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**Evidence of solar wind irradiation on mineral grains embedded in matrix of the
Northwest Africa 801 CR chondrite**

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ABSTRACT

Solar wind (SW) is preserved in meteorites as abundant solar noble gases. We performed *in situ* ^4He isotope imaging of mineral grains in the CR2 chondrite matrix of Northwest Africa 801 (NWA 801) using time-of-flight secondary neutral mass spectrometry with strong-field post-ionization. $^4\text{He}^+$ signals were detected along the surfaces of individual grains of Fe-Ni metal, ferrihydrite, olivine, pyroxene, and troilite. The high ^4He concentrations along the surfaces indicate implantation of SW into the mineral grains. We determined the SW- ^4He fluence of eight mineral grains from the line profiles across the grain boundaries. SW- ^4He fluence ranged from 2.7×10^{16} to 58×10^{16} atoms cm^{-2} . These fluences were then used to calculate the SW irradiation durations. Assuming irradiation occurred at 4 astronomical units, the durations ranged from 3.8 to 82 kyr. These durations correspond to the residence time of individual mineral grains on the surface of the parent body. The variation in residence time for the mineral grains suggests variable durations for local mixing and burial processes on the parent body. The SW exposure ages provide insights into the gardening rate driven by small-scale impact mixing processes on the parent body.

1. INTRODUCTION

Evidence for solar wind (SW) implantation was first observed in the Khor Temiki aubrite meteorite (Eberhardt et al., 1965), classified as a gas-rich meteorite. Analysis of this meteorite revealed a negative correlation between noble gas concentrations and grain sizes for minerals in the matrix. Noble gas concentrations in different grain-size fractions of the aubrite were proportional to the surface-to-volume ratio, except for cosmogenic ^3He and ^{21}Ne . Similar correlations were observed in lunar samples (e.g., Hintenberger et al., 1970, 1971), suggesting that individual grains are enriched with SW noble gases on their surfaces.

Another approach to confirming SW implantation in grain surfaces is closed system step etching (CSSE) gas extraction, which employs acid leaching to extract noble gases from the surface layers of minerals. CSSE-based noble gas analysis of gas-rich chondrites has shown that He- and Ne-rich components were released in the early etching steps, indicating their presence in the surface layers (Becker et al., 1998; Wieler et al., 2000; Riebe et al., 2017), and similar isotopic compositions to SW for He and Ne (Geiss et al., 1972; Heber et al., 2009). The $^4\text{He}/^{36}\text{Ar}$ and $^{20}\text{Ne}/^{36}\text{Ar}$ ratios in most gas-rich meteorites are lower than the SW ratios, indicating escape of He and Ne from the meteorites after SW implantation (Wieler et al., 1989; Lipschutz et al., 1993). On the other hand, the $^4\text{He}/^{36}\text{Ar}$ and $^{20}\text{Ne}/^{36}\text{Ar}$ ratios of Fe-Ni metal in the gas-rich chondrites are nearly identical to the SW ratios, indicating that SW-He and -Ne in Fe-Ni metal are particularly retentive than those in silicates (Murer et al., 1997). These CSSE

results provide evidence that SW noble gases are trapped in the near-surface layers of minerals via ion implantation.

In situ analysis can clarify the location and concentration of SW noble gases, and their distribution can be compared with petrographic observations. The conventional *in situ* noble gas analysis approach with laser ablation has achieved sub-millimeter-scale analysis of SW noble gases in meteorites. Nakamura et al. (1999) showed that SW noble gases are enriched in the clastic matrix but not in the pristine matrix of CM chondrites. *In situ* analysis with secondary neutral mass spectrometry (SNMS) achieved micrometer-scale ^4He analysis in solid materials (Bajo et al., 2015, 2016). Through laser ionization mass nanoscope (LIMAS) analysis, Bajo et al. (2024) showed that SW- ^4He was highly heterogeneously distributed grain by grain in the clastic matrix of a gas-rich meteorite, Northwest Africa 801 (NWA 801) CR2 chondrite. However, *in situ* analyses have not yet demonstrated that the SW-implanted ^4He resides at the very grain surface. NWA 801 contains a large amount of SW noble gases, about one order of magnitude higher than in other solar-gas-rich CR chondrites (Bischoff et al., 1993; Scherer and Schultz, 2000; Obase et al. 2021). Bajo et al. (2024) showed high ^4He concentration spots in the polished section of NWA 801. Nakashima et al. (2009) mentioned that NWA 852, the paired meteorite of NWA 801, was exposed to galactic cosmic rays on the parent body for several tens of million years. The previous studies suggest that NWA 801 preserves the regolith environments of the CR chondrite parent body.

In this study, we conducted *in situ* analysis of the ^4He isotope distribution using LIMAS in individual mineral grains embedded in the NWA 801 section at a sub-micrometer spatial resolution, revealing the distribution of SW- ^4He on the surface region of individual mineral grains. Furthermore, the SW irradiation duration for each mineral and the implications for the history of regolith formation on the parent body are discussed.

2. MATERIALS AND METHODS

2.1. Sample preparation and electron microscopy

A polished section of the NWA 801 meteorite was used in this study. A chip of the chondrite, embedded in an epoxy resin disk, was polished using an automatic polishing machine (MA-200e, Musashino Denshi) at Hokkaido University. The polished section was then cut into a $5 \times 7 \times 1 \text{ mm}^3$ cuboid using a low-speed diamond saw (IsoMet Low Speed, Buehler) for placement in a sample holder. To minimize degassing, the cut specimen was immersed in acetone for 1 day to remove the epoxy resin. Finally, the specimen surface was coated with an approximately 10 nm thick gold film via vacuum deposition.

Backscattered electron (BSE) images of the sample were obtained using a field emission scanning electron microscope (FE-SEM; JSM-7000F, JEOL) at Hokkaido University to characterize the petrography prior to isotope-imaging analysis. Quantitative elemental analysis and X-ray elemental mapping were performed using an energy-dispersive X-ray

spectrometer (EDS; X-Max 150, Oxford) attached to the FE-SEM. A 15 keV electron beam probe was used with a current of 4 nA.

2.2. Helium isotope analysis

He isotope analysis was conducted using an SNMS (LIMAS, JEOL) at Hokkaido University (Bajo et al., 2024). In this study, we used a finer primary $^{69}\text{Ga}^+$ beam of 1 nA than that of Bajo et al. (2024) of 40 nA to achieve *in situ* analysis with a sub-micrometer spatial resolution. The primary ion beam with an energy of 30 keV was pulsed in 200 ns and focused onto the sample surface. The sample surface was inclined at 55° to the primary beam, resulting in a vertical beam diameter 1.7 times larger than the horizontal diameter.

Sputtered neutrals were photoionized under strong-field ionization conditions using a focused femtosecond laser (~ 5.5 mJ, Astrella, Coherent) installed in LIMAS and operating at a 1 kHz repetition rate. The laser was focused at a 30 μm diameter spot approximately ~ 50 μm above the sample surface. Ions with m/z 4 ($^4\text{He}^+$), 12 ($^{24}\text{Mg}^{2+}$), 16 ($^{32}\text{S}^{2+}$), 18.6 ($^{56}\text{Fe}^{3+}$), and 28 ($^{56}\text{Fe}^{2+}$, $^{28}\text{Si}^+$) were measured using a multi-turn time-of-flight (TOF) mass spectrometer, “MULTUM II” (Okumura et al., 2004; Tonotani et al., 2016), with an acceleration voltage of -5 kV. The ion-injection system parameters for MULTUM II were optimized for high ion-transmission efficiency based on the study by Tonotani et al. (2016). The flight time was adjusted to achieve a mass resolving power of $\sim 20,000$ (calculated using $T/2\Delta T$) for m/z 4. Interference ions, such as $^{12}\text{C}^{3+}$ and $^{16}\text{O}^{4+}$ for $^4\text{He}^+$, were removed using ion

gates in the MULTUM II ion trajectory. A typical TOF mass spectrum from the sample is shown in Fig. 1. To obtain a 2D isotope image, the primary beam was scanned horizontally and vertically at 100 nm intervals. Each point in the image represents a pixel, and 600–6000 TOF mass spectra were collected for each pixel. The analytical conditions for the isotope-imaging analysis are summarized in Table 1.

The background signal of ^4He (hereafter referred to as blank ^4He) originates from residual ^4He gas above the sample surface. This was measured without the primary beam pulse after each isotope-imaging run. The $(1\text{--}2) \times 10^6$ TOF spectra were collected to determine the blank ^4He intensity, which varied with analytical conditions, ranging from 1×10^{-4} to 1×10^{-5} ion counts per mass spectrum scan (cpms). The background signals of ^{56}Fe , ^{24}Mg , and ^{32}S are negligible compared to their signals from the sample.

