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Author(s)	Gao, Lei; Narita, Koji; Ozaki, Taro et al.
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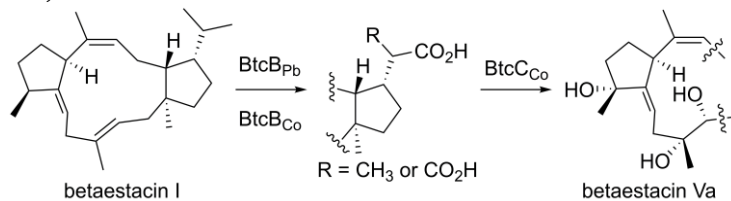
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Identification of novel sesterterpenes by genome mining of phytopathogenic fungi *Phoma* and *Colletotrichum* sp.

Lei Gao^{a,b}, Koji Narita^a, Taro Ozaki^{a,*}, Naoyoshi Kumakura^c, Pamela Gan^c, Atsushi Minami^a, Chengwei Liu^a, Xiaoguang Lei^b, Ken Shirasu^c, and Hideaki Oikawa^{a,*}

^aDivision of Chemistry, Graduate School of Science, Hokkaido University, Sapporo 060-0810, Japan

^bCollege of Chemistry and Molecular Engineering, Peking University, Haidan District, Beijing 100871, China

^cPlant Immunity Research Group, RIKEN Center for Sustainable Resource Science, Kanagawa 230-0045, Japan

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ABSTRACT

Two homologous gene clusters for the biosynthesis of sesterterpenes betaestacins were identified from two phytopathogens, *Phoma betae* and *Colletotrichum orbiculare*. Heterologous expression of identified oxidation enzymes with previously-characterized PbTS1 (BtCA_{Pb}) resulted in the production of seven novel sesterterpenes. Although both strains possessed homologous enzymes, oxidation state of corresponding products were different from each other, suggesting that structural diversification of sesterterpene skeletons might be achieved by these homologous enzymes with different functions.

