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Author(s)	Chen, Binjie; Kang, Kungwan; Jeen, Hyoungjeen; Zhang, Yuqiao; Lin, Jinghuang; Feng, Bin; Ikuhara, Yuichi; Hoshino, Sena; Matsunaga, Katsuyuki; Ohta, Hiromichi
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២ Binjie Chen, ២ Kungwan Kang, ២ Hyoungjeen Jeen, et al.



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Binjie Chen,^{1,a)} ⁽¹⁾ Kungwan Kang,² ⁽¹⁾ Hyoungjeen Jeen,² ⁽¹⁾ Yuqiao Zhang,^{3,4} ⁽¹⁾ Jinghuang Lin,⁵ Bin Feng,⁵ ⁽¹⁾ Yuichi Ikuhara,⁵ ⁽¹⁾ Sena Hoshino,⁶ ⁽¹⁾ Katsuyuki Matsunaga,⁶ ⁽¹⁾ and Hiromichi Ohta^{7,a)} ⁽¹⁾

AFFILIATIONS

¹Graduate School of Information Science and Technology, Hokkaido University, N14W9, Kita, Sapporo 060-0814, Japan

²Department of Physics, Pusan National University, Busan 46241, South Korea

³Institute of Quantum and Sustainable Technology (IQST), School of Chemistry and Chemical Engineering, Jiangsu University, Zhenjiang 212013, China

⁴Foshan (Southern China) Institute for New Materials, Foshan 528200, China

⁵Institute of Engineering Innovation, The University of Tokyo, 2-11-16 Yayoi, Bunkyo, Tokyo 113-8656, Japan

⁶Department of Materials Physics, Nagoya University, Furo, Chikusa, Nagoya 464-8603, Japan

⁷Research Institute for Electronic Science, Hokkaido University, N20W10, Kita, Sapporo 001-0020, Japan

^{a)}Authors to whom correspondence should be addressed: binjie.chen.t7@elms.hokudai.ac.jp and hiromichi.ohta@es.hokudai.ac.jp

ABSTRACT

Titanium dioxide (TiO₂) is an excellent photocatalyst and transparent conducting oxide. It has two major crystal structures: rutile and anatase. Anatase TiO₂ is valuable from an industrial point of view because it typically displays better photocatalytic and electronic transport properties than rutile TiO₂. To further extract the functional properties of TiO₂, understanding the correlation between the electronic structure and the crystal structure is essential. Because the electronic structure strongly depends on the crystal structure, introducing lattice distortion to rutile TiO₂ should effectively modulate its electronic structure. Here, we show that Nb-doped rutile TiO₂ epitaxial films on (1100) α -Al₂O₃ substrates exhibit anatase-like optoelectronic properties due to orthorhombic lattice distortions and experimentally observe the change in the electronic structure. Reducing the film thickness increases the orthorhombic distortion ratio (*b/a*) up to 3.4%. As *b/a* increases, the carrier effective mass decreases from 35 to 3 *m*₀ (*m*₀: electron mass) and the optical bandgap significantly increases. The present observations provide insight into regulating the TiO₂ physical properties and should be beneficial for designing TiO₂-based photocatalysts and transparent conducting electrodes.

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I. INTRODUCTION

Titanium dioxide (TiO₂) is an active material for photocatalytic water splitting. The optical bandgap (E_g) of TiO₂ is 3.1–3.3 eV, which is close to the maximum energy of visible light (~3.1 eV).^{1–3} Fundamentally, the E_g of TiO₂ is composed of a Ti 3d-based conduction band minimum (CBM) and an O 2p-based valence band maximum (VBM). When TiO₂ absorbs light (photon energy > E_g), the valence electron is excited to the conduction band, generating a positive hole in the valence band. The conduction electron reduces water while the hole oxidizes it.^{4–7} In addition, the photocatalytic activity strongly depends on the crystal structure of TiO₂ because the electronic structure is affected by the crystal structure.^{8–12}

Thus, understanding the correlation between the electronic and crystal structures is essential to extract the functional properties of TiO_2 .

TiO₂ has a series of stable polymorphs in nature, e.g., rutile, anatase, brookite, etc. Among them, rutile and anatase are the most thermodynamically stable and representative. Rutile is the most stable phase at high temperatures. Anatase is stable at low temperatures but changes into the rutile phase when heated up above 600 °C.^{13,14} The photocatalytic activity of the anatase phase is an order of magnitude higher than that of the rutile phase, ^{11,15} possibly due to higher mobility and the nature of indirect bandgap. Still due to the conduction electrons of the anatase phase having a higher mobility than those of the rutile phase, ^{16,17} it is more

suitable as transparent conducting oxide (TCO) when donor dopants are doped. Therefore, anatase TiO_2 is more valuable than rutile TiO_2 from an industrial point of view.

