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Ice core records of biomass burning tracers (levoglucosan and dehydroabietic, vanillic and *p*-hydroxybenzoic acids) and total organic carbon for past 300 years in the Kamchatka Peninsula, Northeast Asia

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Abstract

We successfully detected biomass burning tracers including levoglucosan and vanillic, *p*-hydroxybenzoic and dehydroabietic acids in an ice core (153 m long, ca. 300 years old) taken from Ushkovsky ice cap (altitude, 3903 m), the Kamchatka Peninsula, Northeast Asia. Concentrations of total organic carbon (TOC) were also determined in the ice core. Levoglucosan, which is produced by pyrolysis of cellulose and hemicellulose and thus is a general tracer of biomass burning, showed sporadic peaks in the years of 1705, 1759, 1883, 1915, 1949 and 1972, with the largest peak in 1949. However, its concentrations did not show a systematic increase in the last century although the concentration peaks seemingly corresponded to the higher ambient temperatures in the northern high latitudes. In contrast, dehydroabietic acid, a specific tracer of the pyrolysis of conifer resin, showed a gradual increase from the early 1900s to 1990s with a significant peak in 1970. Contributions of dehydroabietic acid to TOC also showed an increasing trend for the 20th century. Similarly, vanillic and *p*-hydroxybenzoic acids presented higher concentrations in the last half-century with sporadic peaks in 1705, 1759 and 1949. This study showed that general biomass burning tracers such as levoglucosan have been sporadically transported over the glacier of the Kamchatka Peninsula. In contrast, the ice core record of dehydroabietic acid indicated that fires of boreal conifer forest have more frequently and increasingly occurred in Far East and Siberia during the last century and transported to the Northwestern Pacific. The present study demonstrates that organic tracers of biomass burning preserved in ice core could provide historical records of biomass burning and boreal forest fires.

1. INTRODUCTION

In polar regions and high mountain areas, atmospheric chemical species are scavenged by wet (snowflakes) and dry deposition processes, and are preserved in ice sheets and snow fields, providing important information on the atmospheric composition of trace gases and aerosol particles in the past (Legrand and Mayewski, 1997). Ice core studies from polar regions often provide global information with long time scales whereas those from mid-latitudes give more information on regional environmental changes for more recent years (Schwikowski et al., 1999). Currently, organic aerosols are recognized as an important fraction of fine particles in the atmosphere (Kanakidou et al., 2005), which contains various organic components derived from anthropogenic, biogenic and photochemical sources (Simoneit, 1986; Kawamura et al., 1996). However, most ice core analyses in the previous studies are focused on inorganic species (e.g., Legrand and Mayewski, 1997) and organic species from ice core or snow have rarely been reported (Gambaro et al., 2008; Kawamura et al., 1996, 1994; Kawamura et al., 2001; Kehrwald et al., 2012; Legrand and DeAngelis, 1996; Makou et al., 2009; Nishikiori et al., 1997). Because the composition of organic aerosols can provide useful information on the source and source regions (Simoneit, 1986), the ice core studies for organic compounds are potentially important to better understand the past changes in atmospheric composition.

Biomass burning, which has a significant impact on a changing climate (Keywood et al., 2011), is an important source of organic aerosols (Simoneit, 2002). Forest fires frequently occur in many regions of the globe. Domestic burning of bio-fuels and field burning of agricultural wastes are also common in developing countries (Gustafsson et al., 2009). Large-scale forest fires emit enormous amounts of gases and carbonaceous aerosols to the atmosphere via direct vaporization/steam stripping and by thermal alteration of the organic molecules under high temperature conditions (Simoneit, 2002). The biomass burning products are long-range transported via the atmosphere to the Polar Regions and high mountains. The emissions from biomass burning are estimated to contribute up to 38% of tropospheric O₃, 39% of the particulate organic carbon, and >86% of the elemental carbon (black soot) (Fang et

al., 1999; Levine et al., 1995). The majority of the total biomass burning is taking place in the tropical regions (Levine et al., 1995). In particular, the smoke pollution derived from the forest fires of Indonesia is becoming a social problem (Fang et al., 1999). Although the high latitudinal forest fires in the Northern Hemisphere contribute less than 3% of global carbon emission (Levine et al., 1995), the tropospheric chemistry of high latitudes may greatly influence the atmospheric oxidizing capacity and radiation characteristics (Browell et al., 1994; Wong and Li, 2002). The forest fire in Siberian taiga is critical because the regeneration of the damaged forest under which permafrost exists may be difficult, in contrast to the fires occurring in a tropical region. The forest fires in boreal taiga lead to the melting of permafrost under the forests and the release of methane (a strong greenhouse gas) stored under the frozen soil.

Biopolymers such as cellulose, lignin, hemicellulose, suberin, sporopollenin, chitin, etc. are the major constituents of biomass. When they are subjected to burning processes, various organic molecules are emitted to the atmosphere (Simoneit, 2002). Levoglucosan (1,6-anhydro- β -D-glucopyranose) is a major pyrolysis product ($>300\text{ }^{\circ}\text{C}$) of cellulose and hemicellulose that typically comprise 50-70% of dry wood (Andreae and Merlet, 2001; Gao et al., 2003), although levoglucosan was detectable in lower temperature char samples ($150\text{-}350\text{ }^{\circ}\text{C}$) with maximum yield at $250\text{ }^{\circ}\text{C}$ (Kuo et al., 2008). This compound is emitted to the atmosphere by combustion of a woody and grassy part of biomass (Simoneit, 2002; Simoneit and Elias, 2001, 2000; Simoneit et al., 1999). Levoglucosan has been reported in the aerosol samples from urban (Fraser and Lakshmanan, 2000; Kubátová et al., 2002), remote continental (Fu et al., 2008; Gao et al., 2003) and marine sites (Fu et al., 2011; Mochida et al., 2003; Simoneit and Elias, 2000; Simoneit et al., 2004) as well as from the Arctic (Fu et al., 2009). It is also detected in lacustrine sediments from Carajas in Amazon (Elias et al., 2001) and in marine sediments from Puget Sound in Washington (Kuo et al., 2011b). Phenolic compounds such as vanillic and hydroxybenzoic acids are also emitted to the atmosphere during combustion of lignin. Lignin pyrolysis products are detected in aerosols collected from urban and rural areas (Nolte et al., 2001). Dehydroabietic acid is produced during the burning

process of conifer resins (Simoneit, 2002). These biomass-burning products are predicted to be transported to the Polar Regions and stored in ice sheets. However, these biomass-burning products have never been reported in an ice core, except for vanillic acid in Greenland ice core (McConnell et al., 2007) and levoglucosan in Antarctic ice (Gambaro et al., 2008), although other organic species were reported in polar ice (Gambaro et al., 2008; Kawamura et al., 1996, 1994; Kawamura et al., 2001; Legrand and de Angelis, 1996; Makou et al., 2009; Nishikiori et al., 1997).

In this paper, we present, for the first time, a set of specific organic tracers of biomass burning in an ice core (ca. 300 years old for upper 152 m) collected from the glacier filling, Ushkovsky volcanic craters, Kamchatka Peninsula, the Russian Federation. This site is located in the outflow region of air masses from Siberia and Northeast Asia, where biomass burning and forest fires occur frequently. Here, we report ice core records of levoglucosan as well as dehydroabietic, vanillic and *p*-hydroxybenzoic acids, and discuss their historical records and variability. Although low molecular weight dicarboxylic acids including oxalic acid have been reported in biomass burning products (Narukawa et al., 1999) and reported in ice core samples (Kawamura et al., 2001; Legrand and Mayewski, 1997), they are also produced in the atmosphere by photochemical processes (Kawamura and Ikushima, 1993; Kawamura et al., 1996). Therefore, diacids cannot be considered as specific tracers of biomass burning and are not discussed here. Although we did not measure black carbon in the present study, black carbon has been studied and discussed in ice cores (e.g., Kehrwald et al., 2010; McConnell et al., 2007; Thevenon et al., 2009).

