T., Ito, T., and Okabe, S. (2004) Ecophysiological interaction between nitrifying 696 bacteria and heterotrophic bacteria in autotrophic nitrifying biofilms as determined by 697 microautoradiography-fluorescence in situ hybridization. Appl Environ Microbiol 70: 698 1641-1650. 699 Kitzinger, K., Marchant, H.K., Bristow, L.A., Herbold, C.W., Padilla, C.C., Kidane, A.T., et 700 al. (2020) Single cell analyses reveal contrasting life strategies of the two main 701 nitrifiers in the ocean. Nat Commun 11: 767. 702 Koch, H., Galushko, A., Albertsen, M., Schintlmeister, A., Gruber-Dorninger, C., Lücker, S., 703 et al. (2014) Growth of nitrite-oxidizing bacteria by aerobic hydrogen oxidation. 704 Science 345: 1052-1054.

- Koch, H., Lücker, S., Albertsen, M., Kitzinger, K., Herbold, C., Spieck, E., et al. (2015)
- Expanded metabolic versatility of ubiquitous nitrite-oxidizing bacteria from the genus
- 707 *Nitrospira. Proc Natl Acad Sci U S A* **112**: 11371–11376.
- To Lancaster, K.M., Caranto, J.D., Majer, S.H., and Smith, M.A. (2018) Alternative bioenergy:
- updates to and challenges in nitrification metalloenzymology. *Joule* 2: 421–441.
- 710 Larsen, P., Nielsen, J.L., Svendsen, T.C., Nielsen, P.H. (2008) Adhesion characteristics of
- 711 nitrifying bacteria in activated sludge. *Water Res.* **42**: 2814–2826.
- 712 Lawson, C.E., Mundinger, A.B., Koch, H., Jacobson, T.B., Weathersby, C.A., Jetten, M.S.M.,
- et al. (2021) Investigating the chemolithoautotrophic and formate metabolism of
- 714 Nitrospira moscoviensis by constraint-based metabolic modeling and ¹³C-tracer
- 715 analysis. mSystems; 6: e00173-21.
- 716 Lebedeva, E.V., Off, S., Zumbrägel, S., Kruse, M., Shagzhina, A., Lücker, S., et al. (2011)
- 717 Isolation and characterization of a moderately thermophilic nitrite-oxidizing bacterium
- from a geothermal spring: Moderately thermophilic *Nitrospira*-cultures from hot
- 719 springs. *FEMS Microbiol. Ecol.* **75**: 195–204.
- 720 Li, D., Liu, CM., Luo, R., Sadakane, K., and Lam, TW. (2015) MEGAHIT: an ultra-fast
- single-node solution for large and complex metagenomics assembly via succinct de
- 722 Bruijn graph. *Bioinformatics* **31**: 1674–1676.

723 Lin, H.H., and Liao, Y.C. (2016) Accurate binning of metagenomic contigs via automated 724 clustering sequences using information of genomic signatures and marker genes. Sci 725 Rep 6: 24175. 726 Lücker, S., Wagner, M., Maixner, F., Pelletier, E., Koch, H., Vacherie, B., et al. (2010) A 727 Nitrospira metagenome illuminates the physiology and evolution of globally 728 important nitrite-oxidizing bacteria. Proc Natl Acad Sci USA 107: 13479–13484. 729 Madigan, M.T., Bender, K.S., Buckley, D.H., Sattley, W.M., and Stahl, D.A. (eds). (2019) 730 Brock biology of microorganisms, 15th. edn. New York, USA: Pearson Education. 731 Mee, M.T., Collins, J.J., Church, G.M., and Wang, H.H. (2014) Syntrophic exchange in 732 synthetic microbial communities. Proc. Natl. Acad. Sci. U. S. A. 111: E2149–E2156. 733 Minh, B.Q., Schmidt, H.A., Chernomor, O., Schrempf, D., Woodhams, M.D., von Haeseler, 734 A., et al. (2020) IQ-TREE 2: new models and efficient methods for phylogenetic 735 inference in the genomic era. Mol. Biol. Evol. 37: 1530-1534. 736 Mobarry, B.K., Wagner, M., Urbain, V., Rittmann, B.E., and Stahl, D.A. (1996) Phylogenetic 737 probes for analyzing abundance and spatial organization of nitrifying bacteria. Appl. 738 Environ. Microbiol. 62: 2156-2162. 739 Okabe, S., Satoh, H., and Watanabe, Y. (1999) In situ analysis of nitrifying biofilms as 740 determined by in situ hybridization and the use of microelectrodes. Appl Environ

741

Microbiol 65: 3182-3191.