2.3. Spatial resolution of isotope-imaging analysis

The spatial resolution for individual ion images was determined using line scan profiles of $^{56}\text{Fe}^{2+}$ and $^{24}\text{Mg}^{2+}$ (Figs. S1–S6) and $^4\text{He}^+$ (Figs. 3–8) across the interface between individual mineral grain and fine-grained groundmass, defined by the 16–84% criterion (Nagata et al., 2019). This is equivalent to the $\pm 1\sigma$ criterion for a Gaussian beam. The spatial resolutions for the individual analyses are listed in Table 2. Because the spatial resolutions calculated from the $^{56}\text{Fe}^{2+}$ and $^{24}\text{Mg}^{2+}$ profiles are broadened by the interface–surface angle, and the ^4He layer is narrower than 100 nm (see Section 4.2), the spatial resolution of the image

in this study is defined as the width of the ^4He layer. The resolutions varied with individual images: 100 nm for Areas 1, 2 and 3, 150 nm for Area 6 and 240 nm for Areas 4 and 5.

2.4. Quantification of ^4He concentration

To determine the ^4He concentration in the sample, a ^4He -implanted Fe-Ni alloy ($\text{Fe}_{0.55}\text{Ni}_{0.45}$) plate was prepared as a standard, with an implantation dose of 3×10^{15} atoms cm^{-2} at an energy of 20 keV. The ^4He -implanted Fe-Ni standard was analyzed using LIMAS with a 30 keV and 32 nA $^{69}\text{Ga}^+$ primary ion beam, delivered in 400 ns pulses. A 30×30 pixel area ($12 \times 20 \mu\text{m}^2$) was scanned, and 100 TOF mass spectra were collected per pixel. Depth profiles of $^4\text{He}^+$, $^{56}\text{Fe}^{2+}$, and $^{58}\text{Ni}^{2+}$ were obtained from the central 5×5 pixels to minimize crater edge effects. Other measurement conditions were identical to those used for sample analysis. Following depth profiling, the crater depth was measured using a confocal laser microscope (VK-X200, Keyence).

Since the ion intensity is proportional to the primary beam current, we calculate the ^4He concentration using the ratio of the primary beam currents between the isotope imaging and the Fe-Ni standard measurement to correct for the ^4He concentration calculations in the sample. Because photoionization by a focused fs laser is less sensitive to matrix effects caused by differences in ionization efficiency depending on the sample material, the ^4He concentration was calculated using the following equation, adapted from Bajo et al. (2024):

$$[^4\text{He}] = (^4\text{He}^+)_{meas} / I_{p1} \times SF^{\text{He}} \quad (1)$$

where $[^4\text{He}]$, $^4\text{He}^+_{\text{meas}}$, I_{p1} , and SF^{He} denote the ^4He concentration in the measuring volume of a single pixel (atoms cm^{-3}), measured ^4He intensities (cpms), primary beam currents for sample, and sensitivity factor for ^4He ($\text{atoms cm}^{-3} \text{ cpms}^{-1}$), respectively. The SF^{He} is derived from the standard analysis as follows:

$$SF^{\text{He}} = \frac{\phi \times n}{\sum_{k=1}^n ((^4\text{He}^+)_{\text{meas}} - (^4\text{He}^+)_{\text{bg}})_k / I_{p2} \times d} \quad (2)$$

where Φ , k , and I_{p2} , d denote the ^4He fluence (atoms cm^{-2}) for the standard, number of measurement layer (from 1 to n), primary beam currents for standard, and sputtered crater depth (cm), respectively. Additionally, $(^4\text{He}^+)_{\text{bg}}$ denotes the blank ^4He .

3. RESULTS

3.1. Petrography of measurement areas

NWA 801 is classified as a CR2 chondrite (Connolly et al., 2007) and is highly enriched in SW noble gases (Obase et al., 2021). BSE and X-ray elemental images of the polished section of NWA 801 are shown in Fig. 2. SEM analysis reveals that the NWA 801 section contains millimeter-sized chondrules and their fragments embedded in matrix. Pervasive rust veins, primarily composed of ferrihydrite, which are typical terrestrial weathering products, are evident throughout the sample. The matrix contains mineral grains of Fe-Ni metal, ferrihydrite, olivine, pyroxene, and troilite, ranging from approximately 1 to 50 μm . These mineral grains are surrounded by a fine-grained groundmass typically less than 0.1

μm in size (Fig. 2c). Fe-Ni metal and troilite also occur within chondrule interiors and rims. Some Fe-Ni metal grains have been partially or completely replaced by ferrihydrite. In this study, we examined the mineral grains and the surrounding fine-grained groundmass in Areas 1–6, as shown in Fig. 2a.

3.2. ⁴He distribution of mineral grains

Enlarged BSE and ⁴He⁺ images of Areas 1–6 are shown in Figs. 3–8. The ion images reveal that the mineral grains (<50 μm in size) of Fe-Ni metal (Figs. 3–5), ferrihydrite (Figs. 4–7), olivine (Figs. 3, 6, and 8), pyroxene (Figs. 3, 6, and 7), and troilite (Fig. 3–8) are enriched in ⁴He along part of their grain-surface contours. These mineral grains are embedded in a fine-grained groundmass. Line scans of ⁵⁶Fe²⁺ and ²⁴Mg²⁺ across a few selected grain boundaries illustrate the relationships between the boundaries and ⁴He-concentrated regions. These line scans were obtained from a location with a high ⁴He⁺ intensity to determine the largest fluence of SW irradiation of the grain.

Fe-Ni metal

Three Fe-Ni metal grains can be observed (Figs. 3–5) in Areas 1–3 (Fig. 2). In Area 1, the Fe-Ni metal grain is approximately 20 × 20 μm² (Fig. 3a). ⁴He⁺ signals were detected along the grain edge, except for the lower right edge (Fig. 3b). Intensity changes in the line profiles for ⁵⁶Fe²⁺ and ²⁴Mg²⁺ delineate the grain boundary (Fig. 3c). Similar profile changes are observed for the other Fe-Ni metal grains (Figs. 4c, 4d, and 5c). The ⁴He⁺ signal

appears as a peak at the metal–groundmass interface with a width of 100 nm, defined as the full width at $e^{-\frac{1}{2}}$ height of the $^4\text{He}^+$ peak maximum, equivalent to $\pm 1\sigma$ for a Gaussian distribution. The peak ^4He concentration is 6.4×10^{21} atoms cm^{-3} (Fig. 3c), and the $^4\text{He}^+$ signals are below the detection limit in the metal's interior and the surrounding groundmass.

In Area 2, the Fe-Ni metal grain is approximately $5 \times 7 \mu\text{m}^2$ (Fig. 4a). $^4\text{He}^+$ signals are detected along the grain edge, except for the center-left edge (Fig. 4b). The $^4\text{He}^+$ profile exhibits a single peak with a 100 nm width ($\pm 1\sigma$) at the narrow distribution of the metal–groundmass interface. The peak ^4He concentration is 3.1×10^{21} atoms cm^{-3} (Fig. 4c), and the $^4\text{He}^+$ signals are below the detection limit in the metal's interior and the surrounding groundmass. The $^4\text{He}^+$ peak width in Line C–D (Fig. 4d) is wider than that in Line A–B (Fig. 4c), and a ^4He peak is observed within the Fe-Ni metal. This wider $^4\text{He}^+$ distribution in Line C–D is attributed to the diagonal polishing of the grain edge.

In Area 3, the Fe-Ni metal grain is approximately $20 \times 30 \mu\text{m}^2$ (Fig. 5a). $^4\text{He}^+$ signals can be partially observed along the grain edge (Fig. 5b), whereas other areas are below the detection limit. The $^4\text{He}^+$ profile exhibits a single 100 nm wide peak ($\pm 1\sigma$). The peak ^4He concentration is 2.2×10^{21} atoms cm^{-3} (Fig. 5c).

Ferrihydrite

Two ^4He -rich ferrihydrite grains can be observed (Figs. 6 and 7) in Areas 4 and 5 (Fig. 2). In Area 4, the grain is approximately $10 \times 15 \mu\text{m}^2$ (Fig. 6a), and $^4\text{He}^+$ signals were

detected along its edge (Fig. 6b). Line profiles of $^4\text{He}^+$, $^{56}\text{Fe}^{2+}$, and $^{24}\text{Mg}^{2+}$ across the grain boundary exhibit significant intensity changes at the boundary (Fig. 6c). The $^4\text{He}^+$ profile shows a single 250 nm wide peak ($\pm 1\sigma$), and the peak ^4He concentration is 17.0×10^{21} atoms cm^{-3} (Fig. 6c). The $^4\text{He}^+$ signals are below the detection limit in the ferrihydrite's interior, whereas weak ^4He intensities are detectable in the groundmass, with small ^4He -enriched spots (Fig. 6b).