2. EXPERIMENTAL METHODS

An ice core (211.7 m long) was taken from the ice cap of Gorshkov crater at Ushkovsky volcano (56°04'N, 160°28'E, 3903 m a.s.l.) of the central part of the Kamchatka Peninsula, Russian Federation (Shiraiwa et al., 2001). The sampling site is shown in Fig.1. The ice core chronology was determined by counting the annual layers of a seasonal oxygen isotopic signal ($\delta^{18}\text{O}$) from the surface to 103.58 m in depth whereas the chronology for

deeper layers was determined by using a two-dimensional thermodynamic coupled model (Salamatin et al., 2000; Shiraiwa et al., 2001). The error in the age determination of the upper 103 m was estimated to be ± 2 years using four identified volcanic ash layers contained in the core (Murav'ev et al., 2007; Shiraiwa and Yamaguchi, 2002). The errors in the deeper section are considered to be larger than those of the upper 103 m. Seventy-five sections were cut off using a band saw. Ice core sections (50 cm long, 1/4 cut) were taken at every one meter for the upper 25 m and at every 4 to 5 m for the layers deeper than 25 m in depth. Approximately 1.0 cm thickness of the outer core surface was mechanically removed using a pre-cleaned ceramic knife in a cold clean room to avoid a potential contamination.

The sample sections were melted in a pre-cleaned Pyrex beaker (2 L). The samples were poisoned with HgCl_2 to prevent potential microbial degradation of organic compounds and stored at 4 °C in a pre-cleaned brown glass bottle (800 ml) with a Teflon-lined screw cap prior to analysis. The presence of sand layers was observed in the deeper sections (>153 m), suggesting that volcanic ash layers of snow might have melted and refrozen in the deeper layers in the past. In this study, we use 60 samples that were collected from 1.32-152.6 m in depth (1997-1693 year) and the data of deeper sections (>153 m) were not used because of the presence of many sand layers although the deeper sections were also analyzed for total organic carbon (TOC) and organic tracers.

After the homogenization (mixing) of the samples in the 800 ml glass bottle, melt water samples (80-250 ml) were transferred in a pear shape flask (300 ml) and concentrated to almost dryness using a rotary evaporator under a vacuum. The total (dissolved + particulate) organic matter in the dried samples was extracted with $\text{CH}_2\text{Cl}_2/\text{CH}_3\text{OH}$ (2:1) mixture using ultrasonic bath. A similar extraction technique has been used to separate organic tracers present in aerosols (e.g., Simoneit et al., 2004; Fu et al., 2008). Although the biomass burning tracers are present as aerosol particles in the atmosphere, they may partition into dissolved and particulate forms when ice samples that contain aerosol particles are melted. Using the dryness and extraction technique, we can recover both the dissolved plus particulate organic matter in the melt water with reasonable recoveries as discussed later. When volcanic ash was present in the sample, the extracts were concentrated and passed through a glass column packed with

quartz wool to remove the ash particles. The ash particles on the quartz wool were further eluted with CH_2Cl_2 and methanol to extract the organic compounds potentially adsorbed on the particles. The eluents were combined with the extracts. The concentrated extracts were transferred to 1.5 ml glass vial and then dried under a pure N_2 stream.

Biomass burning tracers in the extracts were derivatized with 99% *N,O*-bis-(trimethylsilyl) trifluoroacetamide (BSTFA) + 1% trimethylchlorosilane (TMCS) by heating at 70 °C for 2 h in a sealed glass vial (1.5 ml). This fraction was then dissolved in *n*-hexane and analyzed by gas chromatography/mass spectrometry (GC/MS): a Hewlett-Packard Model 5973 MSD coupled to a HP 6890 GC using a capillary column (HP-5MS, 30 m x 0.32 mm I.D. x 0.25 μm film thickness) installed with a split/splitless injector. The GC oven temperature was programmed from 50 °C (2 min) to 120 °C at 30 °C/min, and then to 300 °C at 6 °C/min and maintained at 300 °C for 20 min. Helium was used as a carrier gas. The mass spectrometer was operated on an electron impact mode at 70 eV. Organic compounds in the ice samples were identified by comparing the mass spectra with those of authentic standards and the mass spectral data in the NIST/Wiley library. Levoglucosan, dehydroabietic acid, vanillic acid and *p*-hydroxybenzoic acid were quantified using the authentic standards as external standards (Fu et al., 2008; Simoneit et al., 2004).

Spiked experiments were conducted in triplicate to check the recoveries of target compounds during the analytical procedure. An aliquot of standard solution (10 μl) of authentic standards containing levoglucosan, dehydroabietic acid, vanillic acid and syringic acid (3.4-5.5 ng/ μl each) were spiked to Milli-Q water (200 ml) placed in the pear shape flask (300 ml). The water sample was concentrated and dried by the procedure described above. The concentrates were derivatized with BSTFA and injected to GC-FID (Agilent 6890 GC) and GC/MS. The triplicate analyses of the spiked samples show that the recoveries of levoglucosan, dehydroabietic acid, vanillic acid and syringic acid were 73 \pm 10% (ave. \pm s.d.), 67 \pm 40%, 78 \pm 15% and 78 \pm 18%, respectively. Laboratory blanks was measured using Milli-Q water (200 ml). The results showed no detectable peaks of these compounds. Detection limits of these species were 0.003-0.005 ng/g-ice. Procedural blanks using Milli-Q water showed no detectable amounts of these compounds.

Total organic carbon (TOC) was measured by high temperature catalytic oxidation (HTCO) using a Shimadzu TOC-5000 (Sempéré and Kawamura, 1994). Aliquots of the melt water (ca. 10 ml) were taken from the 800 ml glass bottle into 20 ml glass vial and then acidified with 1.2 M HCl solution (80 μ l). The samples were bubbled with ultra high purity air to remove inorganic carbon and then injected to the oxidation column maintained at 680 °C. The reported TOC data are corrected for the procedural and system blanks.

3. RESULTS AND DISCUSSION

3.1. Total organic carbon

Total organic carbon (TOC) concentrations in the ice core ranged from 410 to 2600 ng/g-ice with an average of 1200 ng/g-ice (Table 1). The averaged TOC value in the Ushkovsky ice core (elevation of site; 3903 m) is slightly lower than that (700-5700 μ g/g-ice: average 1700 ng/g-ice) (Kawamura et al., 1992) reported in ice core samples collected at Site-J in Greenland (66°51.9'N, 46°15.9'W, elevation; 2030 m) and twice lower than that (290-4900 ng/g-ice: average 2300 ng/g-ice) obtained for the H-15 ice core (69°04'S, 40°46'E; elevation; 1057 m) near the coast of Antarctica (Kasukabe, 1995).