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Introduction

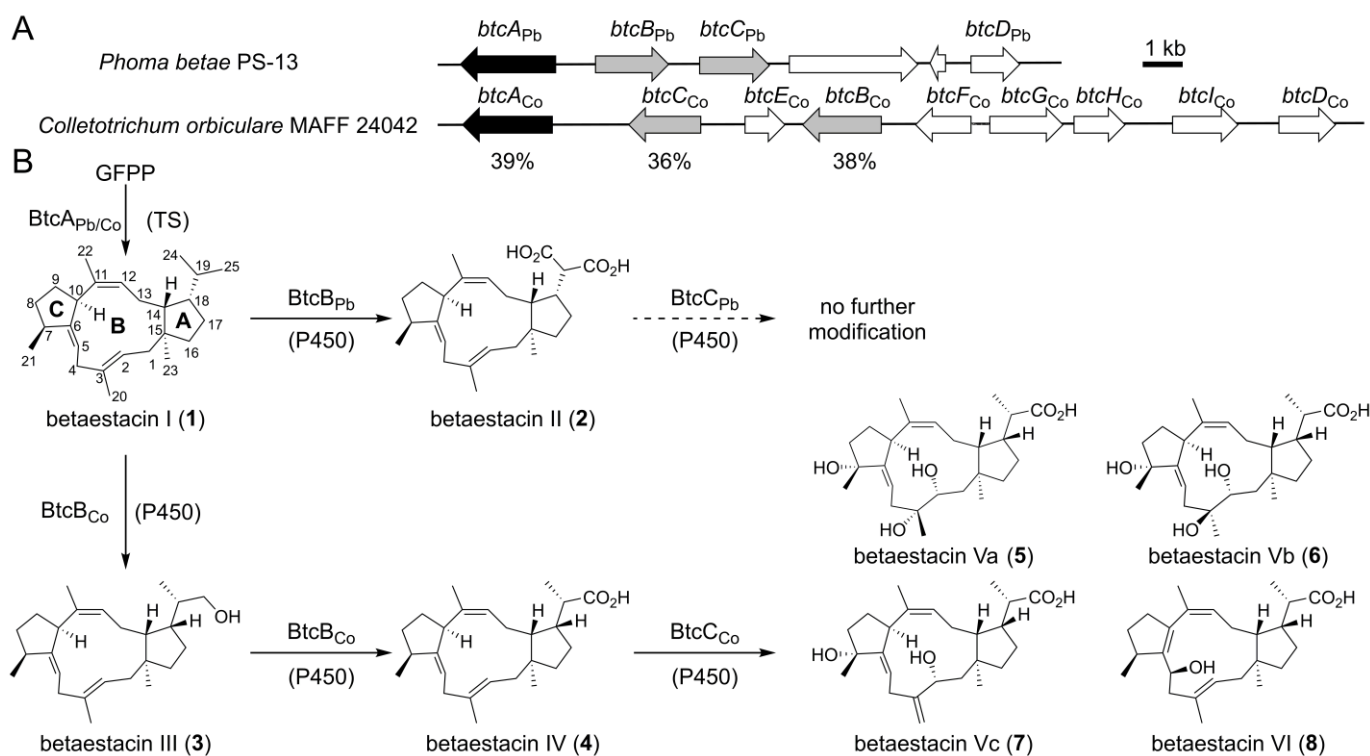
Bifunctional terpene synthases (BFTSs) are the multifunctional enzymes composed of two distinct domains, an N-terminal terpene synthase (TS) domain and a C-terminal prenyltransferase domain. The first example of this enzyme class is a fusicoccadiene synthase (PaFS), which was isolated from *Phomopsis amygdali*¹. Interestingly, a second BFTS, a phomopsene synthase (PaPS), was identified in this strain as well². Previously, we employed the *A. oryzae* expression system to characterize cryptic terpene synthases in fungal genomes and identified the first example of a sesterterpene synthase (AcOS) that catalyzed geranylarnesyl diphosphate (GFPP) synthesis followed by its cyclization to yield ophiobolin F³. This finding showed that BFTSs synthesize not only diterpenes but also the C25 sesterterpenes. After this finding, genome mining has become a standard method for the identification of novel sesterterpenes as in the case of sesterfisherol synthase (NfSS)⁴. To date, more than 10 BFTSs have been characterized by several research groups^{5–10}. In addition, a robust *A. oryzae* expression system also allowed us to perform co-expression of modification enzymes with BFTS, which functionalize sesterterpene skeletons^{4,11}. Herein, we report the identification of seven novel sesterterpenes by genome mining of BFTS against two phytopathogens, *Phoma betae* and *Colletotrichum orbiculare*.

Results and discussion

We have proposed that the initial cyclization modes of BFTSs were closely correlated with the amino acid sequences of the TS domains⁴. For example, BFTSs classified into Clade A catalyze

the A-ring formation at the C1-C15/C14-C18 position of GFPP to give a 5-15 ring system. Recently, focused expression of Clade A BFTSs enabled us to characterize four sesterterpene synthases, BmTS1, BmTS2, BmTS3, and PbTS1 (BtCA_{Pb}), that produce the structurally-related sesterterpenes, Bm1, Bm2, Bm3, and Pb1 (**1**, named betaestacin I in this study, Scheme 1). These products were proposed to be derived from the diastereomeric cationic intermediates in the NfSS reaction based on the determination of their absolute configurations and DFT calculations¹². Because a *btcA_{Pb}* gene was clustered with the three genes encoding oxidation enzymes (two P450s, BtCB_{Pb} and BtCC_{Pb}, and one α -ketoglutarate-dependent dioxygenase BtCD_{Pb}), we expected that heterologous expression of these enzymes could result in production of oxidized derivative(s) of the hydrocarbon **1**. On the other hand, we employed independent study for genome mining of the uncharacterized gene clusters expressed during the infection by *Colletotrichum orbiculare*. Based on transcriptomic analysis of this strain, we selected a cluster that consisted of another Clade A BFTS, BtCA_{Co}. Heterologous expression of this enzyme resulted in the production of the same hydrocarbon **1** (Figure S1 & S2), despite its relatively low homology with BtCA_{Pb} (39% amino acid identity). *btcA_{Co}* was also clustered with *btcB_{Pb}* and *btcC_{Pb}* homologs (36–38% sequence identity), suggesting that both gene clusters could give the same product(s). Therefore, BtCB_{Pb}, BtCC_{Pb}, and BtCD_{Pb} were initially expressed in AO-*btcA_{Pb}*, which produced **1**¹².

