



Title	Comparative Thermodynamic Studies of the Micellization of Amphiphilic Block Copolymers before and after Cyclization
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Supporting Information

Comparative thermodynamic studies of the micellization of amphiphilic block copolymers before and after cyclization

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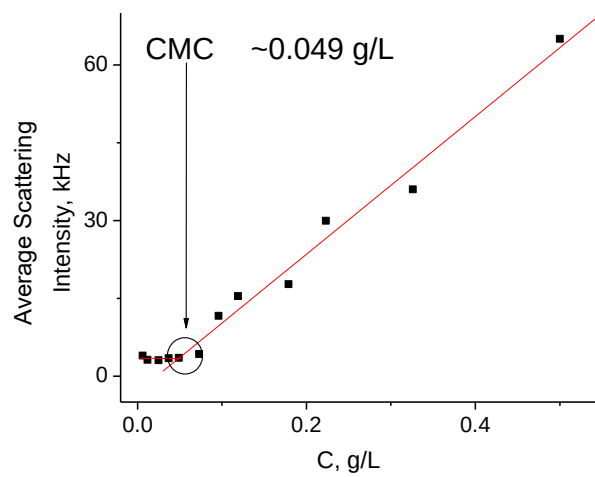


Figure S1. Plots of the average scattering intensity as a function of the copolymer concentration of **L1**.

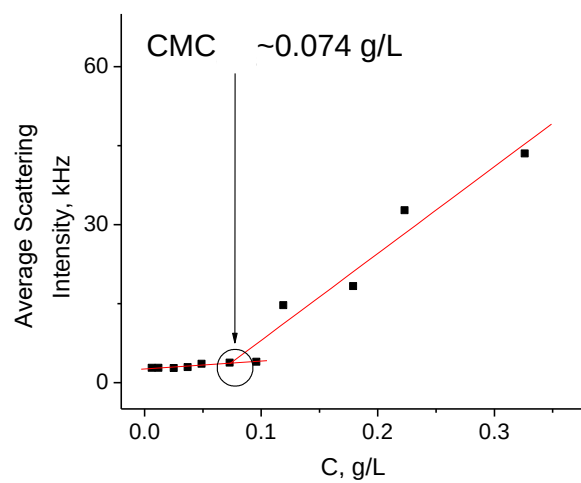


Figure S2. Plots of the average scattering intensity as a function of the copolymer concentration of C1.

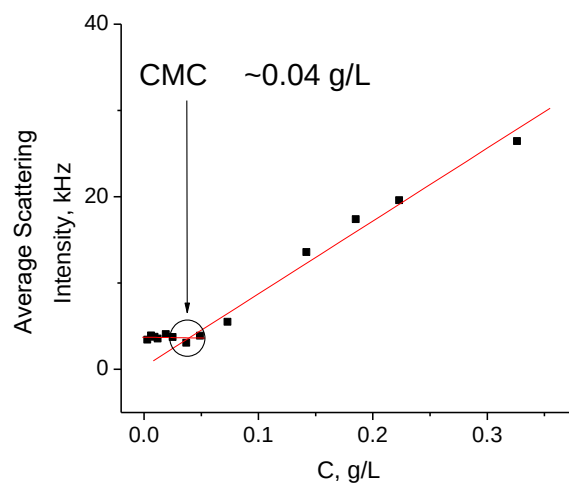


Figure S3. Plots of the average scattering intensity as a function of the copolymer concentration of **L2**.

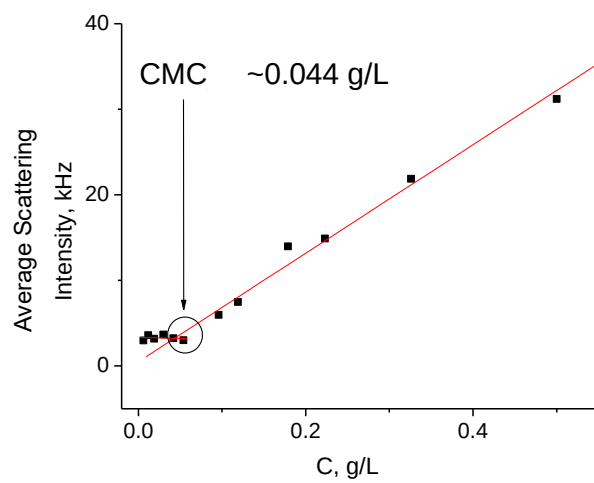


Figure S4. Plots of the average scattering intensity as a function of the copolymer concentration of **C2**.

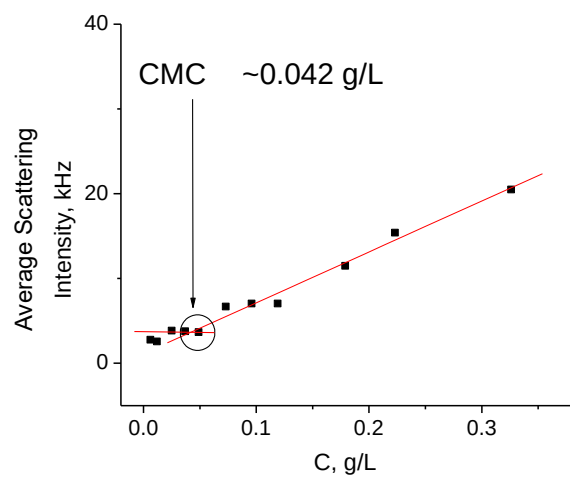


Figure S5. Plots of the average scattering intensity as a function of the copolymer concentration of **L3**.