To clarify the difference, high-quality epitaxial films of TiO₂ are useful. Although high-quality anatase TiO₂ can be heteroepitaxially grown on several perovskite oxide substrates (SrTiO₃ and LaAlO₃) at high temperatures ($T_g \sim 700 \,^{\circ}\text{C}$),^{18–21} fabricating largearea films is difficult due to the size limitation of the substrates. This study focuses on the anisotropic crystal structure of α -Al₂O₃ substrates. Figure 1 depicts the lattices of (001) rutile TiO₂ and (1100) α -Al₂O₃ crystals. The epitaxial relationship is (001) [100] TiO₂ || (1100) [0001] α -Al₂O₃ in the in-plane directions, the rutile TiO₂ lattice sustains compressive strain along [100] (lattice mismatch: + 5.9%), whereas tensile strain occurs along [010] (lattice mismatch: -3.5%). Because the electronic structure strongly depends on the crystal structure, introducing lattice distortion to rutile TiO₂ should effectively modulate its electronic structure.

Here, we show that electron-doped rutile TiO_2 epitaxial films on $(1\overline{1}00) \alpha$ -Al₂O₃ substrates exhibit anatase-like optoelectronic properties, presumably due to orthorhombic lattice distortions.



FIG. 1. Hypothesis of the orthorhombic lattice distortion of the rutile-structured TiO₂ epitaxial film on an M-plane sapphire substrate. Schematic of the crystal lattice of rutile TiO₂ on (1100) α -Al₂O₃ (sublattice of oxide ions, dotted lines). Since lattice parameter *a* of TiO₂ is longer than lattice parameter *c* of α -Al₂O₃ (lattice mismatch: -5.9%), compressive strain is assumed, whereas the tensile strain is assumed because lattice parameter *a* of α -Al₂O₃ is larger than that of the lattice parameter *b* of TiO₂ (lattice mismatch: +3.5%). Therefore, we hypothesized that orthorhombic distortion is introduced in the TiO₂ lattice. For convenience, the TiO₂ lattice is simplified to the sublattice of oxide ions, which is shown by the dotted lines.

In addition, we demonstrate that the TiO_2 film thickness controls the degree of orthorhombic distortion. The present observations provide insight into regulating the physical properties of TiO_2 and should be beneficial for designing TiO_2 -based photocatalysts and transparent conducting electrodes.

II. EXPERIMENTAL SECTION

A. Sample preparation

The target was a 0.5%-Nb-doped TiO₂ ceramic disk. Pulsed laser deposition (PLD) with a KrF excimer laser (λ = 248 nm, laser fluence ~1 J cm⁻² pulse⁻¹, and repetition rate = 10 Hz) was used to fabricate 0.5%-Nb-doped TiO₂ films on (1100) α -Al₂O₃ single crystal substrates, where the slight doping is to provide the donor dopant to facilitate further electronic measurements. Such low-level doping does not affect the lattice structure significantly.^{25,26} During film growth, the substrate temperature and oxygen pressure were kept constant at 500 °C and 5 × 10⁻³ Pa, respectively. Adjusting the deposition time controlled the film thickness.

B. Crystallographic analyses

The film thickness, crystallographic orientations, and lattice parameters were determined using high-resolution x-ray diffraction (XRD, ATX-G, Rigaku) with Cu K α_1 ($\lambda = 0.154059$ nm) radiation. Both out-of-plane and in-plane measurements were performed to extract the lattice parameters. The atomic arrangement of the resultant films was visualized using STEM (ARM200CF, JEOL Co. Ltd) operated at 200 keV. HAADF-STEM images were taken with a detection angle of 68–280 m rad.

All measurements were conducted at room temperature.

C. Electrical and optical properties' measurements

The electrical resistivity (ρ), carrier concentration (n), and Hall mobility (μ_{Hall}) at room temperature were measured by the dc four-probe method with the van der Pauw electrode configuration in air. The temperature-dependent electrical resistivity ($\rho - T$), carrier concentration (n - T), and Hall mobility ($\mu_{\text{Hall}} - T$) were performed using the same configuration in a vacuum and a variable temperature of 10–300 K. The thermopower (S) of the films in the in-plane direction was measured by the conventional steady-state method. The transmission and reflection spectra in the 200–1600 nm wavelength range were measured by a spectrophotometer (SolidSpec-3700, Shimadzu).