In Area 5, the ferrihydrite grain is approximately $7 \times 8 \mu\text{m}^2$ (Fig. 7a), and $^4\text{He}^+$ signals are detected along its edge (Fig. 7b). Line profiles of $^{56}\text{Fe}^{2+}$ and $^{24}\text{Mg}^{2+}$ (Fig. 7c) show similar changes at the grain boundary as those observed in the ferrihydrite grain in Area 4. The $^4\text{He}^+$ profile shows a single 240 nm wide peak ($\pm 1\sigma$), and the peak ^4He concentration is 0.6×10^{21} atoms cm^{-3} (Fig. 7g). The $^4\text{He}^+$ signals in the ferrihydrite's interior are below the detection limit.

Although ferrihydrite grains were also present in Areas 2 and 3, detectable $^4\text{He}^+$ were not observed at the grain boundaries or within the grains (Figs. 4 and 5).

Olivine and pyroxene

^4He -rich olivine and pyroxene grains were observed in the $^4\text{He}^+$ images of Areas 4 and 5, as shown in Figs. 6 and 7, respectively. In Area 4, the olivine grain is approximately $3 \times 2 \mu\text{m}^2$ (Fig. 6a), and $^4\text{He}^+$ signals are detected along its edge (Fig. 6b). The line profiles show a ^4He peak at the grain boundary (Fig. 6d) with a width of 230 nm ($\pm 1\sigma$) and a peak

concentration of 0.9×10^{21} atoms cm^{-3} (Fig. 6d). The $^4\text{He}^+$ signals in the olivine's interior were below the detection limit, whereas the pyroxene grains in this area exhibited no detectable $^4\text{He}^+$ signals.

In Area 5, the pyroxene grain is approximately $2 \times 1 \mu\text{m}^2$ (Fig. 7a), and ^4He is observed along its edge (Fig. 7b). The $^4\text{He}^+$ peak in the line profile appears at the grain boundary with a width of 240 nm ($\pm 1\sigma$) (Fig. 7c) and a peak concentration of 0.7×10^{21} atoms cm^{-3} (Fig. 7h). The $^4\text{He}^+$ signals in the pyroxene's interior were below the detection limit and no ^4He was detected within other pyroxene grains in Area 5.

Troilite

Four ^4He -rich troilite grains were observed in the $^4\text{He}^+$ images of Areas 2, 4, 5, and 6, as shown in Figs. 4, 6, 7, and 8. The largest troilite grain of approximately $5 \times 5 \mu\text{m}^2$ (Fig. 8a) was observed in Area 6, and $^4\text{He}^+$ signals were detected along its edge, except for the upper and left edges (Fig. 8b). Line scans of $^4\text{He}^+$, $^{56}\text{Fe}^{2+}$, and $^{24}\text{Mg}^{2+}$ across the grain boundary were conducted to analyze the ^4He -concentrated region. As shown in Fig. 8c, a $^4\text{He}^+$ peak appears at the grain boundary with a width of 150 nm ($\pm 1\sigma$) and concentration of 2.5×10^{21} atoms cm^{-3} (Fig. 8c). However, ^4He is not detected within the grain or the surrounding fine-grained groundmass. The ^4He concentrations of the troilite grains in Areas 2, 4, and 5 are 2.9×10^{21} , 10.9×10^{21} , and 2.7×10^{21} atoms cm^{-3} , respectively. These grains are attached to ferrihydrite grains. Because the sizes of these grains were smaller than the spatial resolution of

the analysis, the spatial distributions of ^4He within them could not be determined.

Fine-grained groundmass

All areas imaged in this study contained a fine-grained groundmass surrounding the mineral grains. In Areas 4 and 5, $^4\text{He}^+$ signals were detected from the groundmass (Figs. 6 and 7), with a typical concentration of approximately 0.6×10^{21} atoms cm^{-3} . ^4He -rich spots with concentrations of $1.8\text{--}8.4 \times 10^{21}$ atoms cm^{-3} were also observed, often corresponding to S and Fe enrichments (Figs. 6 and 7), suggesting that they represent fine-grained iron sulfides. The areas with $\sim 0.6 \times 10^{21}$ atoms cm^{-3} (blue areas in Fig. 6b, 6f, and 7b) appeared to be composed mainly of fine-grained silicates, having a concentration is similar to the peak concentrations of the olivine and pyroxene grains (Figs. 6d and 7h). By contrast, $^4\text{He}^+$ signals were not detected from the groundmass in Areas 1–3 and 6 (Figs. 3–5, 8). Therefore, we conclude that ^4He content varies significantly within the groundmass.

4. DISCUSSION

4.1. Evidence of SW implantation

The line profiles of the mineral grains showed that the $^4\text{He}^+$ signals characteristically originated from the grain boundaries (Figs. 3–8). These distributions suggest that the ^4He is trapped either in the crystal edges or in the fine-grained groundmass attached to the crystals. In the former case, the trapped ^4He would be located at shallower depths than the

spatial resolution of the image (~100 nm) from the crystal surface. Otherwise, the fine-grained groundmass, smaller than 100 nm, would contain ^4He .

The $^4\text{He}^+$ signals on the diagonally polished edge of the Fe-Ni metal grain demonstrated that the peak position is within the metal (Fig. 4d), providing evidence that ^4He is trapped within the metal itself. We first observe abundant ^4He at the grain surface due to the SW implantation into the grain of the meteorite. The ^4He images shown in Figs. 6 and 7 also validate that the ^4He peaks are within the minerals rather than the groundmass. By contrast, no observations showed groundmass containing ^4He concentrations comparable to the ^4He peaks at the grain boundaries. The ^4He concentration in the groundmass was one to two orders of magnitude lower than that in the ^4He peaks at the grain boundaries. Therefore, it is plausible that ^4He is trapped at shallow depths from the surfaces of the mineral grains. The spatial resolution of the isotope images is not sufficient to observe the shallow SW implantation depth of approximately 20–40 nm from the surface of the minerals, which is expected by the TRIM (the Transport of Ions in Matter) code (Ziegler et al., 2012).

The ^4He -enriched layer beneath the grain surface, less than 100 nm thick, can be produced through ion implantation with kinetic energies of ~keV/nucleon (Futagami et al., 1990; Kuhlman et al., 2015; Wiens et al., 2004; Grimberg et al., 2008; Bajo et al. 2015). In interplanetary space, such low-energy ions are delivered by the SW. The ^4He enrichment observed within 100 nm for the Fe-Ni metal grains (Fig. 3c) is consistent with the typical SW

range of 20–40 nm from the surface. Therefore, we conclude that the ^4He line profiles in Figs. 3–8 provide direct evidence of SW implantation into the mineral grains. These grains were exposed to SW on the surface of the parent body and mechanically mixed with other meteorite precursor materials (Hashiguchi et al., 2013) to form the current structure of NWA 801 by subsequent compaction in the parent body (Charles et al., 2018).

Additionally, abundant SW- ^4He was observed at the edges of the ferrihydrite grains (Figs. 6 and 7). Ferrihydrite is a typical terrestrial weathering product (Noguchi, 1994), and trapped light noble gases are considered to be released during terrestrial weathering (Scherer et al., 1994; Stelzner et al., 1999; Bland et al., 2006). The presence of SW- ^4He in ferrihydrite suggests that the implanted ^4He remained in the ferrihydrite even though the original phase, likely metal, weathered to ferrihydrite on Earth. Another interpretation is that the ferrihydrite formed during aqueous alteration on the parent body, similar to that suggested for CI chondrites by Tomeoka and Buseck (1988). In this scenario, the altered ferrihydrite grains were exposed to SW on the parent body after formation. The results of this study do not allow us to determine which process is more plausible for the formation of ferrihydrite grains. However, in either case, the ^4He distributions in the ferrihydrite grains were produced by SW irradiation.

Bajo et al. (2024) using micrometer ^4He analysis firstly showed that ^4He is highly concentrated and sparsely scattered in the matrix in a gas-rich meteorite. This study advanced

their findings through sub-micrometer ^4He analysis and is the first to identify the ^4He concentrated on the surfaces of individual mineral grains. Thus, it provides direct evidence for SW implantation into the constituent minerals of the meteorite. Furthermore, the observed heterogeneity of SW- ^4He implantation may be related to the regolith formation dynamics on the parent body.

4.2. SW- ^4He fluence in mineral grains

Because the ^4He was implanted by SW irradiation (with a typical energy of ~ 1 keV/nucleon), the actual peak width should be several tens of nanometers. The observed line profiles (Figs. 3–8) were likely broader than the true profiles and exhibited lower peak concentrations because of the ~ 100 nm spatial resolution of the images. However, the integral of the ^4He concentrations in each line profile represents the total amount of ^4He in the sputtered volume. Therefore, the SW- ^4He fluence, Φ_{sample} , can be calculated from the line profile as follows:

$$\Phi_{\text{sample}} = \sum_{k=1}^n (C_{\text{sample}} \times L)_k \quad (3)$$

where C_{sample} is the ^4He concentration (atoms cm^{-3}) and L is the pixel width of ^4He (cm). This method enables direct quantification of the SW- ^4He fluence to the grain surface.