P450 genes *btcB_{Pb}* and *btcC_{Pb}* were cloned into the plasmid pUSA2 to construct pUSA2-*btcB_{Pb}* and pUSA2-*btcC_{Pb}*, while *btcD_{Pb}* was cloned into the plasmid pAdeA2 to construct

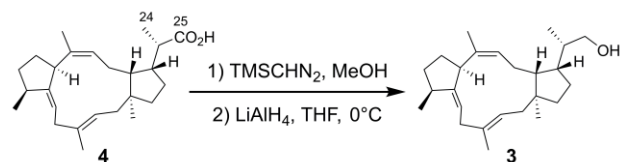


pAdeA2-*btcD_{Pb}*. Subsequently, resulting plasmids were introduced into the transformant AO-*btcA_{Pb}*. The metabolite profile of the AO-*btcA_{Pb}B_{Pb}* cultured on a solid medium was examined, and a new metabolite **2** was readily identified (Figure 1A). Based on HR-ESI-MS analysis, the molecular formula of **2** was determined to be C₂₅H₃₅O₄ (*m/z* 399.2504 [M-H]). In the ¹H NMR spectrum of **2**, a pair of characteristic doublet signals (H24 and H25) of isopropyl group was absent. Instead, new peaks with δ_c 174.1 and 174.0 appeared in ¹³C NMR, indicating presence of the geminal diacid moiety. This speculation was further confirmed by HMBC correlations between C24/C25 and both H19 and H18 (Figure S3). Hereafter, we named this compound betaestacin II (**2**). Other related metabolites are noted in similar way. Isolation of **2** indicated that BtcB_{Pb} is unequivocally responsible for the six-step oxidation of **1**. Because AO-*btcA_{Pb}B_{Pb}C_{Pb}* (Figure 1A) and AO-*btcA_{Pb}B_{Pb}C_{Pb}D_{Pb}* (data not shown) did not produce further oxidized metabolites, BtcC_{Pb} and BtcD_{Pb} were likely inactive in our current experimental conditions. In the case of the biosynthesis of ophiobolins, P450 derived from the producer strain catalyzed the expected oxidation, while a homologous enzyme from the cryptic gene cluster catalyzed unusual oxidations¹¹. Considering the possibility that enzymes encoded in the cryptic *btcX_{Pb}* gene cluster lost their original function, we next examined the heterologous expression of corresponding oxidation enzymes from the gene cluster derived from *C. orbiculare*.

The plasmids pUSA2-*btcB_{Co}* and pUSA2-*btcB_{Co}C_{Co}*, constructed by stepwise cloning of *btcB_{Co}* and *btcC_{Co}* into pUSA2, were introduced to the transformant AO-*btcA_{Pb}* that synthesized the same hydrocarbon **1** to yield AO-*btcA_{Pb}B_{Co}* and AO-*btcA_{Pb}B_{Co}C_{Co}*. In the LC-MS analysis of AO-*btcA_{Pb}B_{Co}C_{Co}* extracts, at least five metabolites were detected (compounds **4**–**8**) (Figure 1B). Their molecular formulae were determined to be C₂₅H₃₇O₂, C₂₅H₃₉O₅, C₂₅H₃₉O₅, C₂₅H₃₇O₄, and C₂₅H₃₉O₃, respectively, by HR-ESI-MS analysis, suggesting that these compounds were the oxidized derivatives of **1** and shared the same carbon skeleton. Therefore, the structural analyses of these

compounds were mainly performed by comparing NMR spectra of these compounds with those of **1** (Table S5).