Fig. 2 gives the historical record of TOC in the Ushkovsky ice core. The TOC concentrations gradually decrease from 1700s to 1900 with peaks in 1715, 1768, 1818, 1847 and 1883. Then they fluctuate with peaks in 1928, 1949, 1966, 1975, and 1986. Although the concentrations of TOC and organic species are affected by dry deposition of atmospheric aerosols and the amounts of snowfall, these trends and fluctuations may be associated with natural variability of the emission of organic matter to the atmosphere and long-range transport to the sampling site. Interestingly, we found an abrupt increase in TOC concentrations from the late 1970s toward the early 1990s. This abrupt increase may be caused by an increased anthropogenic activity in East Asia, followed by long-range atmospheric transport to Kamchatka. The development of east Russia (e.g., Siberia and Far East) started at the beginning of the 20th century. For instance, the construction of the Siberia railroad was completed in 1905. The economic growth in east Russia and northeast China has accelerated in recent years. This suggests that an enhanced atmospheric emission of organic matter of

anthropogenic origins may have begun when the industrial development started in the Siberia since the 1910s. However, such an increase in TOC has not been observed in the Site-J ice core from southern Greenland (Kawamura et al., 1992). This may be reasonable because the Ushkovsky site is much closer to the source regions in East Asia and Siberia whereas Greenland is much further away from the sources.

3.2. Biomass burning tracers

3.2.1. Levoglucosan (pyrolysis product of cellulose and hemicellulose)

Levoglucosan is a pyrolysis product of cellulose and hemicellulose (Simoneit et al., 1999), and it is also produced by combustion of lignite or brown coal (Fabbri et al., 2008; Fabbri et al., 2009; Oros and Simoneit, 2000). However, it is not produced either by hydrolysis or microbial degradation of sugar (Locker, 1988; Simoneit et al., 1999). Levoglucosan has been used as a general tracer of biomass burning in atmospheric aerosols (e.g., Fu et al., 2008; Simoneit, 2002; Simoneit et al., 1999). Although levoglucosan may not be as stable as previously thought in the atmosphere (Hoffmann et al., 2010), its concentrations are not seriously influenced during transport for several days in winter (Fraser and Lakshmanan, 2000; Hoffmann et al., 2010; Mochida et al., 2010). In the Ushkovsky ice core, levoglucosan was detected as the most abundant organic species in the total ion current (TIC) chromatogram (Fig. 3a). Fig. 3b presents the mass spectrum of levoglucosan as TMS derivative, which shows a weak molecular ion at $m/z=378$ and three dominant fragment ions at $m/z=204$, 217 and 333. Concentrations of levoglucosan ranged from less than the detection limit (0.005 ng/g-ice) to 19 ng/g-ice with an average of 0.75 ng/g-ice (Table 1).

Galactosan and mannosan are also emitted to the atmosphere as a smoke by the combustion of hemicellulose (Nolte et al., 2001; Simoneit, 2002). These anhydrosugars were detected as less abundant species than levoglucosan in the Ushkovsky ice core. Although the combustion of lignite that contains fossilized cellulose emits levoglucosan significantly, its isomers (galactosan and mannosan) are not emitted or emitted at relatively low concentrations (Fabbri et al., 2008; Fabbri et al., 2009; Kuo et al., 2011a). Thus, levoglucosan to its isomer

ratios may provide a clue to differentiate the contributions of biomass and lignite burning to the ice core as discussed below.

Ice core record of levoglucosan is shown in Fig. 4a. This biomass-burning tracer showed sporadic peaks in around 1705, 1759, 1883, 1915, 1949, 1972, 1981 and 1989. These peaks may be related to enhanced forest fires on a large scale in Siberia and Far East and/or enhanced atmospheric transport of biomass and/or lignite burning plumes. These events may have been caused by forest fires in Siberia because the sampling site is located downwind of Siberia (see Fig. 1), in which boreal forest fires are frequently occurring in spring, summer and autumn (Ivanova et al., 2010). In particular, the enhanced concentration of levoglucosan in the ice core around 1949 suggest enormous fire must have occurred at around that time, although there is no reported peak of forest fires at around 1949 in the southern Siberia ice core (Eichler et al., 2011). Eichler et al. (2011) reported ice core records of K^+ , NO_3^- and charcoal in southern Siberia since AD 1250 and found a high forest fire activity between AD 1600 and 1680. We could not detect any significant increase in levoglucosan concentration in the period of 1600-1680 in the Ushkovsky ice core. Further, this may suggest that the source region of the Ushkovsky ice core peak of levoglucosan is not southern Siberia, but other regions in Siberia. Interestingly, a large forest fire in Siberia was reported in 1915, which caused a significant cooling due to larger quantities of smoke particles emitted from the fire (Veltishchev et al., 1988), and the peak of levoglucosan in 1915 in the ice core (Figure 4a) matches with this event in Siberia at this time.

On the other hand, brown coal burning and subsequent transport of the plumes may be responsible to the high levels of levoglucosan. For the ice sections where high levels of levoglucosan were detected, we calculated levoglucosan/mannosan (L/M) ratios and found that L/M ratios for the annotated peaks in Figure 4a range from 3.3 to 5.0, which are much lower than those (30-90) reported for lignite burning at 200 °C (Kuo et al., 2011a) and are even lower than those (15-40, av. 21) reported for springtime Asian aerosols collected at Gosan site, South Korea (Simoneit et al., 2004). The L/M ratios (ave. 4.0) in the Ushkovsky ice core are lower than those (7.8 for winter and 6.4 for summer) for the urban aerosols from

Belgium (Pashynska et al., 2002) and those (4.7 for winter and 5.5 for summer) from Ireland (Kourtchev et al., 2011). Interestingly, the L/M ratios obtained for the ice core are typical of gymnosperm (conifer) burning (Engling et al., 2006; Fabbri et al., 2009). These comparisons suggest that lignite burning does not influence the vertical profile of levoglucosan, rather levoglucosan in the Ushkovsky ice core is most likely derived from biomass burning of conifer trees that are a major type of vegetation in Siberia.

We could not detect any clear trend of the recent increase in levoglucosan concentrations (Fig. 4a). This is in contrast to the profile of TOC, which showed a significant and gradual increase from 1970s to 1990s. However, concentrations of levoglucosan seem to increase slightly from the mid 1970s to the mid 1990s as seen in Fig. 4a. A similar weak increase was also seen for mannosan, an isomer of levoglucosan. This may suggest that contributions of biomass burning to the atmospheric aerosols have increased recently due to the frequent occurrence of boreal forest fires in Siberia (Soja et al., 2007; Valendik, 1996) and due to biomass burning of agriculture wastes in Northeast China (e.g., Fu et al., 2008; Lee et al., 2005).

In the Russian Far East area, the frequency of forest fires increased since the development by forest industry was initiated in 1930s (Sheingauz, 1998). However, such an increasing trend was not detected in the historical record of levoglucosan in the Ushkovsky ice core (Fig. 4a). This apparent mismatch may suggest a sporadically enhanced atmospheric transport of levoglucosan rather than the general increase of forest fires in the Far East. In addition, the profile of levoglucosan shows that remarkable enhancement occurred in 1705, 1759, 1883, 1915, 1949 and 1972 (Fig. 4a). Interestingly, terrestrial plant-derived biomarkers such as n-alkanes, n-alkanols and n-fatty acids showed a similar trend in the Ushkovsky ice core (Izawa, 2004). These compounds can also be emitted to the air by vaporization of leaf waxes during biomass burning processes (Nolte et al., 2001; Schauer et al., 2001). Once they are emitted to the atmosphere, they are subjected to long-range atmospheric transport by westerly winds possibly to the western North Pacific rim where Kamchatka Peninsula is located.