The SW- ^4He fluences of the mineral grains are summarized in Table 3. The SW- ^4He fluences for the ferrihydrite grains in Areas 4 and 5 are $(58.4 \pm 2.2) \times 10^{16}$ and $(3.0 \pm 0.2) \times 10^{16}$ atoms cm^{-2} , respectively. Two ferrihydrite grains in Areas 4 and 5 were analyzed in the

same analytical session, with one exhibiting the highest $^4\text{He}^+$ intensity and the other a relatively low $^4\text{He}^+$ intensity. Similar $^{56}\text{Fe}^{2+}$ signal emissions, a major element of ferrihydrite, were measured from the two grains (Figs 6, 7). Therefore, the differences of $^4\text{He}^+$ intensity are not artifact and calculated ^4He concentration in Fig. 6c is 26 times higher than that in Fig. 7g. These results suggest that the ferrihydrite in Area 4 has the highest SW- ^4He fluence. The SW- ^4He fluences for the Fe-Ni metal grains range from (3.7 ± 0.8) to $(11.5 \pm 1.7) \times 10^{16}$ atoms cm^{-2} . The troilite grain in Area 6 has a fluence of $(5.6 \pm 0.9) \times 10^{16}$ atoms cm^{-2} . The pyroxene and olivine have fluences of (2.7 ± 0.2) and $(2.9 \pm 0.5) \times 10^{16}$ atoms cm^{-2} , respectively. These values range from 2.7×10^{16} to 58.4×10^{16} atoms cm^{-2} , representing an approximately 20-fold difference in the SW- ^4He fluence. Because SW- ^4He in minerals may not be quantitatively retained in regolith samples (e.g., Signer et al. 1977), it should be noted that the ^4He fluences obtained in this study represent lower limits. Fe-Ni metals in gas-rich meteorites have the $^4\text{He}/^{20}\text{Ne}$ ratio of ~ 630 and $^4\text{He}/^{36}\text{Ar}$ ratio of $\sim 30,000$ (Murer et al. 1997), nearly identical to the SW ratios of 648 and 27,300, respectively (Heber et al. 2009). The bulk $^4\text{He}/^{20}\text{Ne}$ ratio in NWA 801 is 458 (Obase et al. 2021), which indicates at least $\sim 30\%$ of ^4He escaped from NWA 801. Thus, the ^4He loss of the individual grains, especially Fe-Ni metal grains, is relatively small so we do not consider ^4He loss in this discussion.

Based on these large variations in SW- ^4He fluence, we can estimate the SW irradiation duration for each mineral grain. Assuming $1/\pi$ SW exposure owing to the rotation

of the parent body, the SW irradiation duration, T (yr), is calculated as follows:

$$T = \Phi_{sample} / (\varphi_{pb} \times \frac{1}{\pi}) \quad (4)$$

The SW flux φ_r (atoms $\text{cm}^{-2} \text{yr}^{-1}$) decreases with the square of the distance, r (astronomical unit: au), from the Sun (Adhikari et al., 2014). The SW flux at the parent body φ_{pb} (atoms $\text{cm}^{-2} \text{yr}^{-1}$) is calculated as

$$\varphi_{pb} = \varphi_{1\text{au}} \times \frac{1}{r^2} \quad (5)$$

Obase and Nakashima (2023) showed that the SW flux recorded in solar-gas-rich meteorites is comparable to the current levels. The present SW- ^4He flux at 1 au ($\varphi_{1\text{au}}$) is 3.57×10^{14} atoms $\text{cm}^{-2} \text{yr}^{-1}$ (Heber et al., 2009). The heliocentric distance of the CR chondrite parent body during SW irradiation might have been between the current main asteroid belt (2.2–3.3 au; DeMeo and Carry, 2014) and the presumed formation region of the CR chondrite parent body (4 au; Desch et al., 2018). The calculated SW irradiation durations range from 1.1 to 25 kyr at 2.2 au, 2.6 to 56 kyr at 3.3 au, and 3.8 to 82 kyr at 4 au (Fig. 9, Table 3).

The SW irradiation durations represent the residence time of the individual mineral grains on the top surface of the parent body, reflecting the timescale of local mixing on a millimeter-scale in its regolith. If irradiation occurred at 4 au, the residence times of the seven out of the eight individual mineral grains are 3.8–16 kyr, which is consistent with the ranges observed for Ryugu samples (≤ 10 kyr; Okazaki et al., 2023) and lunar regolith (2.0–13 kyr; Signer et al., 1977; Li et al., 2023). These estimated SW irradiation durations suggest that

the typical timescale of millimeter-scale regolith mixing on the parent body was about ten thousand years. The high ^4He fluence for the ferrihydrite grain in Area 4 (Fig. 6) suggests a SW irradiation duration of 82 kyr, which is more than ten times longer than that of the other grains. This exceptionally high ^4He fluence suggests that the local mixing process lasted longer than that for the other grains. Additionally, the long residence time may indicate that this is an exotic regolith grain incorporated from a location where the mixing process is less frequent. If this is indeed the case, this implies that there is a horizontal transportation mechanism for regolith (e.g., electrostatic levitation; Lee, 1996). Such large variations in regolith mixing frequencies, over several tens of times, have been observed on the surface of the asteroid Vesta (Denevi et al., 2016).

5. CONCLUSIONS

We presented a novel technique for observing past SW-implanted ^4He from individual mineral grains in meteorites and a method for quantifying SW- ^4He fluence within them. *In situ* ^4He isotope imaging with LIMAS, applied to the CR2 chondrite NWA 801, provided the first direct evidence of SW implantation in individual grains of a solar-gas-rich meteorite. The isotope images of Fe-Ni metal, ferrihydrite, olivine, pyroxene, and troilite grains showed that ^4He is detected along grain interfaces, indicating its presence at depths shallower than 100 nm from the grain surfaces. These results demonstrated that SW implantation

occurred in the individual grains. Additionally, the existence of SW- ^4He in the ferrihydrite suggested two possibilities: either the implanted ^4He has survived in the ferrihydrite formed by terrestrial weathering of metal, or that formed through aqueous alteration on the parent body before SW exposure. The SW- ^4He fluence of the mineral grains ranged from $(2.7 \text{ to } 58) \times 10^{16}$ atoms cm^{-2} . Assuming irradiation occurred at 4 au, this corresponds to SW irradiation durations ranging from 3.8 to 82 kyr, representing the residence time of the individual mineral grains on the surface of the parent body. The residence times of most mineral grains suggest that the local mixing process on the parent body typically occurred with a frequency of about ten thousand years.

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534

535

FIGURES

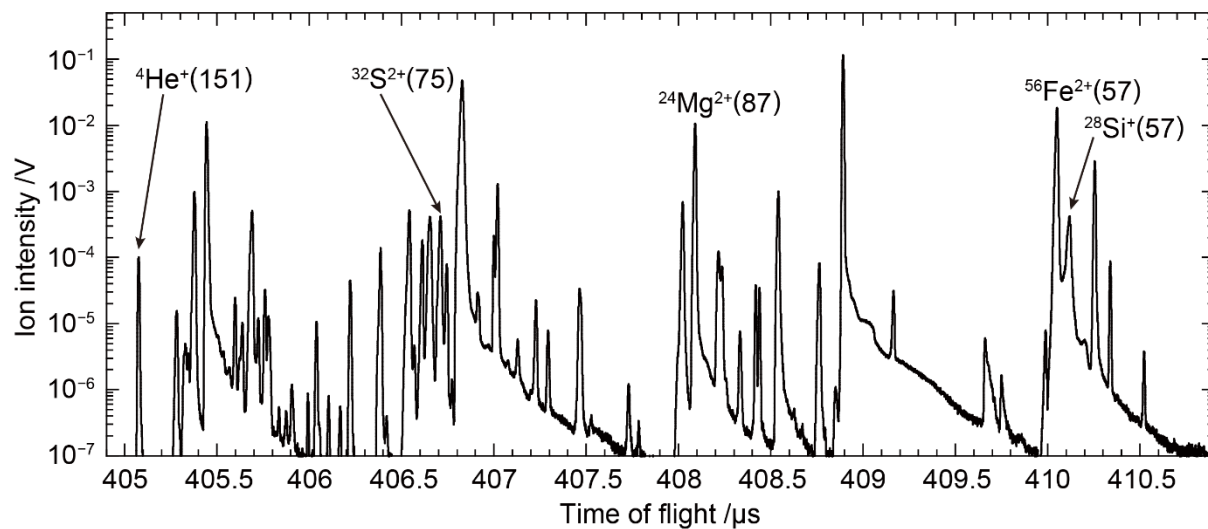


Figure 1. TOF mass spectrum of Area 4 of the NWA 801 sample shown in Fig. 6b. The arrows indicate the measured ion species, where the numbers in parentheses indicate the numbers of laps for each ion in MULTUM-II.