Okabe, S., Kindaichi, T., Ito, T. (2005). Fate of ¹⁴C-labeled microbial products derived from 742 743 nitrifying bacteria in autotrophic nitrifying biofilms. Appl Environ Microbiol 71: 744 3987-3994. Oshiki, M., Shimokawa, M., Fujii, N., Satoh, H., and Okabe, S. (2011) Physiological 745 746 characteristics of the anaerobic ammonium-oxidizing bacterium 'Candidatus Brocadia 747 sinica'. Microbiology 157: 1706-1713. 748 Oshiki, M., Aizuka, T., Netsu, H., Oomori, S., Nagano, A., Yamaguchi, T., et al. (2020) Total 749 ammonia nitrogen (TAN) removal performance of a recirculating down-hanging 750 sponge (DHS) reactor operated at 10 to 20°C with activated carbon. Aquaculture 520: 751 734963. 752 Parks, D.H., Imelfort, M., Skennerton, C.T., Hugenholtz, P., and Tyson, G.W. (2014) 753 Assessing the quality of microbial genomes recovered from isolates, single cells, and 754 metagenomes. Genome Res 25: 1043-1055. 755 Park, S-J., Andrei, A-Ş., Bulzu, P-A., Kavagutti, V.S., Ghai, R., and Mosier, A.C. (2020) 756 Expanded diversity and metabolic versatility of marine nitrite-oxidizing bacteria 757 revealed by cultivation- and genomics-based approaches. Appl Environ Microbiol 86: 758 e01667-20.

760 encoding the beta subunit of nitrite oxidoreductase as functional and phylogenetic 761 marker for nitrite-oxidizing Nitrospira. Environ Microbiol 16: 3055–3071. 762 Pritchard, L., Glover, R.H., Humphris, S., Elphinstone, J.G., Toth, I.K. (2016) Genomics and 763 taxonomy in diagnostics for food security: soft-rotting enterobacterial plant pathogens. 764 Anal. Methods 8: 12-24. 765 Rice, E.W., Baird, R.B., Eaton, A.D., and Clesceri, (eds). (2012) Standard methods for the 766 examination of water and wastewater, 22th edn. Washington, USA: American public 767 health association. 768 Richter, M., and Rosselló-Móra, R. (2006) Shifting the genomic gold standard for the 769 prokaryotic species definition. *Proc Natl Acad Sci U S A* **106**: 19126–19131. 770 Rickard, A.H., McBain, A.J., Stead, A.T., and Gilbert, P. (2004) Shear rate moderates 771 community diversity in freshwater biofilms. Appl Environ Microbiol 70: 7426–7435. 772 Rochex, A., Godon, J-J., Bernet, N., and Escudié, R. (2008) Role of shear stress on 773 composition, diversity and dynamics of biofilm bacterial communities. Water Res 42: 774 4915-4922. 775 Rotthauwe, J-H., Witzel, K-P., and Liesack, A.W. (1997) The ammonia monooxygenase 776 structural gene amoA as a functional marker: molecular fine-scale analysis of natural

Pester, M., Maixner, F., Berry, D., Rattei, T., Koch, H., Lücker, S., et al. (2014) NxrB

759

777

ammonia-oxidizing populations. Appl Environ Microbiol 63: 4704–4712.

778 Schramm, A., Beer, D.D., Gieseke, A., and Amann, R. (2000) Microenvironments and 779 distribution of nitrifying bacteria in a membrane-bound biofilm. Environ Microbiol 2: 780 680-686. 781 Sedlacek, C.J., McGowan, B., Suwa, Y., Sayavedra-Soto, L., Laanbroek, H.J., Stein, L.Y., et 782 al. (2019) A physiological and genomic comparison of Nitrosomonas cluster 6a and 7 783 ammonia-oxidizing bacteria. Microbial Ecol 78: 985-994. 784 Sewell, D.L., and Aleem, M.I.H. (1969) Generation of reducing power in chemosynthesis. V. 785 The mechanism of pyridine nucleotide reduction by nitrite in the chemoautotroph 786 Nitrobacter agilis. Biochim Biophys Acta 172: 467–475. 787 Simon, J., and Klotz, M.G. (2013) Diversity and evolution of bioenergetic systems involved 788 in microbial nitrogen compound transformations. Biochim Biophys Acta 1827: 114-789 135. 790 Song, W.Z., and Thomas, T. (2017) Binning refiner: Improving genome bins through the 791 combination of different binning programs. *Bioinformatics* **33**: 1873-1875. 792 Spieck, E., Ehrich, S., Aamand, J., and Bock, E. (1998) Isolation and immunocytochemical 793 location of the nitrite-oxidizing system in Nitrospira moscoviensis. Arch Microbiol 794 **169**: 225–230.