3.2.2. Dehydroabietic acid (pyrolysis product of conifer resin)

In the Ushkovsky ice core, we also detected dehydroabietic acid, which is another tracer of biomass burning. This acid is produced by pyrolytic dehydration of abietic acid, which is a principal component of conifer resin (Simoneit et al., 1993). Hence, dehydroabietic acid can be used as a more specific tracer of the burning of conifer trees. The concentrations of dehydroabietic acid ranged from below detection limit (0.003 ng/g-ice) to 0.47 ng/g-ice with average of 0.054 ng/g-ice (Table 1). Its average concentration is ca. 14 times lower than that (0.75 ng/g-ice) of levoglucosan, which is a general tracer of biomass burning.

Fig. 4b presents historical record of dehydroabietic acid in Ushkovsky ice core. We found a gradual increase in the concentration of dehydroabietic acid from the mid 1900s to the 1990s (Fig. 4b). It is of interest to note that levoglucosan did not show such an increase in the last century, although a weak increase was obtained for the last 30 years (Fig. 4a). Such an increase of dehydroabietic acid demonstrates that the burning of pine trees in Siberia is more frequently occurring and/or the burning plumes are more frequently transported to the sampling site in those years. More frequent forest fires in the boreal forest may have been caused as a result of recent global warming and drier conditions associated with the global change (Soja et al., 2007). The dominant vegetations of boreal forest in Siberia are characterized by larch, pine, spruce and fir (Ivanova et al., 2010). Dehydroabietic acid did not show a peak in the years when levoglucosan showed sporadic peaks although a simultaneous peak was observed in 1949 for two biomass burning tracers (Fig. 4a). For example, dehydroabietic acid showed a very strong peak in 1970/1971 when levoglucosan did not show a peak (Fig. 4a). This strong peak should suggest that biomass burning plumes from boreal conifer forests probably in Siberia were transported over the Kamchatka Peninsula because air masses can be often transported from Siberia to the Northwestern Pacific rim as discussed in section 3.4.

The apparent discrepancy between two biomass-burning tracers may suggest that levoglucosan probably did not originate from the boreal forest fires in Siberia where pine trees are common, whereas dehydroabietic acid was most likely derived from the boreal forest fires in Siberia. One possible explanation is that levoglucosan could have been transported from

other regions further south including mid-latitudes in East Asia where broad-leaf trees are common. However, this interpretation seems to conflict with the lower L/M ratios (ave. 4.0) that suggests gymnosperm sources for anhydrosugars as discussed above. If the burning of deciduous forests of the south were a significant source to the ice core, the L/M ratios should have been much higher. Alternatively, dehydroabietic acid may have been in part decomposed during long-range atmospheric transport because aromatic structure of the compound is relatively unstable compared to levoglucosan due to the higher sensitivity to photodegradation (Shakya et al., 2011; Simoneit, 2002; Simoneit and Elias, 2001). If dehydroabietic acid were significantly decomposed in the atmosphere during long-distance transport prior to arriving and scavenging over the ice sheet, the significance of conifer burning in Siberia should have been more intensive in these days (see Fig. 4b).

3.2.3. Vanillic and *p*-hydroxybenzoic acids (pyrolysis products of lignin)

Vanillic acid and *p*-hydroxybenzoic acid are produced by incomplete combustion of conifers and grasses, respectively (Simoneit, 2002; Simoneit et al., 1993). Vanillic acid and *p*-hydroxybenzoic acids, which have been used as biomass-burning tracers of lignin (Simoneit, 2002), were detected in the Ushkovsky ice core. Mass spectra of TMS derivative of vanillic acid are characterized by specific fragment ions at $m/z=223$, 267 and 297 and a molecular ion of $m/z=312$ (see Fig. 3c). Concentrations of vanillic acid ranged from below detection limit (0.005 ng/g-ice) to 0.13 ng/g-ice with an average of 0.015 ng/g-ice (Table 1). Its average concentration is ca. 56 times lower than that of levoglucosan and 4 times lower than that of dehydroabietic acid. Mass spectra of *p*-hydroxybenzoic acid gave specific mass fragments at $m/z=193$, 223 and 267 and molecular ion at $m/z=282$ (Fig. 3d). The concentrations of *p*-hydroxybenzoic acid ranged from below detection limit (0.005 ng/g-ice) to 1.74 ng/g-ice with an average of 0.24 ng/g-ice (Table 1). The average concentration of *p*-hydroxybenzoic acid is about 4 times lower than levoglucosan, but about 4 times higher than that of dehydroabietic acid and 16 times higher than that of vanillic acid.

Vertical profiles of vanillic acid showed sporadic peaks in 1705, 1759, 1949, 1971 and 1992 (Fig. 4c) whereas *p*-hydroxybenzoic acid showed peaks at 1705, 1759, 1883, 1915,

1949, 1972, 1984 and 1992 (Fig. 4d). The major peaks of these tracers are consistent with those of levoglucosan (see Fig. 4a), suggesting that these three tracers are emitted from similar combustion sources and source regions. The covariance of these biomass-burning tracers demonstrates that the biomass burning plumes of lignin and cellulose are transported over long distances to the glaciers of the Kamchatka Peninsula, although they may potentially be subjected to OH radical attack during this transport (Hennigan et al., 2010; Hoffmann et al., 2010; Kessler et al., 2010). The similarity in the ice core profiles among the biomass burning tracers (levoglucosan, vanillic acid and *p*-hydroxybenzoic acid) is reasonable because these components are both present in the woody parts of trees. It is of interest to note that these peaks obtained for levoglucosan and two lignin tracers (vanillic and *p*-hydroxybenzoic acids) were not observed in the profile of dehydroabietic acid, except for the peak of 1949 (Fig. 4b).

The different behavior of biomass burning tracer of conifer (dehydroabietic acid) than those of lignin and cellulose suggests that biomass burning plumes with different sources and source regions were delivered to the sampling site. Because dehydroabietic acid is produced by the pyrolysis of resins that are present on the bark surfaces and needle leaves in addition to woody tissues of conifers, its emission by burning process may be different from those of lignin and cellulose, which are located in the woody part. The lack of covariance between resin and lignin/cellulose burning tracers may suggest that the tree barks and needle leaves that contain abundant resin components in the boreal conifer are easily burned to result in the emission of dehydroabietic acid to the atmosphere but the burning of woody parts may be restricted leaving the unburned conifer trees stand. This kind of burning situation of the conifer trees has been often reported in Siberia (Eichler et al., 2011).

3.3. Relative concentrations of levoglucosan and other key biomass burning tracers in ice core and ambient aerosols

Simoneit and his colleagues reported relative concentrations of levoglucosan, dehydroabietic acid and vanillic acid in ambient aerosols from various locations (Simoneit et al., 1999). For winter (mostly December) aerosols, levoglucosan/dehydroabietic acid/vanillic acid ratios were reported to be 100/17/2 for the 1985 Corvallis aerosols from Oregon, 100/14/2

for the 1989 Pasadena aerosols from southern California, and 100/6/1.2 for the 1995 Fresno aerosols from northern California. These comparisons indicate that dehydroabietic acid, a specific tracer of burning of conifer resin, is more enriched in the northern aerosols from Oregon. In Oregon, conifer trees may be more frequently used in wood stoves during winter for house heating compared to broad-leaf trees. Interestingly, the average ratios (100/20/6) calculated for the Ushkovsky ice core (1953-1996) were similar with those of the ambient aerosols from Oregon, although the relative abundances of dehydroabietic acid and vanillic acid were higher in the ice core. The higher relative abundances of these acids in the ice may be involved with their source regions in Siberia where conifer trees are dominant species of vegetation.