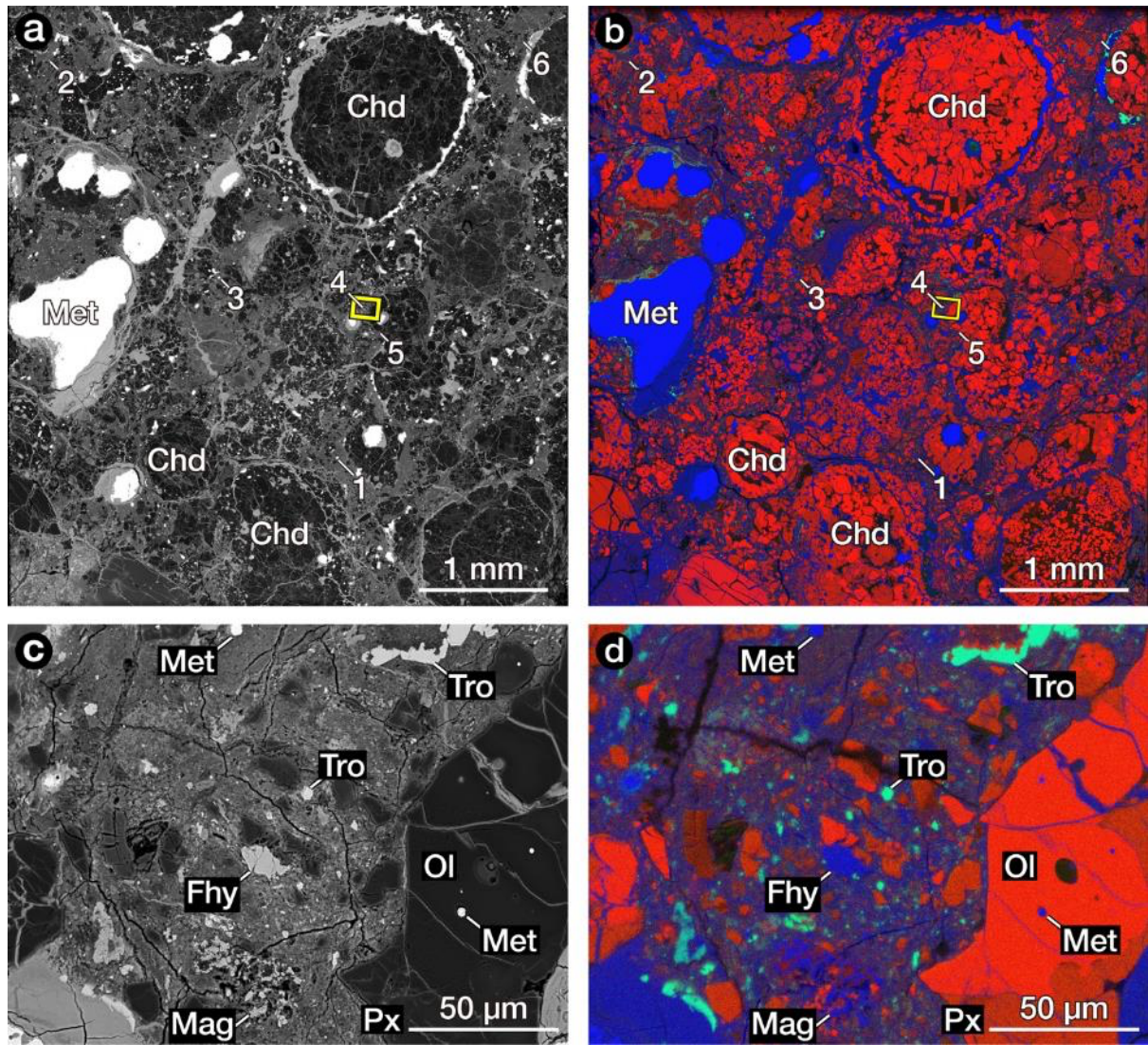


Figure 2. (a) BSE image of a section of NWA 801. (b) Composite X-ray elemental map of the area shown in (a), with Mg (red), Si (green), and Fe (blue). (c) Magnified BSE image of the area indicated by yellow boxes in (a) and (b). (d) Composite X-ray elemental map of the area shown in (c), with Mg (red), Si (green), and Fe (blue). Numbers 1–6 correspond to locations shown in Figs. 3–8, respectively. Abbreviations: Chd, chondrule; Fhy, ferrihydrite; Mag, magnetite; Met, Fe-Ni metal; Ol, olivine; Px, pyroxene; Tro, troilite.

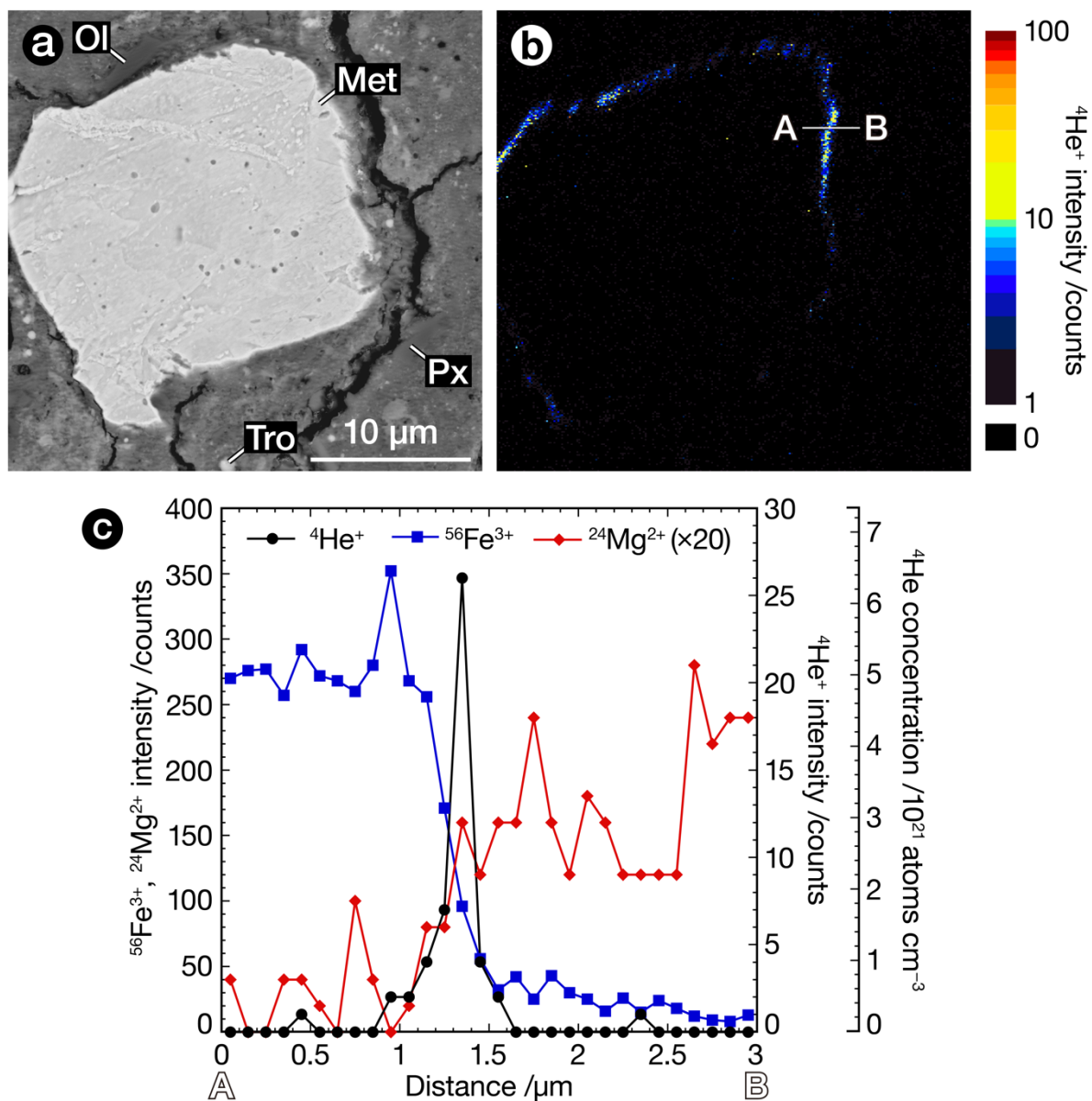


Figure 3. (a) BSE image of Area 1. (b) $^4\text{He}^+$ image of Area 1. (c) Line profiles for $^4\text{He}^+$, $^{56}\text{Fe}^{2+}$, and $^{24}\text{Mg}^{2+}$ along traverse A–B across the metal interface. Abbreviations: Met, Fe–Ni metal; Ol, olivine; Px, pyroxene; Tro, troilite.