D. X-ray absorption spectroscopy (XAS) measurements

XAS measurements were performed at the 2A beamline of the Pohang Accelerator Laboratory (PAL) with the grazing incidence (22.5°) configuration in the total-electron-yield (TEY) mode. The Ti L-edge spectra were measured at room temperature.

E. Density functional theory (DFT) calculations

DFT calculations were performed with the projector-augmented wave method implemented in the Vienna *Ab initio* Simulation Package (VASP) program.^{27,28} For the exchange-correlation term of the electrons, the generalized gradient approximation (GGA) parameterized by Perdew, Burke, and Ernzerhof was used.²⁹ The electrons

TABLE I. Calculated effective masses for electrons in non-distorted (case 1) or orthogonally distorted (cases 2 and 3) rutile TiO_2 .

	Effective mass m^*/m_0			
	$\Gamma \rightarrow X$	$\Gamma { ightarrow} Y$	$\Gamma{\rightarrow}M$	$\Gamma \rightarrow S$
Case 1 $(a_0, b_0 = a_0, c_0)$	1.672		0.588	
Case 2 $(a_0, b_0 = 1.030a_0, c_0)$ Case 3 $(0.995a_0, b_0 = 1.030a_0)$	1.777	1.698		0.656
$\begin{array}{c} \text{Case 5} (0.995u_0, v_0 - 1.050u_0, \\ 0.994c_0) \end{array}$	1.759	1.664		0.742

were $3p^6$, $3d^2$, and $4s^2$ for Ti and $2s^2$ and $2p^4$ for O. To describe the localized features of the Ti 3d orbitals, the GGA + U method was employed, where the onsite Coulomb repulsion was considered with an effective parameter of U = 7.0 eV for Ti $3d^{.30,31}$ Han *et al.* suggested the optimum choice of the U parameter for TiO₂.³¹ The electronic wavefunctions were expanded by plane waves up to a cutoff energy of 500 eV. Brillouin zone integration was performed with the Monkhorst-Pack mesh of $4 \times 4 \times 6$ for the unit cell of rutile TiO₂. The structure was optimized until the atomic forces converged to less than 0.01 eV/Å. The calculated lattice parameters of a = 0.469 nm and c = 0.305 nm slightly overestimated the experimental data,³² but the trends were similar to those in previous GGA calculations.³¹

The present GGA + U calculations indicated that rutile TiO₂ had a direct theoretical $E_{\rm g}$ of 2.30 eV at the Γ point (Fig. S1 in the supplementary material). To evaluate effective masses (m^*) of electrons in rutile TiO₂, the energy dispersion curves at the CBM were obtained around the Γ point for the rutile TiO₂ unit cell (structure-optimized as above) using a k-point mesh density toward a particular symmetric point of more than 4.2 nm. Subsequently, parabolic curve fitting for five k-points centered at Γ was performed, and the m^* values of electrons were obtained from the second partial derivatives of the parabolic fitting curves.³³

Although the theoretical E_g (2.30 eV) of bulk rutile TiO₂ underestimated the experimental value (3.03 eV),³⁴ which is a standard feature of GGA-related DFT calculations, such a small theoretical E_g may also affect the theoretical m^* . To evaluate the effect of the theoretical E_g on m^* , hybrid-functional calculations only at the Γ point were also performed using the hybrid density functional proposed by Heyed, Scuseria, and Ernzerhof (HSE06).^{35,36} The E_g of 3.20 eV calculated by HSE06 reproduced the experimental E_g value better than that by the GGA + U calculations. However, the calculated m^* of 1.168 for Γ to X and 0.414 for Γ to M were slightly smaller than the corresponding GGA + U values of 1.672 and 0.588, respectively (Table 1). However, these differences do not affect the conclusion of this paper. Since hybrid-functional calculations are computationally demanding, the GGA + U calculations were mainly used to investigate m^* in distorted rutile TiO₂.