At first, we determined the structure of the major metabolite **4** (betaestacin IV). Doublet methyl signal (H25) of the isopropyl group in **1** was absent in its ¹H NMR spectra while a signal with δ_c 182.2 appeared in ¹³C NMR, suggesting the presence of one carboxyl group in **4**. HMBC correlations between C25 and both H19 and H24 confirmed the presence of an oxidized moiety derived from the isopropyl group (Scheme 1). Production of **4** in AO-*btcA_{Pb}B_{Co}* (Figure 1B) revealed that BtcB_{Co} was solely responsible for the three-step oxidation of **1**. From the NOE correlations between H19–H23 and H17–H24, a newly formed chiral center was determined to be 19*S* (Figure S3). During isolation of the carboxylic acid **4** by large-scale fermentation a minor component **3** was also isolated. Its HR-ESI-MS analysis showed a signal with *m/z* 357.3178 ([M+H]⁺, calcd. 357.3132 for C₂₅H₄₁O). Based on its molecular formula, **3** was assumed to be a hydroxylated derivative of **1**. Methylation of the carboxylic acid **4** and its subsequent reduction by LiAlH₄ gave **3** (Figure S4), indicating that this compound was a C25-hydroxylated derivative of **1**.



Compounds **5** and **6** (betaestacin Va and Vb) had the same molecular formulae, which had three more oxygens and two more hydrogens compared to **4**, indicating that these were the oxidation products of **4**. Their ¹H and ¹³C NMR spectral data (Table S5) showed that C2–C3 double bond and allylic C7-proton signals were missing while three oxygenated carbon signals were observed in 72–79 ppm, suggesting that these compounds were

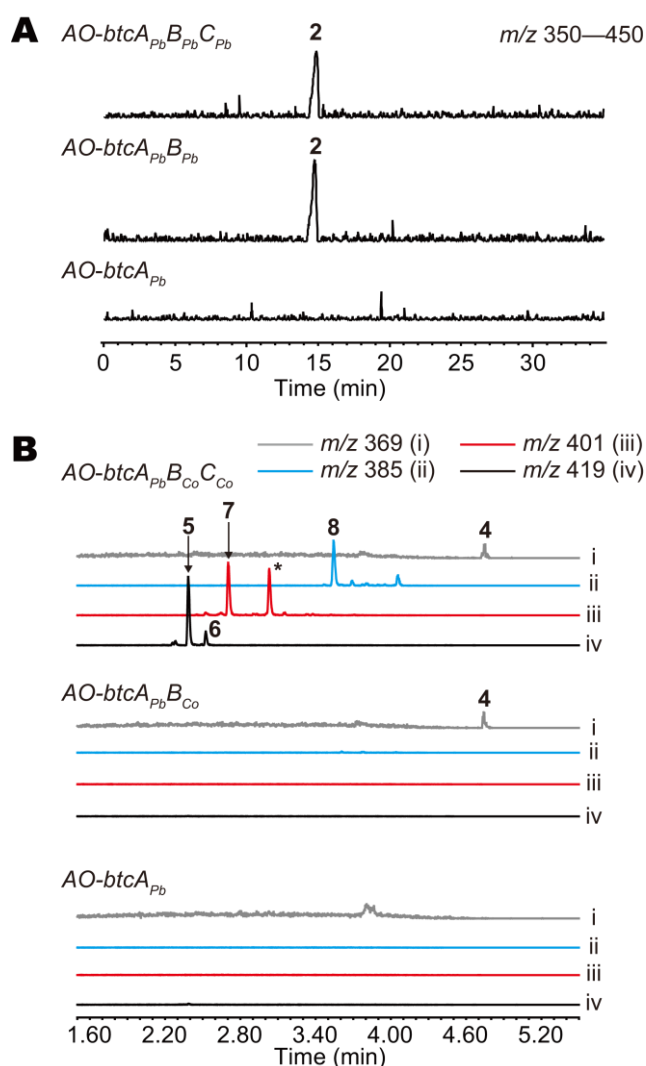


Figure 1. HPLC-MS analysis of the metabolites extracted from (A) AO-*btcA_{Pb}*, AO-*btcA_{Pb}B_{Pb}*, and AO-*btcA_{Pb}B_{Pb}C_{Pb}*. Chromatograms were extracted at *m/z* 350–450. (B) AO-*btcA_{Pb}*, AO-*btcA_{Pb}B_{Co}*, and AO-*btcA_{Pb}B_{Co}C_{Co}*. Chromatograms were extracted at *m/z* 369 (for **4**), *m/z* 385 (**8**), *m/z* 401 (**7**), and *m/z* 419 (**5** and **6**). The peak at 3 min (*) was unlikely sesterterpene derivative based on the TLC behavior