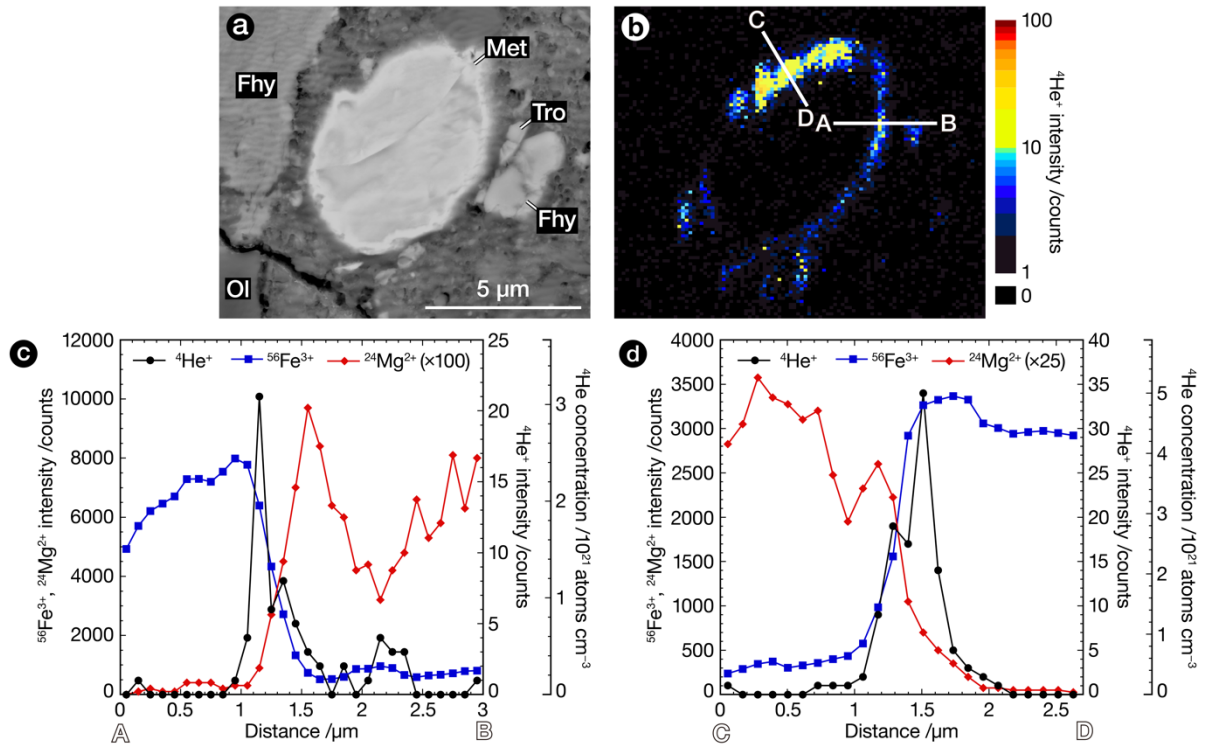


Figure 4. (a) BSE image of Area 2. (b) $^4\text{He}^+$ image of Area 2. (c) Line profiles for $^4\text{He}^+$, $^{56}\text{Fe}^{2+}$, and $^{24}\text{Mg}^{2+}$ along traverse A–B across the metal interface in (b). (d) Line profiles for $^4\text{He}^+$, $^{56}\text{Fe}^{2+}$, and $^{24}\text{Mg}^{2+}$ along traverse C–D across the metal interface in (b). Abbreviations: Fhy, ferrihydrite; Met, Fe-Ni metal; Ol, olivine; Tro, troilite.

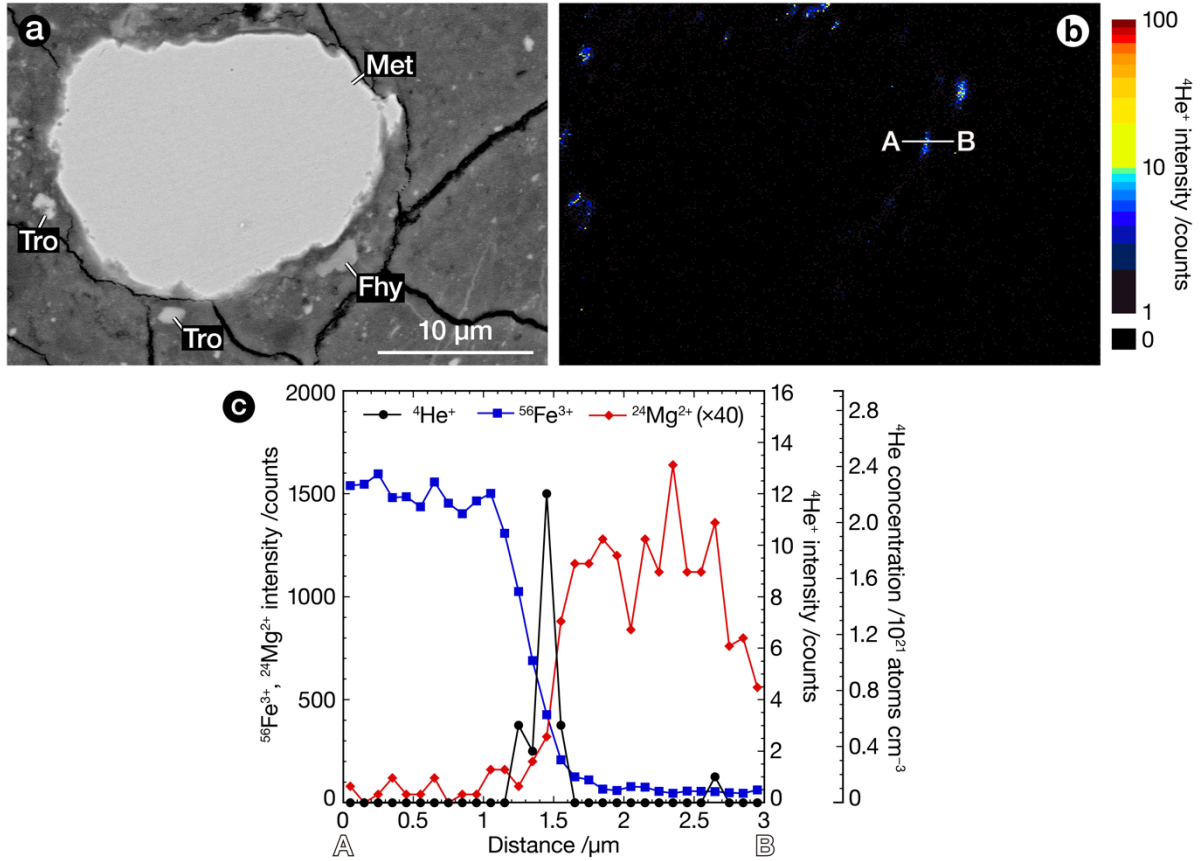


Figure 5. (a) BSE image of Area 3. (b) $^4\text{He}^+$ image of Area 3. (c) Line profiles for $^4\text{He}^+$, $^{56}\text{Fe}^{2+}$, and $^{24}\text{Mg}^{2+}$ along traverse A–B across the metal interface in (b). Abbreviations: Fhy, ferrihydrite; Met, Fe–Ni metal; Tro, troilite.

