



HOKKAIDO UNIVERSITY

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Errata

page	line	error	correct
28	34	three volumes	four volumes
31	10	and k_1^*/k_2^* is	and k_2^*/k_1^* is
32	FIG. 4	in the presence of Ca^{+2} .	in the presence of Ca^{2+}
33	TABLE II	[ATP] 1×10^{-5} mole/lit.	[ATP] 1×10^{-3} mole/lit.
34	FIG. 5	1×10^{-2} mole/lit. (Δ)	1×10^{-2} mole/lit (Δ)
37	8	as measured	was measured
37	10	FIG. 9 a.	Fig. 10 a.
37	11	FIG. 9 b.	Fig. 10 b.
37	13	specially	specially,
39	3	light-scattering	light-scattering
40	2	$\text{MS} \rightarrow \text{M}^*\text{S}$ (2)	$\text{MS} \rightarrow \text{M}^*\text{S}$ (3)
		k_3	k_3
40	7	$-(k_1 + k_2)$ [MS]	$-(k_2 + k_3)$ [MS]
41	TABLE IV	[ATP] 2×10^{-3} mole/lit.	[ATP] 2×10^{-5} mole/lit.
42	TABLE V	The numerical	The numerical
		value $\frac{A-B}{A} \times 100$	value : $\frac{A-B}{A} \times 100$
44	TABLE VI	<u>Reaction</u>	<u>Reaction</u>
		$\text{M}^* + \text{R} \rightarrow \text{M}^*\text{S}$	$\text{M}^* + \text{S} \rightarrow \text{M}^*\text{S}$
44	FIG. 15 a.	※	(#)
45	FIG. 15 b.	$= 2.2 \times 10^{-6}$.	$= 2.2 \times 10^{-8}$
47	21	in rest than in contraction	in contraction than in rest
50	13	than 1.8 moles	than 1.1 moles
50	19	— 1.4 moles	— 0.9 moles