3.4. Source regions of biomass burning tracers in the Ushkovsky ice core

To better understand the source regions of biomass burning tracers detected in the Ushkovsky ice core, we conducted 7 days backward air mass trajectory analyses at the Ushkovsky ice core-sampling site for four seasons in 2001 using the HYSPLIT model of the National Oceanic and Atmospheric Administration (<http://www.ready.noaa.gov/ready/open/hysplit4.html>) (Draxler and Hess, 1997; Draxler and Hess, 1998).

As shown in Fig. 5, the trajectory analyses for different seasons showed that the air mass source regions are mainly Siberia, Far East, North China and Eastern Europe as well as the western Northern North Pacific. These regions, in particular Siberia, are known as areas where forest fires occur frequently (Soja et al., 2007). The aerosols containing the biomass burning products have been transported over the sampling site of Kamchatka Peninsula through the atmosphere and deposited over the glacier by wet and dry processes. However, the trajectory analyses showed that there is no significant air mass contribution from Alaska and North America to the Ushkovsky site (Fig. 5). The backward air mass trajectory analyses suggest that the main source regions of biomass burning tracers detected in the Ushkovsky ice core are located in between Europe and Russia within the latitude band of 40-70 °N (see Fig.

5). The biomass burning tracers emitted from these regions may be transported to the Kamchatka Peninsula by the westerly winds.

Fig. 6 shows an estimated annual emission of total particulate matter (TPM) from fires in the Northern Hemisphere for the year of 2001. The original data were taken from Global Fire Emission Database version 2.1 (GFEDv2.1) (http://daac.ornl.gov/VEGETATION/guides/global_fire_emissions_v2.1.html) (Randerson et al., 2007; van der Werf et al., 2006). Fire emissions of TPM can be seen to be high in western to eastern Russia with the latitude band of 50-60°N and in the northeast Siberia with the latitude band of 60-70°N. A similar emission map was obtained for 2002 (results are not shown as a figure). These areas are the source regions of air masses transported to the Ushkovsky ice cap (see Fig. 5). Thus, the biomass burning tracers detected in the ice core samples could have been transported in the atmosphere from the Russian regions. It is reasonable because the westerly winds are dominant in Northeast Asia. Although Southeast Asia and central Africa are important source regions of fire-emitted TPM (Fig. 6), these regions may not have contributed to the biomass burning tracers detected in the Ushkovsky ice core because air mass trajectories do not support an atmospheric transport from such regions (Fig. 5).

However, the above interpretations based on current fire emission inventories and air mass trajectories may not be applicable because we do not understand the past air circulation patterns that may have transported the biomass burning plumes from the Southeast Asian regions to the Kamchatka Peninsula in the Northeastern Pacific in the last few centuries. In fact, we found that the peaks of levoglucosan (Fig. 4a) correspond to higher ambient temperatures reconstructed in the northern North Hemisphere based on tree rings (Jacoby and Darrigo, 1989). This correlation suggests a possibility that the strong peaks of levoglucosan detected in the Ushkovsky ice core may have been involved with enhanced transport of biomass burning plumes originated from the low and mid latitudes including Southeast Asia when the Arctic air temperature increased. This hypothesis on the meridional transport of biomass burning plumes from tropics to the high latitudes associated with the Arctic warming needs to be verified by future studies.

4. CONCLUSIONS

The present study reported, for the first time, the historical records of biomass burning–derived organic tracers such as levoglucosan and dehydroabietic, vanillic and *p*-hydroxybenzoic acids in the Ushkovsky ice core, Kamchatka Peninsula, Russia. Although levoglucosan, the most abundant biomass-burning tracer in the ice core, showed sporadic peaks, there is no clear increase in the concentrations toward the present. Because levoglucosan is a general tracer of pyrolysis of cellulose, its sporadic peaks suggest that large-scale forest fires may have occurred in Siberia, Far East, Northeast Asia and Southeast Asia in the past. In contrast, concentrations of dehydroabietic acid, which is a specific tracer of pyrolysis of conifer resin, showed a gradual increase from the early 1900s to the present. This historical increase suggests that burning of boreal forests, characterized by conifer trees, has been enhanced significantly during the last one hundred years. This ice core record is consistent with the records of the forest fires and burned forest areas in Siberia (Soja et al., 2007). The present study demonstrated that organic tracers of biomass burning preserved in ice core are useful to reconstruct the historical records of biomass burning and forest fires on a local and regional scale.

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Table 1. Concentrations of total organic carbon (TOC) and biomass burning tracers in the Ushkovsky ice core with ice chronology.