795 Spieck, E., Hartwig, C., McCormack, I., Maixner, F., Wagner, M., Lipski, A., et al. (2006) 796 Selective enrichment and molecular characterization of a previously uncultured 797 Nitrospira-like bacterium from activated sludge. Environ Microbiol 8: 405–415. 798 Tanizawa, Y., Fujisawa, T., and Nakamura, Y. (2018) DFAST: a flexible prokaryotic genome 799 annotation pipeline for faster genome publication. *Bioinformatics* **34**: 1037–1039. 800 Taylor, S., Ninjoor, V., Dowd, D.M., and Tappel, A.L. (1986) Cathepsin B₂ measurement by 801 sensitive fluorometric ammonia analysis. Anal Chem 1974; 60: 153–162. 802 Tsai Y.L., and Tuovinen O.H. Molar growth yield of *Nitrobacter agilis* in batch culture. Can 803 J Microbiol 32: 605-606. 804 Ushiki, N., Fujitani, H., Aoi, Y., and Tsuneda, S. (2013) Isolation of *Nitrospira* belonging to 805 sublineage II from a wastewater treatment plant. *Microbes Environ* **28**: 346–353.00. 806 VanBriesen, J.M. (2001) Thermodynamic yield predictions for biodegradation through 807 oxygenase activation reactions. *Biodegradation* 12: 265–281. 808 van Kessel, M.A.H.J. van Speth, D.R., Albertsen, M., Nielsen, P.H., Camp, H.J.M.O. den, 809 Kartal, B., et al. (2015) Complete nitrification by a single microorganism. *Nature* 528: 810 555-559. 811 Watson, S.W., Book, E., Valois, F.W., Waterbury, J.B., Schlosser, U. (1986) Nitrospira 812 marina gen. nov. sp. nov.: a chemolithotrophic nitrite-oxidizing bacterium. Arch 813 *Microbiol* **144**: 1–7.

814 White, D., Drummond, J., and Fuqua, C. (eds). (2012) The physiology and biochemistry of 815 prokaryotes, 4th edn. New York, USA: Oxford University Press. 816 Whittaker, M., Bergmann, D., Arciero, D., and Hooper, A.B. (2000) Electron transfer during 817 the oxidation of ammonia by the chemolithotrophic bacterium Nitrosomonas 818 europaea. Biochim Biophys Acta 1459: 346-355. 819 Wu, Y.W., Simmons, B.A., and Singer, S.W. (2016) MaxBin 2.0: an automated binning 820 algorithm to recover genomes from multiple metagenomic datasets. Bioinformatics 32: 821 605-607. 822 Xia, F., Wang, J-G., Zhu, T., Zou, B., Rhee, S-K., and Quan, Z-X. (2018) Ubiquity and 823 diversity of complete ammonia oxidizers (commamox). Appl Environ Microbiol 84: 824 e01390-18. 825 Yuan, Z., and VanBriesen, J.M. (2002) Yield prediction and stoichiometry of multi-step 826 biodegradation reactions involving oxygenation. Biotechnol Bioeng 80: 100–113. 827 Zakem, E.J., Al-Haj, A., Church, M.J., van Dijken, G.L., Dutkiewicz, S., Foster, S.Q., et al. 828 (2018) Ecological control of nitrite in the upper ocean. Nat Commun 9: 1206. 829 Zhang, Y., Qin, W., Hou, L., Zakem, E.J., Wan, X., Zhao, Z., et al. (2020) Nitrifier adaptation 830 to low energy flux controls inventory of reduced nitrogen in the dark ocean. Proc Natl Acad Sci U S A 117: 4823-4830. 831

Table 1. Summary of metagenomic bins. *Nitrospira*, *Nitrosomonas*, and *Nitrospina* bins were designated with the label of "NPIRA", "NMNS", and "NPINA", respectively.

Phylogeny of the other DHS20C bins were shown in **Fig. S1**.