III. RESULTS AND DISCUSSIONS

A. Epitaxial film growth of rutile Nb:TiO₂ with orthorhombic distortion

Figure 2 summarizes the crystallographic characterization of the resultant Nb:TiO₂ films on ($1\overline{1}00$) α -Al₂O₃ substrates. The out-of-plane XRD patterns exhibit intense 002 diffraction peaks of rutile Nb:TiO₂ and those of $3\bar{3}00 \alpha$ -Al₂O₃ substrate [Fig. 2(a)], indicating a high *c*-axis orientation. As the thickness increases, the 002 peak shifts toward the smaller scattering vector and approaches the bulk position (dashed line), which suggests an increasing trend of the lattice parameter *c* [Fig. 2(c)]. The in-plane XRD patterns [Fig. 2(b)] have different peak positions for the 400 (upper) and the 040 (lower). The 400 diffraction peaks of Nb:TiO₂ are seen together with those of 00 012 α -Al₂O₃ substrates at $\sim q_x/2\pi = 8.75 \text{ nm}^{-1}$. This is slightly larger than that of bulk (dashed line), but the 400 peaks are independent of thickness. In contrast, the 040 diffraction peaks of Nb:TiO₂, which are seen together with $22\bar{4}0 \alpha$ -Al₂O₃ substrates, shift toward a larger scattering vector as the thickness increases and approaches $\sim q_y/2\pi = 8.67 \text{ nm}^{-1}$. This value is slightly smaller than that of the bulk (dashed line).

Figure 2(d) summarizes the in-plane lattice parameters a and b as functions of the film thickness. Note that the lattice parameters of the 10-nm-thick and 20-nm-thick samples were extracted from the 200 Nb:TiO₂ and 020 Nb:TiO₂ diffraction peaks due to the weak peak intensities of 400 Nb:TiO2 and 040 Nb:TiO2 (Fig. S2 in the supplementary material). Lattice parameter a is ~ 0.457 nm, which is slightly shorter than that of the bulk (0.4594 nm) due to the compressive strain originating from the lattice mismatch (-5.9%). Lattice parameter *a* is independent of the thickness. Unlike lattice parameter a, lattice parameter b shows a clear thickness dependence. For a 10-nm-thick film, lattice parameter b is 0.473 nm, which is ~3% longer than that of bulk rutile TiO_2 . However, it decreases to 0.461 nm, which is slightly longer than that of the bulk when the thickness is 300 nm. This clearly indicates the presence of a strong tensile strain originating from the lattice mismatch (+3.5%) (Table II).

To visualize the orthorhombic lattice distortion, we defined the ratio of lattice parameters (*b/a*) as the degree of orthorhombic distortion [Fig. 2(e)]. The ratio *b/a* gradually decreases as the Nb: TiO₂ thickness increases and ranges from ~1.035 to ~1.01. It should be noted that the Nb:TiO₂ films were grown incoherently on (1100) α -Al₂O₃ substrates (Figs. S3–S8 in the supplementary material). These results confirm that orthorhombic strain is successfully introduced to the rutile TiO₂ and that the strain becomes smaller as the thickness increases. Also note that the change tendency of lattice parameter *b* slowed down when the films are thicker than 100 nm, indicating such tensile strain has been highly weakened under this condition.

To further clarify the orthorhombic distortion, cross-sectional STEM images were acquired for 6- and 200-nm-thick TiO_2 films prepared under the same growth conditions. Figure 3 shows the high-resolution HAADF-STEM images along the [0001]Al₂O₃ zone axis and the corresponding inverse fast Fourier transformation (IFFT) images. HAADF-STEM images [Figs. 3(a) and 3(b)] exhibit a clear Z-contrast for the Ti and Al atoms. Consequently, the interface of TiO_2/M -plane sapphire can be observed, as shown by the dashed line.

Comparing the observed and theoretical structures of TiO_2 demonstrates that both the 6- and 200-nm-thick films show (001) oriented rutile structures, which is consistent with our XRD results. IFFT is an intuitive method for observing dislocations. The lack of dislocations in IFFT of the 6-nm-thick film indicates that the TiO_2



FIG. 2. Crystallographic characterization of the resultant TiO₂ films on an M-plane sapphire substrate. (a) Out-of-plane and (b) in-plane XRD patterns of 5–300-nm-thick TiO₂ films grown on a $(1\overline{1}00) \alpha$ -Al₂O₃ substrate. For the in-plane pattern, the upper and lower figures correspond to the XRD patterns among azimuth [0001] α -Al₂O₃ and azimuth [1120] α -Al₂O₃, respectively. Dotted lines indicate the position of bulk TiO₂. Thickness dependence of the (c) out-of-plane lattice parameter *c* and (d) in-plane lattice parameters *a* and *b*, which are extracted from the XRD patterns. Dotted lines indicate the lattice parameters of bulk TiO₂. (e) Orthorhombic distortion *b/a* of the resultant TiO₂ films.

lattice is strained [Fig. 3(c)]. In contrast, the edge dislocations at the interface of the 200-nm-thick film [Fig. 3(d)] suggest that the strain is released by forming dislocations. The uneven distribution of dislocations may be due to no post-annealing. Based on the dislocation observations, the orthorhombic distortion originates from the elastic deformation of the lattice and was relieved

TABLE II. The comparison of lattice mismatch and distortion along [0001] Al_2O_3 and [1120] Al_2O_3 direction.