| Core depth (m) | Year A.D. | Concentrations (ng/g-ice) | | | | |
|----------------|-----------|---------------------------|--------------|---------------------|---------------|-------------------------------|
| | | TOC | Levoglucosan | Dehydroabietic acid | Vanillic acid | <i>p</i> -Hydroxybenzoic acid |
| 1.1-1.6 | 1997 | 1874 | 0.23 | 0.111 | BDL | BDL |
| 2.2-2.6 | 1996 | 1967 | 0.76 | 0.238 | 0.077 | 0.324 |
| 2.6-3.3 | 1995 | 1695 | 0.19 | 0.096 | BDL | 0.008 |
| 3.3-4.4 | 1994 | 1588 | 0.44 | 0.248 | 0.063 | 0.265 |
| 4.8-5.3 | 1993 | 1713 | 0.41 | BDL | BDL | 0.062 |
| 5.3-5.8 | 1992 | 1980 | 0.55 | 0.248 | 0.105 | 0.381 |
| 5.8-7.1 | 1991 | 1192 | 0.15 | 0.111 | 0.020 | 0.034 |
| 7.1-7.9 | 1990 | 1927 | 0.54 | 0.056 | 0.019 | 0.612 |
| 8.4-8.9 | 1989 | 1475 | 1.12 | 0.037 | 0.011 | 0.101 |
| 8.9-9.8 | 1988 | 1440 | 0.22 | 0.050 | 0.040 | 0.184 |
| 9.8-11.3 | 1987 | 1331 | 0.59 | 0.038 | 0.026 | 0.473 |
| 11.3-12.0 | 1986 | 2123 | 0.20 | 0.108 | 0.013 | 0.220 |
| 12.3-12.8 | 1985 | 1060 | 0.12 | 0.069 | 0.007 | 0.059 |
| 13.0-13.6 | 1984 | 1129 | 0.62 | 0.107 | 0.044 | 0.763 |
| 13.6-14.3 | 1983 | 1492 | 0.51 | 0.067 | 0.015 | 0.178 |
| 14.3-15.4 | 1982 | 757 | 0.38 | 0.093 | 0.022 | 0.205 |
| 15.4-16.5 | 1981 | 961 | 1.52 | 0.045 | 0.025 | 0.599 |
| 16.8-17.3 | 1980 | 1095 | 0.36 | 0.066 | BDL | 0.171 |
| 17.3-17.9 | 1979 | 716 | 0.05 | 0.059 | BDL | 0.046 |
| 17.9-19.1 | 1978 | 730 | 0.31 | 0.092 | 0.030 | 0.239 |
| 19.6-20.1 | 1977 | 1102 | 0.03 | 0.051 | 0.021 | 0.050 |
| 20.7-21.5 | 1976 | 1236 | 0.11 | 0.063 | BDL | 0.048 |
| 21.5-22.6 | 1975 | 1836 | 0.05 | 0.057 | BDL | 0.185 |
| 22.6-23.5 | 1974 | 1213 | 0.05 | 0.034 | 0.009 | 0.117 |
| 23.5-24.1 | 1973 | 750 | 0.11 | 0.049 | 0.024 | 0.117 |
| 24.1-24.6 | 1972 | 1052 | 4.82 | 0.044 | 0.029 | 0.578 |
| 25.0-25.5 | 1971 | 1292 | 0.73 | 0.291 | 0.045 | 0.279 |
| 25.5-25.9 | 1970 | 1209 | 0.36 | 0.475 | BDL | BDL |
| 28.3-28.8 | 1966 | 1404 | 0.71 | 0.027 | 0.013 | 0.170 |
| 32.0-32.5 | 1961 | 843 | 0.06 | 0.016 | BDL | 0.055 |
| 36.3-36.8 | 1954 | 979 | 0.14 | BDL | BDL | 0.224 |
| 39.6-40.1 | 1949 | 1783 | 18.62 | 0.035 | 0.071 | 1.710 |
| 44.0-44.5 | 1942 | 1071 | 0.28 | 0.017 | BDL | 0.069 |
| 48.4-48.9 | 1933 | 999 | 0.44 | 0.014 | BDL | 0.223 |
| 51.9-52.4 | 1928 | 1351 | 0.61 | 0.019 | BDL | 0.127 |
| 56.1-56.6 | 1922 | 1124 | 0.16 | 0.008 | BDL | 0.021 |
| 60.1-60.5 | 1915 | 1068 | 1.81 | 0.007 | BDL | 0.325 |
| 64.9-64.4 | 1908 | 769 | 0.05 | BDL | BDL | 0.037 |
| 67.9-68.4 | 1897 | 639 | 0.05 | BDL | BDL | 0.050 |
| 72.1-72.6 | 1889 | 750 | 0.25 | 0.006 | BDL | 0.139 |
| 76.4-76.9 | 1883 | 876 | 2.20 | 0.007 | BDL | 0.032 |
| 81.0-81.5 | 1871 | 673 | 0.04 | 0.007 | BDL | ND |
| 84.9-85.4 | 1862 | 883 | 0.02 | BDL | BDL | 0.026 |
| 88.8-89.3 | 1854 | 722 | BDL | 0.006 | BDL | BDL |
| 92.4-92.9 | 1847 | 1178 | 0.12 | BDL | BDL | BDL |
| 95.9-96.4 | 1840 | 692 | 0.02 | 0.004 | BDL | BDL |
| 99.9-100.4 | 1832 | 857 | 0.05 | BDL | BDL | BDL |
| 104.2-104.7 | 1818 | 960 | 0.04 | BDL | BDL | BDL |
| 108.1-108.6 | 1809 | 804 | 0.01 | BDL | BDL | BDL |
| 111.6-112.1 | 1800 | 753 | 0.08 | BDL | BDL | 0.130 |
| 115.7-116.2 | 1790 | 761 | 0.78 | BDL | BDL | 0.044 |
| 120.8-121.3 | 1778 | 915 | 0.06 | BDL | BDL | 0.210 |
| 124.7-125.1 | 1768 | 1364 | 0.40 | 0.012 | BDL | 0.270 |
| 128.1-128.6 | 1759 | 1285 | 4.81 | 0.010 | 0.127 | 1.741 |
| 132.3-132.8 | 1748 | 959 | 0.59 | 0.004 | BDL | 0.042 |
| 136.9-137.4 | 1736 | 1127 | 0.18 | BDL | BDL | 0.295 |
| 140.0-140.6 | 1727 | 1109 | 0.43 | 0.005 | BDL | 0.307 |
| 144.6-145.1 | 1715 | 1729 | 0.08 | 0.025 | BDL | 0.113 |
| 148.0-148.6 | 1705 | 1302 | 1.59 | 0.026 | 0.064 | 1.439 |
| 152.1-152.6 | 1693 | 1167 | BDL | BDL | BDL | BDL |
| 155.8-156.3 | | 1163 | 0.10 | BDL | BDL | 0.017 |
| 159.7-160.2 | | 1281 | 0.44 | 0.010 | BDL | 0.013 |
| 163.5-164.0 | | 2604 | 0.34 | 0.029 | BDL | 0.087 |
| 168.9-169.5 | | 1191 | 0.11 | BDL | BDL | 0.020 |
| 171.3-171.8 | | 2241 | 0.02 | BDL | BDL | 0.095 |
| 176.8-177.3 | | 758 | 1.11 | 0.019 | BDL | 0.074 |
| 180.2-180.7 | | 1075 | 2.34 | 0.026 | 0.030 | 0.430 |
| 184.7-185.3 | | 406 | 0.12 | 0.006 | BDL | BDL |
| 188.3-188.7 | | 1714 | 0.06 | 0.015 | BDL | 0.027 |
| 191.0-191.5 | | 871 | 0.05 | 0.006 | BDL | 0.232 |

| | | | | | |
|-------------|------|------|-------|-------|-------|
| 196.4-196.9 | 1004 | 0.29 | 0.012 | 0.009 | 0.038 |
| 199.6-200.0 | 1507 | 0.95 | 0.009 | BDL | 0.016 |
| 203.2-203.7 | 854 | 0.14 | 0.009 | BDL | 0.034 |
| 208.2-208.7 | 1048 | 0.25 | 0.011 | BDL | 0.037 |
| 209.4-209.9 | 1170 | 0.10 | 0.025 | BDL | 0.029 |

BDL: Below detection limit, Detection limits: 0.003 to 0.005 ng/g-ice depending on the species (for more detail, see the text).

Figure captions

Fig. 1. Map of the Kamchatka Peninsula with the sampling site of the Ushkovsky ice core at the ice cap of Ushkovsky volcano.

Fig. 2. Historical change in the concentrations of total organic carbon (TOC) in the Ushkovsky ice core taken from the Kamchatka Peninsula.

Fig. 3. (a) Typical GC/MS trace (total ion current: TIC) of trimethylsilylated total extracts isolated from the Ushkovsky ice core sample (FA: fatty acids, A: alkanes, ●: unknown monosaccharide), (b) mass spectra of levoglucosan-TMS, (c) mass spectra of vanillic acid-TMS and (d) mass spectra of *p*-hydroxybenzoic acid-TMS.

Fig. 4. Historical variations of (a) levoglucosan, (b) dehydroabietic acid, (c) vanillic acid, and (d) *p*-hydroxybenzoic acid in the Ushkovsky ice core.

Fig. 5. Seven days backward trajectory analyses at Ushkovsky for 4 seasons (a) MAM; March, April and May, (b) JJA; June, July and August, (c) SON; September, October and November, and (d) DJF; December, January and February, throughout 2001. The trajectory calculations were performed everyday from the Ushkovsky site (56.07°N and 160.47°E) at 4500 m above sea level. Each color trajectory line indicates tracing time from sampling point, red: 0-3 days, yellow: 3-5 days and blue: 5-7 days.

Fig. 6. Estimated annual emission of total particulate matter (TPM) from fires in the Northern Hemisphere for 2001. Original data are in Global Fire Emission Database version 2.1 (GFEDv2.1) (http://daac.ornl.gov/VEGETATION/guides/global_fire_emissions_v2.1.html). Monthly TPM data are summed for 2001.