C2, C3, C7-trihydroxy derivatives. Detailed 2D NMR analyses enabled us to determine that both compounds had the same planar structure. In ^{13}C NMR spectra of these trihydroxy acids **5** and **6**, apparent differences were only observed around C3. Notably, ^{13}C chemical shifts of the C20 methyl group showed clear differences (δ_{C} 21.8 in **5** and δ_{C} 28.2 in **6**), suggesting that **6** was a C3 epimer of **5**. In NOESY spectra of **5** and **6**, NOE correlations between H2–H4 α and H4 β –H21 were observed, suggesting that their stereochemistries were 2*R*, 7*R* (Figure S3). This observation was further supported by NOESY analyses; NOE between H2 and H20 was not observed for **5**, while the corresponding NOE was observed for **6**. To obtain the corresponding acetonides from **5** and **6**, both compounds were treated with acetone in the presence of *p*-TsOH. Unexpectedly, both compounds gave the same acetonide **9** (Figure S5), which showed similar NOEs to those of **5**, indicating its stereochemistry was 2*R*, 3*R* and thus the stereochemistry of **6** should be 2*R*, 3*S* (Figure S6).

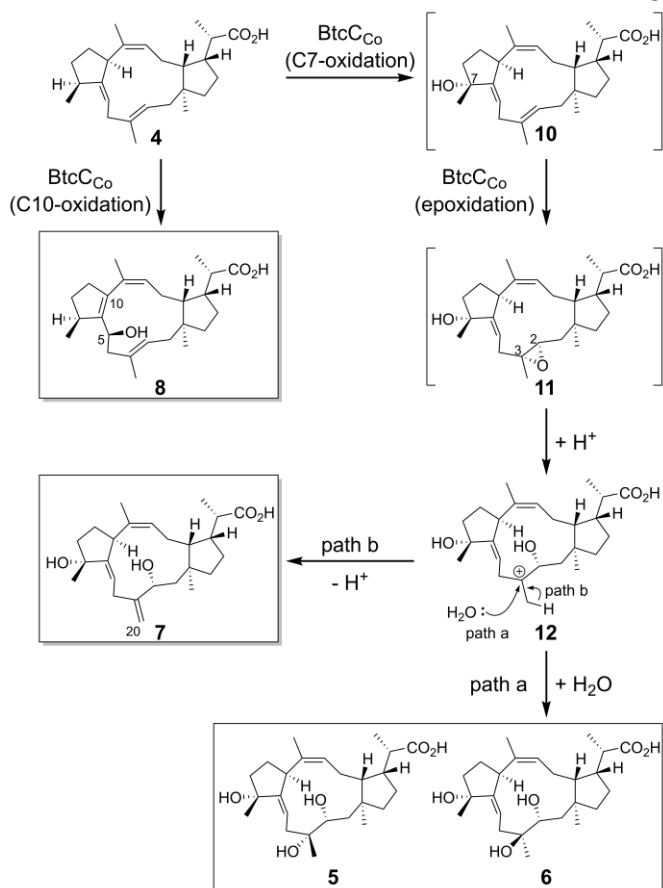
Structures of the minor products **7** and **8** (betaestacin Vc and VI) were also determined in a similar way. Molecular formulae of **7** and **8** suggested that they may be dihydroxy and

monohydroxy derivatives of **4**. In the HMBC spectra of **7**, correlations were observed between the exo-C20-methylene signals and the C2-oxymethine signal and between the C21-methyl group and the C7-oxygenated carbon/C6-olefin carbon signals, suggesting a planar structure. In ^1H NMR of **8** broad singlet of H10 and olefinic proton of H5 were missing, while C5-oxymethine signal was observed at 4.37 ppm. This signal was correlated to C3 and C7-signals in HMBC spectra suggesting that **8** is a 5-hydroxy acid. In NOESY spectra of **7**, NOE correlations between H2–H4 α and H4 β –H21 similar to those in **5** and **6** were observed, suggesting that its stereochemistry was 2*R*, 7*R* (Figure S3). In the NOESY spectra of **8**, H5 showed NOEs between both H22 and H20, indicating that the stereochemistry of C5 was *S* (Figure S3).