Substrate orientation	0001	1120
Film orientation	100	010
Lattice mismatch	+5.9%	-3.5%
Lattice parameter of	0.457 nm (10 nm)	0.473 nm (10 nm)
films	0.457 nm (300 nm)	0.461 nm (300 nm)
Relation between lattice		
parameter and		
thickness	Weak	Strong

through dislocation formation. In addition, we also observed the HAADF-STEM images along the[$11\overline{2}0$]Al₂O₃ zone axis of these two films (Fig. S9 in the supplementary material). The existence of edge dislocations in both films along this direction is also well consistent with the smaller change tendency of lattice parameter *a* [Fig. 2(d)]. For the 200-nm-thick films, we also observed the lattice parameter difference along the out-of-plane. No clear atom spacing difference was observed, indicating that the strain release only occurs around the dislocation at the interface instead of a gradient release from the interface to the surface. Such phenomenon has been experimentally revealed by other reports, ^{57,38} and it also confirms the uniformity of the film.

B. Electrical and optical properties of the rutile Nb:TiO $_2$ films

The electron transport properties of the resultant $Nb:TiO_2$ films were measured at room temperature. Figure 4 summarizes the electron transport properties as a function of the



FIG. 3. Cross-sectional STEM images of the TiO₂ films on an M-plane sapphire substrate. (a) and (b) HAADF-STEM images of (a) 6-nm-thick and (b) 200-nm-thick TiO₂ films grown on ($1\overline{100}$) α -Al₂O₃ substrates along the [0001]Al₂O₃ zone axis. The dashed line indicates the interface of TiO₂/ α -Al₂O₃. (c) and (d) Corresponding IFFT images of (c) 6-nm-thick and (d) 200-nm-thick TiO₂ films. The red marker denotes edge dislocations. Insets are the corresponding FFT patterns.

orthorhombic distortion (i.e., b/a). The electrical conductivity (σ) is 2–3 S cm⁻¹ and is independent of b/a [Fig. 4(a)]. Similar to the Nb concentration ($0.5\% \approx 1.6 \times 10^{20}$ cm⁻³), the carrier concentration (n) is in the range of $10^{19}-10^{20}$ cm⁻³ and shows a gradual decreasing tendency as b/a increases [Fig. 4(b)]. For a semiconductor film depositing on sapphire, the band bending possibly occurs around the interface. Thus, the carriers within a contained thickness would be trapped, resulting in the decreasing of n when thickness decreased. In contrast, μ_{Hall} shows the opposite trend [Fig. 4(c)]. For a small b/a ratio (~1.01), the μ_{Hall} value is ~0.3 cm² V⁻¹ s⁻¹. μ_{Hall} increases with b/a and exceeds 1 cm² V⁻¹ s⁻¹ when b/a = 1.034. The variety of μ_{Hall} should be related to the change of effective mass, which would be revealed later.

It should be noted that the thermopower (-S) gradually decreases with b/a from ~600 to ~430 μ V K⁻¹ [Fig. 4(d)]. According to Mott's formula, -S reflects n for semiconductors; -S decreases as n increases if the Fermi energy increases on the parabolic electronic density of states (DOS) and m^* is constant. Thus, the observed tendencies of -S and n reveal that the m^* becomes lighter when orthorhombic lattice distortion is introduced.

For visualization, m^* is estimated from the relation between n and S as³⁹

$$S = -rac{k_B}{e} \left(rac{(r+2)F_{r+1}(\xi)}{(r+1)F_r(\xi)} - \xi
ight),$$

 $n = 4\pi \left(rac{2m^*k_BT}{h^2}
ight)^{3/2} F_{1/2}(\xi),$

where k_B , r, F_r , ξ , T, and h are the Boltzmann constant, scattering parameter of the relaxation time, Fermi integral, chemical potential, absolute temperature, and Plank's constant, respectively. F_r can be further expressed as

$$F_r(\xi) = {\stackrel{\infty}{_0}} \frac{x^r}{1 + e^{x - \xi}} dx$$

Figure 4(e) plots the estimated m^* as a function of b/a. m^* of the almost relaxed rutile TiO₂ film ($b/a \sim 1.008$) is $\sim 35 m_0$. This value is consistent with the reported m^* of bulk rutile TiO₂ (12–35 m_0).^{40–43} It should be noted that the m^* decreases as b/a