Fig. 1

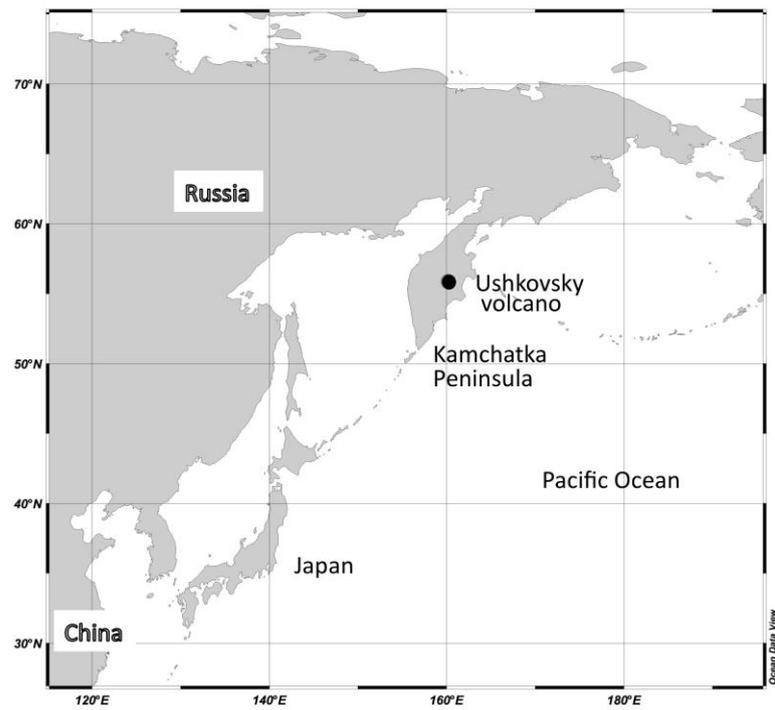


Fig. 2

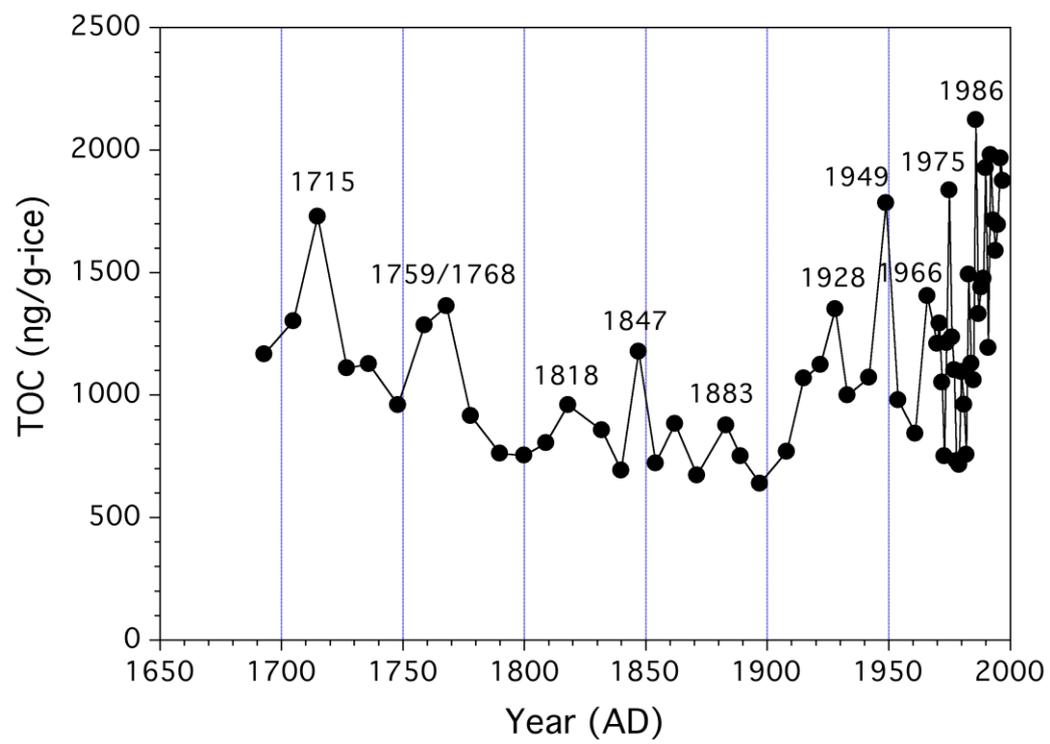


Fig. 3

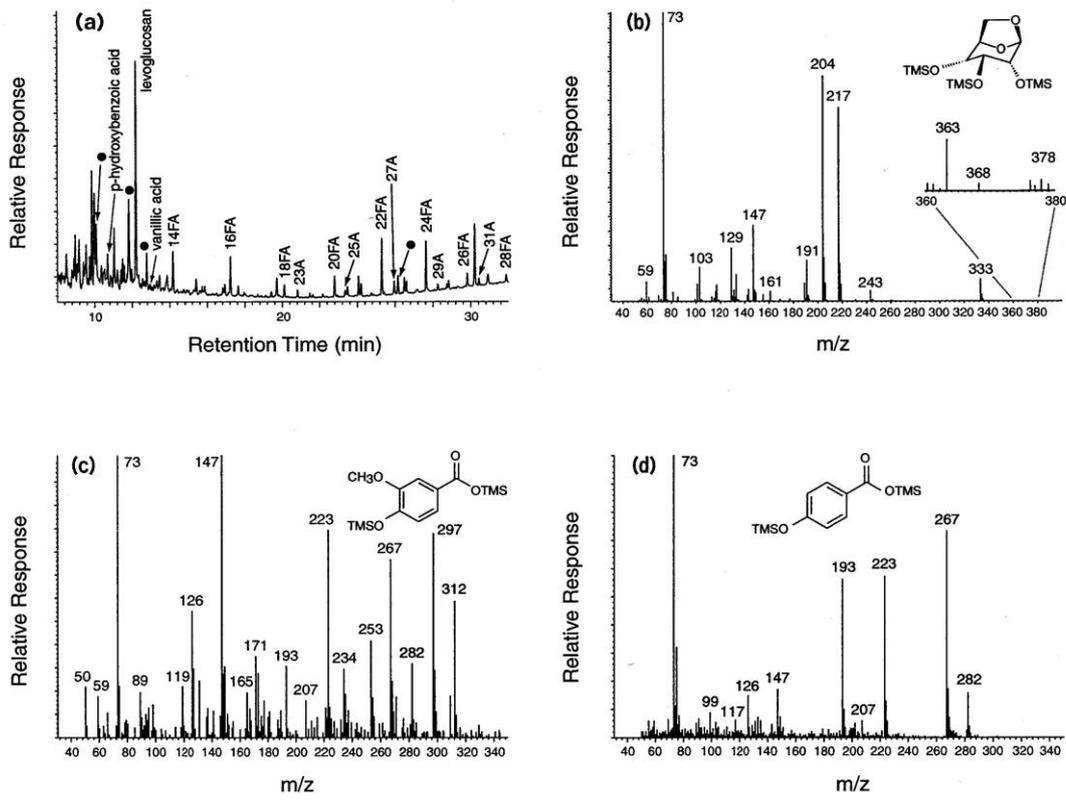


Fig. 4a

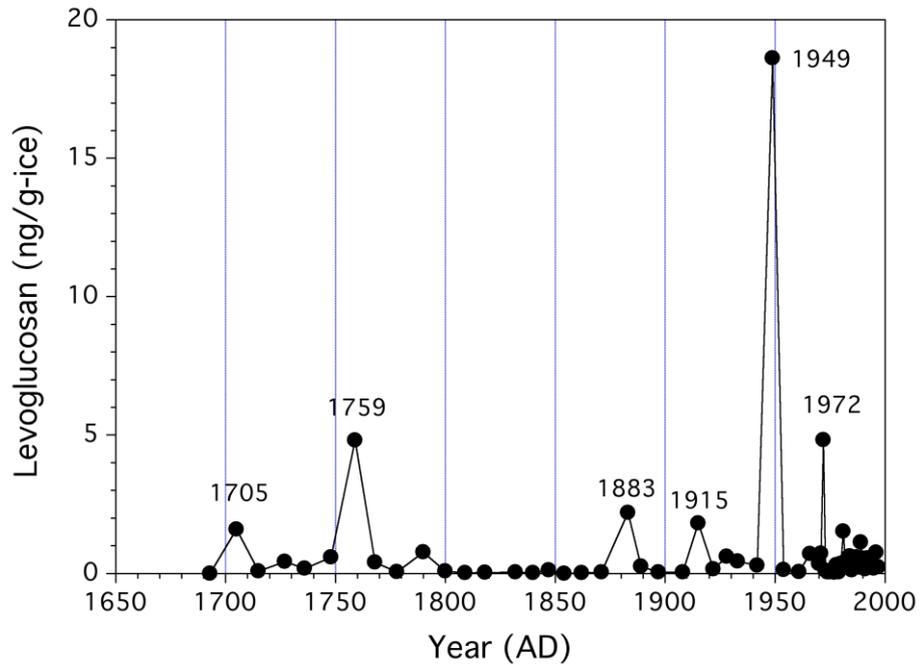


Fig. 4b