Structure determination of a series of betaestacin analogs enabled us to propose the biosynthetic pathway as shown in Scheme 2. *BtcB_{Co}* oxidizes the C25 methyl group to yield the carboxylic acid **4** via the alcohol **3**. On the other hand, *BtcB_{Pb}* catalyzes another three-step oxidation to yield the diacid **2**. The structure of the 5-hydroxy acid **8** suggested that an initial hydrogen abstraction catalyzed by P450 *BtcC_{Co}* may occur at C7 and C10 to obtain a putative C7-hydroxy product **10** and a C5-hydroxy acid **8** rearranged at the C ring, respectively. In the case of the product **8**, *BtcC_{Co}* might not accept **8** with it being accumulated as a plausible shunt product. On the other hand, a C7-hydroxy acid **10** may be further converted to the putative epoxide **11**. Considering the reactivity of the C3-alcohol in the acetonide formation of **5** and **6**, non-enzymatic epoxide opening resulted in the formation of the carbocation **12** under acidic condition, which was quenched by either addition of water or H16 deprotonation to yield **5**–**7**¹³. In the study of heterologous production of ophiobolin in solid medium, similar side reactions occurred to yield by-products.¹² Although an $\text{S}_{\text{N}}1$ -type epoxide opening is rare, co-occurrence of epimeric chaetoglobosins Q and R might be explained by this mechanism¹⁴. Cationic epoxide rearrangements have also been reported recently in the biosynthesis of the alkaloid aspoquinolone¹⁵.

As various secondary metabolite gene clusters were expressed in the infection of *Colletotrichum* species^{16,17}, we examined gene expression of the *btcA_{Co}* gene cluster in infection of *Nicotiana benthamiana*¹⁷. Although expression was highest *in vitro*-grown hyphae (VH), *btcA_{Co}* was also expressed in the early stages of the infection (1 dpi and 3 dpi) at lower levels (Figure S7A). This observation indicated that isolated betaestacins might have a physiological functions in this fungus. We then constructed the *btcA_{Co}* disruptant of *C. orbiculare* to examine whether the mutant showed phenotypic changes. Two knockout mutants were successfully obtained by homologous recombination and disrupting the target gene with a hygromycin resistance gene cassette (HPT) (Figure S7B, C). Gene knock-out (KO) was confirmed by PCR. Two *btcA_{Co}* KO mutants, #12 and #46, showed normal growth and pigmentation on plate culture (Figure S7D). Conidial formation of each mutant was also examined on PDA plates. Each mutant produced conidia at a similar level to wild type *C. orbiculare* (Figure S7E). As knock-out mutants did not exhibit any phenotypic changes *in vitro*, the isolated metabolites may function by manipulating host plants. The infection process for this mutant is under investigation and will be reported elsewhere.

In summary, by functional characterization of two similar biosynthetic gene clusters in two phytopathogens we isolated seven previously unknown sesterterpenes. In this study, we found that two homologous P450s, *BtcB_{Pb}* and *BtcB_{Co}*, catalyzed multiple oxidations of isopropyl groups although the oxidation



Scheme 2, Proposed mechanism for the biosynthesis of compounds 5–8.

levels were significantly different. The second set of homologous P450s, BtcC_{Pb} and BtcC_{Co} showed a different behavior; BtcC_{Pb} did not catalyze further oxidation while the corresponding BtcC_{Co} did catalyze the multistep oxidation of **4**. Although the physiological functions of the identified metabolites remain to be elucidated, structural diversification of the same carbon skeleton by similar oxidation enzymes might be a strategy used by these phytopathogens for adaptation to the environmental changes in host plants.

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Supplementary Material

Supplementary data (experimental procedures, supplementary figures and tables, and NMR data) associated with this article can be found in the online version.