FIG. 4. Room temperature electron transport properties of the TiO₂ films as the function of orthorhombic distortion. (a) Electrical conductivity (σ), (b) carrier concentration (n), (c) Hall mobility (μ_{Hall}), (d) Thermopower (S), and (e) DOS effective mass (m^*/m_0). The gray region and red dotted line in (e) indicate the reported effective masses of rutile^{40–43} and anatase TiO₂,^{17,19} respectively. Solid lines are to guide the eye. Note that *n* decreases while |S| increases with *b/a*.

increases and reaches $\sim 3 m_0$ when b/a = 1.034. This smaller m^* is closer to that of anatase TiO₂ ($\sim 1 m_0$).^{17,19} So far, we demonstrated the relation between distortion and effective mass. The decreasing tendency is corresponding to the increase of mobility, revealing the major factor in the change of mobility.

In addition to the electron transport properties, $E_{\rm g}$ is also a property associated with the electronic structure. There is a consensus that the $E_{\rm g}$ of bulk rutile TiO₂ is ~3.06 eV while that of anatase is ~3.20 eV.^{1,41,45} To clarify the effect of the orthorhombic distortion on $E_{\rm g}$, we measured the optical transmission and reflection spectra of the resultant films (Fig. S10 in the supplementary material). It should be noted that, although rutile TiO₂ is a direct bandgap semiconductor, the direct transition is dipole-forbidden and degenerated with the indirect allowed transition.^{46–48} Thus, direct $E_{\rm g}$ is equivalent to indirect $E_{\rm g}$ and was possible to be evaluated by extrapolating the linear region of the Tauc $[(\alpha h v)^{1/2} - hv]$ plot (Fig. S11 in the supplementary material). $E_{\rm g}$ gradually increases from ~3.11 to ~3.48 eV as b/a increases [Fig. 5(a)]. Even a highly relaxed film (300 nm) shows a small distortion ($b/a \sim 1.008$), which may explain why $E_{\rm g}$ is slightly higher than the rutile bulk value (~3.06 eV).⁴⁹

Generally, the E_g of an extremely thin layer is greater than that of the bulk due to the quantum size effect (QSE). QSE occurs



FIG. 5. Denying the quantum size effect of TiO₂ epitaxial films. Change in E_g of the TiO₂ films. (a) E_g increases with *b*/*a*. (b) E_g increases as the thickness decreases. This increasing tendency completely differs from the assumption for the quantum size effect.

when the film thickness is thinner than the de Broglie wavelength. $E_{\rm g}$ gradually increases as the film thickness decreases [Fig. 5(b)]. To clarify the origin of the increased $E_{\rm g}$ we calculated $E_{\rm g}$ assuming that QSE can be expressed as^{50,51}

$$E_g = E_{g, Bulk} + rac{h^2}{2d^2} \left(rac{1}{m_e^*} + rac{1}{m_h^*}
ight),$$

where $E_{g, Bulk}$, h, and d are the bulk bandgap, Planck's constant, and film thickness, respectively. m_e^* and m_h^* are the electron and hole effective masses, which are assumed to be $\sim 20^{41}$ and $\sim 5 m_0^{52}$ respectively. The calculated E_g increases dramatically when the film thickness is thinner than ~ 10 nm. This behavior does not fit with the observed E_g , indicating that the QSE theory cannot explain the observed E_g . These observations confirm that the electronic structure of a rutile TiO₂ film is predominantly modulated via orthorhombic distortion.

Nb:TiO₂ films grown on $(1\bar{1}00) \alpha$ -Al₂O₃ substrate show differentiated properties from the rutile phase when the thickness is sufficiently thin due to orthorhombic lattice distortion but typical properties of the rutile phase appear as the thickness increases. Since rutile Nb:TiO₂ shows interesting transport properties at low temperatures and the temperature-related behavior may provide information about the transport mechanism, we measured the temperature dependence of the electron transport properties of the 300-nm-thick Nb:TiO₂ film (*b/a* ~ 1.008) and the 20-nm-thick Nb:TiO₂ film (*b/a* = 1.034). Both films show semiconductor-like behaviors of ρ [Fig. 6(a)]. As the temperature decreases, ρ gradually increases, *n* gradually decreases [Fig. 6(b)], and μ_{Hall} increases [Fig. 6(c)]. These results suggest that phonons dominate carrier scattering.