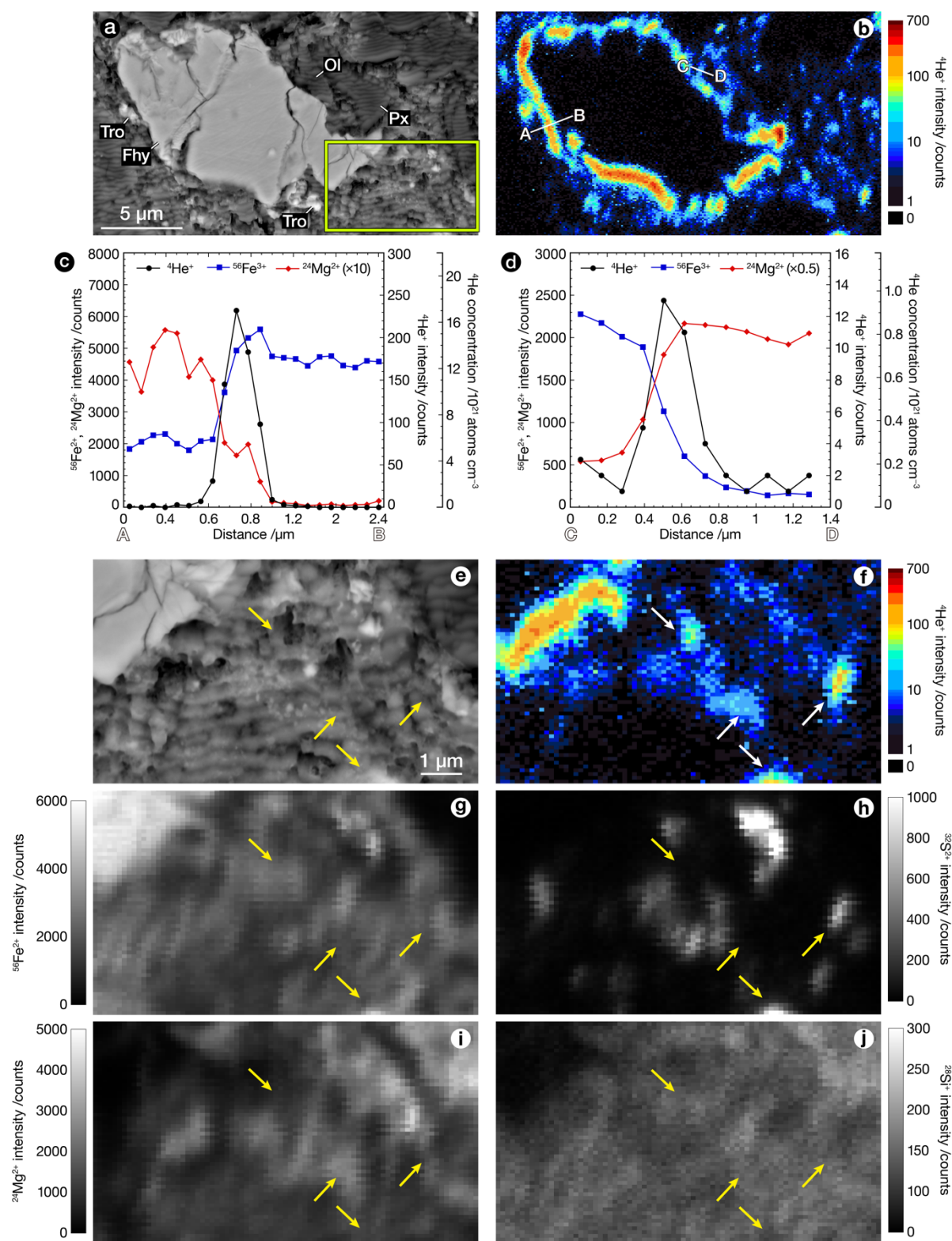


Figure 6. (a) BSE image of Area 4. (b) $^4\text{He}^+$ image of Area 4. (c) Line profiles for $^4\text{He}^+$, $^{56}\text{Fe}^{2+}$, and $^{24}\text{Mg}^{2+}$ along traverse A–B across the ferrihydrite interface in (b). (d) Line profiles for $^4\text{He}^+$, $^{56}\text{Fe}^{2+}$, and $^{24}\text{Mg}^{2+}$ ion intensities along traverse C–D across the pyroxene interface in (b). (e) Magnified BSE image of Area 4 delineated by the yellow box in (a). (f) $^4\text{He}^+$ image of Fig. 6e. (g) $^{56}\text{Fe}^{2+}$ image of Fig. 6e. (h) $^{32}\text{S}^{2+}$ image of Fig. 6e. (i) $^{24}\text{Mg}^{2+}$ image of Fig. 6e. (j) $^{28}\text{Si}^+$ image of Fig. 6e.

567 (j) $^{28}\text{Si}^{2+}$ image of Fig. 6e. The yellow arrows in (e) and (g)–(j) and white arrows in (f)
568 indicate the locations of $^4\text{He}^+$ enriched spots in the groundmass. Abbreviations: Fhy,
569 ferrihydrite; Ol, olivine; Px, pyroxene; Tro, troilite.

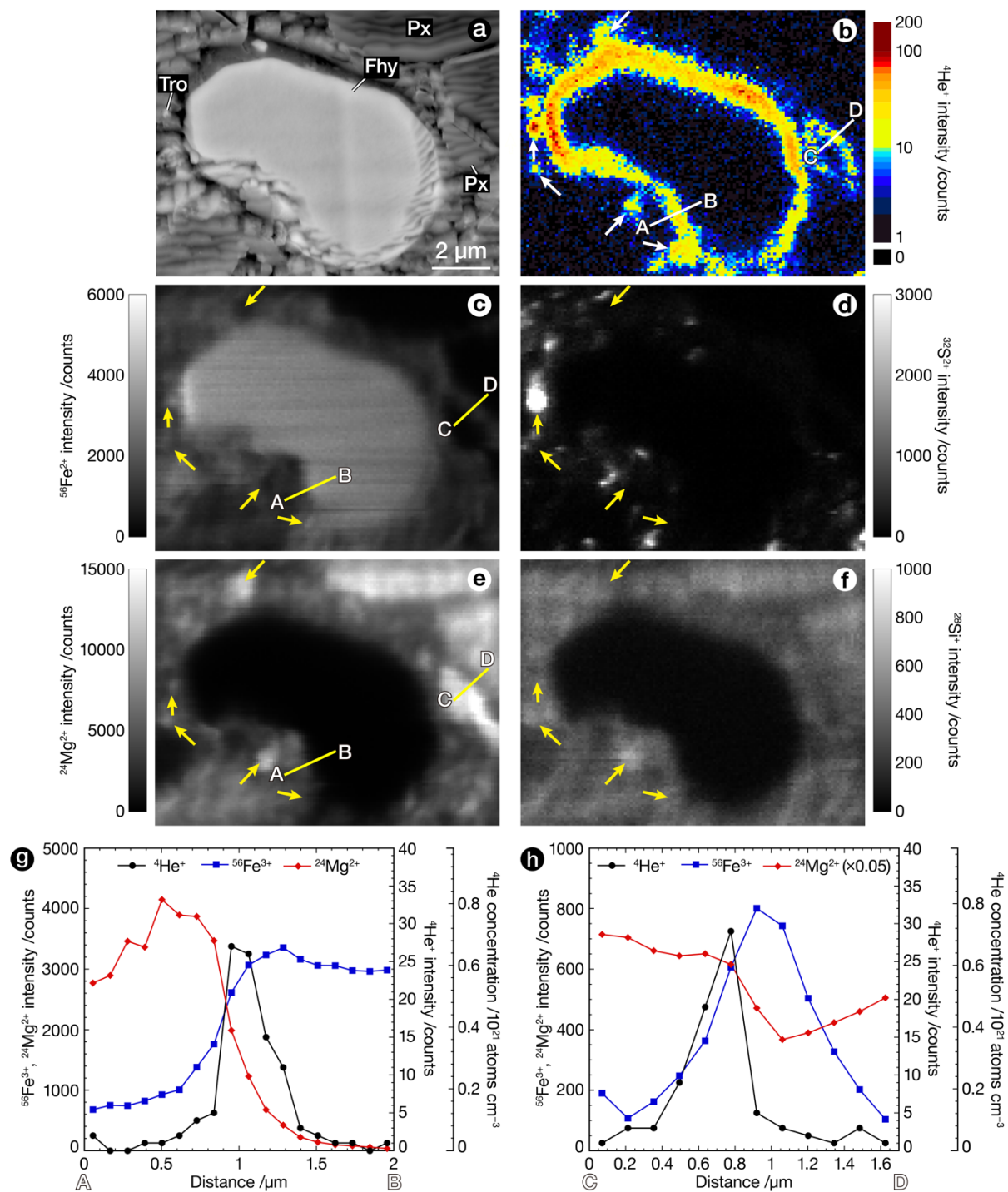


Figure 7. (a) BSE image of Area 5. (b) $^4\text{He}^+$ image of Area 5. (c) $^{56}\text{Fe}^{2+}$ image of Area 5. (d) $^{32}\text{S}^{2+}$ image of Fig. 7b. (e) $^{24}\text{Mg}^{2+}$ image of Fig. 7b. (f) $^{28}\text{Si}^{2+}$ image of Fig. 7b. (g) Line profiles for $^4\text{He}^+$, $^{56}\text{Fe}^{2+}$, and $^{24}\text{Mg}^{2+}$ along traverse A–B in (b). (h) Line profiles for $^4\text{He}^+$, $^{56}\text{Fe}^{2+}$, and $^{24}\text{Mg}^{2+}$ ion intensity along traverse C–D in (b). The yellow arrows in (a)–(f) indicate the locations of $^4\text{He}^+$ enriched spots in the groundmass. Abbreviations: Fhy, ferrihydrite; Px, pyroxene; Tro, troilite.