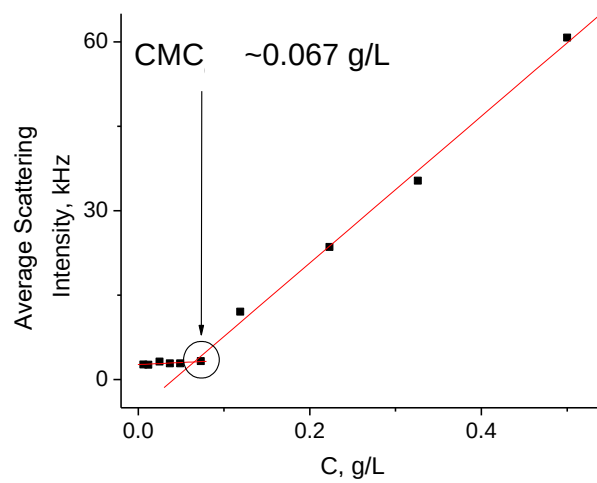


Figure S6. Plots of the average scattering intensity as a function of the copolymer concentration of C3.

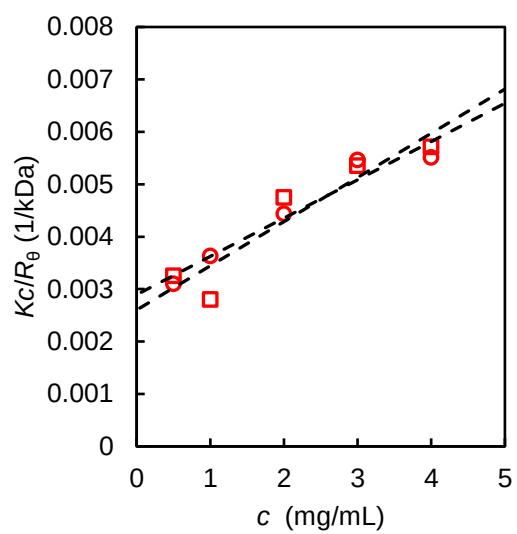


Figure S7. Zimm plots of **L1** and **C1** to determine $M_{w\text{ mic}}$ and A_2 of the micelles.

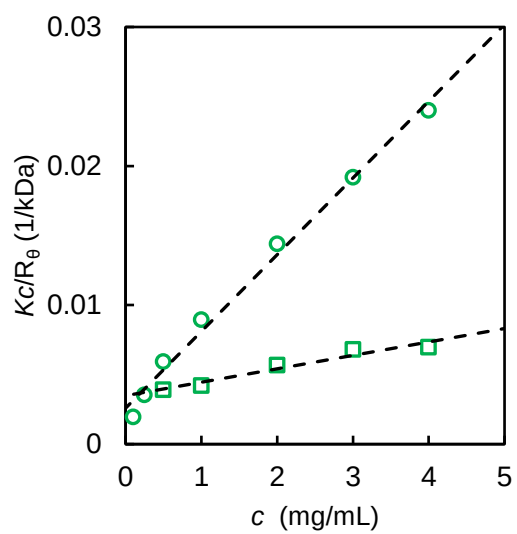


Figure S8. Zimm plots of **L2** and **C2** to determine $M_{w \text{ mic}}$ and A_2 of the micelles.

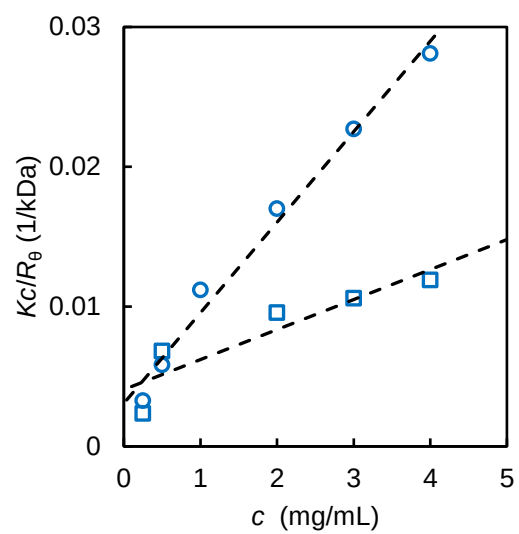


Figure S9. Zimm plots of **L3** and **C3** to determine $M_{w\text{ mic}}$ and A_2 of the micelles.

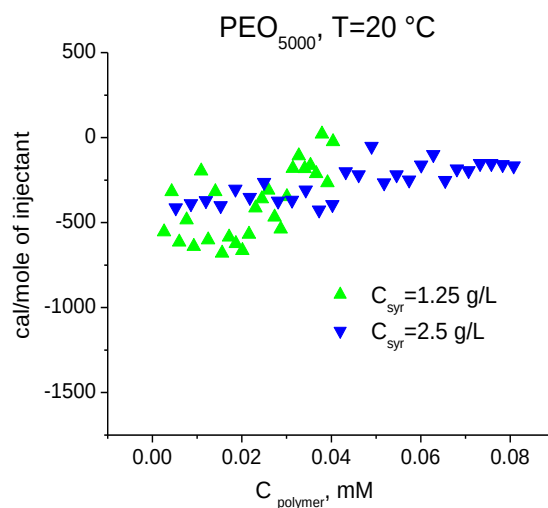


Figure S10. Enthalpy of dilution (ΔH_{dil}) vs. the concentration of PEO ($M_n = 5,000$). The concentration of PEO in the syringe attached to the microcalorimeter (C_{syr}) was 1.25 g/L (green) and 2.5 g/L (blue). $\Delta H_{\text{dil}}(\text{PEO})$ was -2.1 kJ/mol for 2.5 g/L.