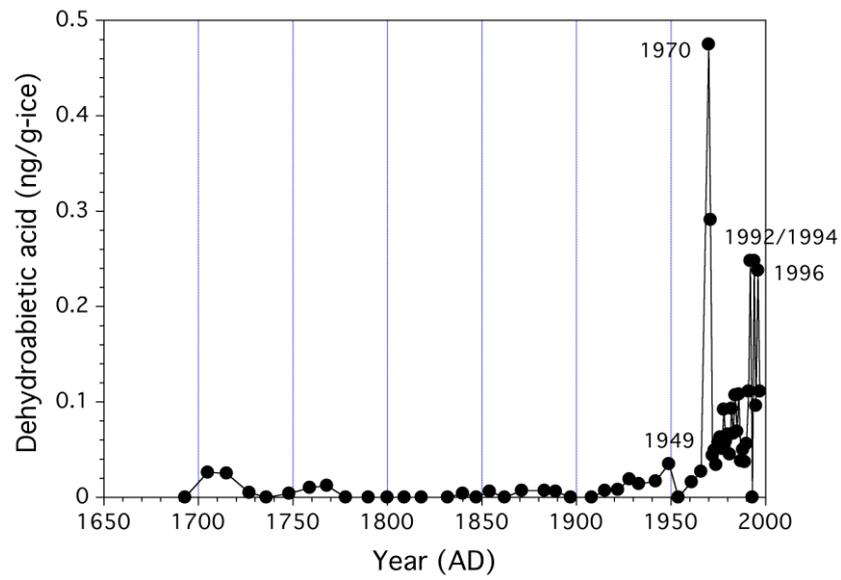


Fig. 4c

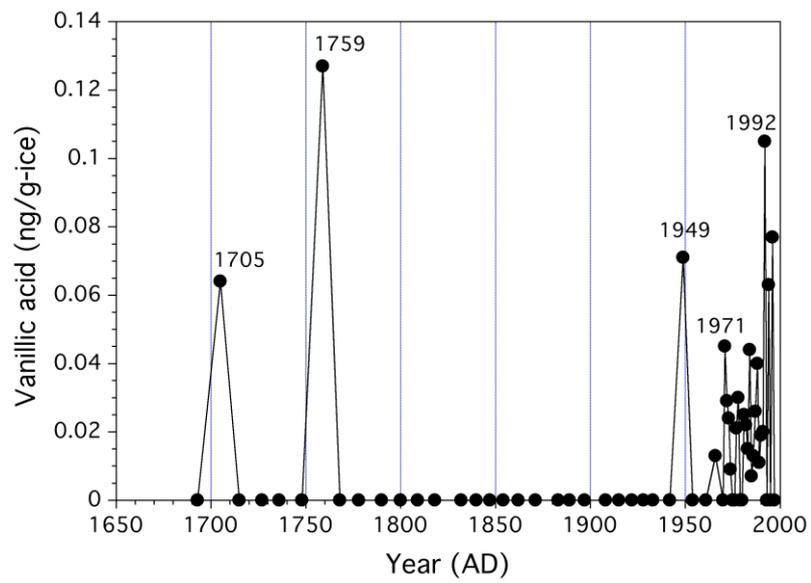


Fig. 4d

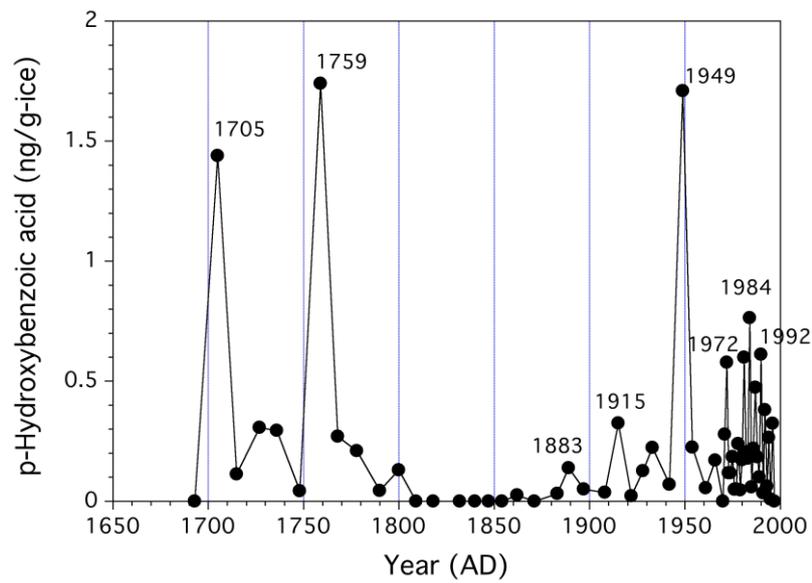


Fig. 5

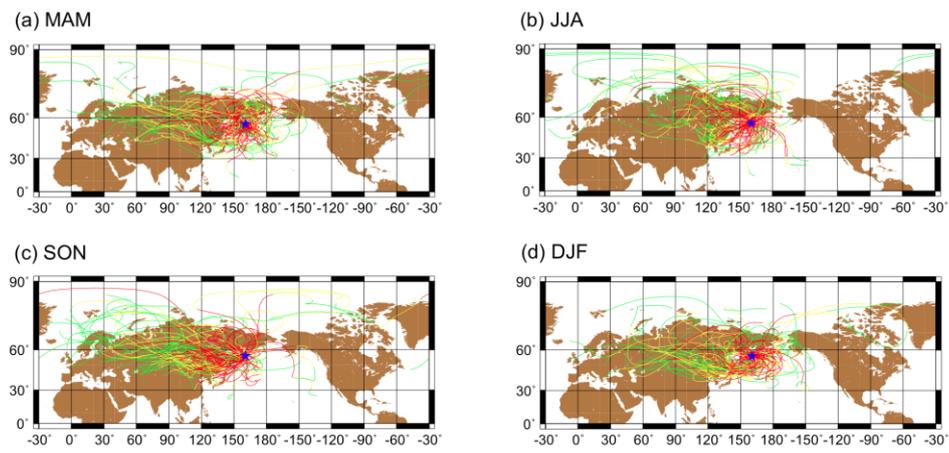


Fig. 6