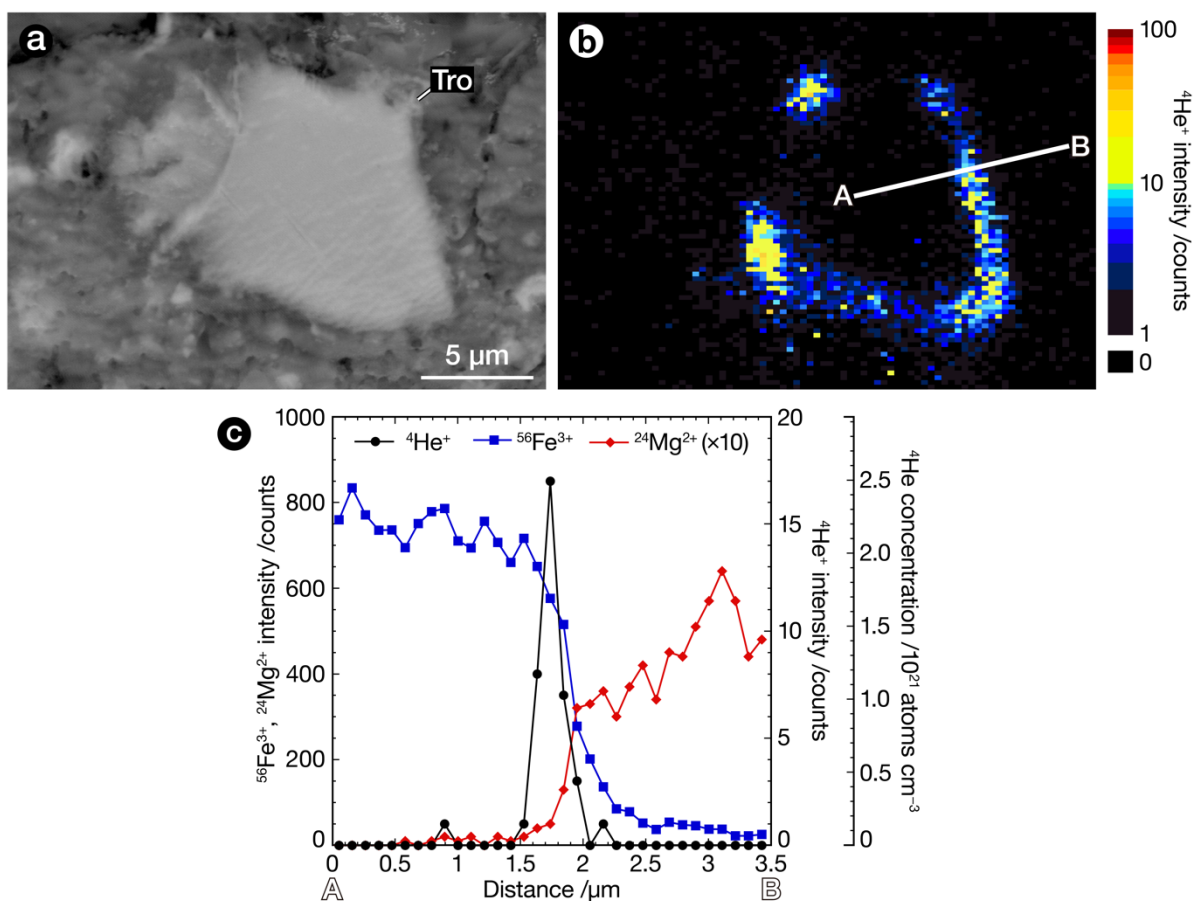


Figure 8. (a) BSE image of Area 6. (b) $^4\text{He}^+$ image of Area 6. (c) Line profiles for $^4\text{He}^+$, $^{56}\text{Fe}^{2+}$, and $^{24}\text{Mg}^{2+}$ along traverse A–B in (b). Abbreviations: Tro, troilite.

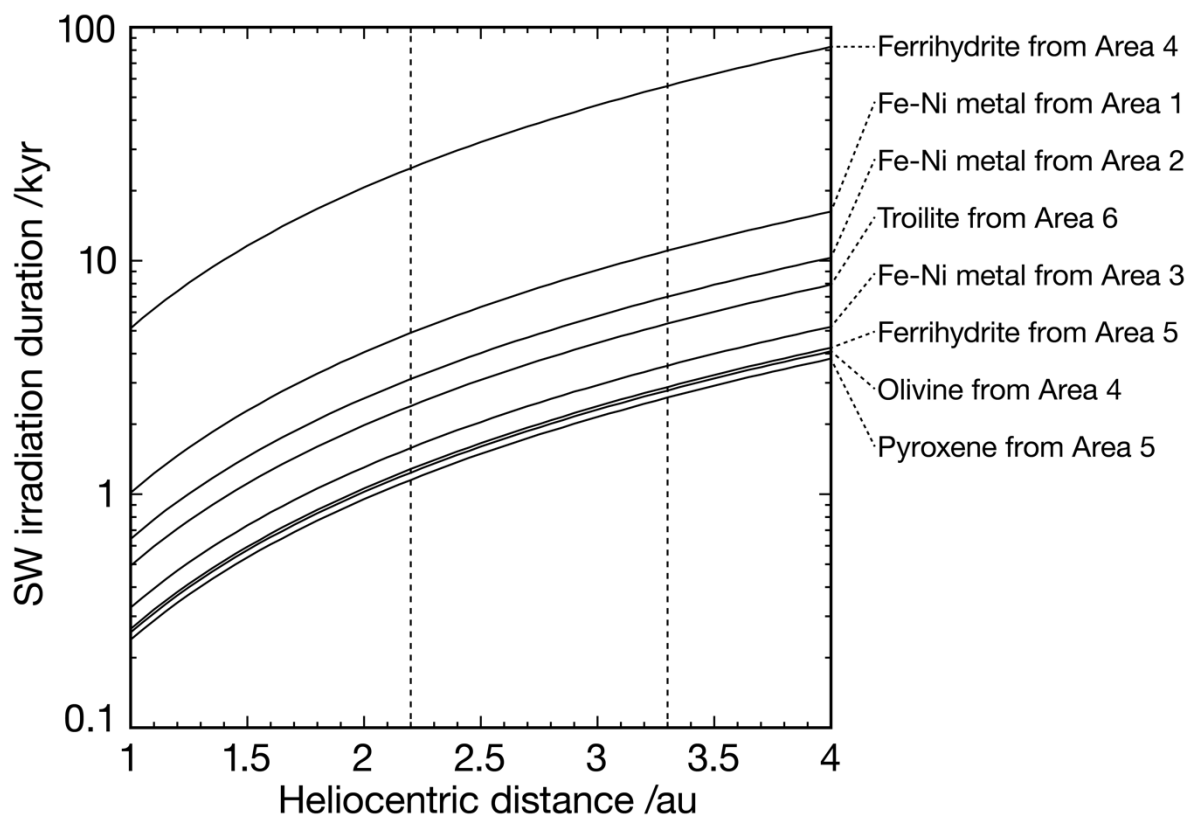


Figure 9. SW irradiation durations for mineral grains.

584 **TABLES**585 **Table 1.** Analytical conditions for isotope imaging.

Area	Image size / μm^2	Pixels	Num. of mass scan per pixel
1	25×30	250×300	600
2	10×10	100×100	1000
3	30×25	300×250	800
4	10×12	100×120	2000
5	15×20	150×200	6000
6	8×8	80×80	1000

586

587 **Table 2.** Spatial resolutions of isotope images.

Area	Mineral	$^4\text{He}^+$ * /nm	$^{56}\text{Fe}^{2+, 3+}$ ** /nm	$^{24}\text{Mg}^{2+}$ ** /nm
1	Fe-Ni metal	100	260	240
2	Fe-Ni metal	100	260	300
3	Fe-Ni metal	100	350	220
4	Ferrihydrite	250	160	440
4	Olivine	230	270	210
5	Ferrihydrite	240	310	330
5	Pyroxene	240	300	170
6	Troilite	150	450	400

*Peak width of $^4\text{He}^+$, **16–84% width from the line scan (Figs. S1–S6)

589 **Table 3.** SW-⁴He fluences and SW irradiation durations for mineral grains.

Area	Mineral	Line profile	⁴ He fluence	SW irradiation duration /kyr		
			/10 ¹⁶ atoms cm ⁻²	2.2 au	3.3 au	4.0 au
1	Fe-Ni metal	A-B	11.5 ± 1.7	4.9 ± 0.7	11.0 ± 1.6	16.2 ± 2.4
2	Fe-Ni metal	A-B	7.3 ± 1.0	3.1 ± 0.4	7.0 ± 1.0	10.3 ± 1.4
3	Fe-Ni metal	A-B	3.7 ± 0.8	1.6 ± 0.3	3.6 ± 0.8	5.2 ± 1.1
4	Ferrihydrite	A-B	58.4 ± 2.2	24.9 ± 0.9	56.1 ± 2.1	82.4 ± 3.1
4	Olivine	C-D	2.9 ± 0.5	1.2 ± 0.2	2.8 ± 0.5	4.1 ± 0.7
5	Ferrihydrite	A-B	3.0 ± 0.2	1.3 ± 0.1	2.9 ± 0.2	4.2 ± 0.3
5	Pyroxene	C-D	2.7 ± 0.2	1.1 ± 0.1	2.6 ± 0.2	3.8 ± 0.3
6	Troilite	A-B	5.6 ± 0.9	2.4 ± 0.4	5.4 ± 0.9	7.9 ± 1.3

Errors in 1σ