The films display different temperature dependences of -S [Fig. 6(d)]. At room temperature, the -S values of the 300- and the 20-nm-film are ~0.6 and 0.45 mV K⁻¹, respectively. As the temperature decreases, the former increases dramatically and exceeds 1.5 mV K⁻¹. In contrast, the latter decreases gradually and approaches zero. It has been observed that rutile TiO₂ shows a colossal -S at low temperature due to the phonon drag effect.^{40,53} This behavior is usually observed in materials with a large $m^{*.54-57}$ As above described, m^* of the 300-nm-film is $35 m_0$ while that of the 20-nm-film is $\sim 3 m_0$. Similarly, the presence or absence of a colossal -S reflects the distortion-induced m^* modulation at room temperature.

These observations reveal that introducing orthorhombic lattice distortion using (1100) α -Al₂O₃ substrates significantly modulates the electronic structure of rutile TiO₂. Orthorhombic distortion realizes changes in E_g , m^* , and -S, indicating that the electronic structure of rutile TiO₂ shifts toward that of anatase TiO₂.

C. Changes in the electronic structure of rutile TiO_2 due to orthorhombic distortion

The XAS spectra of the TiO₂ films were measured to directly clarify the change in the electronic structure of rutile TiO₂ upon orthorhombic distortion. Figure 7 shows the XAS spectra of the 10–200-nm-thick rutile TiO₂ films at the Ti L₃-edge. The single peak around ~456 eV corresponds to the threefold degenerate t_{2g} state, and the double peak around ~459 eV corresponds to the



FIG. 6. Low-temperature electron transport properties of the TiO₂ films. (a) Resistivity, (b) carrier concentration (n), (c) Hall mobility (μ_{Hall}), and (d) thermopower (S) of 20- and 300-nm-thick rutile TiO₂ films grown on (1100) α -Al₂O₃ substrates. Note that the |S|-*T* curve of the 300-nm-thick TiO₂ is similar to that of the reduced rutile TiO₂ single crystal, whereas that of the 20-nm-thick TiO₂ is similar to that of the anatase TiO₂ epitaxial film.

twofold degenerate e_g state. The e_g peak is further split into two peaks (e_{g1} and e_{g2}) due to the octahedral distortion of TiO₆ octahedra. The octahedral distortion differs between rutile and anatase. Rutile is D_{2h} symmetry, while anatase is D_{2d} . This difference is reflected in the XAS spectra as the inversion of the relative intensities of the e_{g1} and e_{g2} peaks. Hence, their intensities can be used as a fingerprint for different TiO₂ polymorphs (Fig. S12 in the supplementary material).⁵⁸ The observed XAS spectra indicate that the TiO₂ films maintain the electronic structure of the rutile phase. Although the anatase phase does not appear even with a



FIG. 7. Ti L₃ XAS spectra of the 10–200-nm-thick rutile TiO₂ films. (a) 10-nm-thick TiO₂, (b) 20-nm-thick TiO₂, (c) 90-nm-thick TiO₂, and (d) 200-nm-thick TiO₂ grown on (1100) α -Al₂O₃ substrates. Peak fitting is processed by the Gaussian distribution. (e) Relative peak area ratio of e_{g2}/e_{g1} as a function of orthorhombic distortion. The inset plots the e_{g2}/e_{g1} ratio as a function of the film thickness.

large orthorhombic distortion, we consider that changes in the lattice parameters result in changes in the degree of octahedral distortion. This small perturbation has a non-negligible effect on the physical properties.^{9,59,60} From this viewpoint, we fitted the XAS data into a Gaussian distribution (the fitting parameters were available in Table S1 in the supplementary material) and calculated the relative peak area ratio of e_{g2}/e_{g1} [Figs. 7(a)–7(d)]. As the thickness increases, e_{g2}/e_{g1} linearly decreases as b/a increases [Fig. 7(e)], clearly indicating that there is a strong correlation between the electronic structure of rutile TiO₂ and the orthorhombic lattice distortion.

D. DFT calculation of TiO_2 with orthorhombic lattice distortion

Finally, DFT calculations of rutile TiO₂ with orthorhombic lattice distortion were performed to investigate the influence of lattice distortion of TiO₂ thin films on electron m^* . Three TiO₂ unit cells were considered: (1) a rutile unit cell with fully optimized *a*- and *c*-axis lengths (a_0 , $b_0 = a_0$, c_0), (2) a rutile unit cell with the *b*-axis length extended 3% from the fully optimized value (a_0 , 1.030 b_0 , c_0), and (3) a rutile unit cell with the *a*-axis length compressed 0.5% and the *b*-axis length extended 3% (0.995 a_0 , 1.03 b_0 , 0.994 c_0). These extensions and compressions of a_0 and b_0 are based on the experimental data shown in Fig. 2. It should be noted that the *c*-axis length in case 3 was optimized by the DFT calculations, and the optimized *c*-axis length of 0.994 c_0 agrees well with that for the thin-film value (about 0.9925 c_0) shown in Fig. 2(c). Moreover, the theoretical E_g value for the distorted unit cell (case 3) is 2.37 eV, which is larger than the fully optimized unit cell value of 2.30 eV. The increased $E_{\rm g}$ trend for the distorted rutile unit cell agrees well with the experimental $E_{\rm g}$ variation against orthogonal distortion shown in Fig. 5(a).