	Relative abo	undance (%)						
Bin(s)	NH ₄ ⁺ -	NO ₂	Size	Contigs	CDSs	Completeness	Contamination	Strain
	enriched	enriched	(Mb)					heterogeneity
NPIRA01	1.44	39.7	4.7	203	4,095	90%	3%	50
NPIRA02	35.0	21.2	4.7	92	4,275	97%	3%	33
NPIRA03	1.08	1.50	4.9	218	4,203	94%	4%	20
NPIRA04	14.1	8.88	4.8	633	3,683	81%	4%	30
NPIRA05	0.59	0.73	3.1	595	2,336	59%	0%	0
NPIRA06	1.52	0.21	4.0	303	3,436	85%	5%	57
NMNS01	1.72	0.58	3.7	199	3,093	98%	1%	0
NMNS02	5.86	0.04	3.6	117	3,117	99%	2%	0
NPINA01	0.13	2.10	3.6	14	3,355	97%	3%	0
DHS20C01	1.01	0.73	4.4	119	3,859	99%	3%	8
DHS20C02	0.00	3.43	2.1	12	2,066	96%	1%	0
DHS20C03	1.47	0.51	4.6	234	4,050	100%	2%	82
DHS20C04	1.63	0.00	3.4	133	3,254	95%	0%	0

	Relative abo	undance (%)						
Bin(s)	NH ₄ ⁺ -	NO ₂	Size	Contigs	CDSs	Completeness	Contamination	Strain
	enriched	enriched	(Mb)					heterogeneity
DHS20C05	0.43	0.00	3.0	386	2,588	84%	1%	17
DHS20C06	0.46	0.10	2.4	271	2,137	88%	0%	0
DHS20C07	2.79	0.07	3.6	27	3,178	100%	1%	0
DHS20C08	2.41	0.75	2.9	6	2,510	98%	1%	0
DHS20C09	0.46	0.07	2.8	420	2,257	84%	8%	3
DHS20C10	1.04	0.03	1.7	65	1,458	89%	1%	67
DHS20C11	0.00	0.81	4.7	210	3,873	96%	1%	0
DHS20C12	0.55	0.03	3.6	282	3,076	91%	11%	39
DHS20C13	1.01	0.14	3.4	173	3,114	97%	7%	9
DHS20C14	0.61	0.27	2.9	273	2,298	91%	3%	33
DHS20C15	1.41	0.03	4.5	11	3,500	96%	1%	0
DHS20C16	0.50	0.02	5.3	478	3,780	96%	9%	0
DHS20C17	0.46	0.19	5.2	462	3,634	97%	3%	0
DHS20C18	0.83	0.05	8.0	161	5,634	100%	1%	0
DHS20C19	0.44	0.17	3.7	589	3,085	80%	3%	0
DHS20C20	0.01	0.68	4.9	1044	3,440	73%	15%	70

	Relative abo	undance (%)						
	NH ₄ ⁺ -	NO ₂	Size		CDG			Strain
Bin(s)	enriched	enriched	(Mb)	Contigs	CDSs	Completeness	Contamination	heterogeneity
DHS20C21	0.01	0.69	3.6	653	2,475	77%	4%	0

Table 2 CO₂ assimilation and free-energy efficiencies of NH₄⁺- and NO₂⁻-enriched biomass, AOB, and NOB. CO₂ assimilation efficiencies examining ¹⁴CO₂ fixation into biomass during NH₃ or NO₂⁻ oxidation are summarized here, and the values are available as (mean ± standard deviations). NA; not available because the NO₂⁻-enriched biomass did not show the activity of aerobic ammonia oxidation. *; the value was calculated by subtracting the CO₂ assimilation efficiency of aerobic ammonia oxidation with that of nitrite oxidation. The CO₂ assimilation efficiencies in this table does not include a fraction of the carbon released as dissolved organic carbon (DOC). The reference data of Bayer *et al.* (2022) are a personal gift from Dr. Barbara Bayer.

Biomass/microorganisms	Reaction	μmol-CO ₂ /μmol-NH ₃ or NO ₂ - (Free-energy efficiency)	Reference
NH ₄ ⁺ -enriched biomass	$NH_3 \rightarrow NO_3^-$	0.13 ± 0.019	this study
	$NH_3 \rightarrow NO_2^-$	0.077* (13%)	this study
	$NO_2^- \rightarrow NO_3^-$	$0.053 \pm 0.013 \ (31\%)$	this study
NO ₂ -enriched biomass	$NH_3 \rightarrow NO_2^-$	NA	this study
	$NO_2^- \rightarrow NO_3^-$	$0.054 \pm 0.019 \ (32\%)$	this study
AOB			
Nitrosomonas marina	$NH_3 \rightarrow NO_2^-$	0.04 – 0.07 (7 – 12%)	Glover (1985)
Nitrosococcus oceanus	$NH_3 \rightarrow NO_2^-$	0.024 - 0.062 (4 - 11%)	Glover et al. (1985)
Nitrosomonas marina C-25	$NH_3 \rightarrow NO_2^-$	$0.043 \pm 0.012 \ (7\%)$	Bayer et al. (2022)

Biomass/microorganisms	Reaction	μmol-CO ₂ /μmol-NH ₃ or NO ₂ - (Free-energy efficiency)	Reference	
Nitrosomonas sp. C-15	$NH_3 \rightarrow NO_2^-$	$0.044 \pm 0.007 \ (8\%)$	Bayer <i>et al.</i> (2022)	
NOB				
Nitrococcus mobilis	$NO_2^- \rightarrow NO_3^-$	0.014 - 0.031 (9 – 21%)	Glover (1985)	
Nitrococcus mobilis Nb-	$NO_2^- \rightarrow NO_3^-$	$0.016 \pm 0.002 \ (11\%)$	Bayer <i>et al.</i> , 2022	
Nitrobacter agilis	$NO_2^- \rightarrow NO_3^-$	0.009 (6%)	Tsai and Tuovinen (1986)	
Nitrospira marina Nb-295	$NO_2^- \rightarrow NO_3^-$	$0.032 \pm 0.005 \ (22\%)$	Bayer et al., 2022	
Nitrospina sp. Nb-3	$NO_2^- \rightarrow NO_3^-$	$0.035 \pm 0.005 \ (23\%)$	Bayer et al., 2022	
Nitrospina gracilis Nb-211	$NO_2^- \rightarrow NO_3^-$	$0.029 \pm 0.002 \ (19\%)$	Bayer et al., 2022	

Figure legends

846

847

848

849

850

851

852

853

854

855

856

857

858

859

860

861

862

863

Fig. 1. Fluorescence in-situ hybridization analysis of NH₄⁺-enriched biomass. The biomass was fixed with 4% paraformaldehyde, hybridized with the oligonucleotide probe Ntspa712 (labeled with Cy3) for Nitrospira (panel a) or Nsm156 (TRITC) for Nitrosomonas (panel b), and stained with DAPI. The cells showing red and cyan color represent *Nitrospira* or *Nitrosomonas* population (red) and total cells (cyan), respectively. Bar = $20 \mu m$. Fig. 2. Genome tree showing the phylogeny of Nitrospira bins. A phylogenetic clade of Nitrospira sublineage IV was shown with a bracket, and the phylogenetic position of the obtained Nitrospira bins affiliated into Nitrospira marina and Nitrospira salsa clades were shown with red color. The scale bar represents 10% sequence divergence. Fig. 3. Genome tree showing the phylogeny of *Nitrosomonas* bins. The phylogenetic position of the obtained Nitrosomonas bins were shown with red color. The scale bar represents 5% sequence divergence. Fig. 4. Nitrification activities of NH₄⁺- and NO₂⁻-enriched biomass (left and right panels, respectively). The biomass was aerobically incubated with the addition of NH₄⁺ or NO₂⁻, and the concentrations of NH₄⁺, NO₂⁻, and NO₃⁻ were determined. The incubation was performed in triplicates, and the symbol and error bars represent the mean value and the range of standard deviation, respectively.

Fig. 5. Aerobic H₂ oxidation by NH₄⁺- and NO₂⁻-enriched biomass. The biomass suspension (2.5 mL) was incubated in closed 10-mL vials with the addition of pure H₂ gas into the head space, and the concentrations of H₂ and O₂ in the head space were monitored.

Both the biomass consumed H₂, while the consumption required more than 4 d of lag phase.

Error bars represent the range of standard deviation derived from triplicate incubations.

Fig. 6. Abundance of AOB and *Nitrospira* in the NH₄⁺-enriched biomass and the reactor effluents discharged from the NH₄⁺-feeding DHS reactor. Copy numbers of bacterial 16S rRNA gene, AOB *amoA*, *Nitrospira nxrB*, and *Nitrospira* 16S rRNA gene were determined by quantitative PCR assay. Genomic DNA extraction was performed with >3 biomass samples, and error bar represent the standard deviation of the copy numbers determined from each DNA extracts.