Table I lists the calculated m^* values at the Γ point toward the X (or Y) point and the M (or S) point (these directions correspond to the in-plane ones to the thin-film surface). The DFT results show that the m^* 's of the fully optimized rutile TiO₂ are around 1.7 and $0.6 m_0$, which are much smaller than the experimentally observed value ($\sim 35 m_0$). Previous GGA calculations provided similar theoretical results.47,61,62 It is thought that electronic conduction in rutile TiO_2 is not simply described by the m^* -like states, but is strongly affected by the formation of polarons, namely, electron coupling with lattice vibrations (phonons). As the orthogonal distortions increase (cases 2 and 3), the calculated m^* values tend to increase. However, the m^* values are still smaller than the experimentally observed one $(3 m_0)$ for the distorted thin films [Fig. 4(e)]. Hence, it is most likely that the original orthogonal distortions of the rutile TiO₂ thin films due to the substrate considerably change the lattice vibrational states in the thin films.

IV. CONCLUSION

Nb-doped rutile TiO₂ epitaxial films on $(1\overline{1}00) \alpha$ -Al₂O₃ substrate exhibit anatase-like optoelectronic properties due to orthorhombic lattice distortions. As the film thickness decreases, b/aincreases up to 3.4%. As b/a increases, the carrier m^* decreases from 35 to 3 m_0 (m_0 : electron mass) but E_g increases. The electronic structure change was observed experimentally. The present observations provide insight into regulating the physical properties

of TiO₂ and should aid in the design of TiO₂-based photocatalysts and transparent conducting electrodes.

SUPPLEMENTARY MATERIAL

See the supplementary material for x-ray reciprocal space mappings (RSMs) around (left) ($3\overline{3}06$) α -Al₂O₃ and (right) ($4\overline{2}\overline{2}0$) α -Al₂O₃ diffraction spots of the TiO₂ film (300, 200, 100, 50, 20, and 10 nm) grown on $(1\overline{1}00) \alpha$ -Al₂O₃ substrates. Optical transmission and reflection spectra of the 5-300-nm-thick TiO2 films grown on (1010) α -Al₂O₃ substrates and Tauc plots ((αhv)^{1/2} – hv) of the 5–300-nm-thick rutile TiO₂ films grown on $(1\overline{1}00) \alpha$ -Al₂O₃ substrates.

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AUTHOR DECLARATIONS

Conflict of Interest

The authors have no conflicts to disclose.

Author Contributions

B.C. prepared the samples and conducted the measurements. K.K. and H.J. performed the XAS measurements and analyses. Y.Z. analyzed the effective mass. J.L., B.F., and Y.I. performed the STEM observations and analyses. S.H and K.M. performed DFT calculations. B.C. and H.O. planned and supervised the project. All authors discussed the results and commented on the manuscript.

Binjie Chen: Conceptualization (equal); Data curation (equal); Formal analysis (equal); Investigation (equal); Validation (equal); Writing - original draft (equal); Writing - review & editing (equal). Kungwan Kang: Investigation (equal); Writing - review & editing (equal). Hyoungjeen Jeen: Investigation (equal); Writing review & editing (equal). Yuqiao Zhang: Investigation (equal); Validation (equal); Writing - review & editing (equal). Jinghuang Lin: Investigation (equal); Writing - review & editing (equal). Bin Feng: Investigation (equal); Writing - review & editing (equal). Yuichi Ikuhara: Investigation (equal); Writing - review & editing

(equal). Sena Hoshino: Investigation (equal); Writing - review & editing (equal). Katsuyuki Matsunaga: Investigation (equal); Writing – review & editing (equal). Hiromichi Ohta: Conceptualization (equal); Data curation (equal); Funding acquisition (equal); Project administration (equal); Supervision (equal); Writing - original draft (equal); Writing - review & editing (equal).

DATA AVAILABILITY

The data that support the findings of this study are available from the corresponding author on reasonable request.

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